

Supporting Information:

A simple and Novel protocol for Li-trapping of POM/MOF nano-composite as a new adsorbent for CO₂ uptake

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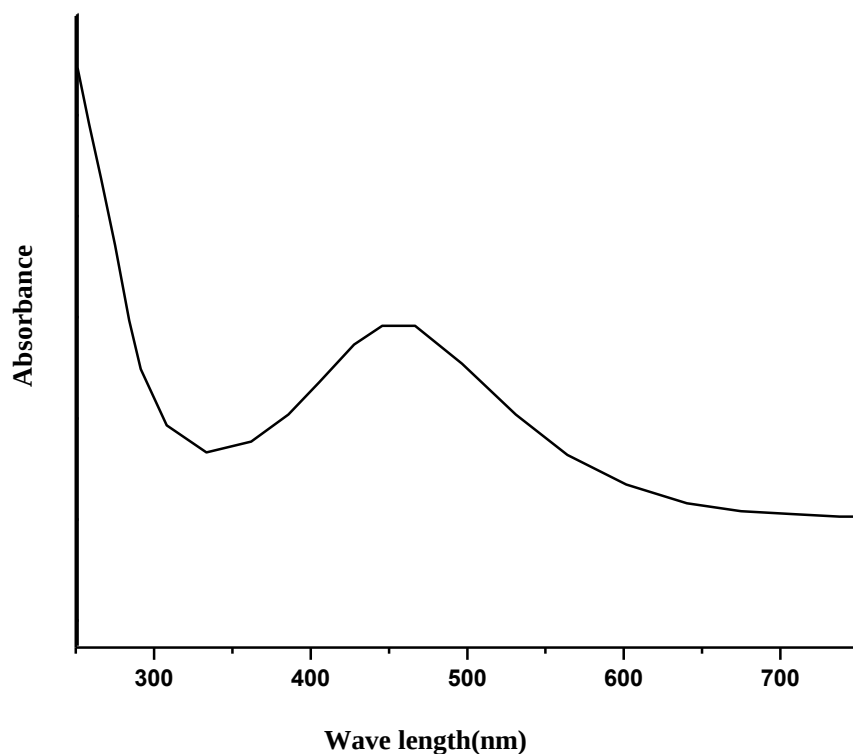


Fig. S1. UV-Vis spectrum of the synthesized {Mo₁₃₂}

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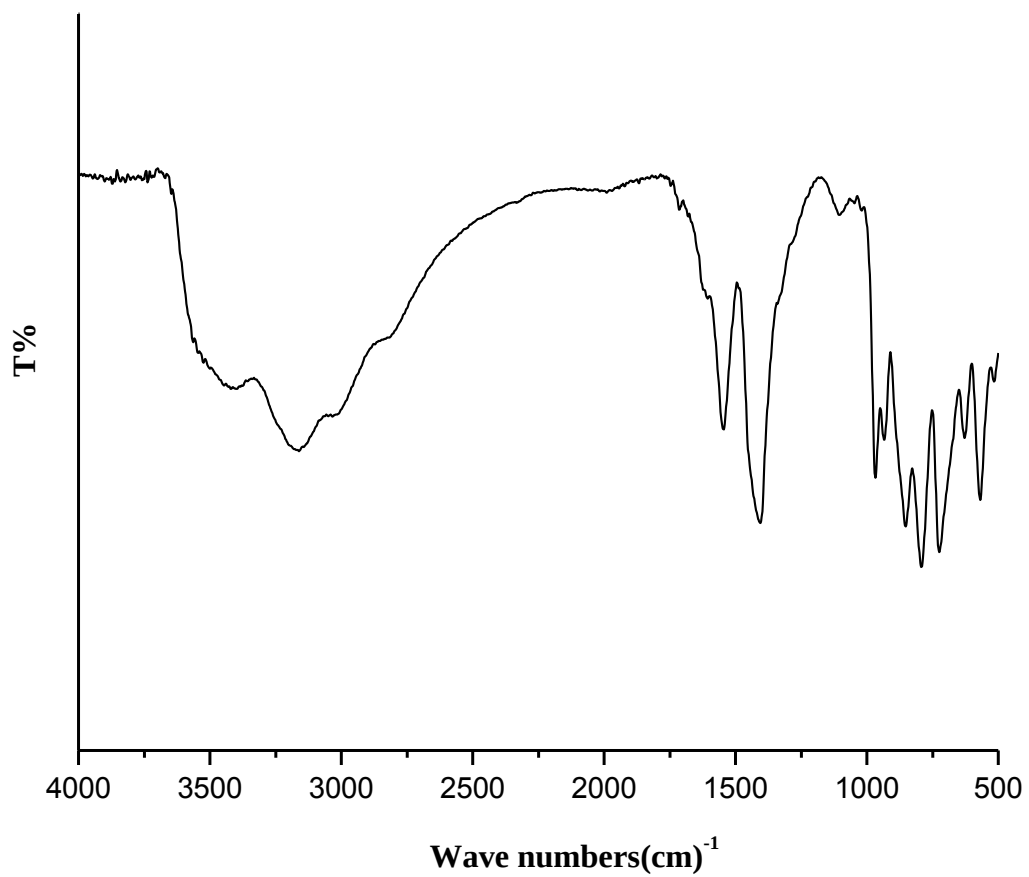


