

Electronic Supporting Information

Mechanism of Generation of *closo*-Decaborato Amidrazones. Intramolecular Non-covalent B–H••• π (Ph) Interaction Determines Stabilization of the Configuration around the Amidrazone C=N Bond

Valeria K. Burianova, Dmitrii S. Bolotin, Alexander S. Mikherdov, Alexander S. Novikov, Pennie Petrus Mokolokolo, Andreas Roodt, Vadim P. Boyarskiy, Dmitry Dar'in, Mikhail Krasavin, Vitalii V. Suslonov, Andrey P. Zhdanov, Konstantin Yu. Zhizhin, and Nikolay T. Kuznetsov

Table of Contents

Analytical and spectroscopy data for (Ph ₃ PCH ₂ Ph)[3a–g]	2
Synthesis and characterization of (Ph ₃ PCH ₂ Ph)[3a–g]	5
Spectra of (Ph ₃ PCH ₂ Ph)[3a–g]	13
Synthesis and characterization of 2c–f	35
Spectra of 2c–f	37
IR identification of dimers of 2e	48
Theoretical studies	49
Kinetic measurements	84
Literature	93

Analytical and spectroscopic data for (Ph₃PCH₂Ph)[**3a–g**]

Clusters (Ph₃PCH₂Ph)[**3a–g**] give satisfactory ICPMS-based B elemental analysis for the proposed formulas. These species were also characterized by HRESI-MS, IR, ¹H{¹¹B}, and ¹¹B{¹H} NMR spectroscopies, molar conductivity and by single-crystal X-ray diffraction. (Ph₃PCH₂Ph)[**3a–d**] and (Ph₃PCH₂Ph)[**3g**] were additionally characterized by ¹³C{¹H} NMR spectroscopy (**Figures 2S–35S**). Molar conductivities of (Ph₃PCH₂Ph)[**3a–g**] in MeCN are in the range 77.7–110.3 S cm² mol⁻¹, which is somehow lower than that expected for 1:1 electrolytes (120–160 S cm² mol⁻¹ in MeCN¹). However, the obtained range of conductivities is typical for 2-substituted *closo*-decaborates.²

Although compounds (Ph₃PCH₂Ph)[**3a–g**] are stable in the solid state at RT, they decompose at 142–238 °C. The negative mode HRESI mass spectra of (Ph₃PCH₂Ph)[**3a–g**] exhibit sets of peaks corresponding to the [M]⁻, [2M + Na]⁻, and [M – R¹R²CN]^{+•-} ions. The [M – R¹R²CN]^{+•-} ion is likely generated via the homolytic fragmentation of the N–N bond. Similar sets of peaks were previously observed for the oxime derived 2-iminium *closo*-decaborates.^{2a, b}

In the IR spectra of (Ph₃PCH₂Ph)[**3a–g**], medium to weak bands were observed at the 3598–3300 and 3061–2912 cm⁻¹ intervals, which are related to the N–H and C–H stretching vibrations, respectively.³ The spectra also exhibit one strong band at 2474–2458 cm⁻¹ from ν(B–H) of the *closo*-decaborate cluster,^{2a, 4} and strong bands at 1652–1600 cm⁻¹, which is characteristic for ν(C=N).³

The ¹H{¹¹B} and ¹¹B{¹H} NMR spectra of (Ph₃PCH₂Ph)[**3a–f**], and ¹³C{¹H} NMR spectra of (Ph₃PCH₂Ph)[**3a–d**], display double set of signals of the corresponding anions due to the existence of the compounds in the (*E*)- and (*Z*)-forms at the iminium C=N bond, respectively. The (*E/Z*)-ratio varies for different products and ranges from 45/55 to 85/15. The benzophenone hydrazone derivative (Ph₃PCH₂Ph)[**3g**] exists exclusively in the (*Z*)-form, because of the availability of attractive non-covalent interactions B–H•••Ph (see later). A characteristic feature of the ¹H{¹¹B} NMR spectra is the presence of a low-field broad singlet at δ 10.74–7.66 ppm due to

the N_{imine}H hydrogen. For the substances (Ph₃PCH₂Ph)[**3c–f**] a broad signal of the NH₂ moiety at δ 5.86–5.65 ppm was observed. Another characteristic feature of the $^1\text{H}\{^{11}\text{B}\}$ NMR spectra is the presence of two broad singlets of one H each at δ 3.52–3.11 ppm, assignable to the axial hydrides B(10)H and B(1)H of the *closo*-decaborate cluster. The signals of the residual hydrides are located at δ 0.94–0.82, 0.46–0.39, 0.25–0.14 ppm (broad singlets, each with integral intensity of 2H from the B(3)H and B(5)H, B(6)H and B(9)H, B(7)H and B(8)H), and δ 0.21–0.01 ppm (broad singlet of 1H from the B(4)H). In the $^1\text{H}\{^{11}\text{B}\}$ NMR spectrum of (Ph₃PCH₂Ph)[**3g**], the signal of B(10)H is located at δ 2.59, whereas the other peaks of the hydrides are located at δ 0.38, 0.26, 0.03, and –0.10 ppm; this high-field shift of the cluster hydrides is associated with the shielding effect of one of the phenyl rings. The $^{11}\text{B}\{\text{H}\}$ NMR spectra of (Ph₃PCH₂Ph)[**3a–g**] are typical for 2-substituted *closo*-decaborates.^{2a, 5} Sets of doublets observed in the $^{13}\text{C}\{^1\text{H}\}$ NMR spectra at δ 135.32–117.46 and 29.88–29.85 ppm are typical for (Ph₃PCH₂Ph)⁺ species.^{2a} Two singlets at δ 172.26–151.51 ppm are specific for the quaternary carbon of the imine moiety of the (*E/Z*)-isomers of (Ph₃PCH₂Ph)[**3a–f**]. In the high-field region, two singlets at δ 29.90–19.92 and 11.47–9.72 ppm, respectively, were observed from the ethyl moiety.

Structure Analysis. (i) *X-ray Structure Determination.* Compounds (Ph₃PCH₂Ph)[**3a–g**] crystallize from MeCN solutions as (*E*)-(Ph₃PCH₂Ph)[**3a**]•MeCN, (*Z*)-(Ph₃PCH₂Ph)[**3b**]•MeCN, (*Z*)-(Ph₃PCH₂Ph)[**3c**], (*E*)-(Ph₃PCH₂Ph)[**3d**]•MeCN, (*E*)-(Ph₃PCH₂Ph)[**3e**], (*E*)-(Ph₃PCH₂Ph)[**3f**], and (*Z*)-(Ph₃PCH₂Ph)[**3g**]•MeCN. In the molecular structures, the *closo*-decaborate cluster exhibits the usual bicapped square bipyramidal geometry (**Figure 1**). The N(1)–B(2) bond lengths (1.516(3)–1.527(16) Å) are typical for *closo*-decaborate bound imines.² The N(1)–C(1) and N(2)–C(1) bonds have intermediate order between single and double bonds (1.2993(15)–1.316(2) and 1.326(2)–1.3490(15) Å, respectively), whereas the N(2)–N(3) distances (1.364(3)–1.414(2) Å) are normal single bonds.⁶ The Ph₃PCH₂Ph⁺ cation exhibits usual bond distances and angles.⁷

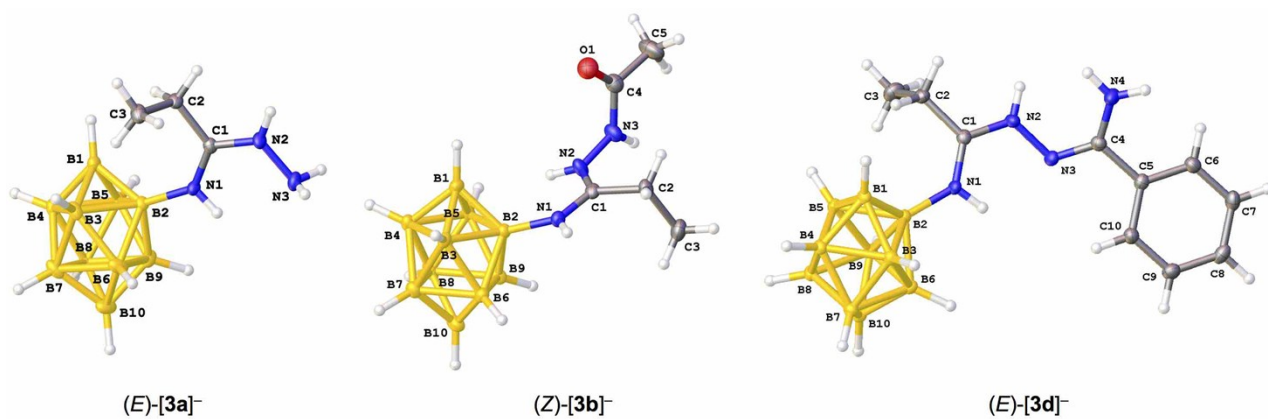
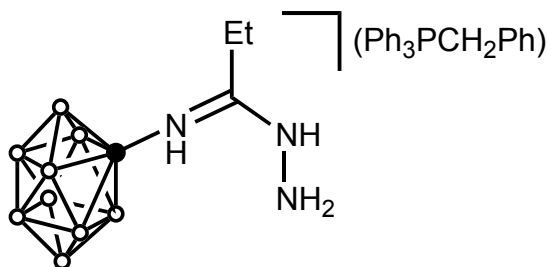


Figure 1S. Molecular structures of (*E*)-[**3a**]⁻ (left), (*Z*)-[**3b**]⁻ (center), and (*E*)-[**3d**]⁻ (right) showing the atomic numbering scheme. Thermal ellipsoids are given at the 50% probability level.

Synthesis and Characterization of (Ph₃PCH₂Ph)[3a–g]

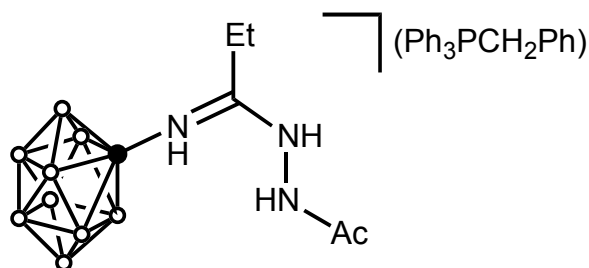
Preparation of (Ph₃PCH₂Ph)[3a]. A solid (Ph₃PCH₂Ph)[1] (190 μmol) was dissolved in CH₂Cl₂ (750 μL) under ultrasound treatment at RT. A solution of **2a** in THF (190 μmol) was added to the solution of (Ph₃PCH₂Ph)[1]. The mixture was kept stirring at RT for 10 min, and a colorless precipitate appears in the range of 1–2 min. The solid was filtered off, washed with five 1-mL portions of Et₂O, and dried at RT for 12 h.



(Ph₃PCH₂Ph)[3a]. Yield: 92% (98 mg). *E:Z* = 45:55. mp: 204–207 °C (dec). Anal. Calcd for C₂₈H₄₀B₁₀N₃P: B, 19.38. Found: B, 19.39. HRESI[–]-MS (*m/z*): 204.2515 ([M][–], calcd 204.2511). Λ_M (CH₃CN, 7 × 10^{–4} M): 93.0 S cm² mol^{–1}. IR (KBr, selected bonds, cm^{–1}): 3315(m) ν(N–H), 3058(w), 2930(w) ν(C–H), 2464(s) ν(B–H), 1652(s) ν(C=N). ¹H{¹¹B} NMR (δ): 9.23 (s, 1H, NH; *E*), 7.90 (t, *J* = 7.9 Hz, 3H, Ph₃PCH₂Ph), 7.75–7.67 (m, 6H, Ph₃PCH₂Ph), 7.66 (s, 1H, NH; *Z*), 7.64–7.56 (m, 6H, Ph₃PCH₂Ph), 7.35 (t, *J* = 7.6 Hz, 1H, Ph₃PCH₂Ph), 7.24 (t, *J* = 7.6 Hz, 2H, Ph₃PCH₂Ph), 7.00 (d, *J* = 7.6 Hz, 2H, Ph₃PCH₂Ph), 6.93 (s, br, 1H, NHNH₂; *Z*), 5.95 (s, br, 1H, NHNH₂; *E*), 4.72 (d, *J* = 14.8 Hz, 2H, Ph₃PCH₂Ph), 4.25 (s, 2H, NHNH₂; *E*), 3.85 (s, 2H, NHNH₂; *Z*), 3.43, 3.32 (s, 1H, B(1)H; *Z* and *E*), 3.14 (s, br, 1H, B(10)H), 2.75 (q, *J* = 7.6 Hz, 2H, CH₂CH₃; *Z*), 2.44 (q, *J* = 7.6 Hz, 2H, CH₂CH₃; *E*), 1.14 (t, *J* = 7.6 Hz, 3H, CH₃; *Z*), 1.05 (t, *J* = 7.6 Hz, 3H, CH₃; *E*), 0.87, 0.82 (s, 1H, B(3)H and B(5)H; *Z* and *E*), 0.42 (s, br, 2H, B(6)H and B(9)H), 0.16, 0.03 (s, br, 1H, B(7)H, B(8)H and B(4)H; *Z* and *E*). ¹¹B{¹H} NMR (δ): –6.11, –3.91, –0.66, 0.97 (B(1) and B(10); *Z* and *E*), –15.04, –16.84 (B(2); *Z* and *E*), –25.41 (B(6), B(9), B(3), B(5)), –28.48, –29.83 (B(7), B(8) and B(4)). ¹³C{¹H} NMR (δ): 171.89, 169.68 (NHC(Et)NH; *E* and *Z*), 135.30 (d, *J* = 5.5 Hz, Ph₃PCH₂Ph), 134.23 (d, *J* = 12.6 Hz, Ph₃PCH₂Ph), 131.00 (d, *J* = 3.2 Hz, Ph₃PCH₂Ph), 130.16 (d, *J* = 3.8 Hz, Ph₃PCH₂Ph), 128.99 (d, *J* = 3.2 Hz, Ph₃PCH₂Ph), 128.72 (d, *J* = 3.8 Hz, Ph₃PCH₂Ph),

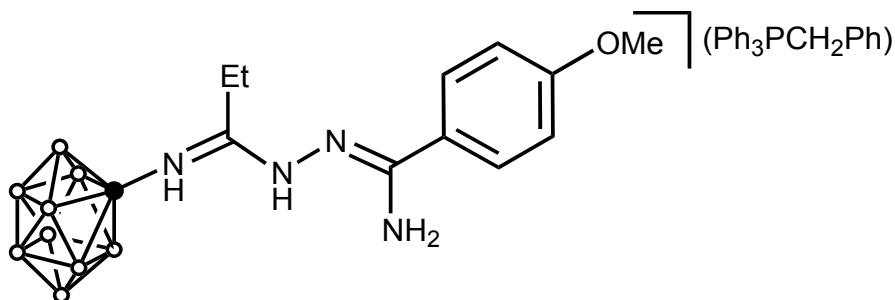
127.31 (d, $J = 8.4$ Hz, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 117.51 (d, $J = 86.2$ Hz, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 29.88 ($\text{Ph}_3\text{PCH}_2\text{Ph}$), 24.02, 21.80 (CH_2CH_3 ; E and Z), 10.86, 10.01 (CH_3 ; E and Z). Crystals of (E)-($\text{Ph}_3\text{PCH}_2\text{Ph}$)[**3a**]•MeCN suitable for X-ray diffraction were obtained by slow evaporation of MeCN solution.

