

## Supporting Information

### Green Luminescent MoS<sub>2</sub>-CdTe Hybrid Nanostructure Synthesized through Surface Charge Interaction

Dhrubaa Haldar<sup>a</sup>, Saptasree Bose,<sup>a,b</sup> Arnab Ghosh<sup>a,b</sup> and Shyamal K Saha<sup>a,\*</sup>

Department of Materials Sciences, Indian Association for the Cultivation of Science

2A and 2B Raja S C Mullick Road, Jadavpur, Kolkata - 700032, India.

\*Email of the corresponding author: [cnssks@iacs.res.in](mailto:cnssks@iacs.res.in)

<sup>b</sup>Authors contribute equally.

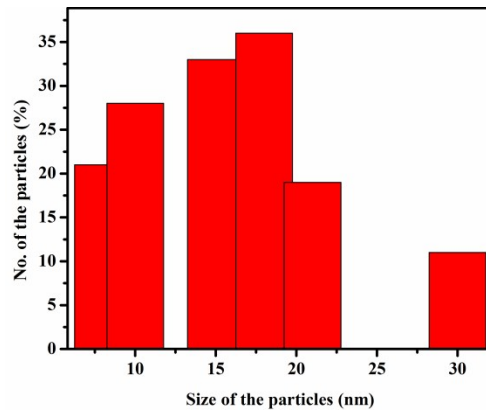


Figure S1: Size distribution histogram of MoS<sub>2</sub> QDs.

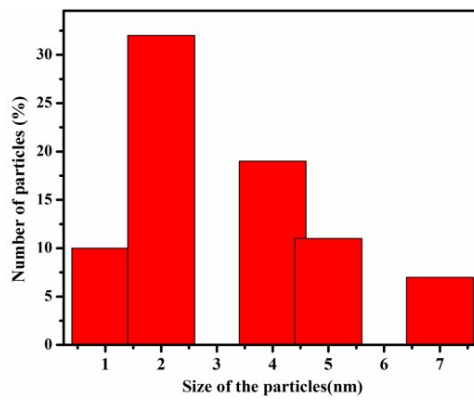


Figure S2: Size distribution histogram of green CdTe QDs.

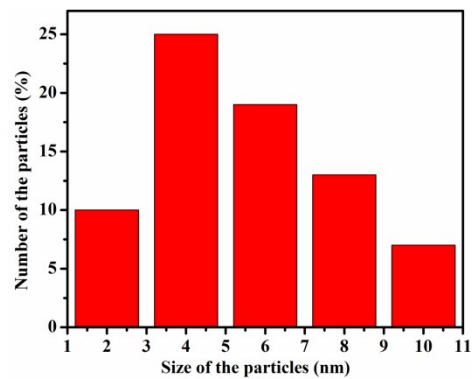


Figure S3: Size distribution histogram of yellow CdTe QDs

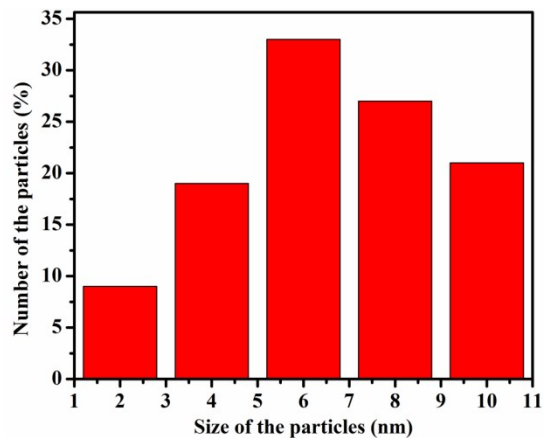


Figure S4: Size distribution histogram of red CdTe QDs.