Fig. S2. FT-IR spectra of the synthesized {Mo₁₃₂}

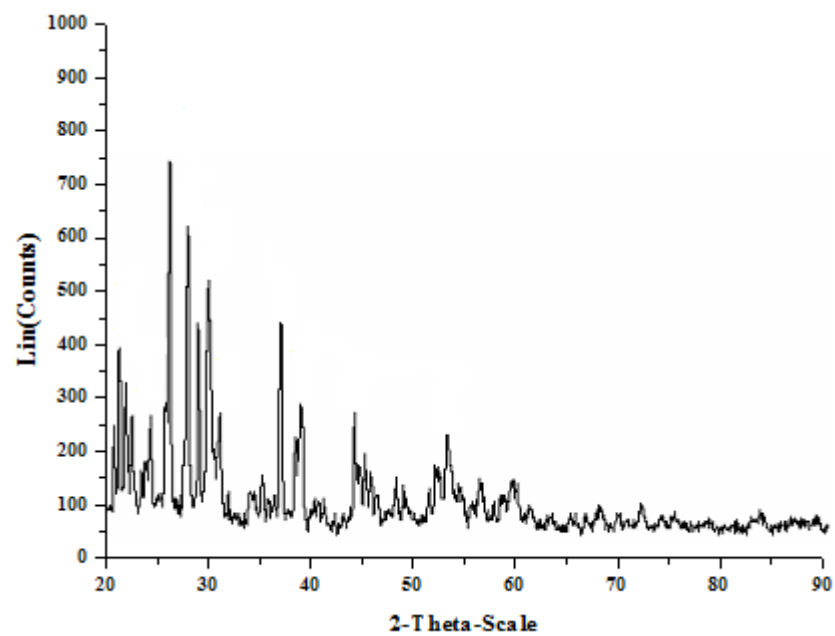


Fig. S3. XRD pattern of the synthesized $\{\text{Mo}_{132}\}$

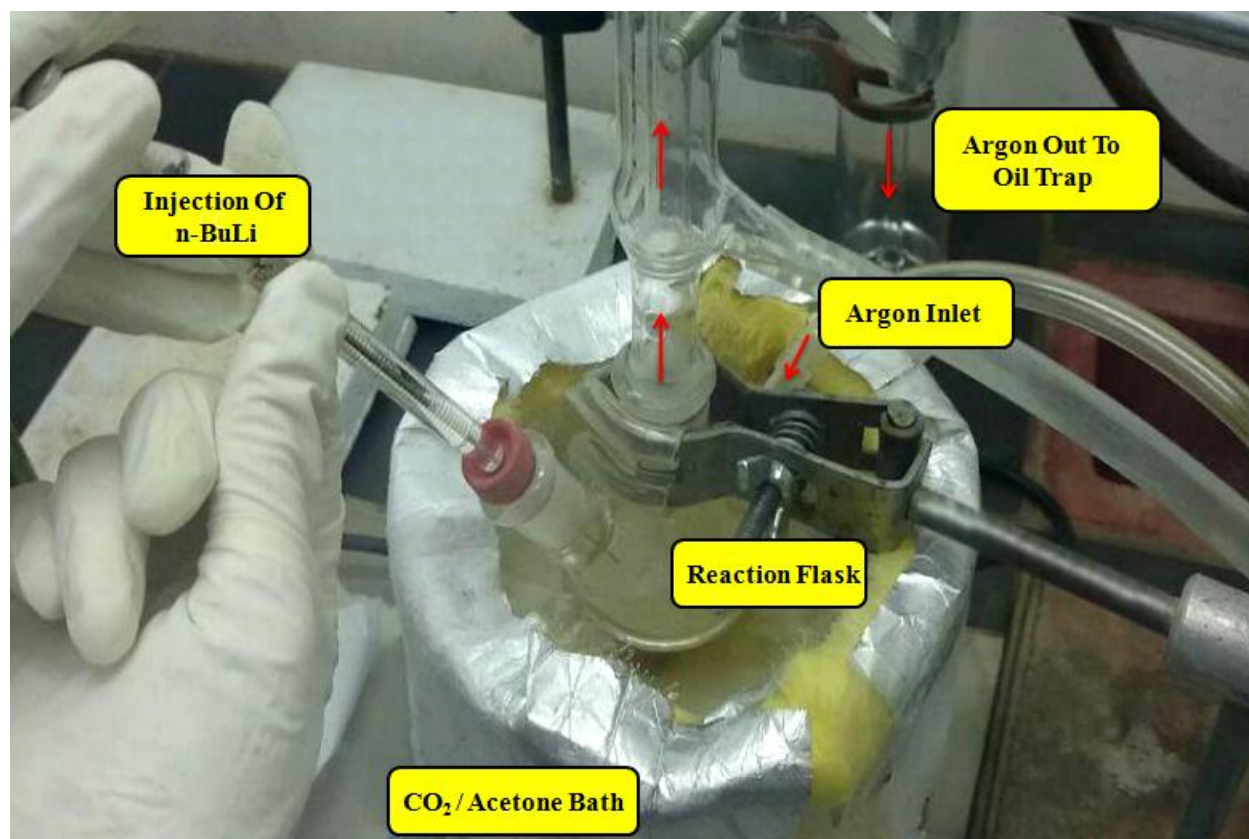


Fig. S4. Experimental equipments used for ZIF-8@{Mo₁₃₂}@Li⁺ nano-composite synthesis