*Preparation of ($\text{Ph}_3\text{PCH}_2\text{Ph}$)[**3b–g**].* A mixture of any one of solid **2b–g** (95.2 μmol) and solid ($\text{Ph}_3\text{PCH}_2\text{Ph}$)[**1**] (95.2 μmol) was dissolved in MeCN (1 mL) under ultrasound treatment at RT. Then colorless (($\text{Ph}_3\text{PCH}_2\text{Ph}$)[**3b–f**]) or yellow (($\text{Ph}_3\text{PCH}_2\text{Ph}$)[**3g**]) crystals were obtained by slow evaporation of MeCN solution at RT in air. The crystals were washed by three 1 mL-portions of Et_2O and dried at RT in air.



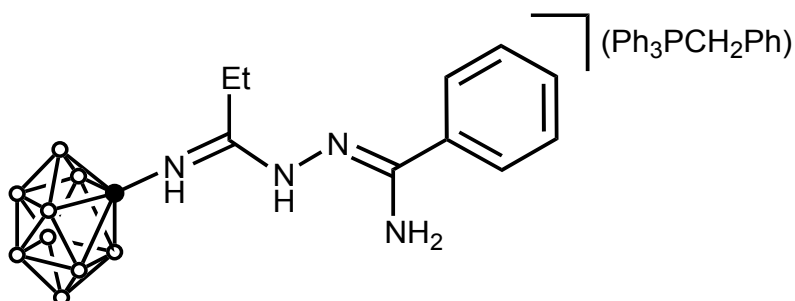
($\text{Ph}_3\text{PCH}_2\text{Ph}$)[**3b**]. Yield: 75% (43 mg). $E:Z = 85:15$. mp: 236–238 °C (dec). Anal. Calcd for $\text{C}_{30}\text{H}_{42}\text{B}_{10}\text{N}_3\text{OP}$: B, 18.03. Found: B, 18.17. HRESI[−]-MS (m/z): 187.2248 ($[\text{M} - \text{AcNH}]^+$, calcd 187.2246), 246.2623 ($[\text{M}]^+$, calcd 246.2618), 515.5127 ($[\text{2M} + \text{Na}]^+$, calcd 515.5129). Λ_{M} (CH_3CN , 7×10^{-4} M): 100.7 $\text{S cm}^2 \text{mol}^{-1}$. IR (KBr, selected bonds, cm^{-1}): 3306(m) $\nu(\text{N-H})$, 3050(w), 2933(w) $\nu(\text{C-H})$, 2474(s) $\nu(\text{B-H})$, 1637(s) $\nu(\text{C=N})$. $^1\text{H}\{^{11}\text{B}\}$ NMR (δ): 9.44 (s, br, 1H, NH), 8.44 (s, br, 1H, NH), 7.90 (t, $J = 7.1$ Hz, 3H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 7.74–7.68(m, 6H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 7.62–7.58 (m, 6H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 7.36 (t, $J = 7.5$ Hz, 1H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 7.25 (t, $J = 7.5$ Hz, 2H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 6.98 (d, $J = 7.5$ Hz, 2H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 6.60 (s, br, 1H, NH; E), 6.47 (s, br, 1H, NH; Z), 4.68 (d, $J = 11.6$ Hz, 2H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 3.44 (s, br, 1H, B(1)H; E), 3.31 (s, br, 1H, B(1)H; Z), 3.15 (s, br, 1H, B(10)H; E), 3.11 (s, br, 1H, B(10)H; Z), 2.87 (q, $J = 7.6$ Hz, 2H, CH_2CH_3 ; Z), 2.26 (q, $J = 7.6$ Hz, 2H, CH_2CH_3 ; E), 1.99 (s, 3H, OCH_3 ; E), 1.93 (s, 3H, OCH_3 ; Z), 1.24 (t, $J = 7.6$ Hz, 3H, CH_2CH_3 ; Z), 2.24 (t, $J = 7.6$ Hz, 3H, CH_2CH_3 ; E), 0.86 (s, br, 2H, B(3)H and B(5)H; Z), 0.82 (s, br, 2H,

B(3)H and B(5)H; *E*), 0.40 (s, br, 2H, B(6)H and B(9)H), 0.18 (s, br, 2H, B(7)H and B(8)H), 0.14 (s, br, 1H, B(4)H; *E*), 0.04 (s, br, B(4)H; *Z*). $^{11}\text{B}\{^1\text{H}\}$ NMR (δ): 1.25, 0.45, 3.82, 6.09, (B(1) and B(10); *E* and *Z*), -14.68 (B(2); *Z*), -16.74 (B(2); *E*), -25.57 (B(3), B(5), B(6) and B(9)), -28.77 (B(7), (B(8) and B(4)). $^{13}\text{C}\{^1\text{H}\}$ NMR (δ): 172.26 (NHC(Et)NH), 169.82 (C(O)CH₃), 135.32 (d, J = 3.0 Hz, *Ph*₃PCH₂Ph), 134.21 (d, J = 9.7 Hz, *Ph*₃PCH₂Ph), 130.95 (d, J = 5.4 Hz, *Ph*₃PCH₂Ph), 130.16 (d, J = 12.1 Hz, *Ph*₃PCH₂Ph), 129.03 (d, J = 3.2 Hz, *Ph*₃PCH₂Ph), 128.76 (d, J = 3.7 Hz, *Ph*₃PCH₂Ph), 127.26 (d, J = 8.4 Hz, *Ph*₃PCH₂Ph), 117.49 (d, J = 86.2 Hz, *Ph*₃PCH₂Ph), 29.85 (d, J = 48.7 Hz *Ph*₃PCH₂Ph), 23.68 (s, C(O)CH₃), 19.92 (s, CH₂CH₃), 9.73 (s, CH₂CH₃). Crystals of (*Z*)-(Ph₃PCH₂Ph)[**3b**]•MeCN suitable for X-ray diffraction were obtained by slow evaporation of MeCN solution.



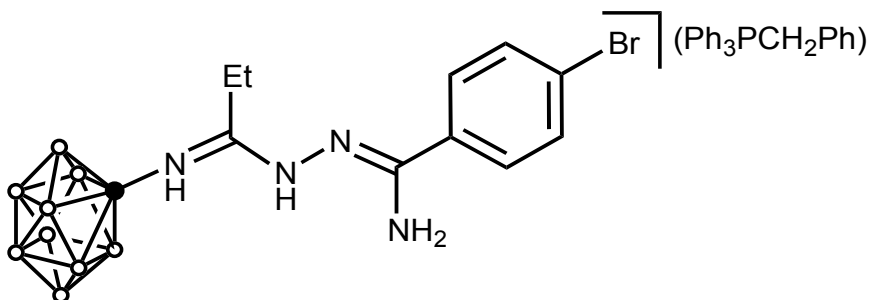
(Ph₃PCH₂Ph)[**3c**]. Yield: 58% (38 mg). *E*:*Z* = 75:25. mp: 177–178 °C (dec). Anal. Calcd for C₃₆H₄₇B₁₀N₄OP: B, 15.65. Found: B, 15.63. HRESI-MS (m/z): 188.2313 ([*M* – MeOC₆H₄C(NH₂)N]⁺, calcd 188.2317), 337.3042 ([*M*]⁺, calcd 337.3042). Λ_M (CH₃CN, 7×10^{-4} M): 88.9 S cm² mol⁻¹. IR (KBr, selected bonds, cm⁻¹): 3449(m), 3316(m) ν (N–H), 3061(w), 2912(w) ν (C–H), 2462(s) ν (B–H), 1610(s) ν (C=N). $^1\text{H}\{^{11}\text{B}\}$ NMR (δ): 10.38 (s, 1H, NH; *E*), 8.39 (s, 1H, NH; *Z*), 7.90 (t, J = 7.6 Hz, 3H, *Ph*₃PCH₂Ph), 7.77 (d, J = 8.8 Hz, 2H, *p*-CH₃OC₆H₄; *E*), 7.74–7.67 (m, 6H, *Ph*₃PCH₂Ph), 7.65 (d, J = 8.8 Hz, 2H, *p*-CH₃OC₆H₄; *Z*), 7.62–7.54 (m, 6H, *Ph*₃PCH₂Ph), 7.36 (t, J = 7.6 Hz, 1H, *Ph*₃PCH₂Ph), 7.25 (t, J = 7.6 Hz, 2H, *Ph*₃PCH₂Ph), 7.04–6.95 (m, 4H, *Ph*₃PCH₂Ph and *p*-CH₃O*Ph*; *Z* and *E*), 6.88 (s, 1H, *p*-CH₃OC₆H₄C(NH₂)NNH; *Z*), 6.21 (s, 1H, *p*-CH₃OC₆H₄C(NH₂)NNH; *E*), 5.65 (s, 2H, *p*-CH₃OC₆H₄C(NH₂)NNH), 4.67 (d, J = 14.4 Hz, 2H, *Ph*₃PCH₂Ph), 3.86 (s, 3H, *p*-CH₃OC₆H₄C(NH₂)NNH; *E*), 3.84 (s, 3H, *p*-CH₃OC₆H₄C(NH₂)NNH; *Z*), 3.50 (s, 1H, B(1)H; *E*), 3.30 (s, 1H, B(1)H; *Z*), 3.25 (s, 1H, B(10)H;

Z), 3.15 (s, 1H, B(10)H; *E*), 2.93 (q, 2H, $J = 7.6$ Hz, CH_2CH_3 ; *Z*), 2.55 (q, $J = 7.6$ Hz, 2H, CH_2CH_3 ; *E*), 1.30 (t, $J = 7.6$ Hz, 3H, CH_2CH_3 ; *Z*), 1.11 (t, $J = 7.6$ Hz, 3H, CH_2CH_3 ; *E*), 0.92 (s, 2H, B(3)H and B(5)H; *E*), 0.88 (s, 2H, B(3)H and B(5)H; *Z*), 0.45 (s, 2H, B(6)H and B(9)H; *E*), 0.41 (s, 2H, B(6)H and B(9)H; *E*), 0.25 (s, 2H, B(7)H and B(8)H; *E*), 0.21 (s, 1H, B(4)H; *E*), 0.15 (s, 2H, B(7)H and B(8)H; *Z*), 0.03 (s, 1H, B(4)H; *Z*). ^{11}B { ^1H } (δ): 1.58, -0.62 , -4.03 , -6.41 (s, B(1) and B(10); *E* and *Z*), -15.05 (s, B(2); *Z*), -16.72 (s, B(2); *E*), -25.36 (s, B(3), B(5), B(6) and B(9)), -28.65 (s, B(7), B(8) and B(4)). ^{13}C { ^1H } NMR (δ): 167.11, 165.44, 161.81, 161.61, 153.73, 151.51 (NHC(Et)NH, NC(NH₂)C₆H₄ and C₆H₄; *E* and *Z*), 135.31 (d, $J = 2.4$ Hz, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 134.19 (d, $J = 9.4$ Hz, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 130.93 (d, $J = 5.2$ Hz, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 130.14 (d, $J = 12.6$ Hz, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 129.03 (d, $J = 3.2$ Hz, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 128.75 (d, $J = 3.8$ Hz, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 128.14, 128.05 (C₆H₄; *Z* and *E*), 127.23 (d, $J = 8.4$ Hz, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 125.50 (C₆H₄), 117.46 (d, $J = 86.2$ Hz, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 113.86, 113.79 (C₆H₄; *Z* and *E*), 55.15 (OCH₃); 29.87 (d, $J = 48.7$ Hz, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 24.50, 22.13 (CH₂; *E* and *Z*), 11.47, 9.72 (CH₂CH₃; *Z* and *E*). Crystals of (*Z*)-(Ph₃PCH₂Ph)[**3c**] suitable for X-ray diffraction were obtained by slow evaporation of MeCN solution.



