

## **Electronic Supplementary Information**

**A pyrene-benzimidazole composed effective fluoride sensor: potential mimicking of a Boolean logic gate**

**Archana Kushwaha<sup>a</sup>, Sagar K. Patil<sup>a</sup> and Dipanwita Das<sup>a,\*</sup>**

<sup>a</sup>*Department of Chemistry, Institute of Chemical Technology, Matunga, Mumbai 400019  
(India)*

Fax: (91)022-3361-1020

E-mail: [dr.das@ictmumbai.edu.in](mailto:dr.das@ictmumbai.edu.in)

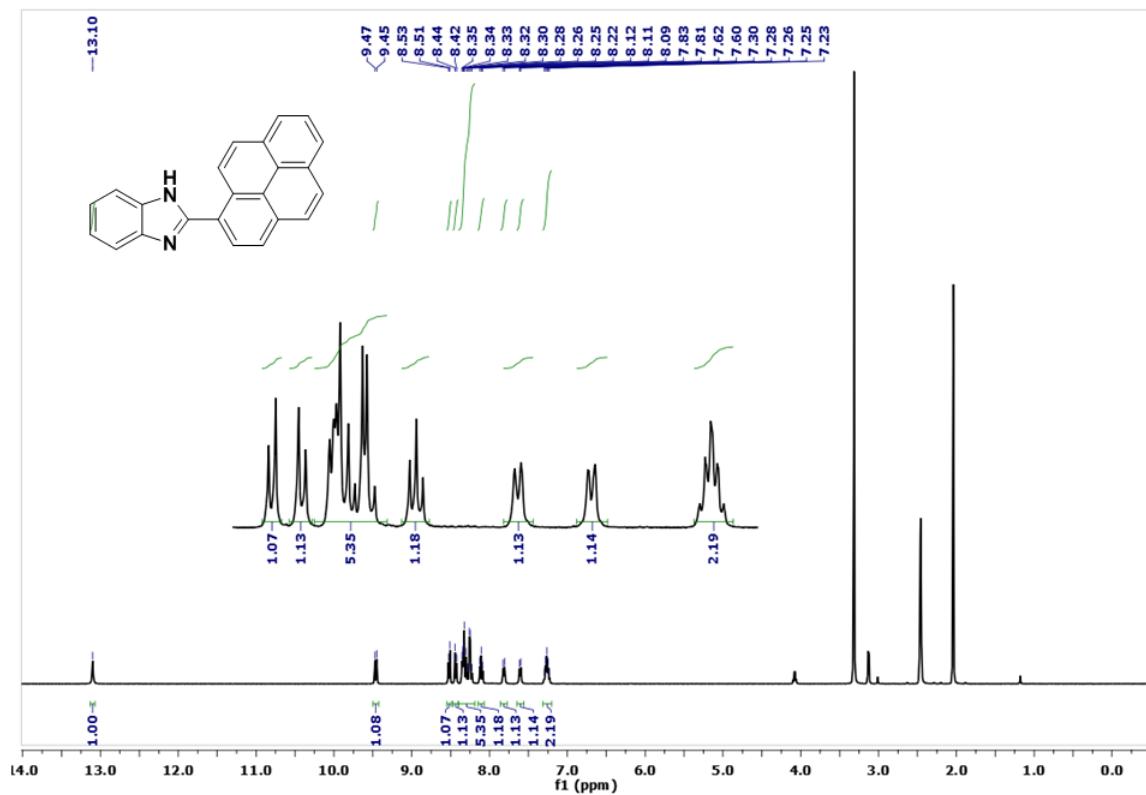
	<b>CONTENT</b>	
<b>1.</b>	Crystallographic data for receptor <b>2</b>	<b>Table S1–S2</b>
<b>2.</b>	$^1\text{H}$ NMR spectrum of receptor <b>2</b> and <b>3</b>	<b>S1–S2</b>
<b>3.</b>	$^{13}\text{C}$ NMR spectrum of receptor <b>2</b>	<b>S3</b>
<b>4.</b>	$^{13}\text{C}$ NMR spectrum of receptor <b>3</b>	<b>S4</b>
<b>5.</b>	High resolution mass spectrum of receptor <b>2</b>	<b>S5</b>
<b>6.</b>	High resolution mass spectrum of receptor <b>3</b>	<b>S6</b>
<b>7.</b>	HOMO-LUMO composition and transition energy of receptor <b>2</b>	<b>S7</b>
<b>8.</b>	Job plot for the determination of the stoichiometry of receptor <b>2</b> with $\text{F}^-$ ions	<b>S8</b>
<b>9.</b>	Benesi-Hildebrand plot for receptor <b>2</b> with $\text{F}^-$ ions	<b>S9</b>
<b>10.</b>	Competitive selectivity of <b>2</b> ( $2 \times 10^{-5}$ M in DMSO) towards $\text{F}^-$ .	<b>S10</b>
<b>11.</b>	Changes in UV-vis spectra of <b>1</b> (a) and <b>3</b> (b) with different TBA salts	<b>S11</b>
<b>12.</b>	$\text{pK}_\text{a}$ of receptor <b>2</b> and receptor <b>3</b> in aqueous solution	<b>S12</b>
<b>13.</b>	Emission spectra of receptor <b>3</b> with addition of different anions as their TBA salts	<b>S13</b>
<b>14.</b>	Plot of absorbance versus anion concentration for receptor <b>2</b> for detection limit calculation	<b>S14</b>
<b>15.</b>	UV-vis titration spectra of receptor <b>2</b> with TBAOH	<b>S15</b>
<b>16.</b>	Reversibility of receptor <b>2</b> with alternate addition of $\text{F}^-$ and $\text{H}^+$	<b>S16</b>