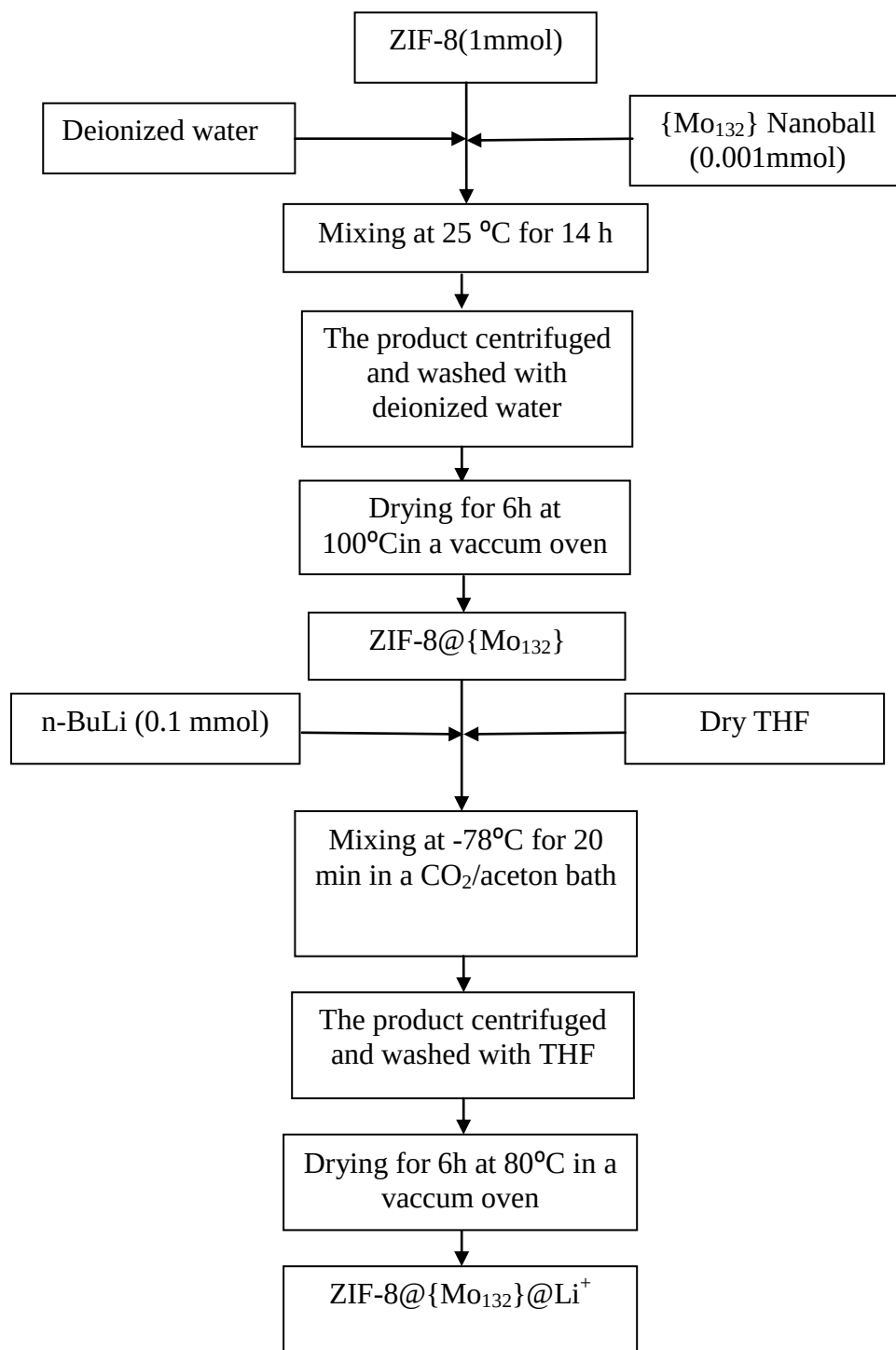


Fig. S5. The flow chart for preparation of ZIF-8@{Mo₁₃₂}@Li⁺ nano-composite