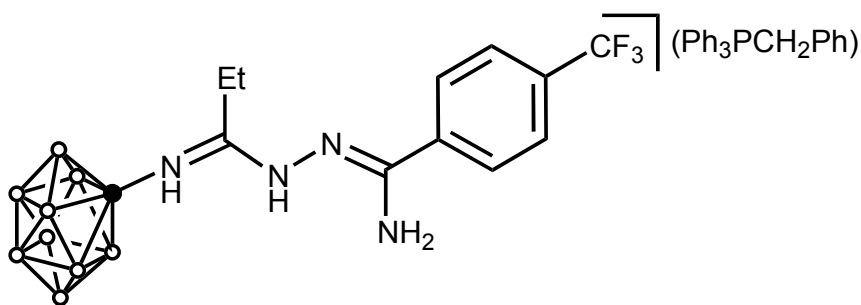
(Ph₃PCH₂Ph)[**3d**]. Yield: 84% (53 mg). *E*:*Z* = 80:20. mp: 159–162 °C (dec). Anal. Calcd for C₃₇H₄₈B₁₀N₅P: B, 15.40. Found: B, 15.36 ((Ph₃PCH₂Ph)[**3d**]•MeCN). HRESI-MS (m/z): 307.2939 ([M]⁺, calcd 307.2936). Λ_M (CH₃CN, 7×10^{-4} M): 77.7 S cm² mol⁻¹. IR (KBr, selected bonds, cm⁻¹): 3485(m), 3356(m), 3250(m) ν (N–H); 3050(w), 2930(m) ν (C–H); 2465(s) ν (B–H); 1622(s), 1597(s) ν (C=N). ^1H { ^{11}B } NMR (δ): 10.43 (s, 1H, NH; *E*), 8.63 (s, 1H, NH; *Z*), 7.90 (t, $J = 7.5$ Hz, 3H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 7.83 (d, $J = 6.8$ Hz, 2H, Ph; *E*), 7.74–7.67 (m, 6H, $\text{Ph}_3\text{PCH}_2\text{Ph}$ and Ph; *Z*), 7.63–7.54 (m, 6H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 7.54–7.44 (m, 3H, Ph), 7.37 (t, $J = 7.6$ Hz, 1H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 7.25 (t, $J = 7.6$ Hz, 2H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 6.97 (d, $J = 7.6$ Hz, 2H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 6.29 (s, br, 1H, PhC(NH₂)NNH),

5.78 (s, 2H, PhC(NH₂)NNH; *Z*), 5.74 (s, 2H, PhC(NH₂)NNH; *E*), 4.66 (d, 2H, Ph₃PCH₂Ph), 3.50 (s, 1H, B(1)H; *E*), 3.28 (s, 1H, B(1)H; *Z*), 3.25 (s, 1H, B(10)H; *E*), 3.14 (s, 1H, B(10)H; *Z*), 2.95 (q, *J* = 7.6 Hz, 2H, CH₂CH₃; *Z*), 2.58 (q, *J* = 7.6 Hz, 2H, CH₂CH₃; *E*), 1.31 (t, *J* = 7.6 Hz, 3H, CH₂CH₃), 1.13 (t, *J* = 7.6 Hz, 3H, CH₂CH₃), 0.92 (s, 2H, B(3)H and B(5)H; *E*), 0.86 (s, 2H, B(3)H and B(5)H; *Z*), 0.44 (s, 2H, B(6)H and B(9)H; *E*), 0.39 (s, 2H, B(6)H and B(9)H; *Z*), 0.24 (s, 2H, B(7)H and B(8)H; *E*), 0.21 (s, 1H, B(4)H; *E*), 0.13 (s, 2H, B(7)H and B(8)H; *Z*), 0.01 (s, 1H, B(4)H; *Z*). ¹¹B{¹H} NMR (δ): -1.57, -0.46, -3.97, -6.39 (B(1) and B(10)), -14.91 (B(2); *Z*), -16.65 (B(2); *E*), -25.35 (B(3), B(5), B(6) and B(9)), -28.59 (B(4), B(7) and B(8)). ¹³C{¹H} NMR (δ): 167.30, 151.52 (NHC(Et)NH and NC(NH₂)Ph), 135.31 (d, *J* = 3.1 Hz, Ph₃PCH₂Ph), 135.18 (d, *J* = 79.0 Hz, Ph₃PCH₂Ph), 133.37 (NC(NH₂)Ph), 130.91 (d, *J* = 5.4 Hz, Ph₃PCH₂Ph), 130.50 (NC(NH₂)Ph), 130.14 (d, *J* = 12.6 Hz, Ph₃PCH₂Ph), 129.03 (d, *J* = 3.2 Hz, Ph₃PCH₂Ph), 128.77 (d, *J* = 3.7 Hz, Ph₃PCH₂Ph), 128.57, 128.50 (NC(NH₂)Ph; *Z* and *E*), 127.21 (d, *J* = 8.4 Hz, Ph₃PCH₂Ph), 126.52 (NC(NH₂)Ph), 117.46 (d, *J* = 86.2 Hz, Ph₃PCH₂Ph), 29.86 (d, Ph₃PCH₂Ph), 29.88, 24.51 (CH₂CH₃; *Z* and *E*), 9.73 (CH₃; *Z* and *E*). Crystals of (*E*)-(Ph₃PCH₂Ph)[**3d**]•MeCN suitable for X-ray diffraction were obtained by slow evaporation of MeCN solution.



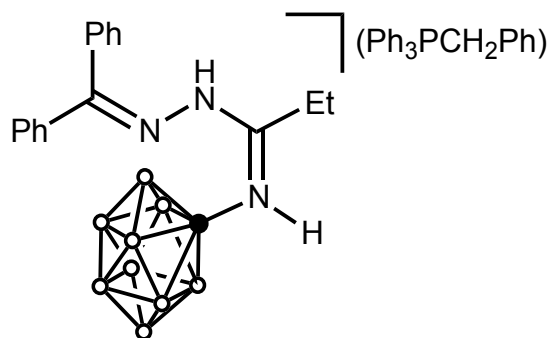
(Ph₃PCH₂Ph)[**3e**]. Yield: 68% (48 mg). *E*:*Z* = 80:20. mp: 164 °C (dec). Anal. Calcd for C₃₅H₄₄B₁₀BrN₄P: B, 14.61. Found: B, 14.60. HRESI⁺-MS (*m/z*): 386.2025 ([M]⁺, calcd 386.2027). Λ_M (CH₃CN, 7 × 10⁻⁴ M): 110.3 S cm² mol⁻¹. IR (KBr, selected bonds, cm⁻¹): 3447(m), 3352(m) ν(N-H), 3059(w), 2927(w) ν(C-H), 2464(s) ν(B-H), 1600(s) ν(C=N). ¹H{¹¹B} NMR (δ): 10.45 (s, 1H, NH; *E*), 8.51 (s, 1H, NH; *Z*), 7.90 (t, *J* = 7.5 Hz, 3H, Ph₃PCH₂Ph), 7.77–7.55 (m, 16H, Ph₃PCH₂Ph and *p*-BrC₆H₄C(NH₂)NNH), 7.36 (t, *J* = 7.5 Hz, 1H, Ph₃PCH₂Ph), 7.25 (t, *J* = 7.7 Hz, 2H, Ph₃PCH₂Ph), 6.98 (dd, 2H, Ph₃PCH₂Ph), 6.91 (s, 1H, *p*-BrC₆H₄C(NH₂)NNH; *Z*) 6.33 (s, 1H, *p*-

BrC₆H₄C(NH₂)NNH; *E*), 5.73 (s, 2H, *p*-BrC₆H₄C(NH₂)NNH; *Z* and *E*), 4.65 (d, *J* = 14.8 Hz, 2H, Ph₃PCH₂Ph), 3.50 (s, 1H, B(1)H; *E*), 3.28 (s, 1H, B(1)H; *Z*), 3.24 (s, 1H, B(10)H; *E*), 3.14 (s, 1H, B(10)H; *Z*), 2.95 (q, *J* = 7.5 Hz, 2H, CH₂CH₃; *Z*), 2.57 (q, *J* = 7.5 Hz, 2H, CH₂CH₃; *E*), 1.31 (t, *J* = 7.5 Hz, 3H, CH₂CH₃; *Z*), 1.12 (t, *J* = 7.5 Hz, 3H, CH₂CH₃; *E*), 0.91 (s, 2H, B(3)H and B(5)H; *E*), 0.86 (s, 1H, B(3)H and B(5)H; *Z*), 0.43 (s, 1H, B(6)H and B(9)H; *E*), 0.39 (s, 1H, B(6)H and B(9)H; *Z*), 0.24 (s, 1H, B(7)H and B(8)H; *E*), 0.21 (s, 1H, B(4)H; *E*), 0.14 (s, 1H, B(7)H and B(8)H; *Z*), 0.01 (s, 1H, B(4)H; *Z*). ¹¹B{¹H} NMR (δ): 1.55, -0.57, -3.97, -6.49 (B(1) and B(10); *E* and *Z*), -14.94 (B(2); *Z*), -16.69 (B(2); *E*), -25.41 (B(6), B(9), B(3) and B(5)), -28.67 (B(4), B(7) and B(8)). ¹³C{¹H} NMR spectrum was not measured due to poor solubility of (Ph₃PCH₂Ph)[**3e**] in MeCN. Crystals of (*E*)-(Ph₃PCH₂Ph)[**3e**] suitable for X-ray diffraction were obtained by slow evaporation of MeCN solution.



(Ph₃PCH₂Ph)[**3f**]. Yield: 59% (41 mg). *E*:*Z* = 70:30. mp: 211–212 °C (dec). Anal. Calcd for C₃₆H₄₄B₁₀F₃N₄P: B, 14.83. Found: B, 14.49. HRESI⁺-MS (*m/z*): 375.2814 ([M]⁺, calcd 375.2811). Λ_M (CH₃CN, 7 × 10⁻⁴ M): 93.9 S cm² mol⁻¹. IR (KBr, selected bonds, cm⁻¹): 3458(m), 3352(m) ν(N–H), 3036(w), 2930(w) ν(C–H), 2463(s) ν(B–H), 1602(s) ν(C=N). ¹H{¹¹B} NMR (δ): 10.51 (s, 1H, NH; *E*), 8.64 (s, 1H, NH; *Z*), 8.00 (d, *J* = 8.4 Hz, 2H, *p*-CF₃C₆H₄C(NH₂)NNH; *E*), 7.90 (t, 3H, Ph₃PCH₂Ph), 7.87 (d, 2H, *p*-CF₃C₆H₄C(NH₂)NNH; *Z*), 7.79 (d, *J* = 8.4 Hz, 2H, *p*-CF₃C₆H₄C(NH₂)NNH; *E*), 7.76 (d, *J* = 8.4 Hz, 2H, *p*-CF₃C₆H₄C(NH₂)NNH; *Z*), 7.74–7.67 (m, 6H, Ph₃PCH₂Ph), 7.64–7.54 (m, 6H, Ph₃PCH₂Ph), 7.36 (t, *J* = 7.5 Hz, 1H, Ph₃PCH₂Ph), 7.24 (t, *J* = 7.5 Hz, 2H, Ph₃PCH₂Ph), 6.98 (d, *J* = 7.5 Hz, 2H, Ph₃PCH₂Ph), 6.39 (s, 1H, *p*-CF₃C₆H₄C(NH₂)NNH; *E*), 5.86 (s, 2H, *p*-CF₃C₆H₄C(NH₂)NNH; *Z*), 5.80 (s, 2H, *p*-CF₃C₆H₄C(NH₂)NNH; *E*), 4.68 (d, *J* = 14.8 Hz, 2H, Ph₃PCH₂Ph), 3.52 (s, 1H, B(1)H; *E*), 3.31 (s,

1H, B(1)H; *Z*), 3.26 (s, 1H, B(10)H; *E*), 3.16 (s, 1H, B(10)H; *Z*), 2.95 (q, $J = 7.5$ Hz, 2H, CH_2CH_3 ; *Z*), 2.59 (q, $J = 7.5$ Hz, 2H, CH_2CH_3 ; *E*), 1.31 (t, $J = 7.5$ Hz, 3H, CH_2CH_3 ; *Z*), 1.13 (t, $J = 7.5$ Hz, 3H, CH_2CH_3 ; *E*), 0.94 (s, 2H, B(3)H and B(5)H; *E*), 0.89 (s, 2H, B(3)H and B(5)H; *Z*), 0.46 (s, 2H, B(6)H and B(9)H; *E*), 0.42 (s, 2H, B(6)H and B(9)H; *Z*), 0.26 (s, 2H, B(7)H, B(8)H and B(4)H; *E*), 0.17 (s, 2H, B(7)H and B(8)H; *Z*), 0.04 (s, 2H, B(4)H; *Z*). $^{11}\text{B}\{^1\text{H}\}$ NMR (δ): 1.63, -0.49 , -3.92 , -6.42 (s, B(1) and B(10); *E* and *Z*), -14.92 (s, B(2); *Z*), -16.68 (s, B(2); *E*), -25.35 (s, B(6), B(9), B(3) and B(5)), -28.64 (s, B(7), B(8) and B(4)). $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum was not measured due to poor solubility of $(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3f}]$ in MeCN. Crystals of (*E*)- $(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3f}]$ suitable for X-ray diffraction were obtained by slow evaporation of MeCN solution.