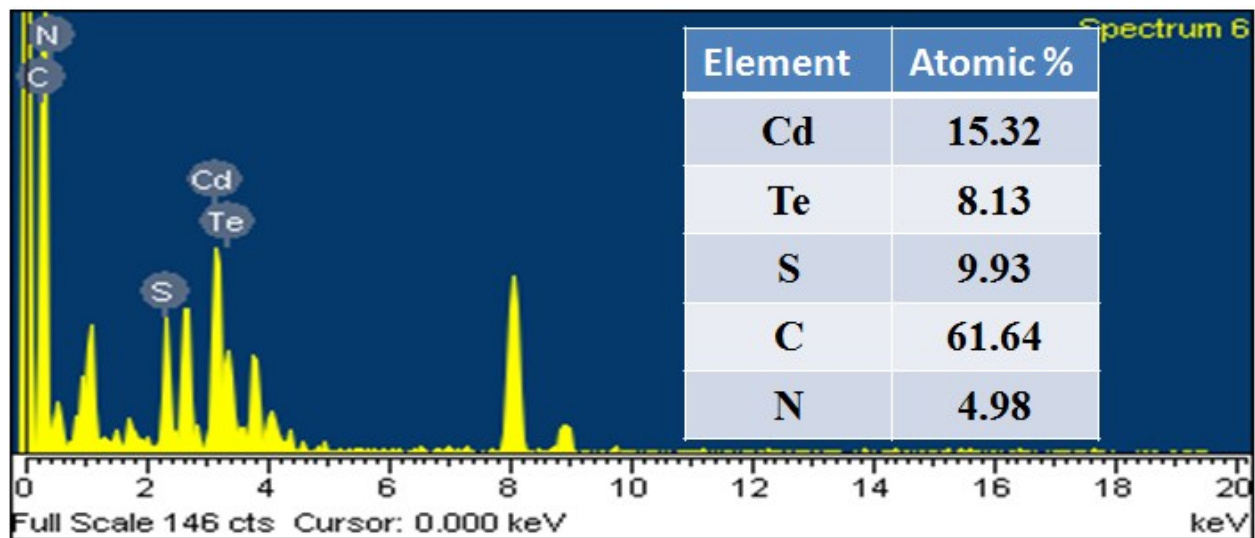


Figure S5: EDX analysis of cysteamine capped CdTe QDs.

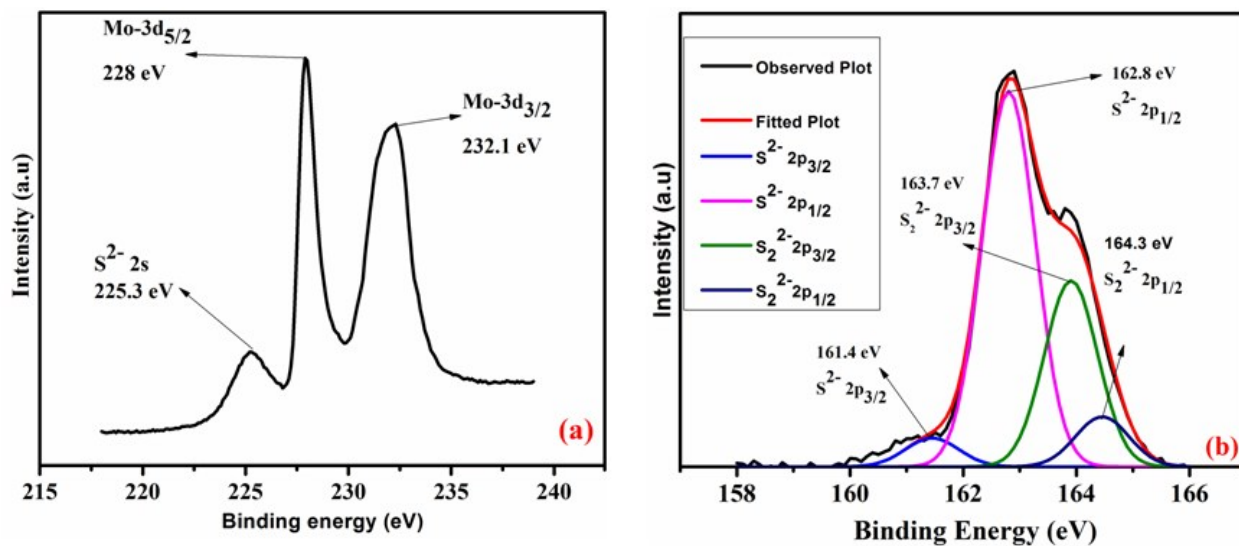


Figure S6. XPS spectra of S<sup>2-</sup>-2s, Mo(IV)-3d<sub>5/2</sub> and 3d<sub>3/2</sub> in bare amorphous MoS<sub>2</sub> QDs and (b) Deconvoluted XPS spectra of S<sup>2-</sup>-2p<sub>3/2</sub> and 2p<sub>1/2</sub> and S<sub>2</sub><sup>2-</sup> 2p<sub>3/2</sub> and 2p<sub>1/2</sub> in bare amorphous MoS<sub>2</sub> QDs.