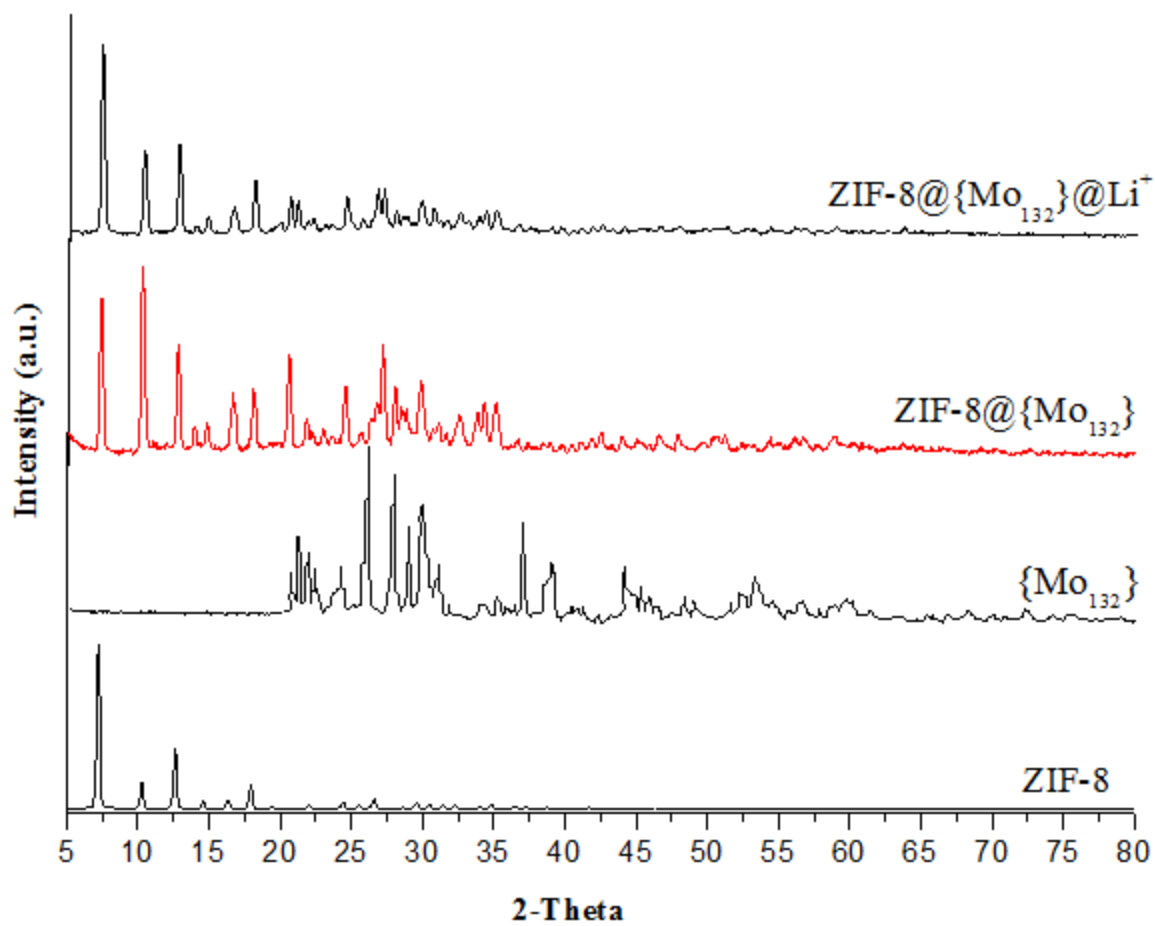


Fig. S6. XRD patterns of ZIF-8, {Mo₁₃₂}, ZIF-8@{Mo₁₃₂} and ZIF-8@{Mo₁₃₂}@Li⁺ nano-composite