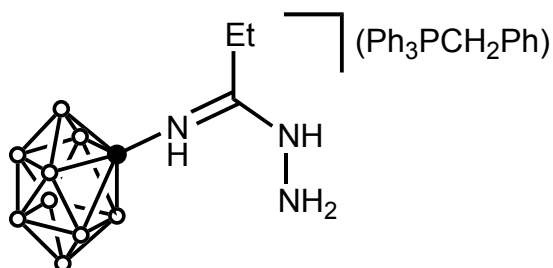


(*Z*)- $(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3g}]$. Yield: 78% (54 mg). mp: 136–139 °C (dec). Anal. Calcd for $\text{C}_{43}\text{H}_{51}\text{B}_{10}\text{N}_4\text{P}$: B, 14.17. Found: B, 14.07. HRESI-MS (m/z): 368.3139 ($[\text{M}]^-$, calcd 368.3143). Λ_{M} (CH_3CN , 7×10^{-4} M): 78.4 $\text{S cm}^2 \text{ mol}^{-1}$. IR (KBr, selected bonds, cm^{-1}): 3598(m), 3300(m) $\nu(\text{N-H})$, 3059(w), 2936(w) $\nu(\text{C-H})$, 2458(s) $\nu(\text{B-H})$, 1641(s) $\nu(\text{C=N})$. $^1\text{H}\{^{11}\text{B}\}$ NMR (δ): 10.74 (s, 1H, NH), 7.90 (t, $J = 7.5$ Hz, 3H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 7.74–7.54 (m, 17H, $\text{Ph}_3\text{PCH}_2\text{Ph}$ and $\text{Ph}_2\text{C=NNH}$), 7.48–7.39 (m, 6H, $\text{Ph}_2\text{C=NNH}$ and $\text{Ph}_3\text{PCH}_2\text{Ph}$), 7.36 (t, 1H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 7.25 (t, $J = 7.6$ Hz, 2H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 7.00 (d, $J = 7.6$ Hz, 2H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 6.74 (s, 1H, NHNCPh_2), 4.67 (d, $J = 14.4$ Hz, 2H, $\text{Ph}_3\text{PCH}_2\text{Ph}$), 3.32 (s, br, 1H, B(1)H), 2.78 (q, $J = 7.5$ Hz, 2H, CH_2CH_3), 2.59 (s, br, 1H, B(10)H), 1.22 (t, $J = 7.5$ Hz, 3H, CH_2CH_3), 0.38 (s, br, 2H, B(3)H and B(5)H), 0.26 (s, br, 2H, B(6)H and B(9)H), 0.03 (s, br, 2H, B(7)H and B(8)H), -0.10 (s, br, 1H, B(4)H). $^{11}\text{B}\{^1\text{H}\}$ NMR (δ): 0.80 (B(1)), -5.72 (B(10)), -17.07 (B(3) and B(5)), -25.48 (B(6) and B(9)), -26.07 (B(7) and B(8)), -28.78 (B(4)). $^{13}\text{C}\{^1\text{H}\}$ NMR (δ): 168.72, 155.02 ($\text{NHC}(\text{Et})\text{NH}$ and NHNCPh_2), 137.47 (NHNCPh_2),

135.32 (d, $J = 3.0$ Hz, Ph_3PCH_2Ph), 134.19 (d, $J = 9.7$ Hz, Ph_3PCH_2Ph), 131.86 (NHNC Ph_2), 130.93 (d, $J = 5.5$ Hz, Ph_3PCH_2Ph), 130.15 (d, $J = 12.6$ Hz, Ph_3PCH_2Ph), 130.06, 130.02, 129.85 (NHNC Ph_2), 129.03 (d, $J = 3.2$ Hz, Ph_3PCH_2Ph), 129.01 (NHNC Ph_2), 128.76 (d, $J = 3.8$ Hz, Ph_3PCH_2Ph), 128.42, 127.70 (NHNC Ph_2), 127.23 (d, $J = 8.5$ Hz, Ph_3PCH_2Ph), 117.47 (d, $J = 86.2$ Hz, Ph_3PCH_2Ph), 29.87 (d, $J = 48.7$ Hz, Ph_3PCH_2Ph), 25.90 (CH_2CH_3), 10.26 (CH_3). Crystals of (Z)-(Ph₃PCH₂Ph)[**3g**]•MeCN suitable for X-ray diffraction were obtained by slow evaporation of MeCN solution.

Spectra of (Ph₃PCH₂Ph)[3a-g]

(Ph₃PCH₂Ph)[3a]



Acquisition Parameter

Source Type	ESI	Ion Polarity	Negative	Set Nebulizer	1.0 Bar
Focus	Active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	3500 V	Set Dry Gas	4.0 l/min
Scan End	1200 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Source

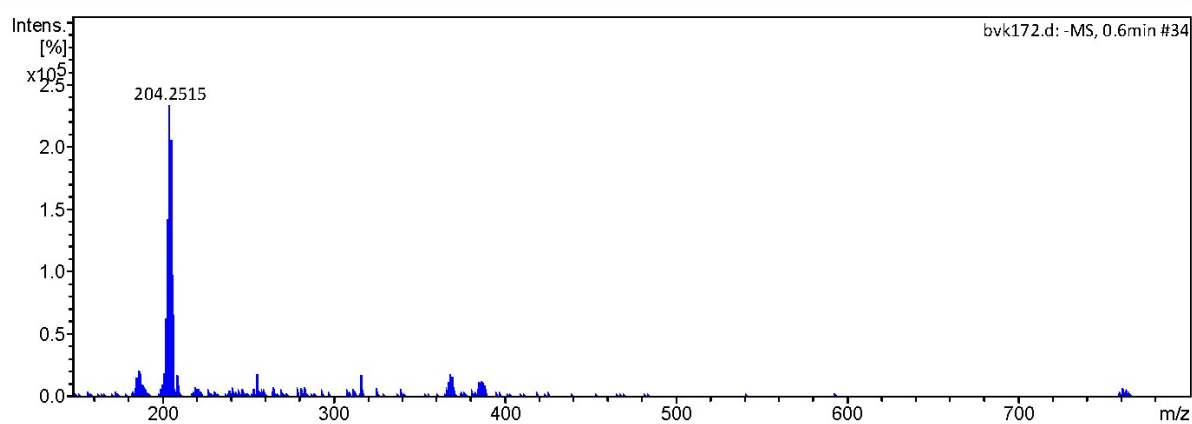


Figure 2S. HRESI⁻-mass spectrum of (Ph₃PCH₂Ph)[3a].

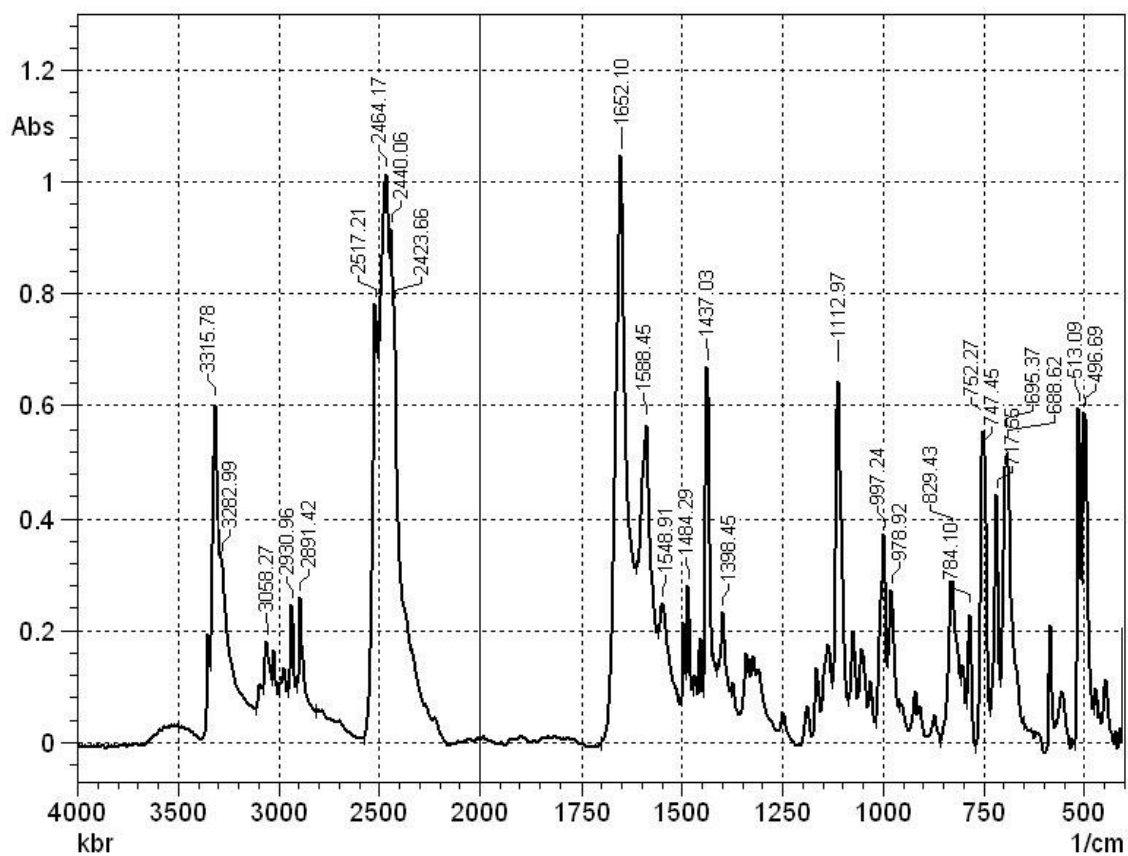


Figure 3S. IR spectrum of $(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3a}]$.

$(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3a}]$, 400.13 MHz, 295 K

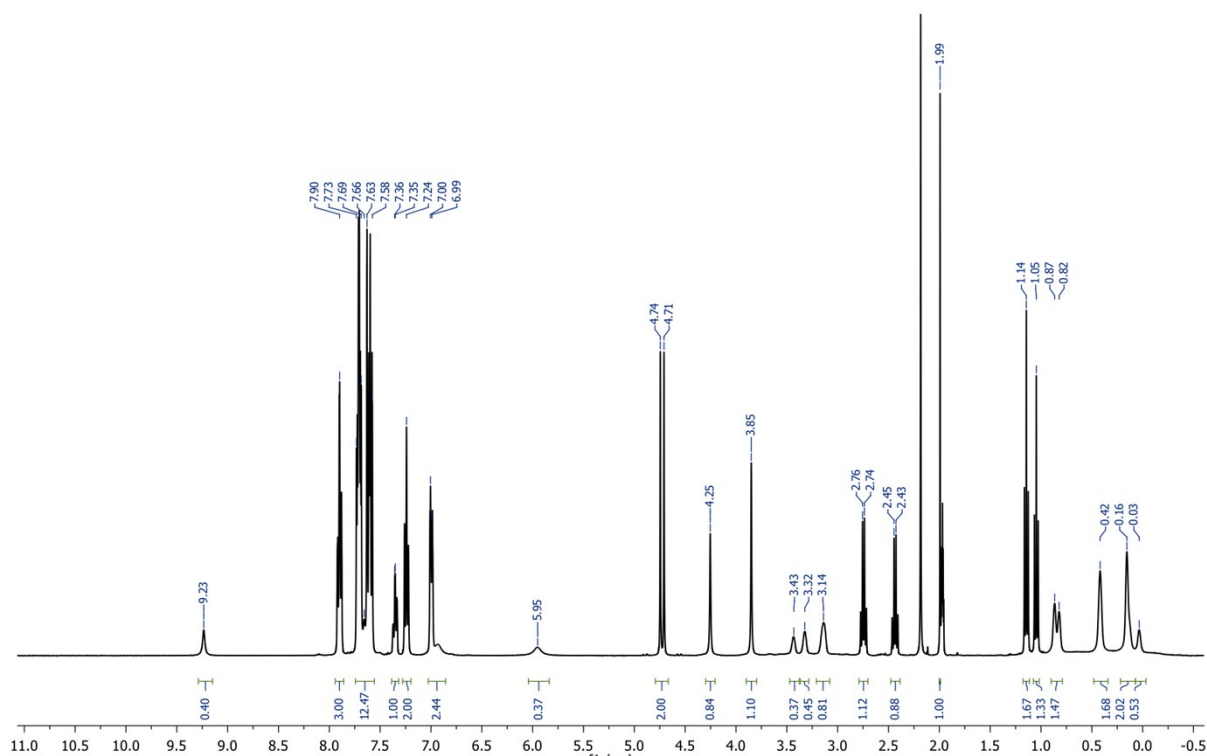


Figure 4S. $^1\text{H}\{^{11}\text{B}\}$ NMR spectrum of $(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3a}]$.

