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Colorimetric and Fluorometric Turn-on Sensor for Selective Detection of Fluoride Ion:

Sol-Gel Transition Studies and Theoretical Insights

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Contents

- 1. Fig. S1 FT-IR spectrum of receptor R1
- 2. Fig. S2 ¹H-NMR spectrum of receptor R1
- 3. Fig. S3 ESI-MS spectrum of receptor R1
- 4. Fig. S4 ¹³C-NMR spectrum of receptor R1
- 5. Fig. S5 Job's plot for R1-TBAF complex
- 6. Fig. S6 Color change of R1 with the addition of 2 equiv. of TBAOH
- 7. Fig. S7 Fluorometric response of R1 with the addition of 2 equiv. of TBAOH
- 8. Fig. S8 UV-Vis spectra of R1 with the addition of 2 equiv. of TBAOH
- 9. Fig. S9 PL spectra of R1 with the addition of 2 equiv. of TBAOH
- 10. Fig. S10 DFT derived UV-Vis spectra of the R1 and R1+ F^- complex
- 11. Calculation of binding constant and detection limit from UV-Vis studies
- 12. Table S1 Gelation properties of R1



Fig. S1 FT-IR spectrum of receptor R1



Fig. S2 ¹H-NMR spectrum of receptor R1



. S3 ESI-MS spectrum of receptor R1



Fig. S4 ¹³C NMR spectrum of receptor R1



Fig. S5 Job's plot for R1-TBAF complex



Fig. S6 Color change of R1 with the addition of 2 equiv. of TBAOH



Fig. S7 Fluorometric response of R1 with the addition of 2 equiv. of TBAOH



Fig. S8 UV-Vis spectra of R1 with the addition of 2 equiv. of TBAOH



Fig. S9 PL spectra of R1 with the addition of 2 equiv. of TBAOH



Fig. S10 DFT derived UV-Vis spectra of the R1 and R1+ F⁻ complex

Table S1 Gelation properties of R1

| Solvent | State |
|---|--------------|
| DMSO | G (26 mg/mL) |
| DMF | G (26 mg/mL) |
| Cyclohexane | S |
| Chloroform | S |
| Methanol | S |
| Acetonitrile | G (26 mg/mL) |
| n-Hexane | S |
| Diethyl ether | S |
| DCM | S |
| THF | S |
| DMF : H ₂ O (1:2 , v/v) | Ι |
| DMSO: H ₂ O (1:2 , v/v) | Ι |
| Propyl alcohol | S |
| Ethanol | S |
| Dioxane | S |
| S = solution; $G =$ gel (minimum gelatination concentration); $I =$ insoluble | |

Calculation of binding constant from UV-Vis studies:

Binding constant has been calculated using Benesi-Hildebrand equation¹ as given below;

$1/(A-A_0) = 1/(A_{max} - A_0) + 1/K [X_]^n (A_{max} - A_0)$

where, A_0 , A, A_{max} are the absorption considered in the absence of anion, at an intermediate, and at a concentration of saturation. K is binding constant, $[X^-]$ is concentration of anion and n is the stoichiometric ratio.

Calculation of Detection limit:

The limit of detection was found using this equation.² DL = $C_L \times C_T$ C_L = Conc. of receptor; C_T = Conc. of Titrant at which change was observed.

References:

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- 2. V. Bhalla, A. Gupta, M. Kumar., Chem. Commun., 2012, 48, 11862.