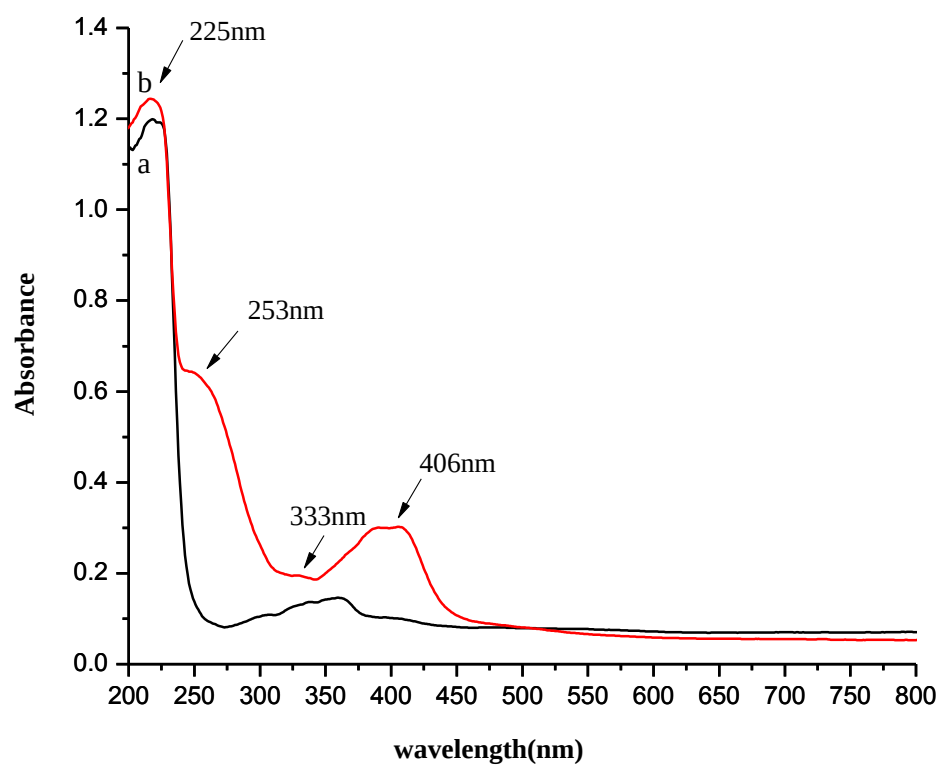


Fig. S7. UV-Vis-spectra of (a) ZIF-8 and (b) ZIF-8@{Mo₁₃₂} nano-composite

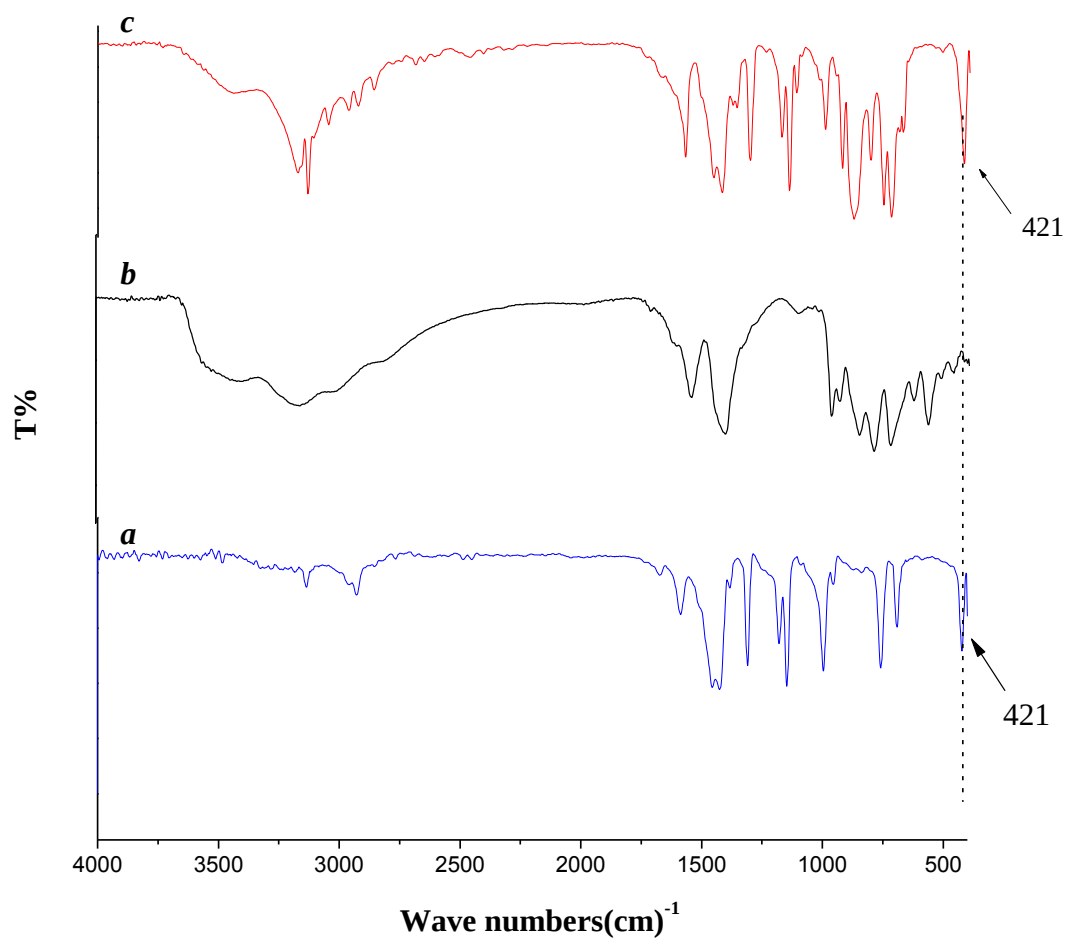


Fig. S8. FT-IR spectra of (a) ZIF-8 (b) {Mo₁₃₂} (c) ZIF-8@{Mo₁₃₂}

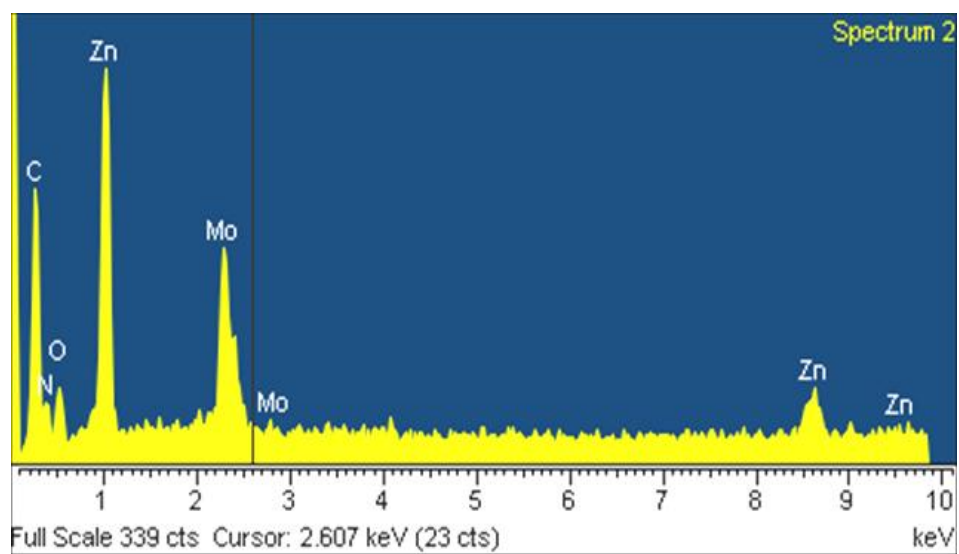


