

# Are Octahedral Clusters Missing on the Carbon Energy Landscape?

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## Supporting Information Available

### Computational Details

The main GULP parameters using Tersoff (G1) and Dreiding (G2) IP:

Table 1: Potentials and the main computational settings used within the GULP package.

Name	Potential	Convergence criterion based on			
		xtol (Å)	gtol	ftol (eV/Å)	switch rfo
G1	Tersoff	10 <sup>-6</sup>	10 <sup>-4</sup>	10 <sup>-8</sup>	10 <sup>-2</sup>
G2	Dreiding	10 <sup>-6</sup>	10 <sup>-4</sup>	10 <sup>-8</sup>	10 <sup>-2</sup>

The main FHI-aims parameters with respect to functionals used in the study:

Table 2: Functionals and the main computational settings used within the FHI-aims package. “-” notes that the default value or algorithm was used.

Name	Functional	Convergence criterion based on				
		Basis	charge density (electrons)	eigenvalues (eV)	total energy (eV)	forces (eV/Å)
A1	LDA	light	10 <sup>-4</sup>	10 <sup>-3</sup>	10 <sup>-5</sup>	10 <sup>-3</sup>
A2		tight	10 <sup>-4</sup>	10 <sup>-3</sup>	10 <sup>-5</sup>	10 <sup>-3</sup>
A3	GGA	light	10 <sup>-2</sup>	10 <sup>-1</sup>	10 <sup>-2</sup>	10 <sup>-1</sup>
A4		tight	10 <sup>-4</sup>	10 <sup>-3</sup>	10 <sup>-5</sup>	10 <sup>-3</sup>
A5	GGA+MDB	tight	-	-	-	-
A6	Hybrid	light	10 <sup>-2</sup>	10 <sup>-1</sup>	10 <sup>-2</sup>	10 <sup>-1</sup>

Vibrational frequencies have been calculated with FHI-aims using the LDA functional and *light* basis set. We have used the script-based finite difference approach from the FHI-aims package with the step size of  $\delta=0.0025$  Å.

## Geometry Optimization

We report the summary of geometry optimisation calculations in Table 4 with respect to the graphite energy calculated at the same level of theory as it is given in Table 3. The lowest energy structure of 24 atoms ( $L_{24}$ ) at the DFT level of theory is given in Figure 1.

Table 3: Total energy per atom of graphite at different levels of theory and computational settings, as described in Table 2.

Theory	Energy (eV)	Theory	Energy (eV)
G1	0.4324	A3	-1036.9925
G2	-7.3395	A4	-1037.0579
A1	-1029.0728	A5	-1037.2029
A2	-1029.1394	A6	-1037.0007

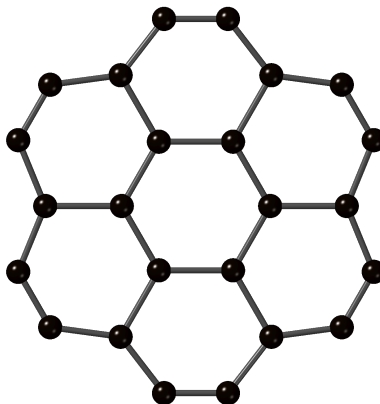


Figure 1: The lowest energy structure of 24 atoms -  $L_{24}$  - a one-layer convex structure containing 7 six-membered rings (terminated by two-coordinated atoms).

## Density of States

The averaged vibrational density of states (DOS) have been calculated using the standard DOS expression with a Gaussian smearing function:

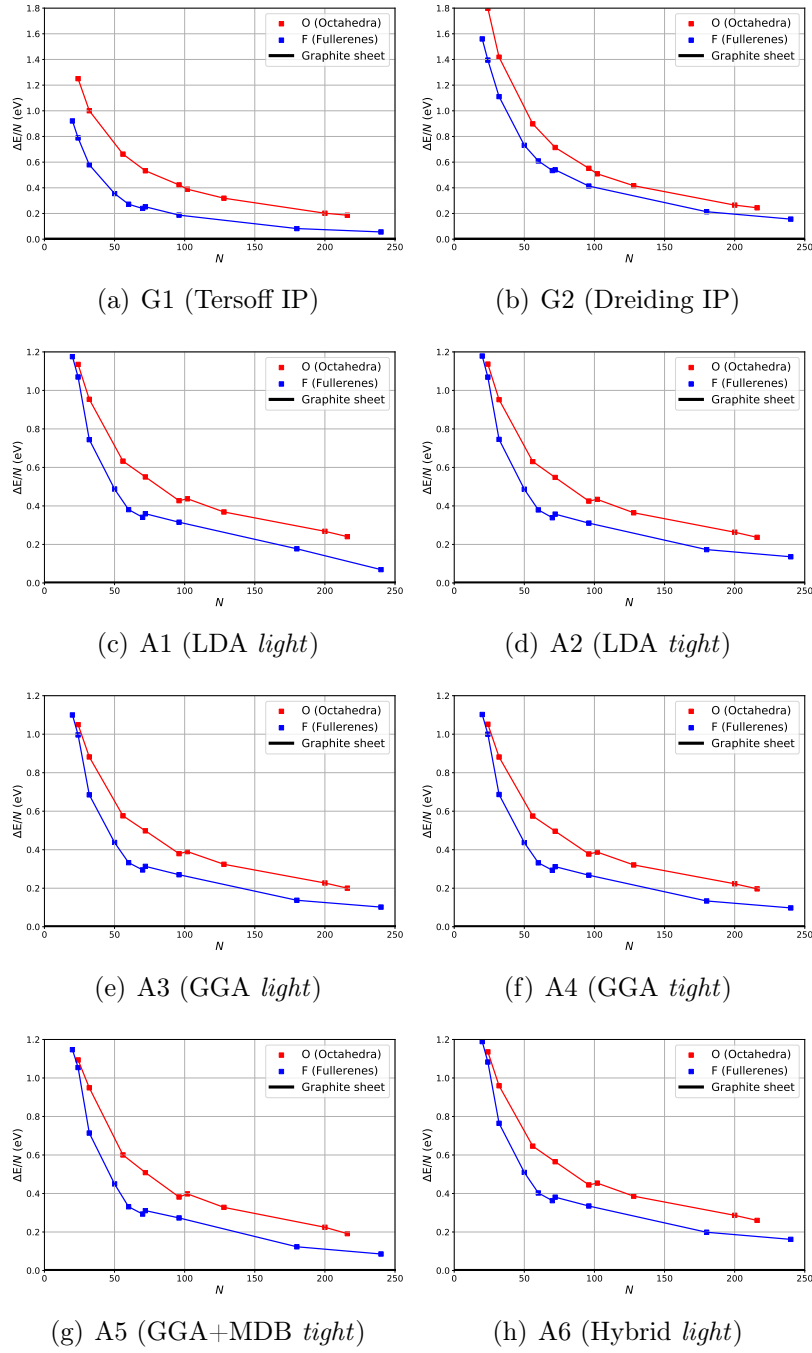


Figure 2: Energy differences (Table 4) per atom between graphene sheet and the cluster structures with respect to the cluster size  $N$ . Yellow line represent layer structure, green - double bubbles, red - single bubbles and blue lines represent fullerene structures.

Table 4: Summary of energy differences with respect to the graphene sheet of optimised clusters' using different functionals and settings (as given in Tables 1 and 2). The subscript in name for layer, single bubble and fullerene configurations indicates the total number of atoms  $N$ , where for the double bubble configurations, this number indicates the number of atoms of the outer and the superscript - for the inner bubble atom counts respectively; "\*" marks structures in spin-polarized triplet state. S is the cluster's symmetry point group,  $\Delta E/N$  is the energy difference with respect to the graphite energy calculated at the same level of theory and using the same settings (Table 3) per atom, H-L is the HOMO-LUMO gap.

Name	S	G1	G2	A1		A2	A3	A4	A5	A6
		$\Delta E/N$ (eV)	$\Delta E/N$ (eV)	$\Delta E/N$ (eV)	H-L (eV)	$\Delta E/N$ (eV)	$\Delta E/N$ (eV)	$\Delta E/N$ (eV)	$\Delta E/N$ (eV)	$\Delta E/N$ (eV)
Layer										
L <sub>24</sub>	D6h	-0.2064	1.1154	1.1451	2.205	1.2354	1.0845	0.9930		1.0635
Octahedral										
O <sub>24</sub>	Oh	1.2506	1.7986	1.1349	1.232	1.2366	1.0498	1.0518	1.1356	1.0937
O <sub>52</sub>	Oh	1.0006	1.4175	0.9538	0.729	1.0522	0.8820	0.8814	0.9597	0.9491
O <sub>56</sub>	O	0.6624	0.8984	0.6327	1.078	0.7306	0.5761	0.5748	0.6459	0.6003
O <sub>72</sub>	Oh	0.5328	0.7141	0.5506	1.569	0.6478	0.4981	0.4961	0.5651	0.5086
O <sub>96</sub>	Oh	0.4227	0.5519	0.4274	1.475	0.5252	0.3796	0.3778	0.4446	0.3818
O <sub>104</sub>	O	0.3891	0.5092	0.4371	0.294	0.5339	0.3901	0.3876	0.4539	0.3980
O <sub>128</sub>	Oh	0.3191	0.4164	0.3686	0.889	0.4645	0.3243	0.3209	0.3861	0.3279
O <sub>200</sub>	Oh	0.2021	0.2652	0.2682	0.142	0.3635	0.2271	0.2230	0.2866	0.2246
O <sub>216</sub>	Oh	0.1864	0.2444	0.2405	0.984	0.3367	0.2003	0.1967	0.2601	0.1915
Fullerenes										
*F <sub>20</sub>	D3d	0.9209	1.5600	1.1732	0.423	1.2785	1.0998	1.1023	1.1891	1.1471
*F <sub>24</sub>	D6d	0.7889	1.3951	1.0657	0.228	1.1690	0.9963	1.0000	1.0830	1.0544
F <sub>32</sub>	D3	0.5795	1.1108	0.7443	1.448	0.8454	0.6851	0.6868	0.7645	0.7139
F <sub>50</sub>	D5h	0.3554	0.7311	0.4879	0.441	0.5868	0.4375	0.4371	0.5094	0.4498
F <sub>60</sub>	Ih	0.2721	0.6093	0.3810	1.657	0.4794	0.3328	0.3319	0.4026	0.3312
F <sub>70</sub>	D5h	0.2390	0.5348	0.3410	1.734	0.4388	0.2950	0.2935	0.3629	0.2933
F <sub>72</sub>	D6d	0.2521	0.5402	0.3594	1.442	0.4573	0.3135	0.3119	0.3809	0.3111
*F <sub>96</sub>	D6d	0.1872	0.4136	0.3138	0.072	0.4107	0.2701	0.2677	0.3349	0.2731
F <sub>180</sub>	Ih	0.0826	0.2137	0.1773	1.466	0.2732	0.1374	0.1338	0.1988	0.1232
F <sub>240</sub>	Ih	0.0562	0.1561	0.1406	1.241	0.2359	0.1019	0.0977	0.1617	0.0858