**Table S1** Selected bond distances ( $\text{\AA}$ ) and bond angles ( $^\circ$ ) for receptor **2**.

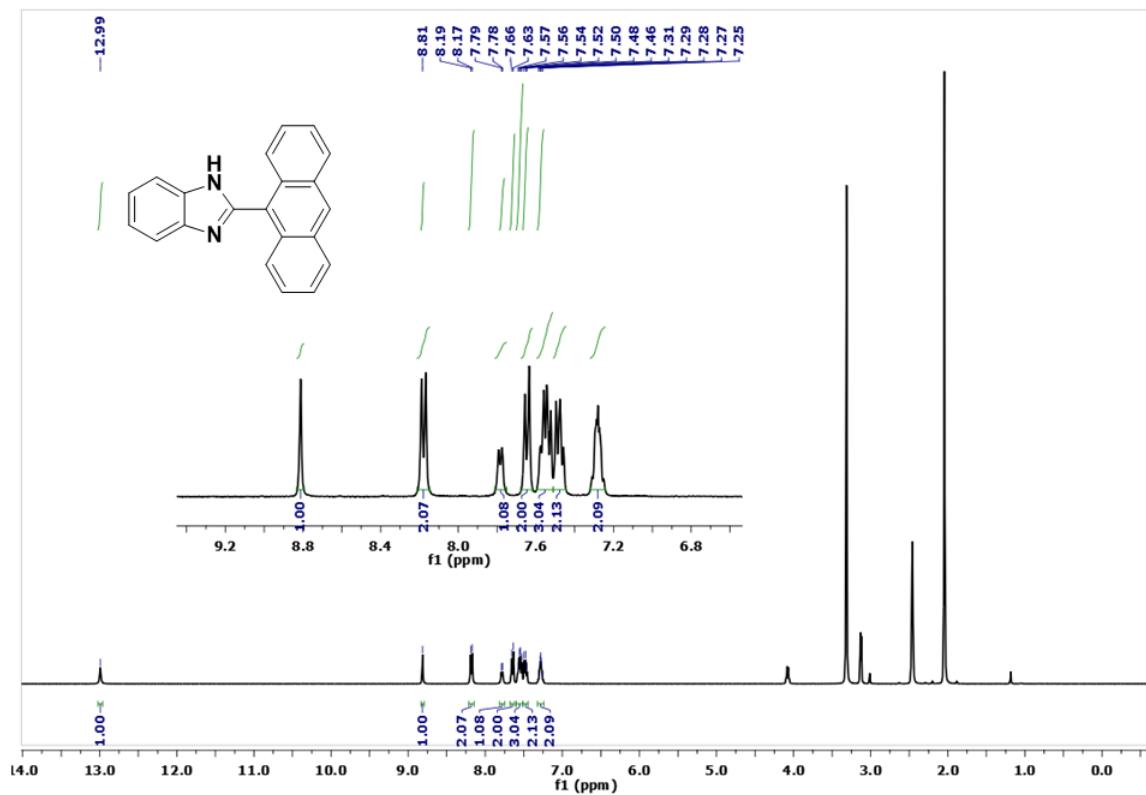
2	
C(4)–C(5)	1.389(2)
C(5)–N(2)	1.374(19)
C(5)–C(6)	1.399(19)
C(6)–N(1)	1.389(18)
C(7)–N(1)	1.326(18)
C(7)–N(2)	1.358(17)
C(7)–C(8)	1.472(2)
C(8)–C(9)	1.394(2)
C(8)–C(21)	1.410(2)
N(2)–H(1B)	0.931(17)
C(6)–C(1)–H(1)	121.0
C(5)–C(4)–H(4)	121.6
N(2)–C(5)–C(4)	132.12(13)
N(2)–C(5)–C(6)	105.58(12)
C(4)–C(5)–C(6)	122.27(14)
C(1)–C(6)–N(1)	130.43(13)
C(1)–C(6)–C(5)	119.99(13)
N(1)–C(6)–C(5)	109.55(12)
N(1)–C(7)–N(2)	112.56(12)
N(1)–C(7)–C(8)	125.57(12)
N(2)–C(7)–C(8)	121.65(13)
C(9)–C(8)–C(21)	119.66(14)
C(9)–C(8)–C(7)	117.51(13)
C(21)–C(8)–C(7)	122.77(13)
C(8)–C(21)–C(20)	123.34(14)
C(7)–N(1)–C(6)	104.99(11)
C(7)–N(2)–C(5)	107.32(12)
C(7)–N(2)–H(1B)	125.9(10)

**Table S2.** Intermolecular hydrogen bond distances ( $\text{\AA}$ ) and bond angles ( $^\circ$ ) in receptor **2**.

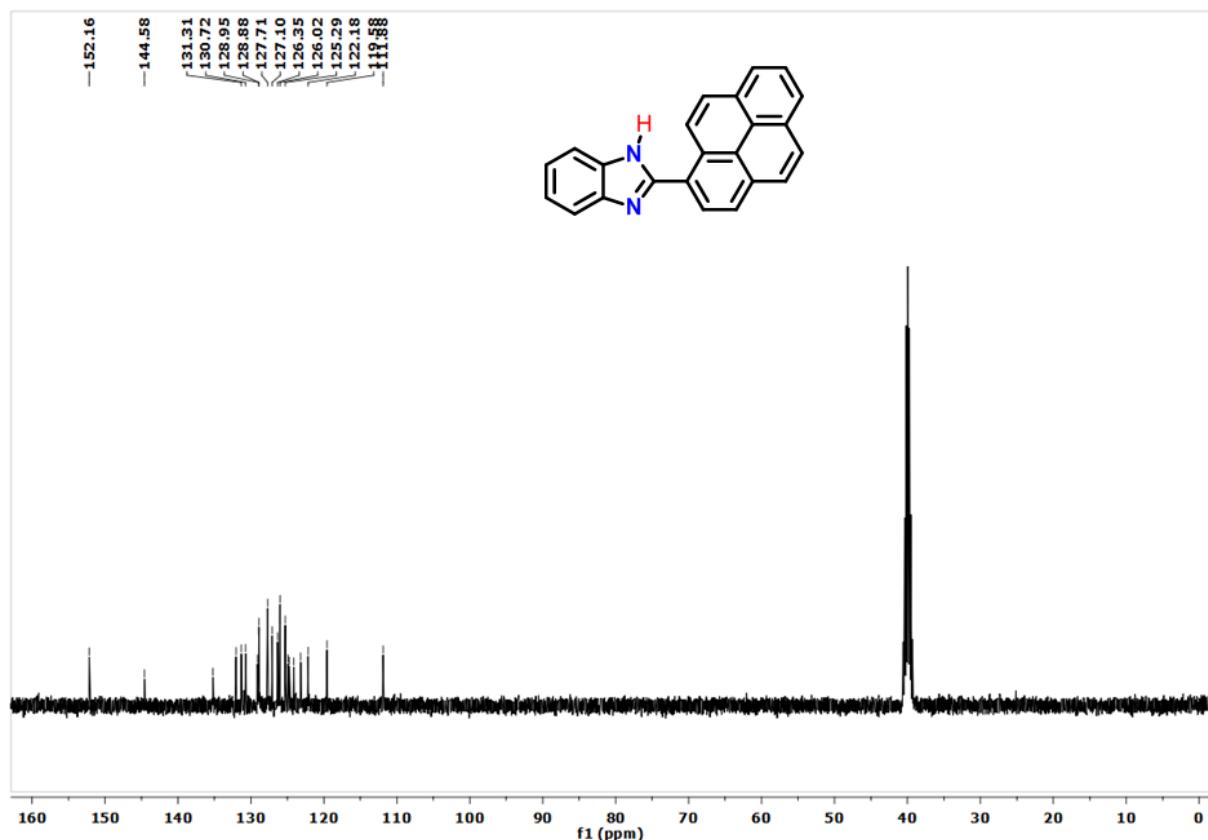
D–H $\cdots$ A	d(D–H)	d(H $\cdots$ A)	D $\cdots$ A	$\angle$ (DHA)
N(2)–H(1B) $\cdots$ N(1)	0.931(17)	1.954(18)	2.8638(16)	165.4(15)