Fig. S9. EDX analysis of the ZIF-8@{Mo₁₃₂} nano-composite

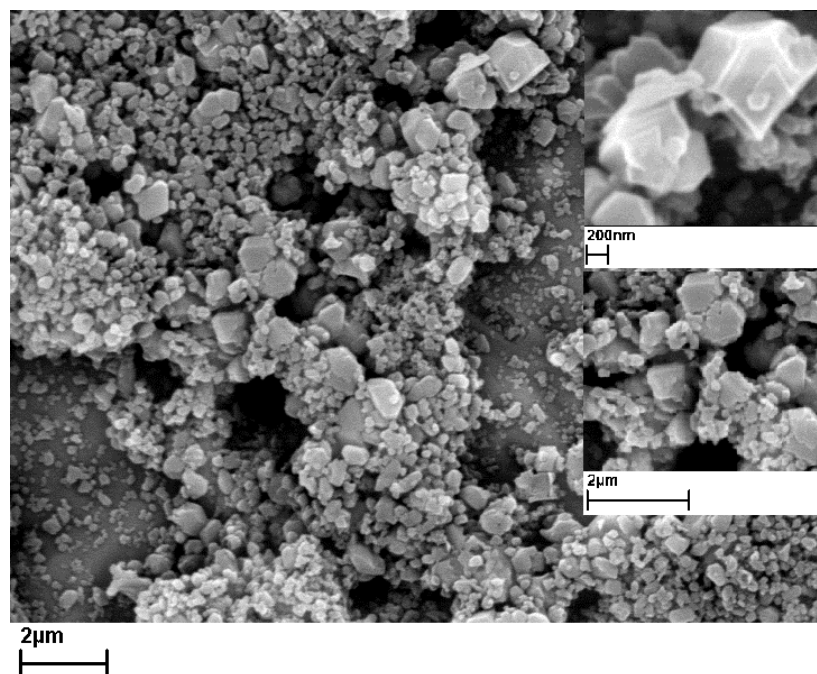


Fig. S10. SEM images of ZIF-8

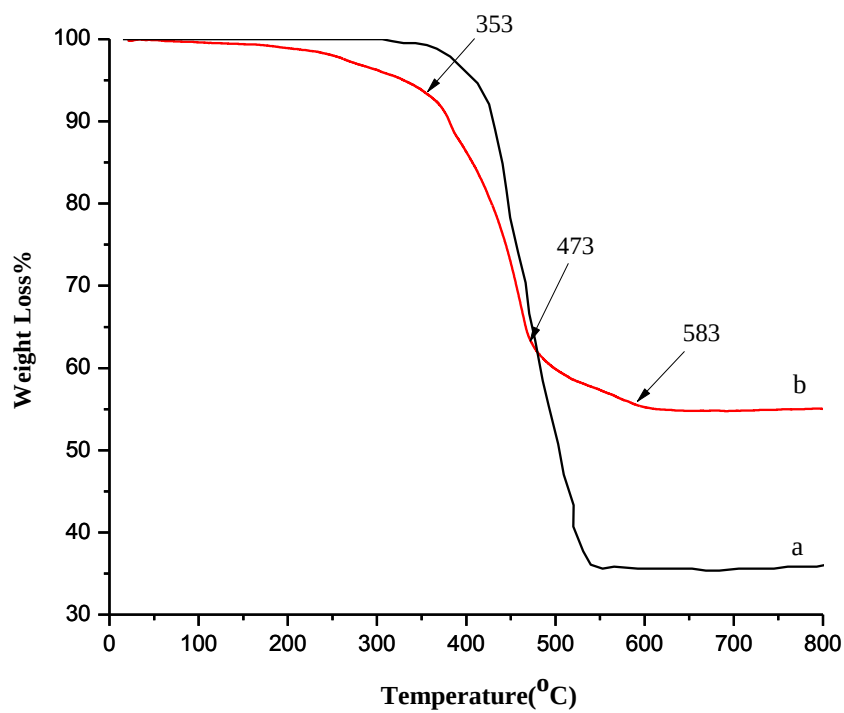


Fig. S11. TGA curve (a) of the ZIF-8 (b) as-synthesized ZIF-8@{Mo₁₃₂} nano-composite