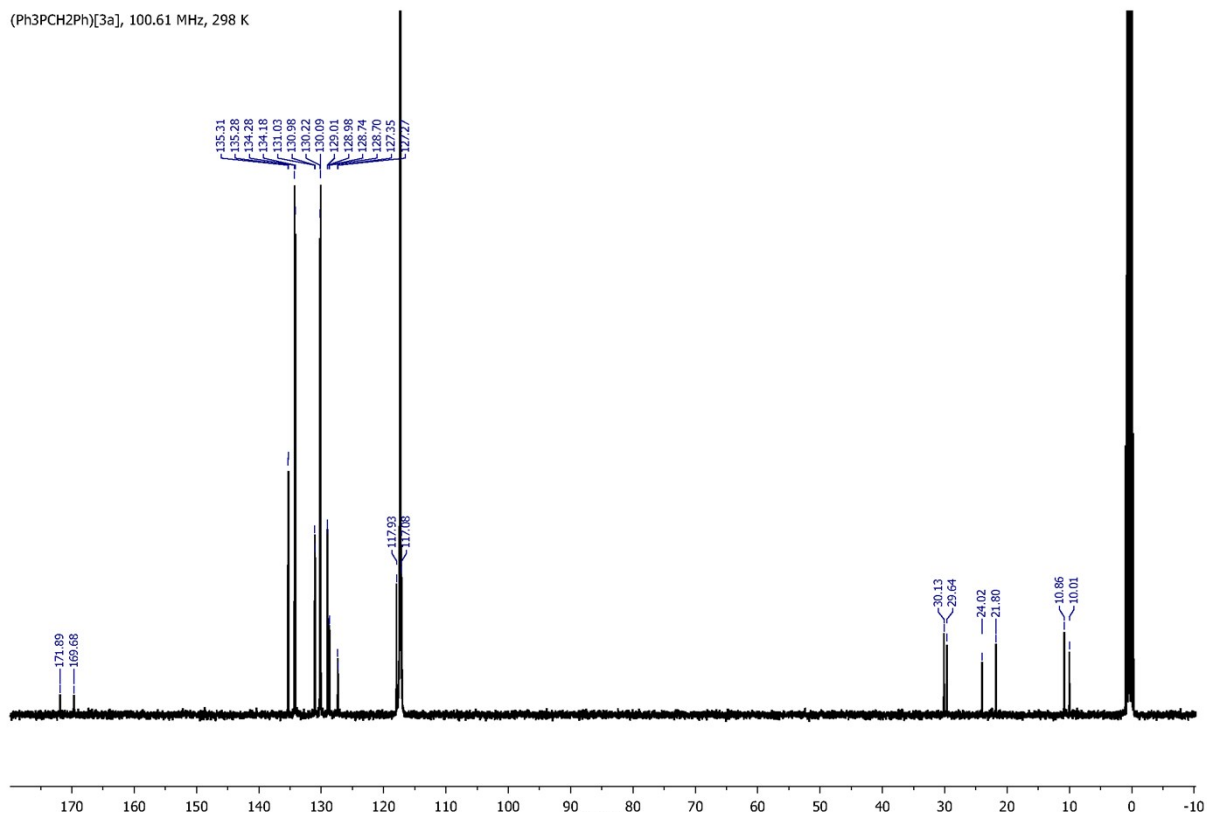


Figure 5S. $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of (Ph₃PCH₂Ph)[3a].

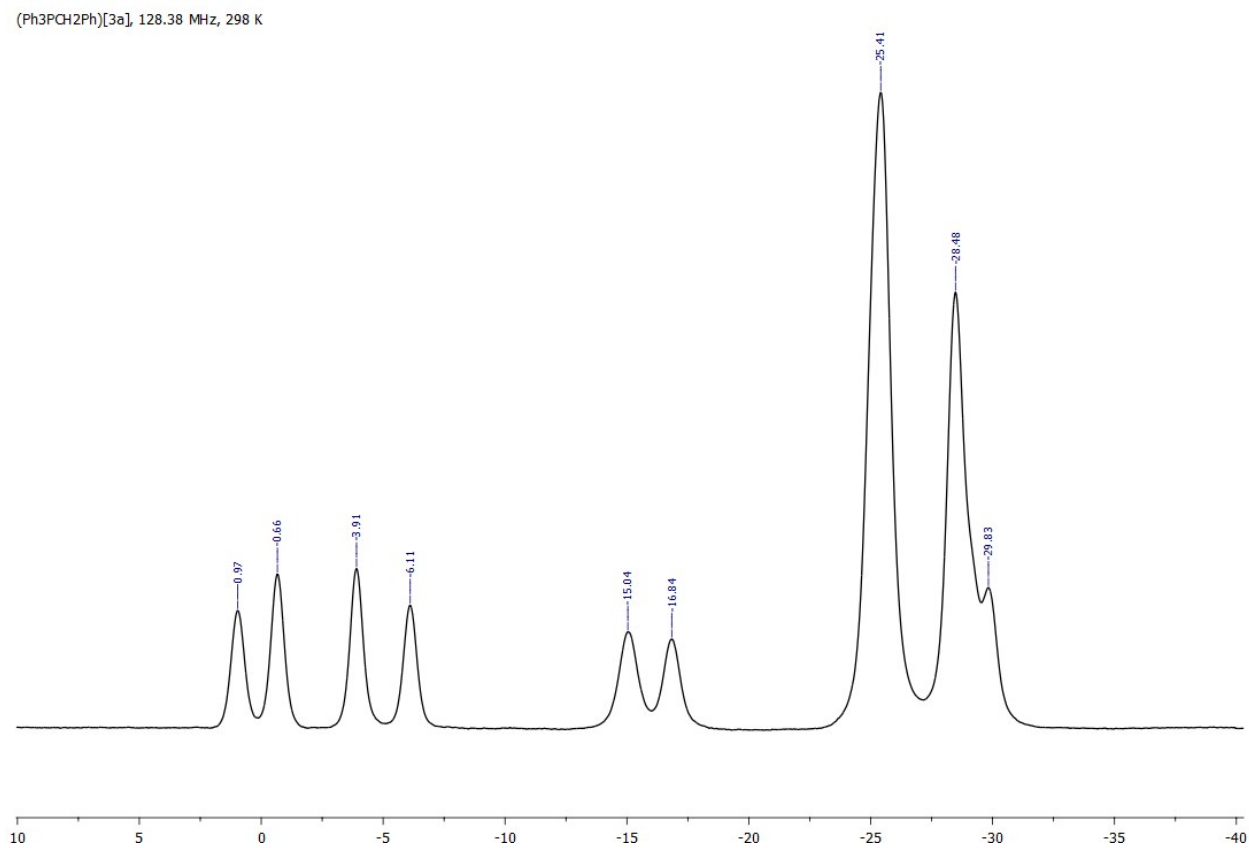


Figure 6S. $^{11}\text{B}\{^1\text{H}\}$ NMR spectrum of (Ph₃PCH₂Ph)[3a].

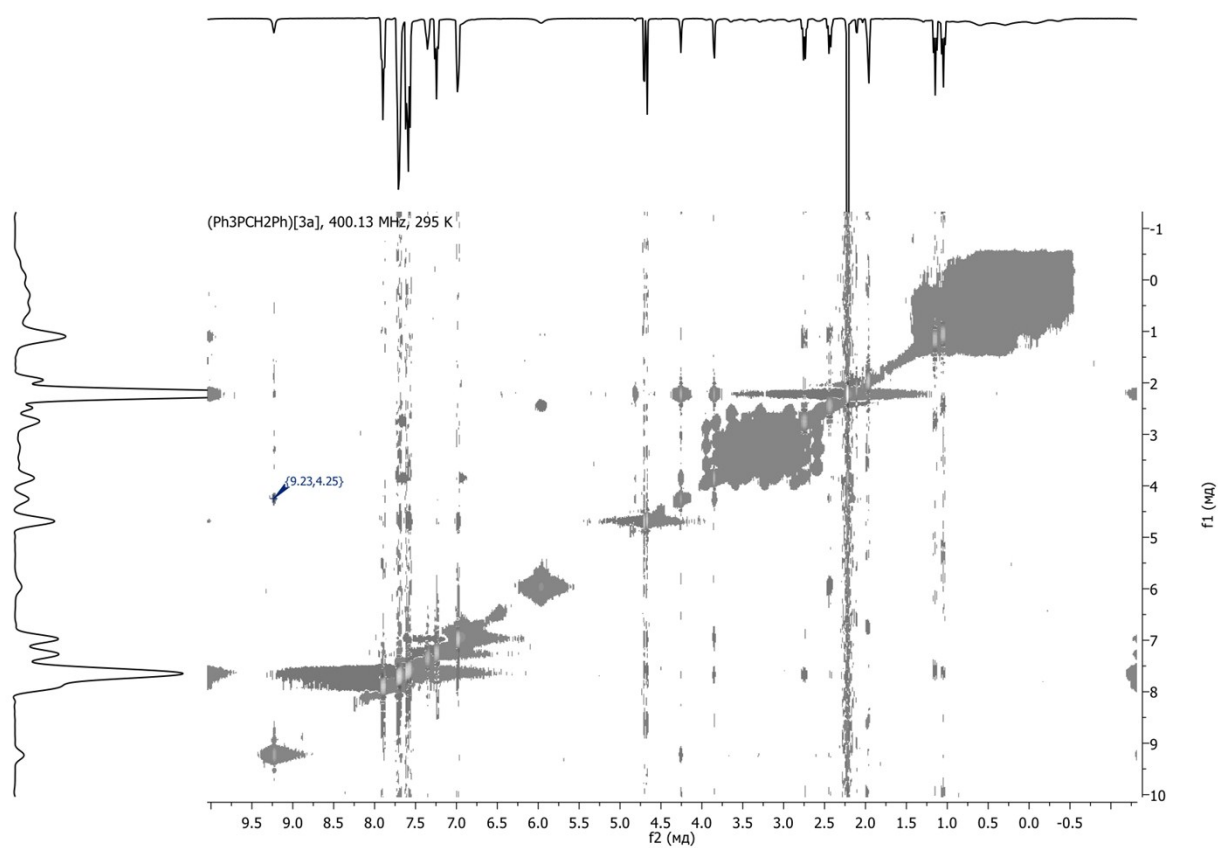
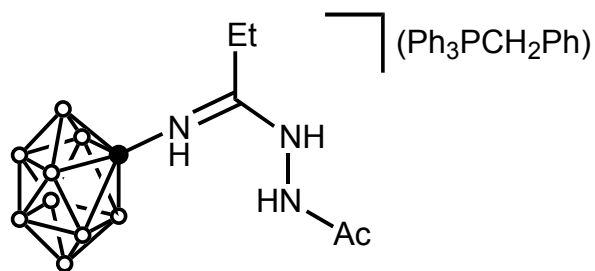


Figure 7S. NOESY-spectrum of (Ph₃PCH₂Ph)[**3a**].

(Ph₃PCH₂Ph)[**3b**]



Acquisition Parameter

Source Type	ESI	Ion Polarity	Negative	Set Nebulizer	1.0 Bar
Focus	Active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	3500 V	Set Dry Gas	4.0 l/min
Scan End	1200 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Source

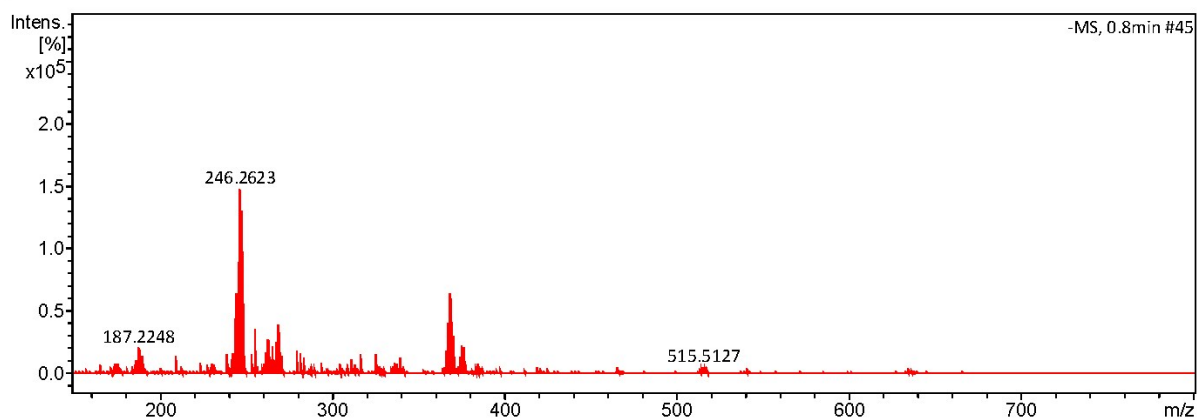


Figure 8S. HRESI-mass spectrum of (Ph₃PCH₂Ph)[**3b**].

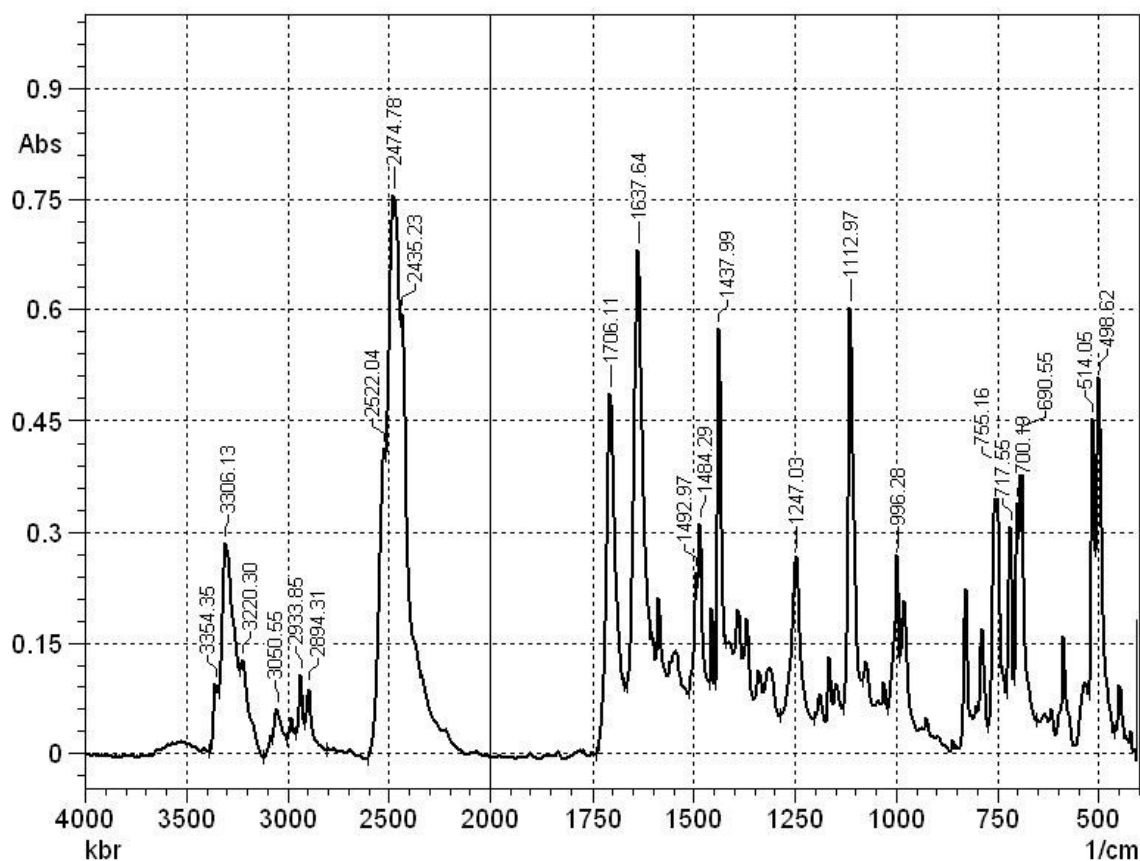


Figure 9S. IR spectrum of (Ph₃PCH₂Ph)[3b].