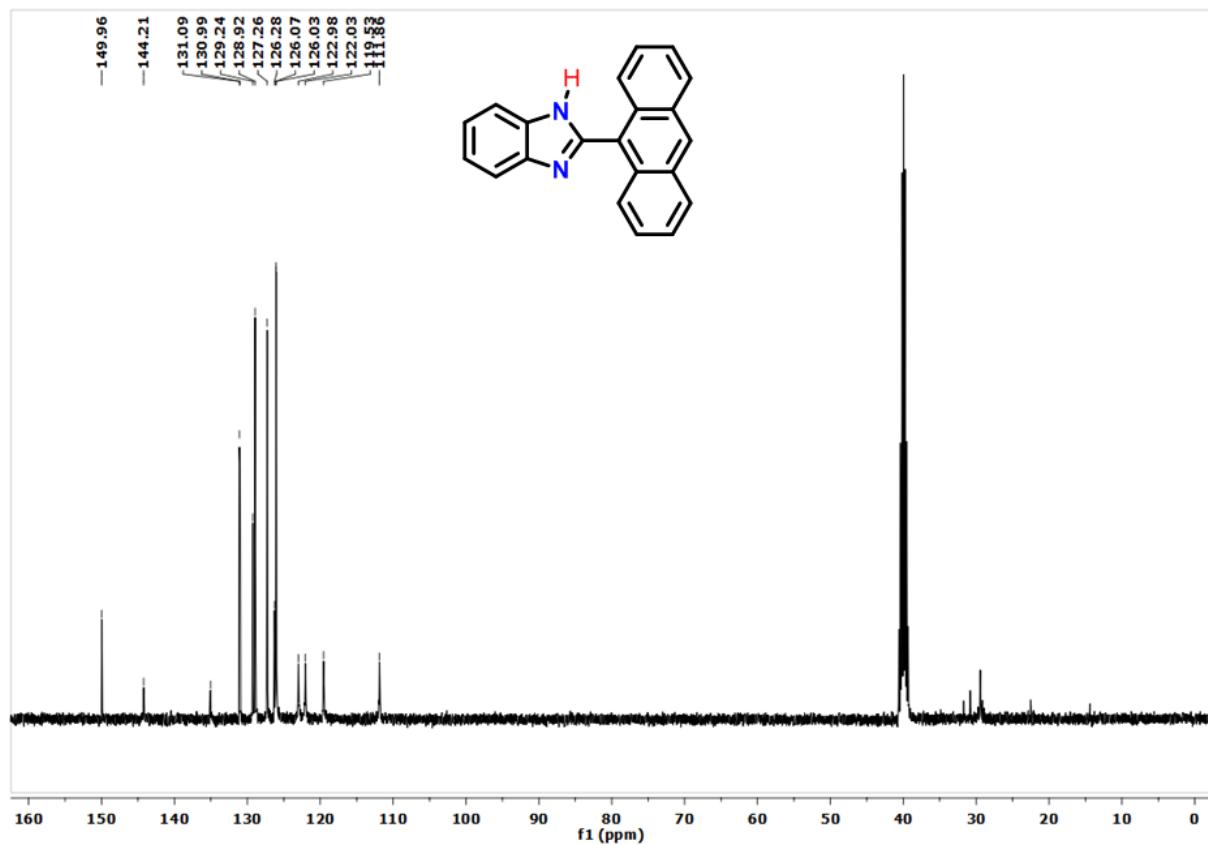
**Fig. S1**  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ ) spectrum of receptor **2**.