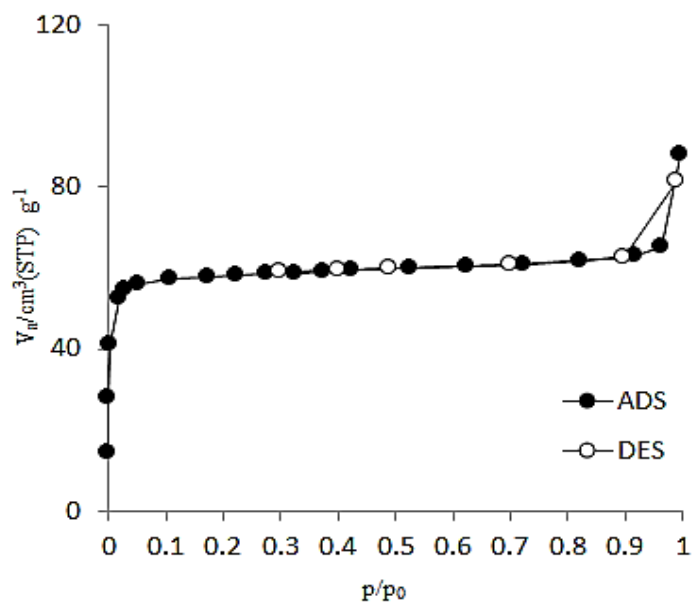


Fig. S12. The N₂ adsorption-desorption isotherms for ZIF-8@{Mo₁₃₂} nano-composite at 77K.

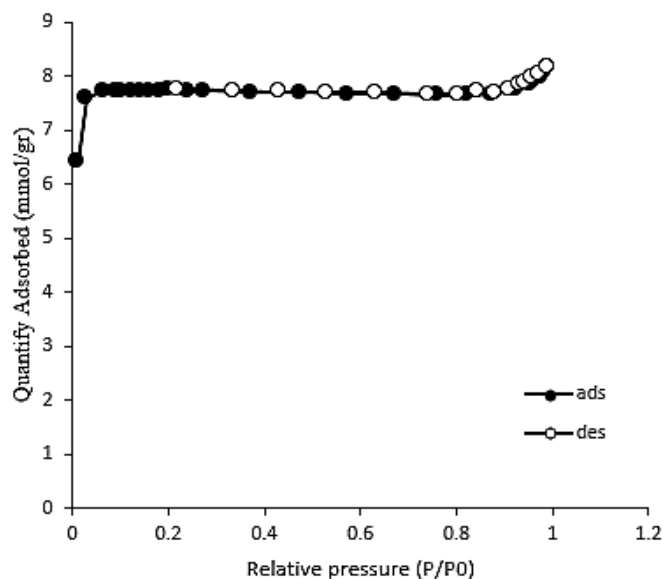


Fig. S13. The N_2 adsorption-desorption isotherms for ZIF-8@{Mo₁₃₂}@Li⁺ nano-composite at 77K.