(Ph₃PCH₂Ph)[3b], 400.13 MHz, 295 K

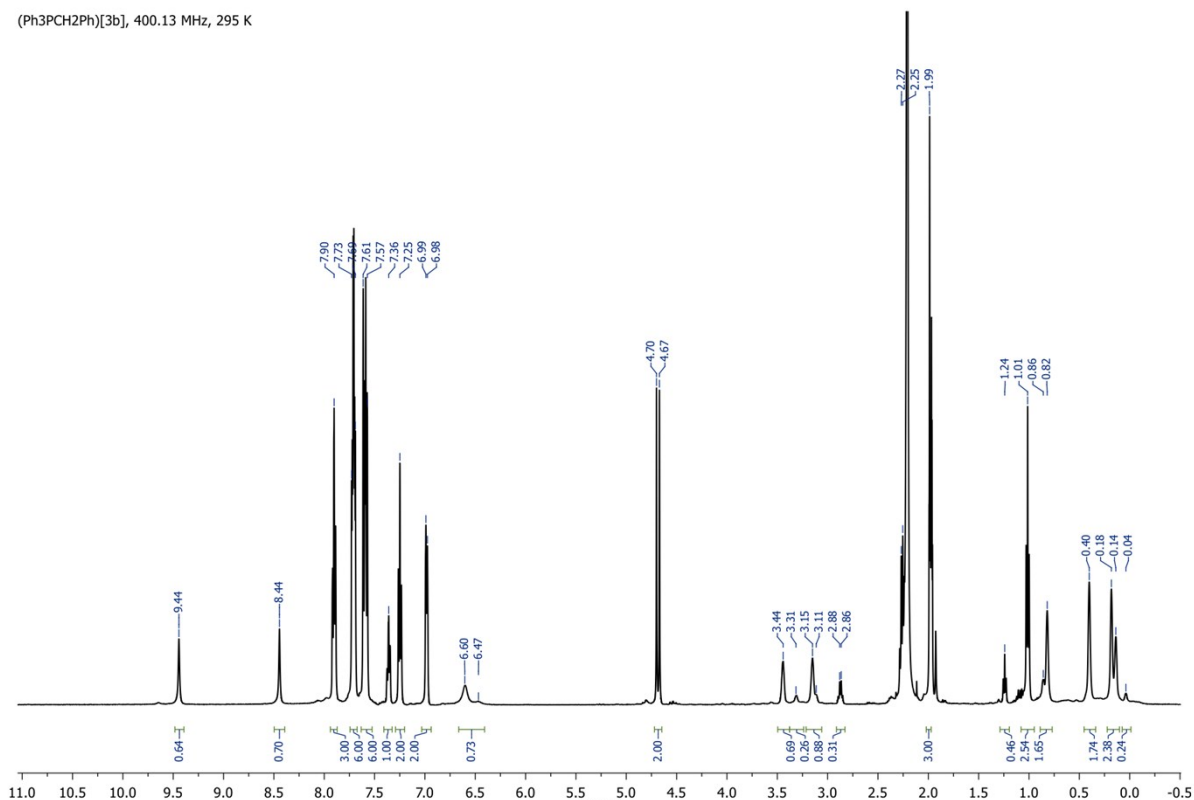


Figure 10S. ¹H{¹¹B} NMR spectrum of (Ph₃PCH₂Ph)[3b].

(Ph₃PCH₂Ph)[3b], 125.73 MHz, 297 K

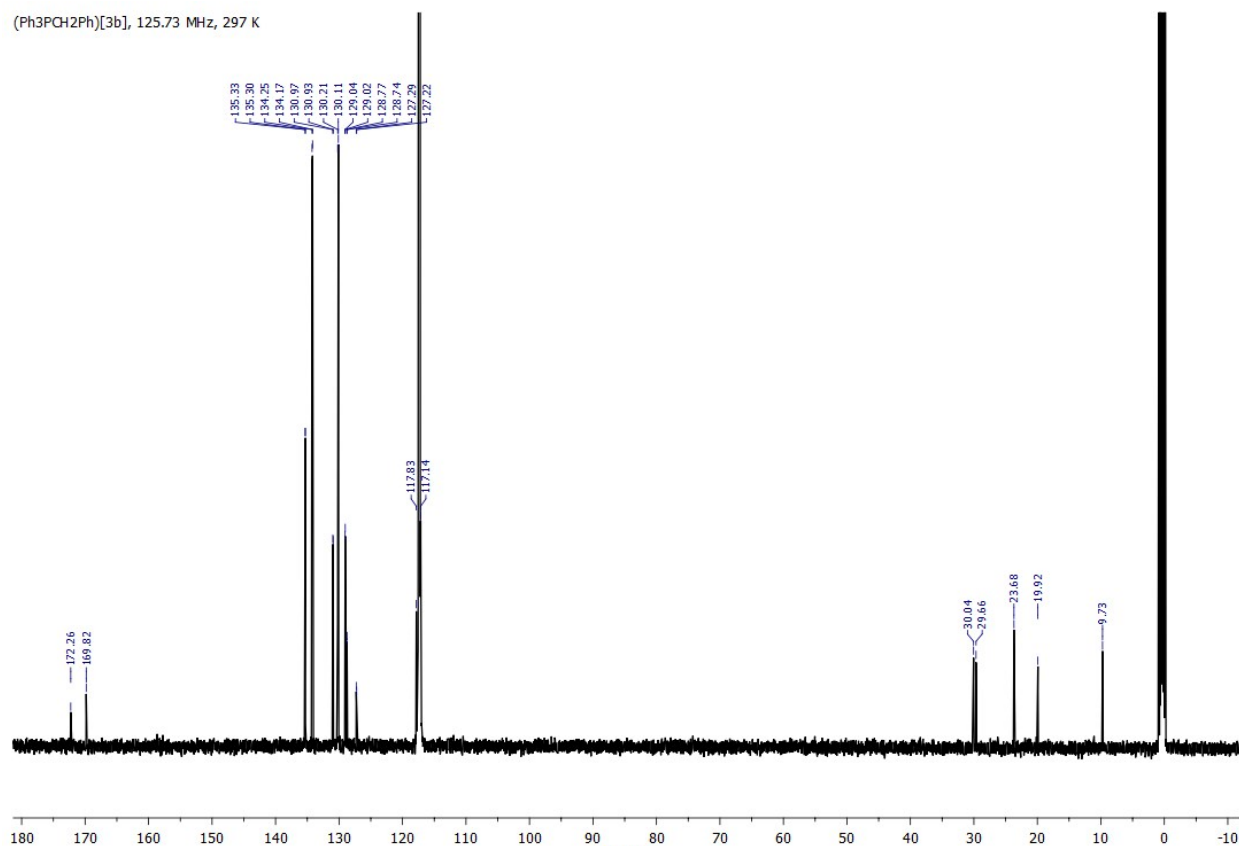


Figure 11S. ¹³C{¹H} NMR spectrum of (Ph₃PCH₂Ph)[3b].

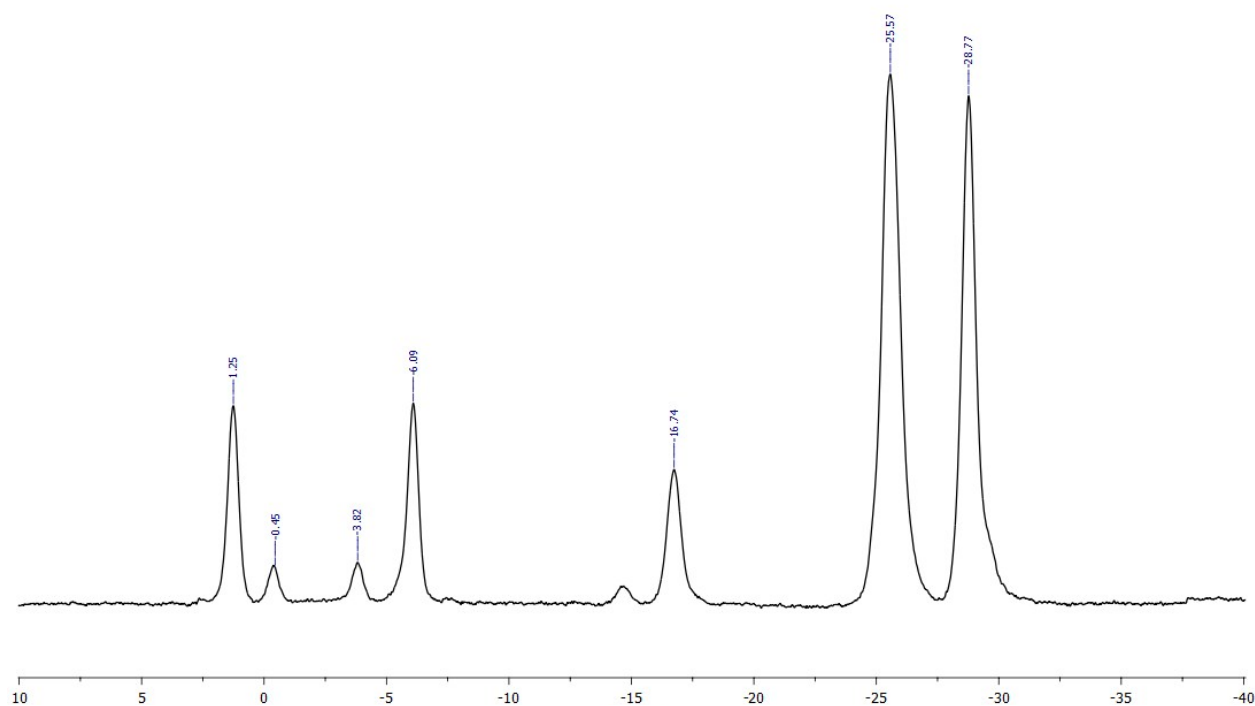
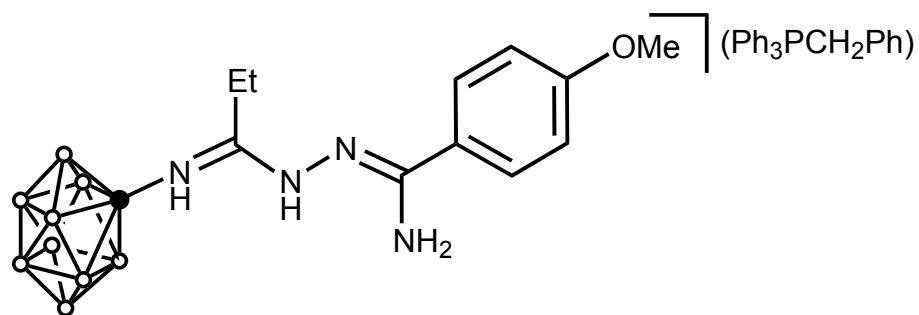


Figure 12S. $^{11}\text{B}\{^1\text{H}\}$ NMR spectrum of (Ph₃PCH₂Ph)[3b].

(Ph₃PCH₂Ph)[3c]



Acquisition Parameter

Source Type	ESI	Ion Polarity	Negative	Set Nebulizer	1.0 Bar
Focus	Active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	3500 V	Set Dry Gas	4.0 l/min
Scan End	1200 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Waste

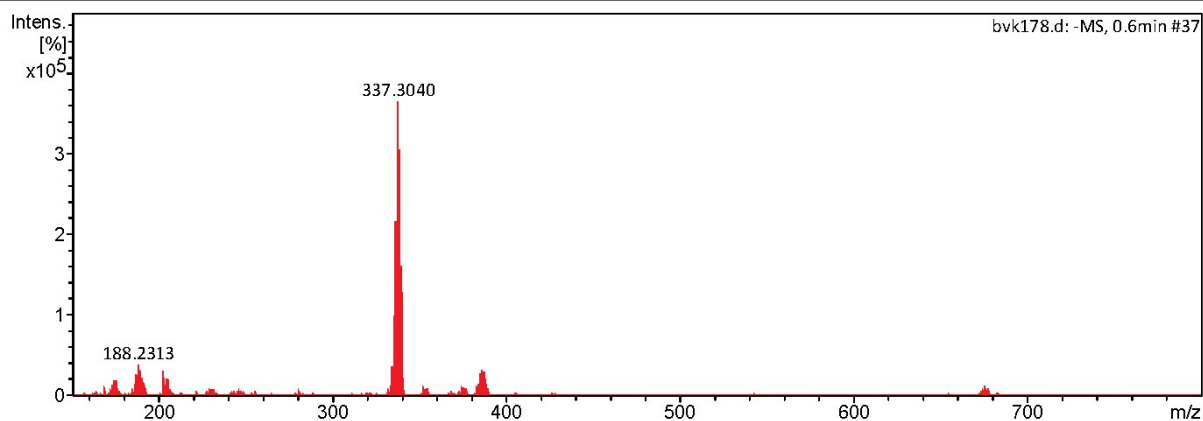


Figure 13S. HRESI-mass spectrum of (Ph₃PCH₂Ph)[3c].