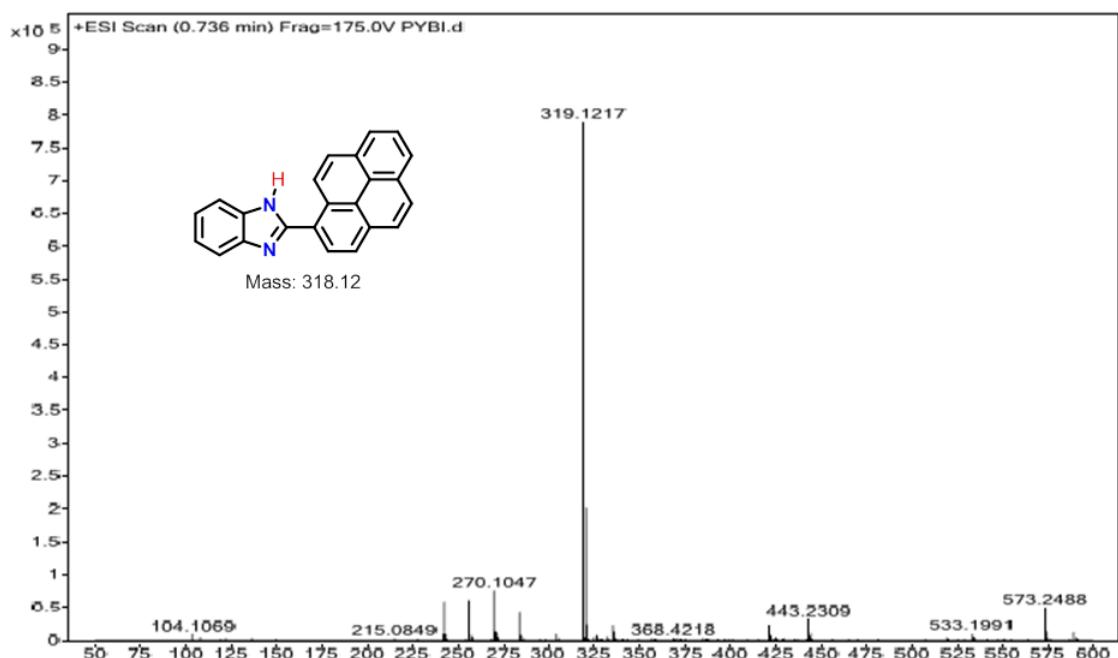
**Fig. S2**  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ ) spectrum of receptor **3**.



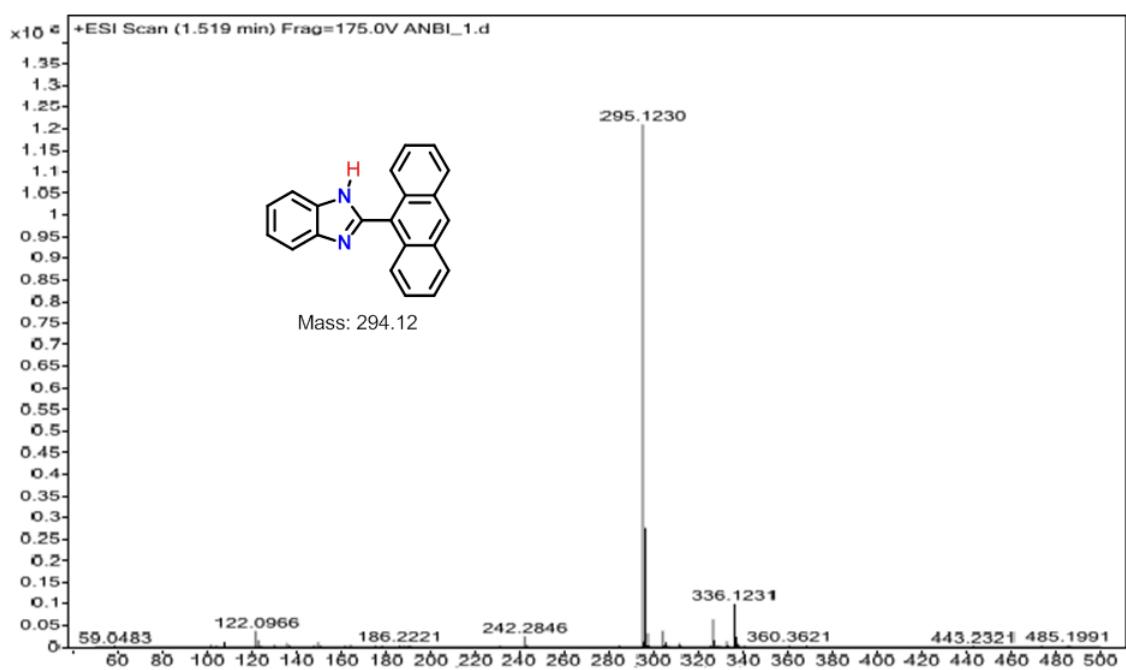
**Fig. S3**  $^{13}\text{C}$  NMR (101 MHz,  $\text{DMSO}-d_6$ ) spectrum of receptor 2.



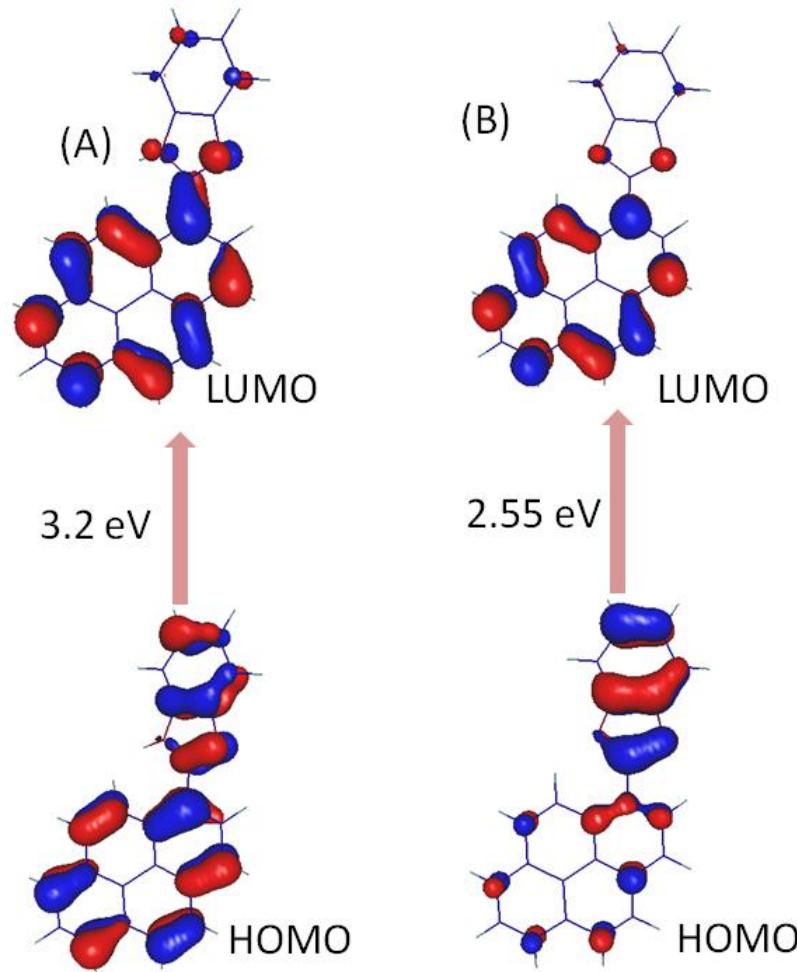
**Fig. S4**  $^{13}\text{C}$  NMR (101 MHz,  $\text{DMSO}-d_6$ ) spectrum of receptor 3.