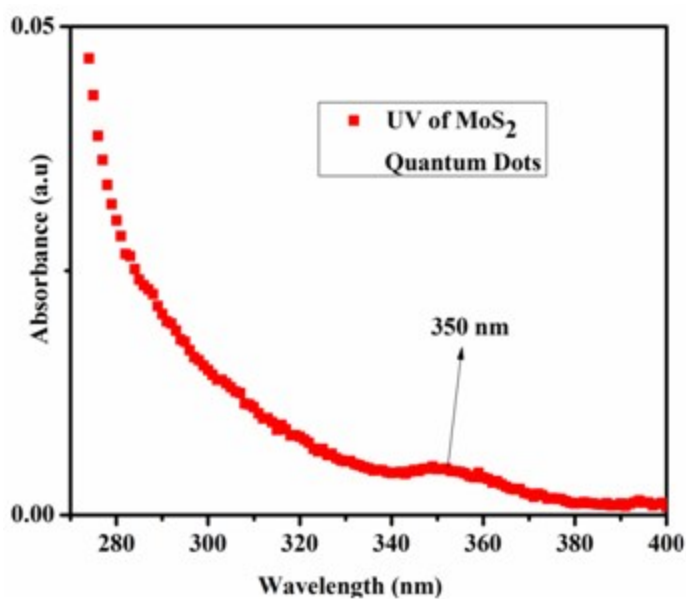


Figure S7: UV-visible absorption spectroscopy of MoS<sub>2</sub> QDs.

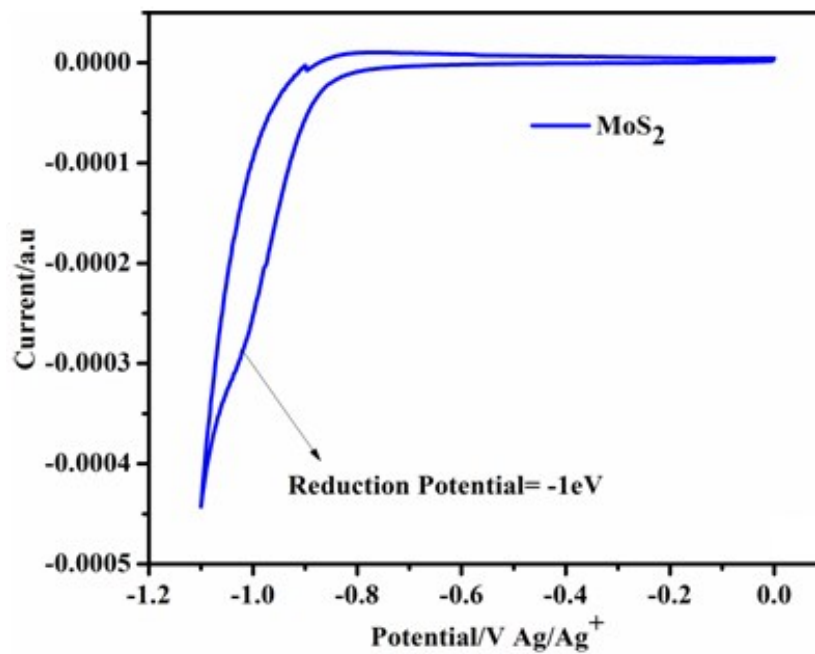


Figure S8: Cyclic voltametry of amorphous MoS<sub>2</sub> QDs.

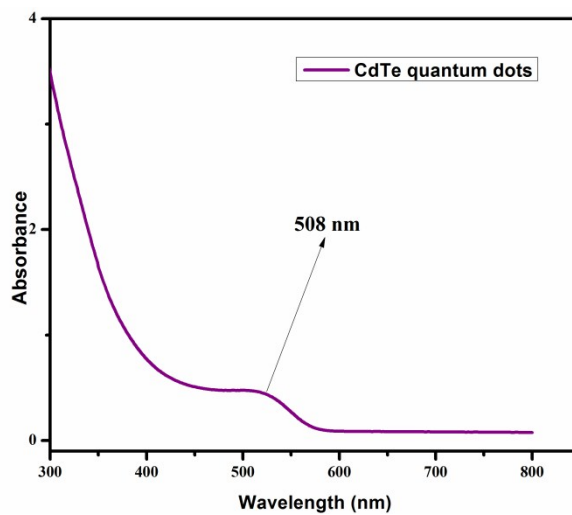


Figure S9: UV-visible absorption spectroscopy of cysteamine capped CdTe QDs.

