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Supporting Information

for

Synthesis of stable and phase-adjustable CsPbBr₃@Cs₄PbBr₆ nanocrystals via novel anion-cation reaction

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Experiment Section

Materials

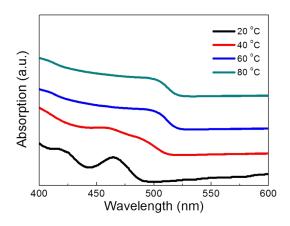
PbBr₂, CsBr, PDBr, cesium acetate, lead acetate, DMF, DMSO, DDA-Cl, acetonitrile (from Macklin reagent) Cesium stearate (CsSt, 98% pure, from J&K reagent), oleylamine (OAm, 80–90% pure, from Aladdin-reagent), oleic acid (OA, from Aladdin-reagent), 1-octadecene (ODE, >80% pure, from Aladdin-reagent), OAm-I (from Xi'an Baolaite) were used without further purification.

Synthesis of cubic CsPbBr₃ nanocrystals

PbBr₂ (0.6 mmol) and CsBr (0.6 mmol) were dissolved in 10 mL dimethyl sulfoxide (DMSO) solution. OA (0.5 mL) and OAm (0.5 mL) were added to stabilize the precursor solution. Then, 1 mL of the precursor solution was quickly added into 20 mL toluene under vigorous stirring. Strong green-mission crude production was observed immediately after the injection. After reaction for 1 min, 10 mL acetonitrile was added into the green crude solution for purification. The precipitate was dispersed in toluene for further use after centrifugation.

Synthesis of cubic CsPbBr₃ nanocrystals

15 mL of octadecene (ODE), 3 mL of OAm, 1.5 mL of OA, and PbBr2 (0.2 g) were loaded into a 100 mL four-neck flask, degassed at 100 °C for 10 min, mixed at 100 °C for 30 min, and heated to 170 °C in 10 min under Ar fl ow. 0.55 mL of Cesium Stearate (CsSt) solution (0.15 M in ODE) was quickly injected. After 5 s, the reaction mixture was cooled by the ice-water bath. The resultant QDs were precipitated by 20 mL of acetone and separated via centrifugation. The separated QDs were redispersed in in toluene for further use.



 $\textbf{Fig. S1} \ \ \, \text{Absorption spectra of } CsPbBr_3@Cs_4PbBr_6 \ \ \, \text{NCs prepared at different temperature}.$