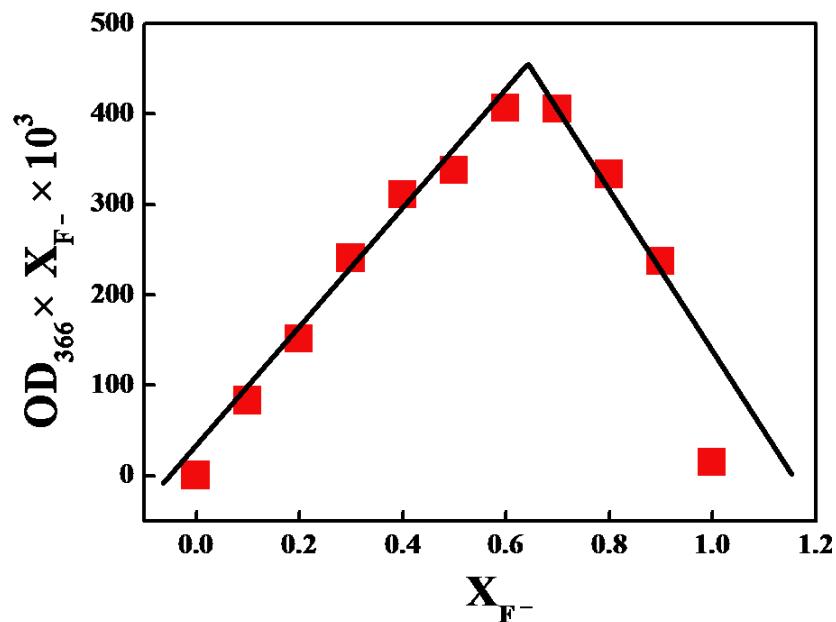
**Fig. S5** High resolution mass spectrum (HRMS) of receptor **2**.



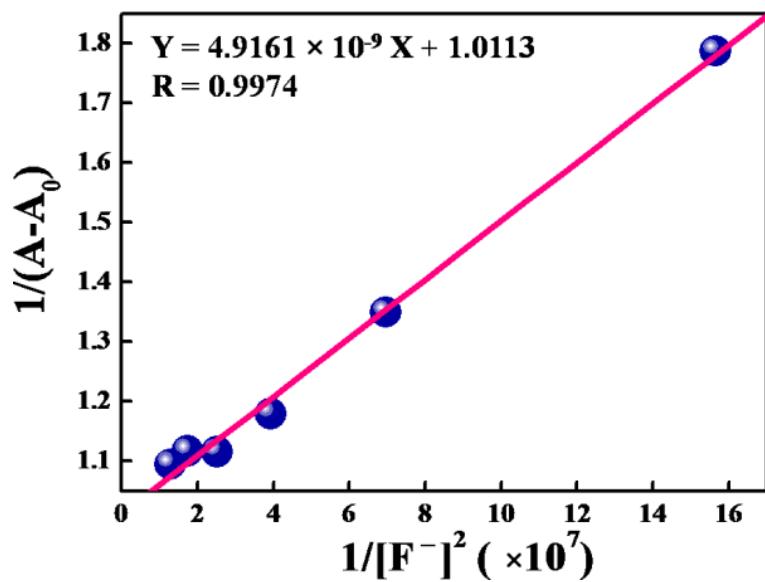
**Fig. S6** High resolution mass spectrum (HRMS) of receptor **3**.



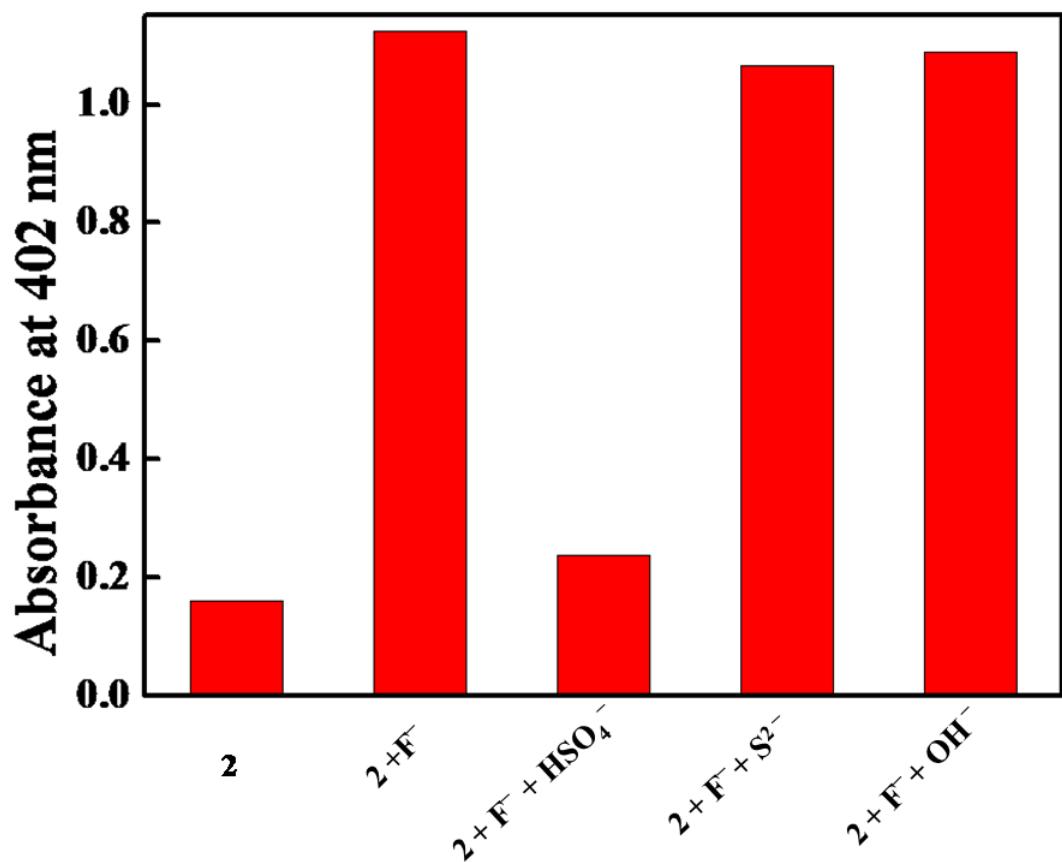
**Fig. S7** HOMO-LUMO composition and transition energy of receptor **2** in neutral form (A) and in F<sup>-</sup> induced deprotonated form (B).