$$\sum_k \left[ \frac{1}{\sigma\sqrt{2\pi}} \sum_{i=1}^N \exp\left(\frac{\omega_i - \Omega_k}{2\sigma^2}\right) \right] / K, \quad (1)$$

where  $\sigma = 10.0$  (Gaussian smearing parameter),  $\omega_i$  is the vibrational frequency interval,  $\Omega_k$  - the set of vibrational frequencies corresponding to a particular cluster  $k$ , and  $K$  - is the number of clusters of the same type, i.e. fullerenes or octahedra.

## Fitting functions of the Gibbs free energy

The inverse power function (Equation 2) and its coefficients used for fitting free Gibbs energies for the O and F types of clusters at 10, 300, 1000, 2500 and 5000 K temperatures.

$$F(N) = \frac{a}{N^b} + c. \quad (2)$$

Table 5: The coefficients for the inverse power function to fit free Gibbs energies for the O and F types of clusters at 10, 300, 1000, 2500 and 5000 K temperatures.

T	Octahedral			Fullerene		
	a	b	c	a	b	c
10 K	8.36	0.61	-1036.95	27.69	1.06	-1036.83
300 K	8.18	0.60	-1036.96	27.90	1.06	-1036.83
1000 K	8.19	0.59	-1037.08	26.27	1.03	-1036.94
2500 K	8.61	0.59	-1037.65	23.05	0.97	-1037.51
5000 K	10.61	0.63	-1039.04	21.11	0.89	-1038.95

The constant  $c$  should correspond to the limit of the graphene sheet, cf. analyses in reference 17.

## Integrated vibrational frequencies

We have individually integrated the vibrational frequencies for the O and F types of clusters in three ranges: 0-850  $\text{cm}^{-1}$ , 850-1075  $\text{cm}^{-1}$ , and 1075-1700  $\text{cm}^{-1}$ .

Table 6: .

Cluster	0-850 $\text{cm}^{-1}$	850-1075 $\text{cm}^{-1}$	1075-1700 $\text{cm}^{-1}$
Octahedral			
O <sub>24</sub>	36.87	4.11	25.00
O <sub>52</sub>	47.00	10.10	32.89
O <sub>56</sub>	83.00	14.16	64.81
O <sub>72</sub>	109.00	14.96	85.92
O <sub>96</sub>	145.75	18.21	117.68
O <sub>104</sub>	160.90	21.60	122.97
O <sub>128</sub>	194.26	28.37	155.23
O <sub>200</sub>	310.62	40.92	242.11
O <sub>216</sub>	332.48	41.35	267.56
Fullerenes			
F <sub>20</sub>	24.70	5.07	24.16
F <sub>24</sub>	32.00	4.46	29.49
F <sub>32</sub>	46.00	5.00	39.00
F <sub>50</sub>	74.91	9.61	59.33
F <sub>60</sub>	91.99	8.00	74.00
F <sub>70</sub>	105.98	9.84	88.07
F <sub>72</sub>	108.86	11.92	89.00
F <sub>96</sub>	146.99	15.88	118.97
F <sub>180</sub>	276.41	35.40	221.98
F <sub>240</sub>	367.86	49.76	295.89