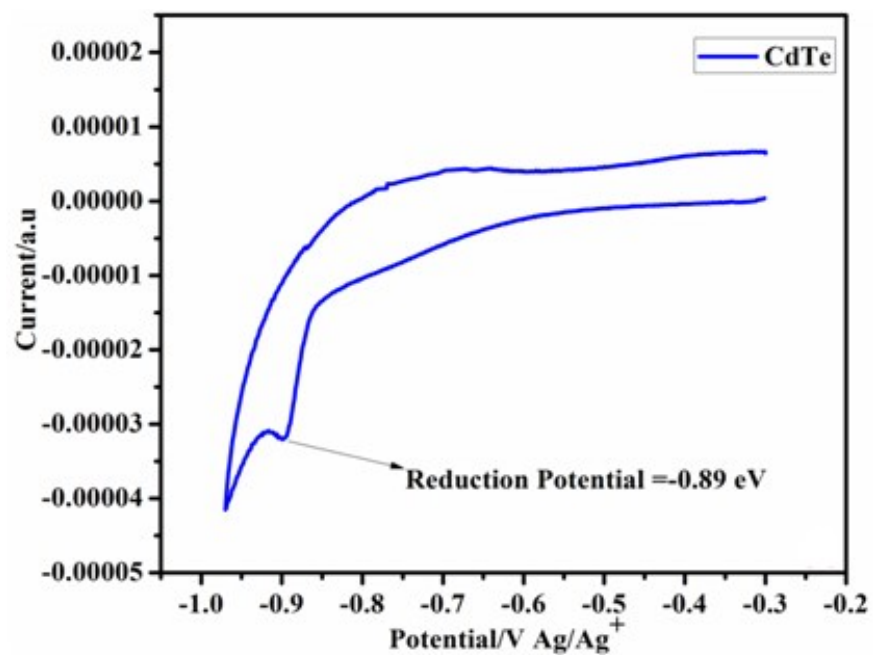


Figure S10: Cyclic voltametry of cysteamine capped CdTe QDs.

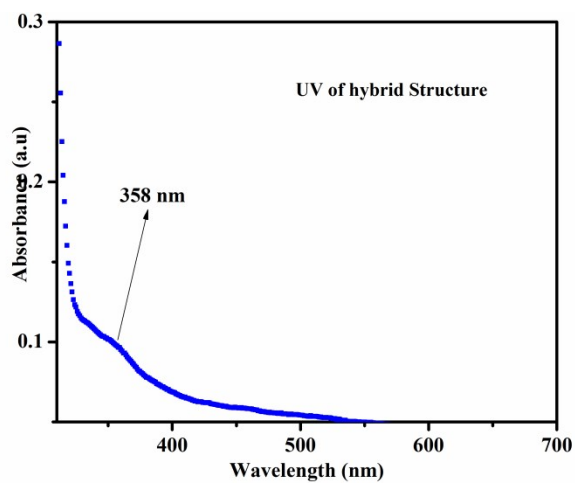


Figure S11: UV-visible absorption spectroscopy of hybrid structure.