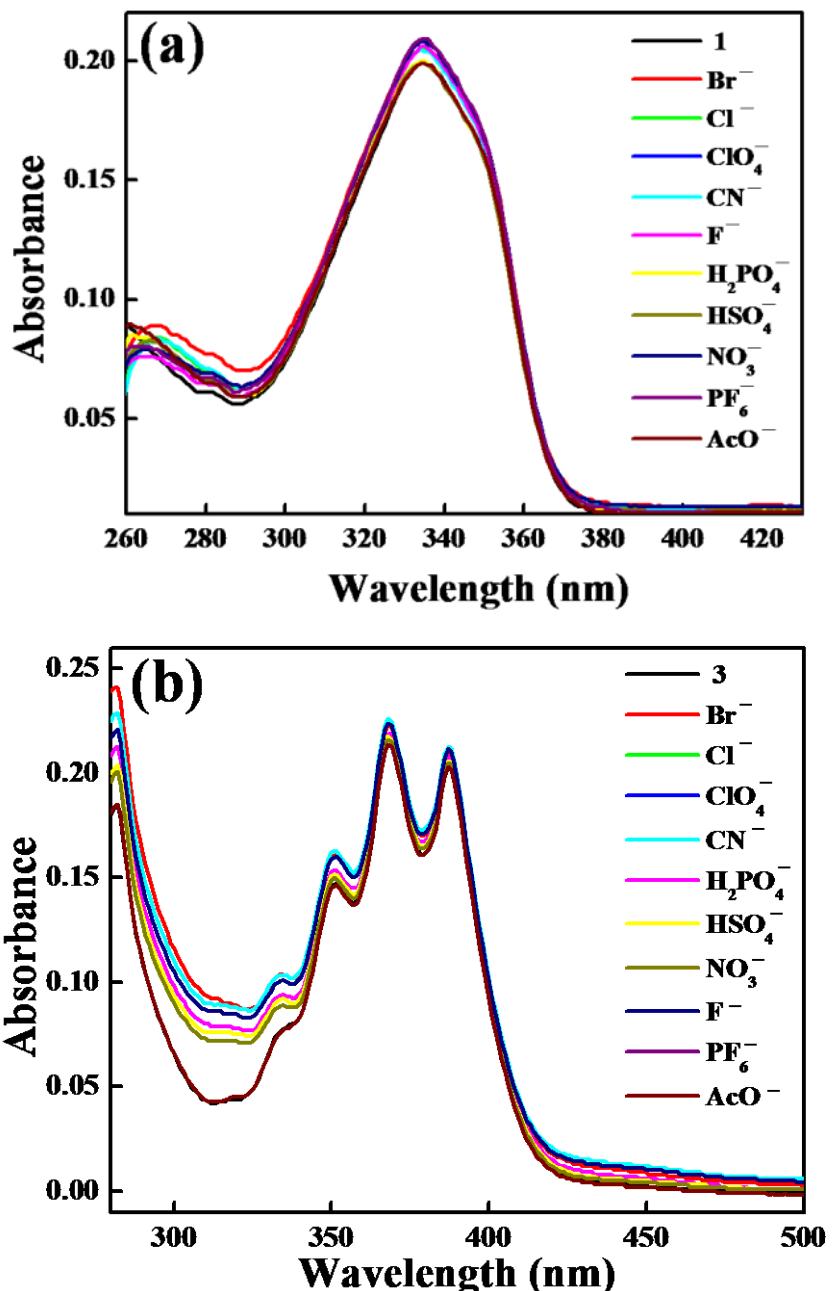
**Fig. S8** Job plot for the determination of the stoichiometry of receptor **2** with F<sup>-</sup> ions.



