A facile synthesis strategy to couple porous nanocube of CeO<sub>2</sub> with Ag nanoparticles: An excellent catalyst with enhanced reactivity for 'click reaction' and carboxylation of terminal alkynes

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**Figure SI-1.** TEM images of 3-D porous  $CeO_2$  nanocube prepared by hot injection method at (A) low magnification, (B) high magnification and (C) the corresponding EDX.



Figure SI-2. TEM image of DMP-modified  $CeO_2$  nanocubes. No deformation due to surface modification with DMP.

**Table SI-1** Synthesis of 1, 4-disubstituted-1, 2, 3-triazole in the presence of different catalyst in the model reaction.

Entry	Catalyst	Time (h)	Yield <sup>b</sup> (%)
1	CeO <sub>2</sub> -Ag nanocomposites	3	98
2	CeO <sub>2</sub>	12	15
3	Ag-nps	8	44
4	Ag-HCP	6	68
5	Ag-MPTiO <sub>2</sub>	6	75
6	AgCl	12	23
7	AgNO <sub>3</sub>	12	18

<sup>a</sup>Reaction conditions: phenyl acetylene (1.2 equiv); aromatic amine (1equiv); Water (10 ml); catalyst (20 mg), room temperature. <sup>b</sup>Isolated yields.

Entry	Solvent	Time (h)	Yield <sup>b</sup> (%)
1	Toluene	6	12
2	DMF	3	88
3	DMSO	3	80
4	THF	5	68
5	ACN	6	47
6	Water	3	98

Table SI-2 Optimization of the solvent for the formation of 1, 4-disubstituted-1, 2, 3-triazole.

<sup>a</sup>Reaction conditions: phenyl acetylene (1.2 equiv); aromatic amine (1equiv); solvent (10 ml); CeO<sub>2</sub>-Ag nanocomposites catalyst (20 mg), room temperature. <sup>b</sup>Isolated yields.

Table SI-3. <sup>1</sup>H NMR data of isolated 1, 4-disubstituted 1, 2, 3-triazoles.





Entry	Base	Solvent	Temperature	Time	Yield <sup>b</sup>
			( <sup>0</sup> C)	(h)	(%)
1	Na <sub>2</sub> CO <sub>3</sub>	DMF	80	12	23
2	K <sub>2</sub> CO <sub>3</sub>	DMF	80	12	49
3	Cs <sub>2</sub> CO <sub>3</sub>	DMF	80	12	98
4	DBU	DMF	80	12	34
5	DBN	DMF	80	12	37
6	Et <sub>3</sub> N	DMF	80	12	Trace
7	KO <sup>t</sup> Bu	DMF	80	12	11
8	Cs <sub>2</sub> CO <sub>3</sub>	DMSO	80	12	62
9	Cs <sub>2</sub> CO <sub>3</sub>	THF	80	12	44
10	Cs <sub>2</sub> CO <sub>3</sub>	Dioxane	80	12	28
11	Cs <sub>2</sub> CO <sub>3</sub>	Toluene	80	12	No reaction
12	Cs <sub>2</sub> CO <sub>3</sub>	DMF	25	12	12
13	Cs <sub>2</sub> CO <sub>3</sub>	DMF	50	12	39
14	Cs <sub>2</sub> CO <sub>3</sub>	DMF	70	12	81
15	Cs <sub>2</sub> CO <sub>3</sub>	DMF	90	12	98
16	Cs <sub>2</sub> CO <sub>3</sub>	DMF	80	4	24
17	Cs <sub>2</sub> CO <sub>3</sub>	DMF	80	8	52
18	Cs <sub>2</sub> CO <sub>3</sub>	DMF	80	10	78
19	Cs <sub>2</sub> CO <sub>3</sub>	DMF	80	14	98

**Table SI-4** Effect of various reaction parameters on carboxylation of 1-phenylethyne.

<sup>a</sup>Reaction conditions: alkyne (1.0 mmol), 30 mg of CeO<sub>2</sub>-Ag nanocomposites, base (1.5 mmol), CO<sub>2</sub> (1.0 atm), solvent (5 ml), <sup>b</sup>GC Yield.



Figure SI-3. (A) TEM and (B) EDX analysis of CeO<sub>2</sub>-Ag nanocomposite catalyst recovered after 5 cycles.



Figure SI-4. XRD of  $CeO_2$ -Ag catalyst recovered after 5 cycles. X (hkl) denotes the (hkl) plane of  $CeO_2$ .



Figure SI-5. XPS spectra of (A) Ce 3d, (B) O 1s, (C) N 1s and (D) Ag 3d from CeO<sub>2</sub>-Ag catalyst recovered after 5 cycles.