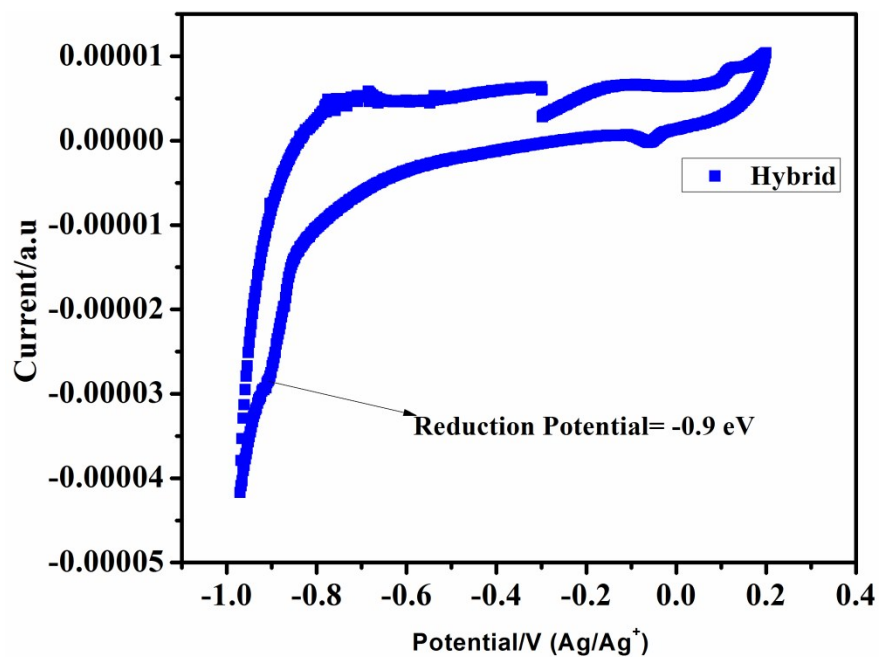


Figure S12: Cyclic voltametry of cysteamine capped CdTe QDs.

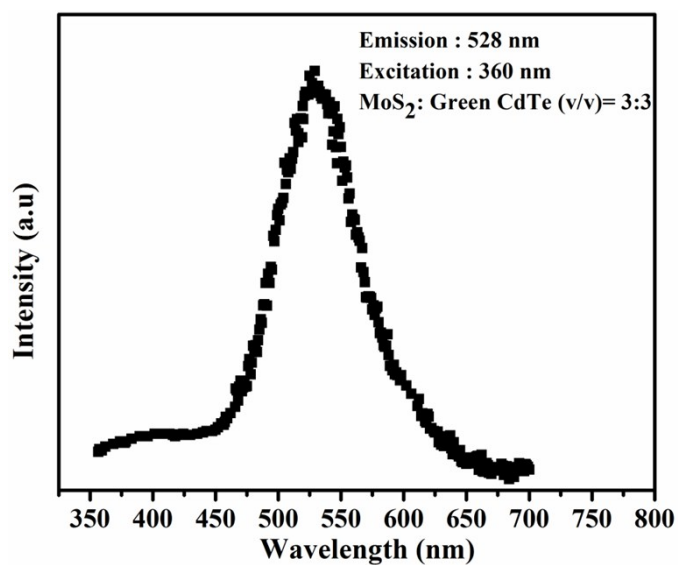


Figure S13: Photoluminescence emission spectra of MoS<sub>2</sub>:Green CdTe =3:3 (v/v).

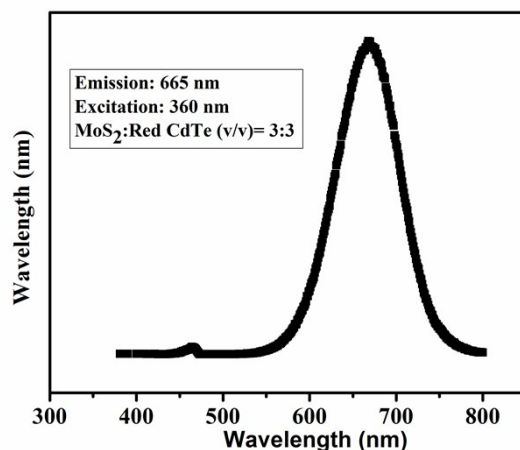


Figure S14: Photoluminescence emission spectra of MoS<sub>2</sub>:Red CdTe =3:3 (v/v).

**15. The quantum yield calculation:** Relative fluorescence quantum yield of the samples are calculated according to the following expression (i)

$$\Phi_s = (I_s/I_r) \times (A_r/A_s) \times (\eta_s^2/\eta_r^2) \times \Phi_r \quad (i)$$

Where the subscripts r and s refer to the reference and the sample respectively,  $\Phi$  is the fluorescence quantum yield, I is the measured integrated fluorescence emission intensity, A is the optical density and  $\eta$  is the refractive index of the solvents.

The quantum yield of MoS<sub>2</sub>, CdTe and hybrid materials are 14.77%, 76.08% and 63.12% respectively in which it is found that the quantum yield of the hybrid materials is much greater than MoS<sub>2</sub> QDs, and comparable to that of CdTe.