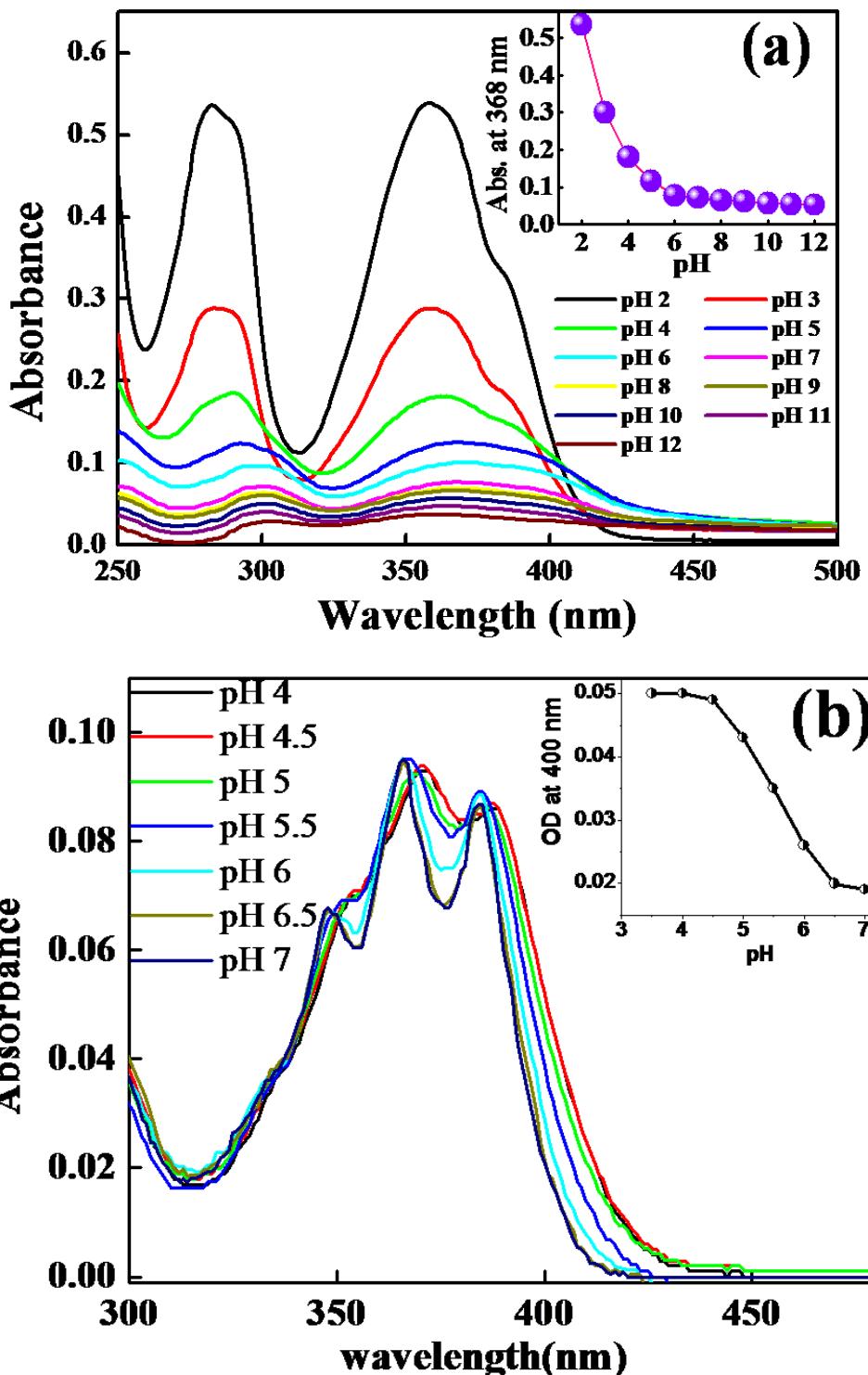
**Fig. S9** Benesi-Hildebrand plot for receptor **2** with  $F^-$  ions.



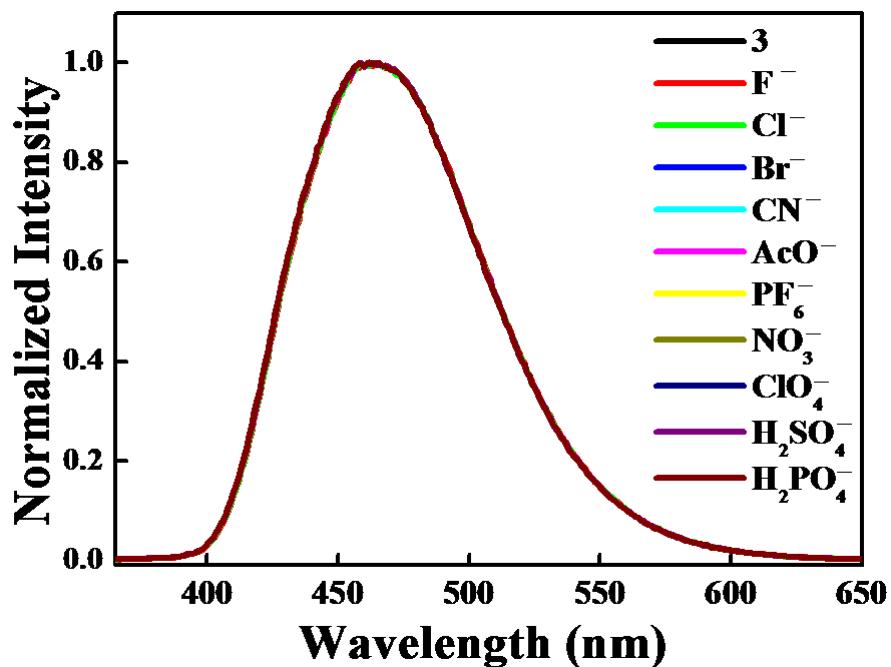
**Fig. S10** Absorption spectral changes (402 nm) of competitive selectivity of 2 ( $2 \times 10^{-5}$  M in DMSO) towards F<sup>-</sup> (20 equiv.) in the presence of HSO<sub>4</sub><sup>-</sup>, S<sup>2-</sup>, OH<sup>-</sup> ions (20 equiv.).



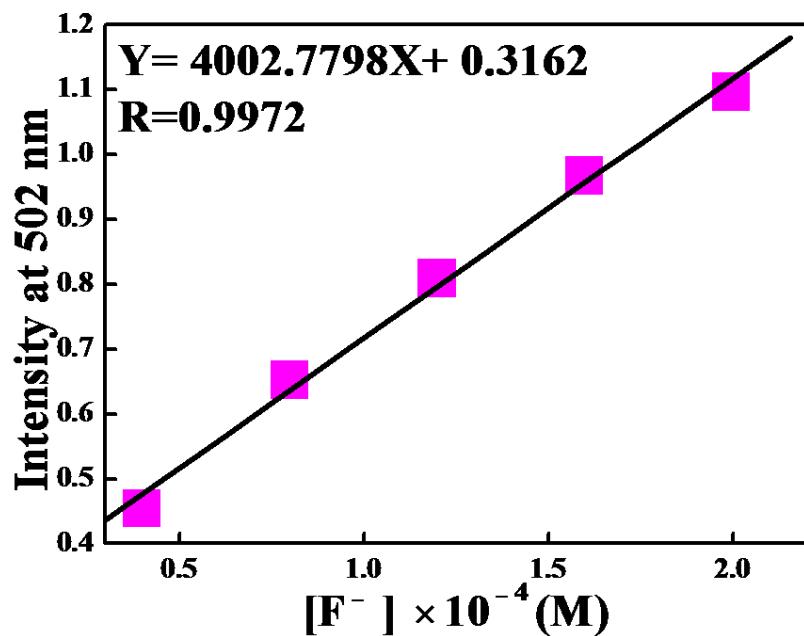
**Fig. S11** Changes in UV-vis spectra of receptor **1**(a) and **3**(b) ( $2 \times 10^{-5}$  M in DMSO) with addition of different anions as their TBA salts.