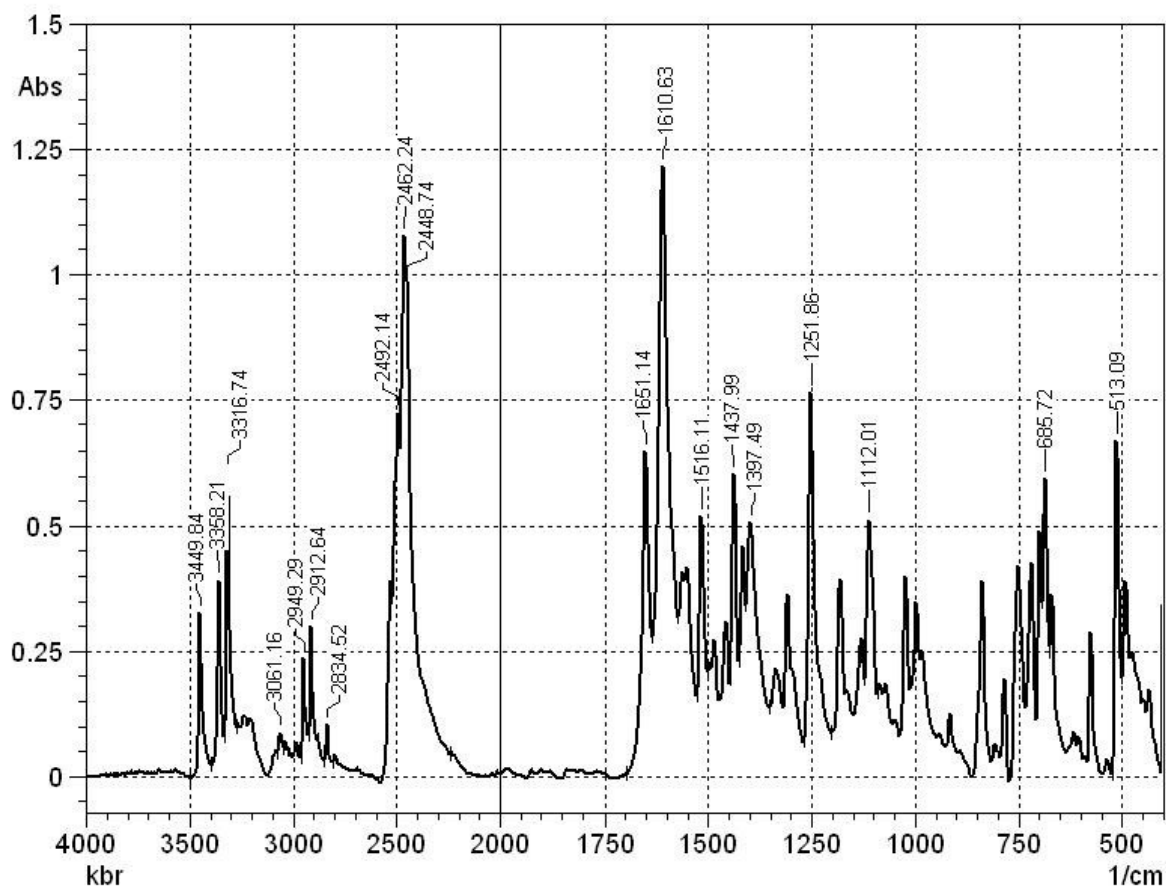


Figure 14S. IR spectrum of $(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3c}]$.

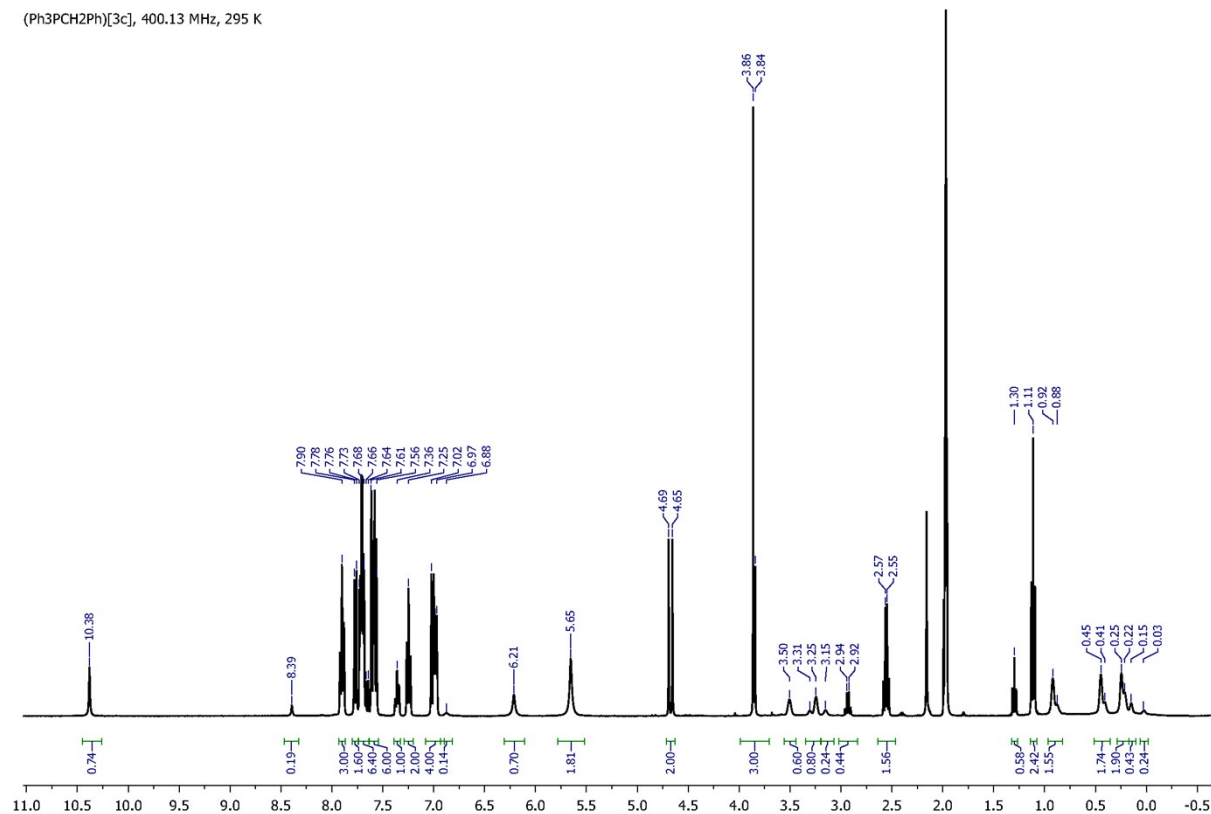


Figure 15S. $^1\text{H}\{^{11}\text{B}\}$ NMR spectrum of $(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3c}]$.

$(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3c}]$, 128.38 MHz, 298 K

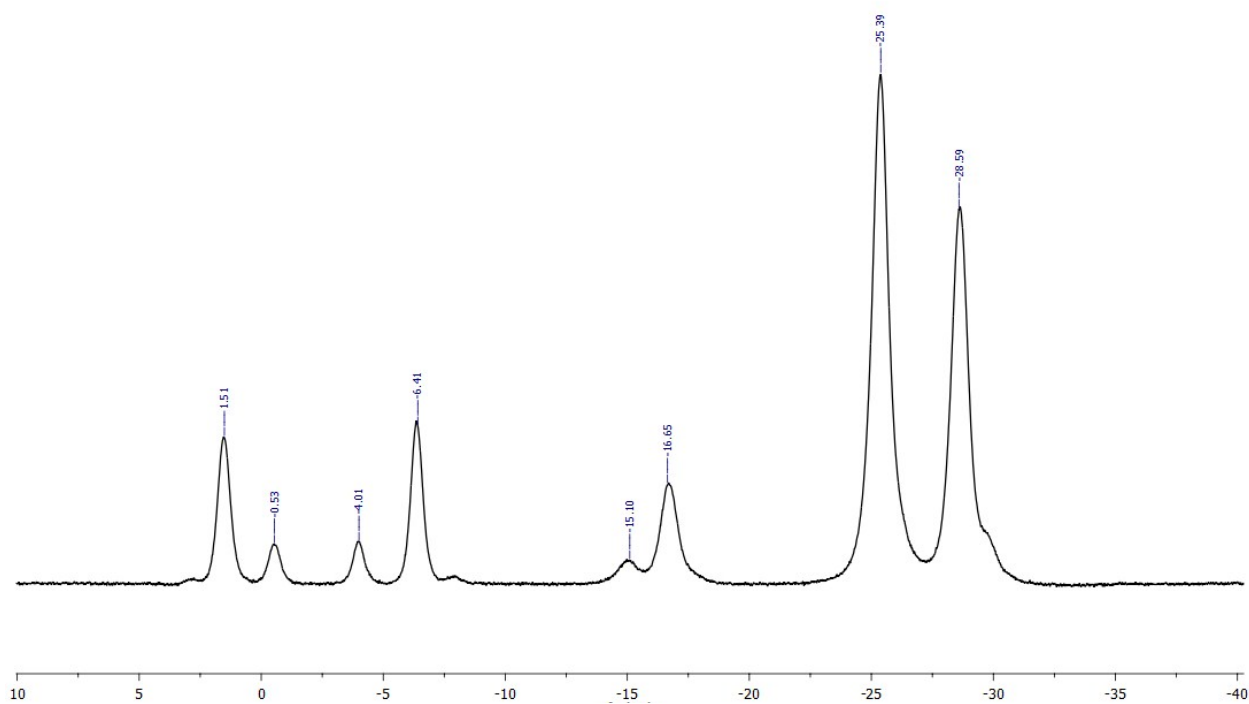


Figure 16S. $^{11}\text{B}\{^1\text{H}\}$ NMR spectrum of $(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3c}]$.

(Ph₃PCH₂Ph)[3c], 100.61 MHz, 298 K

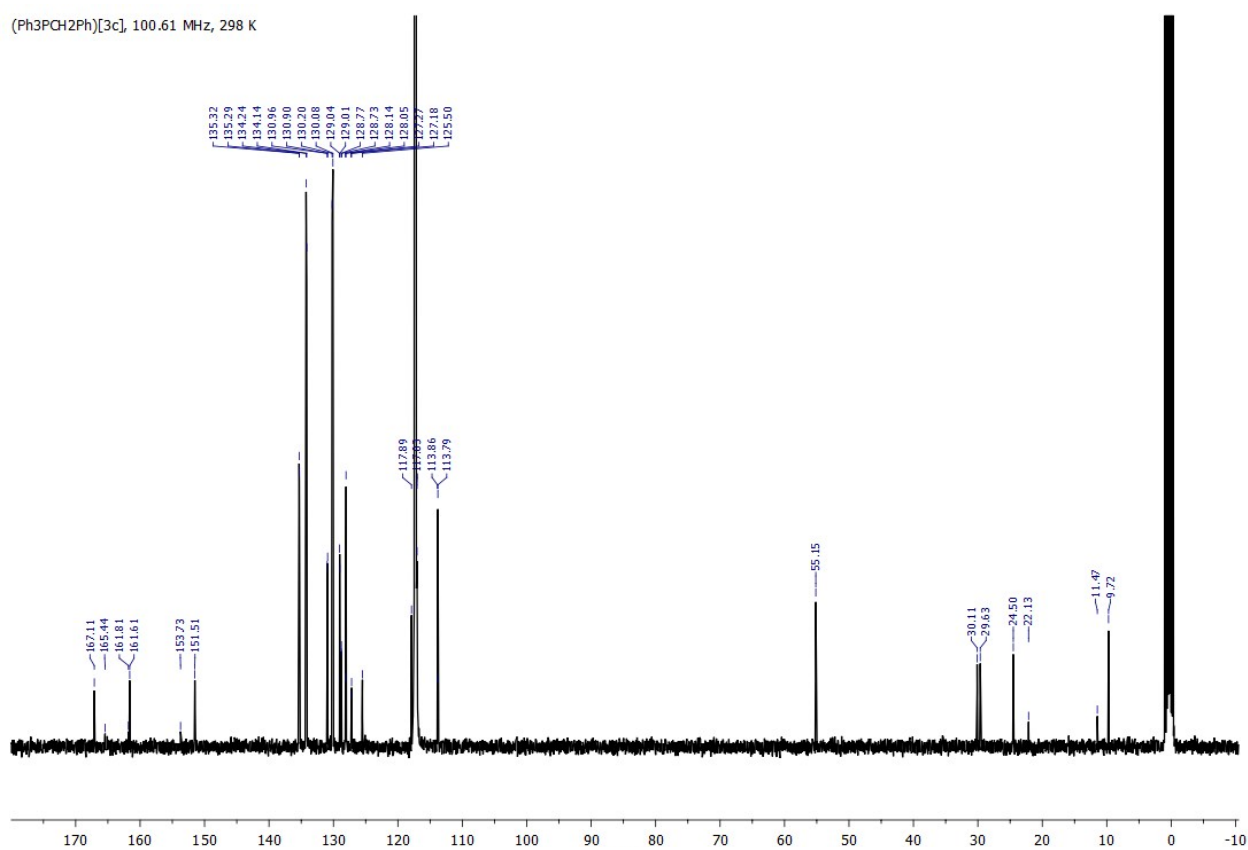
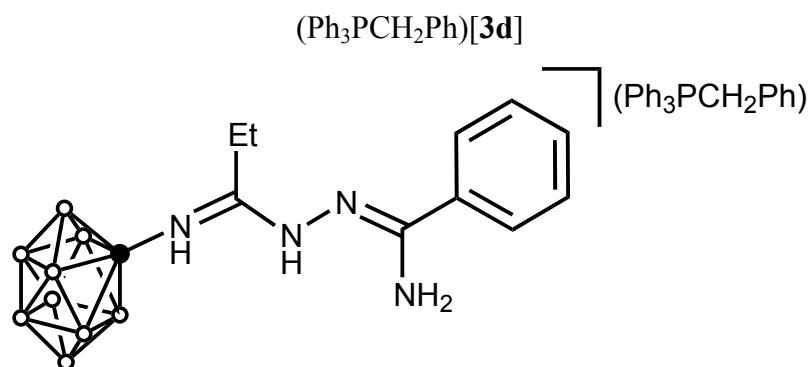


Figure 17S. ¹³C{¹H} NMR spectrum of (Ph₃PCH₂Ph)[3c].