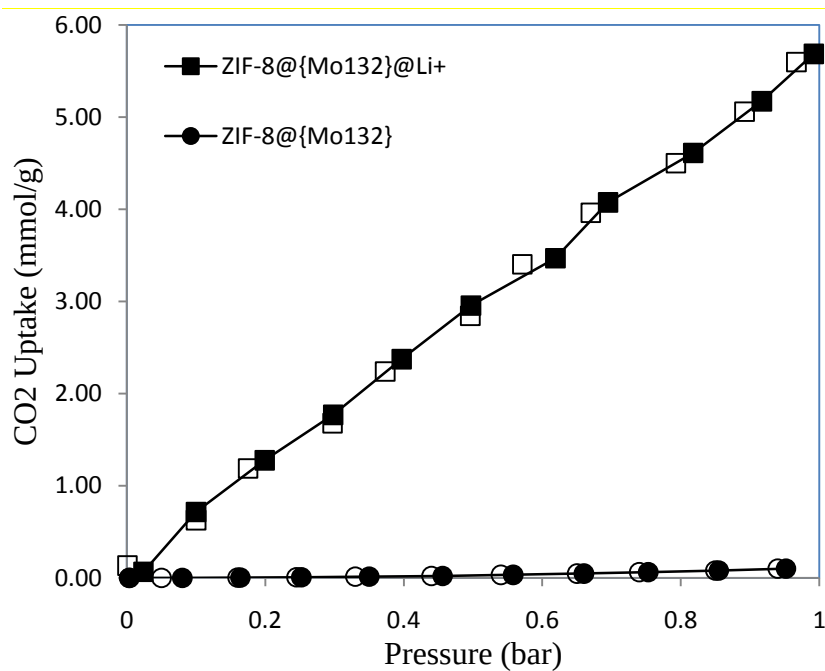


Fig. S14. The CO_2 adsorption-desorption isotherms for ZIF-8@{Mo₁₃₂}@Li⁺ nano-composite in comparison with ZIF-8@{Mo₁₃₂} intermediate composite at 298K.

Table S1. The ICP analysis performance of ZIF-8@{Mo132}@Li+ nano-composite to obtain the actual content of Li+ in the structure

		Printing Date:	07/02/2018 12:08:26
		Current User:	spectro
Method Name:	30MULTIELEMENT METHO	Creation Date:	2018-01-24 09:27:49
Method Autor:	spectro	Last Change:	2018-02-07 12:07:40

Sample Name: ZIF-8@{(Mo132)}@Li+				Sample Type: ZIF-8@{(Mo132)}@Li+					
Measure Date: 2018-02-07 11:4		Recalc. Date: 2018-02-07 12		State: Recalculated		Quality: Recalc Drifted		Total Dilution: 10.000000	
Sample Identification									
Sample Name	DILUTION	VOLUME	WEIGHT						
sample-1:10	10.000000	20	0.029						
	LI.								
Conc 1	6460.94[ppm]								
Conc 2	6490.51[ppm]								
Conc MinRange	---								
Conc Mean	6475.72[ppm]								
Conc MaxRange	---								
Reported	6475.72[ppm]								

Table S2. The ICP analysis performance of ZIF-8@{Mo132}@Li+ nano-composite to obtain the actual content of {Mo₁₃₂} in the structure

		Printing Date:	07/02/2018 08:02:34
		Current User:	spectro
Method Name:	30MULTIELEMENT METHO	Creation Date:	2018-01-24 09:27:49
Method Autor:	spectro	Last Change:	2018-02-07 08:02:18

Sample Name: ZIF-8@{Mo132}@Li+				Sample Type: ZIF-8@{Mo132}@Li+					
Measure Date: 2018-02-06 14:3		Recalc. Date: 2018-02-07 08		State: Recalculated		Quality: Recalc Drifted		Total Dilution: 100.000000	
Sample Identification									
Sample Name	DILUTION	VOLUME	WEIGHT						
SAMPLE MS NIKN	100.000000	1	0.029						
	Mo..								
Conc 1	35398.9[ppm]								
Conc 2	34446.9[ppm]								
Conc MinRange	---								
Conc Mean	34922.9[ppm]								
Conc MaxRange	---								
Reported	34922.9[ppm]								

Table S3. Textural properties of as-synthesized ZIF-8@{Mo₁₃₂} and ZIF-8@{Mo₁₃₂}@Li⁺ composites

Sample	BET surface area (m ² /g)	Langmuir surface area (m ² /g)	Average pore size* (°A)	Pore volume** (cc/g)
ZIF-8@{Mo ₁₃₂ }	591	777	19.4	0.29
ZIF-8@{Mo ₁₃₂ }@Li ⁺	237	312	21.9	0.13

* Calculated by BET (4V/A)

**Calculated at P/P₀= 0.99