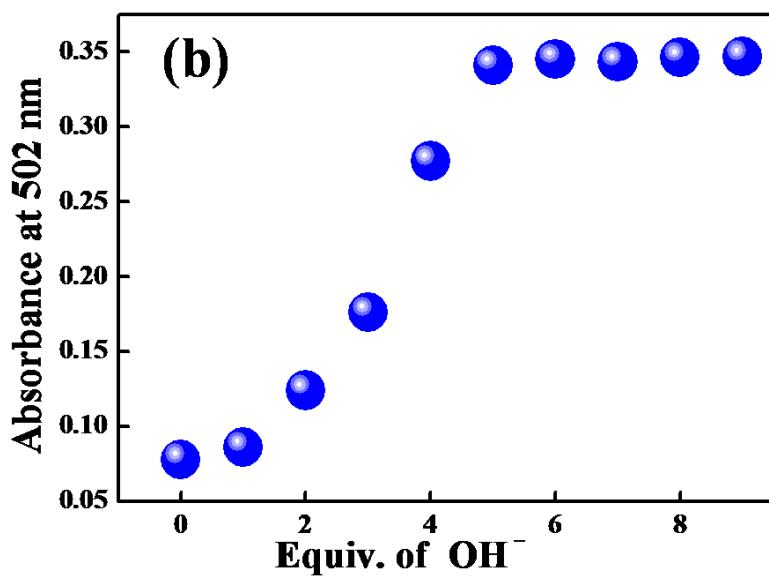
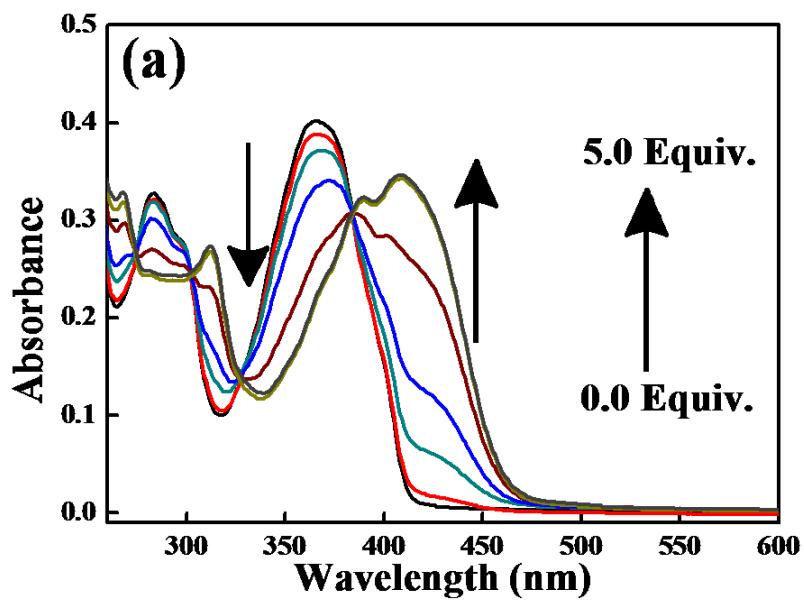
**Fig. S12** Colorimetric pH titrations of receptor **2**(a) and **3**(b) in aqueous solution. Insets show the variation of absorbance at low energy peak as a function of pH which suggests pKa of receptor **2** and receptor **3** to be  $\sim 3.5$  and  $5-6$  respectively.



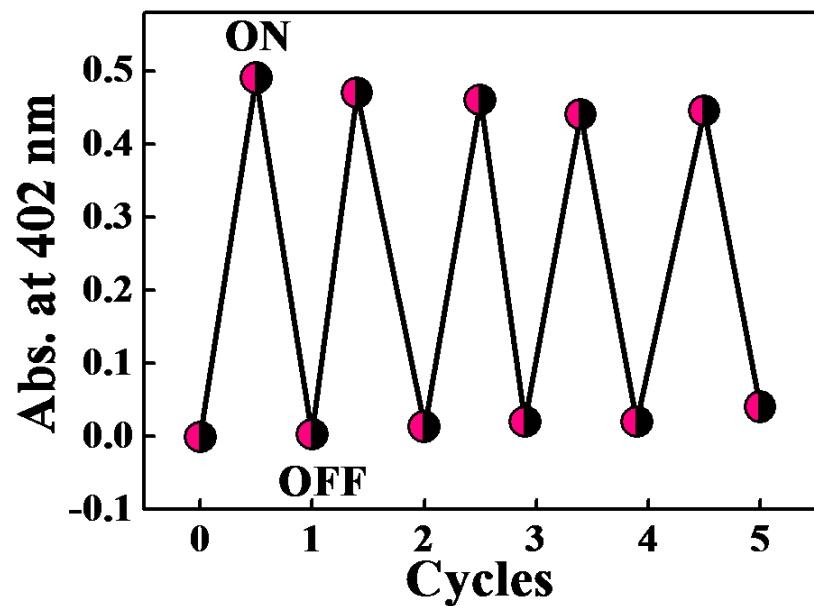
**Fig. S13** Changes in emission spectra of receptor **3** ( $2 \times 10^{-5}$  M) in DMSO with addition of different anions as their TBA salts.



**Fig. S14** Plot of emission intensity of receptor **2** at 502 nm versus concentration of F<sup>-</sup> ions added for detection limit calculation.



**Fig. S15** Changes in absorbance of receptor **2** ( $2 \times 10^{-5}$  M) in DMSO with addition of TBAOH.



**Fig.S16** Reversibility of receptor **2** with alternate addition of  $\text{F}^-$  and  $\text{H}^+$ .