Acquisition Parameter

Source Type	ESI	Ion Polarity	Negative	Set Nebulizer	1.0 Bar
Focus	Active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	3500 V	Set Dry Gas	4.0 l/min
Scan End	1200 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Source

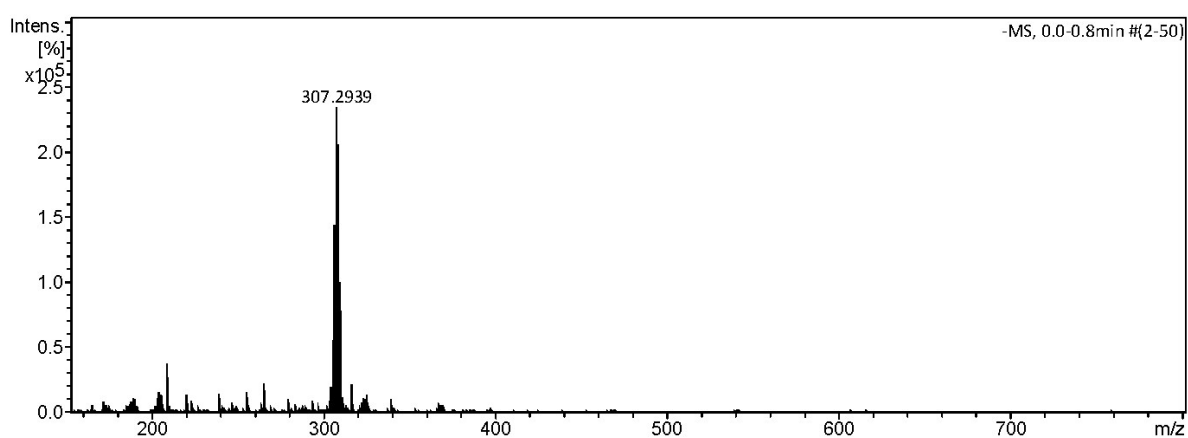


Figure 18S. HRESI⁻-mass spectrum of $(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3d}]$.

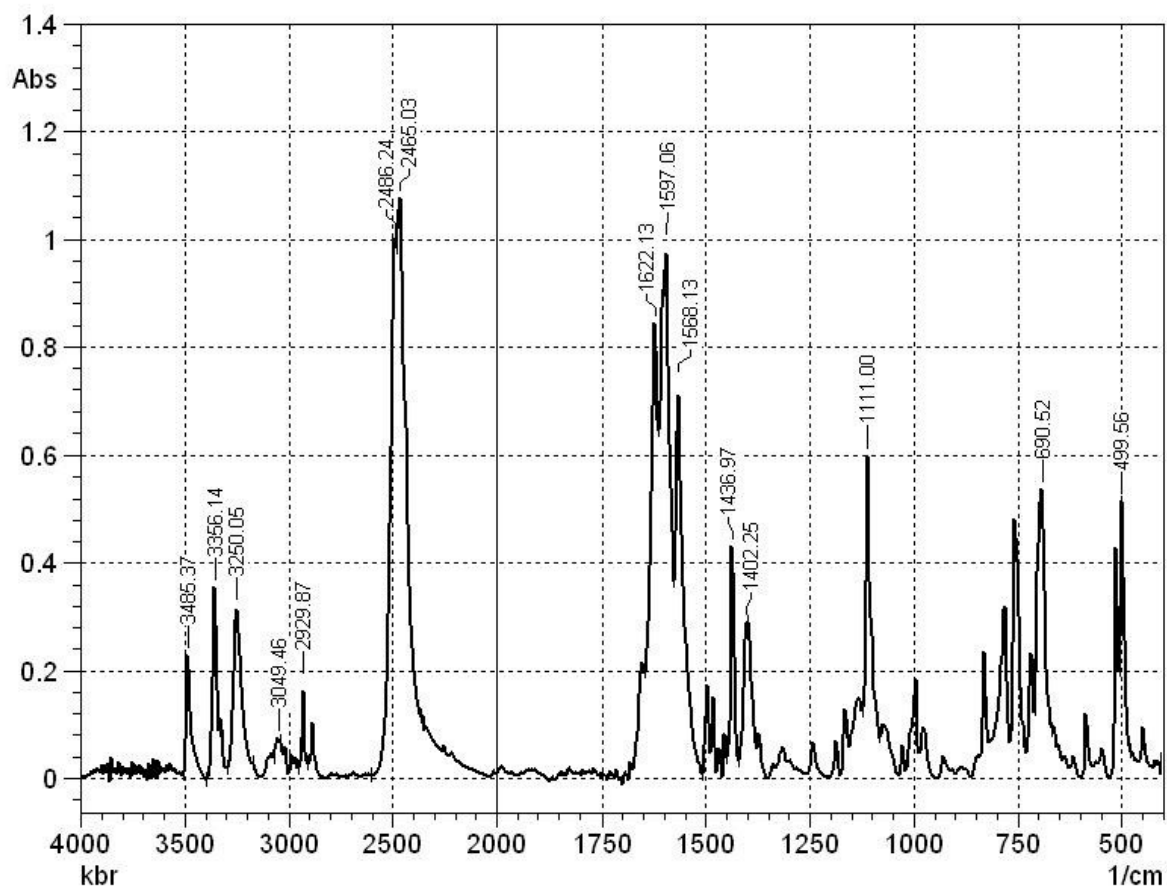


Figure 19S. IR spectrum of (Ph₃PCH₂Ph)[3d].

(Ph₃PCH₂Ph)[3d], 400.13 MHz, 295 K

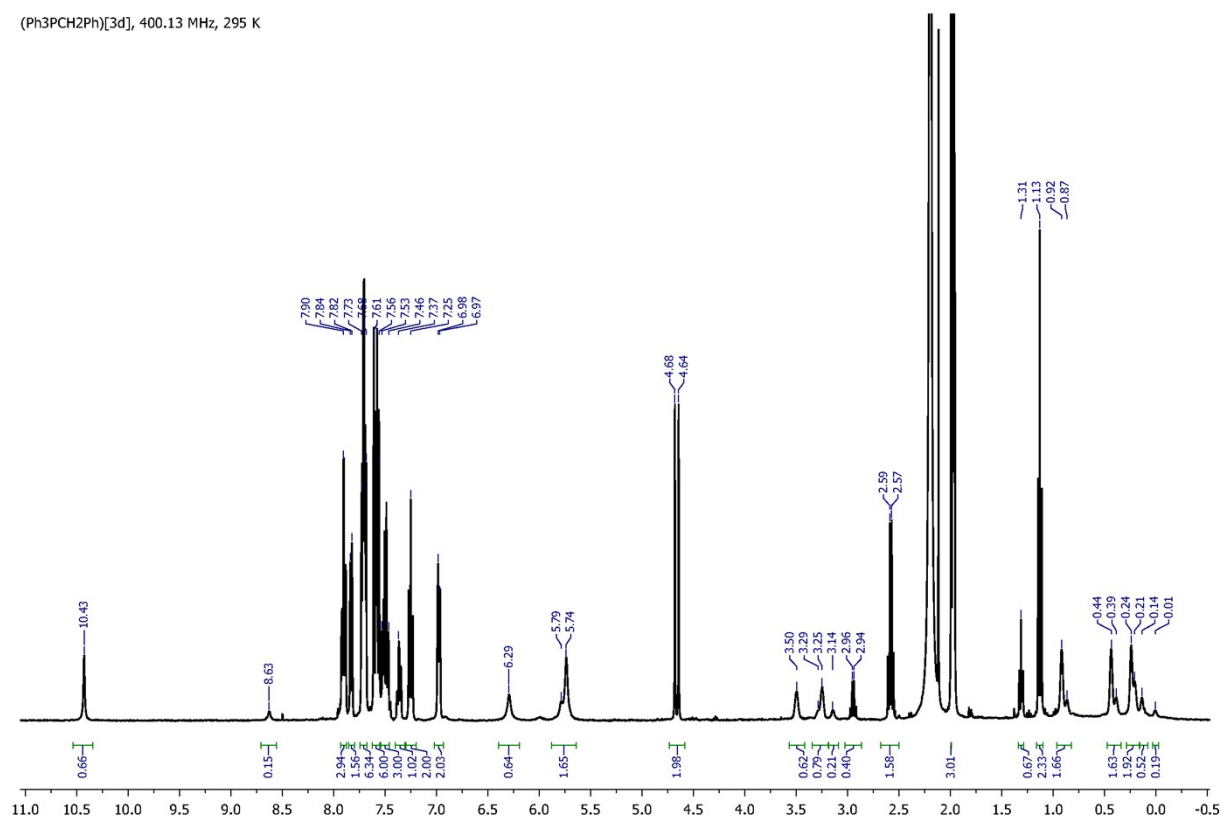


Figure 20S. ¹H{¹¹B} NMR spectrum of (Ph₃PCH₂Ph)[3d].

(Ph₃PCH₂Ph)[3d], 128.38 MHz, 298 K

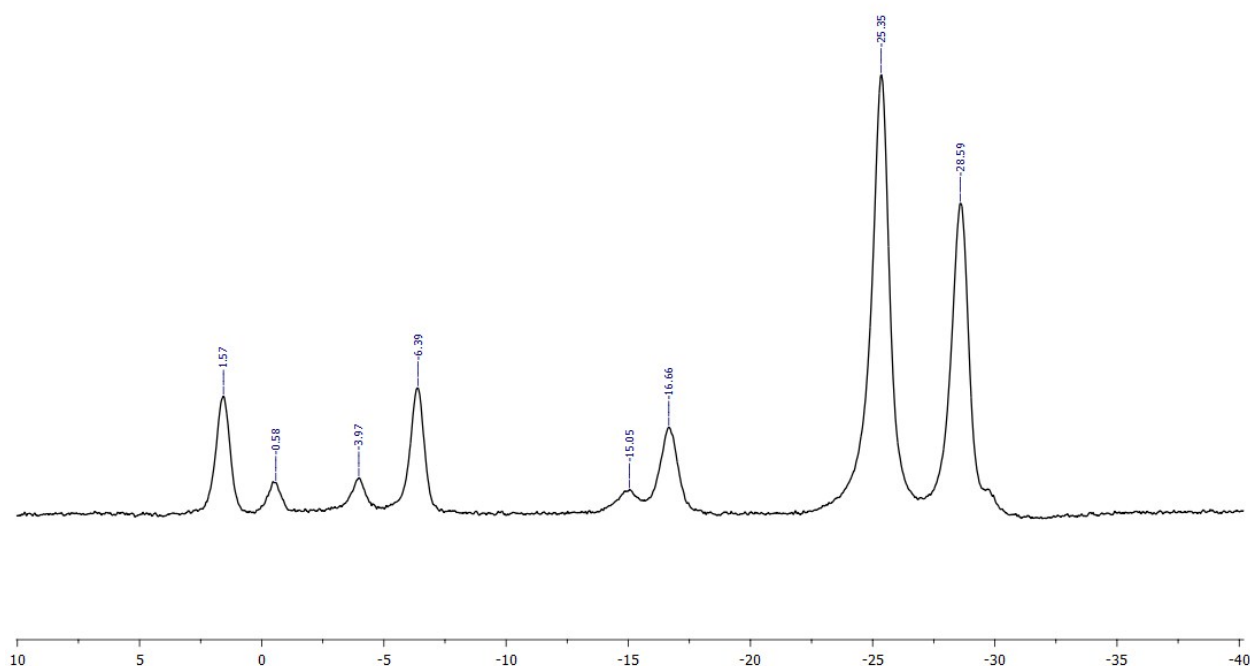


Figure 21S. ¹¹B{¹H} NMR spectrum of (Ph₃PCH₂Ph)[3d].

(Ph₃PCH₂Ph)[3d], 100.61 MHz, 298 K

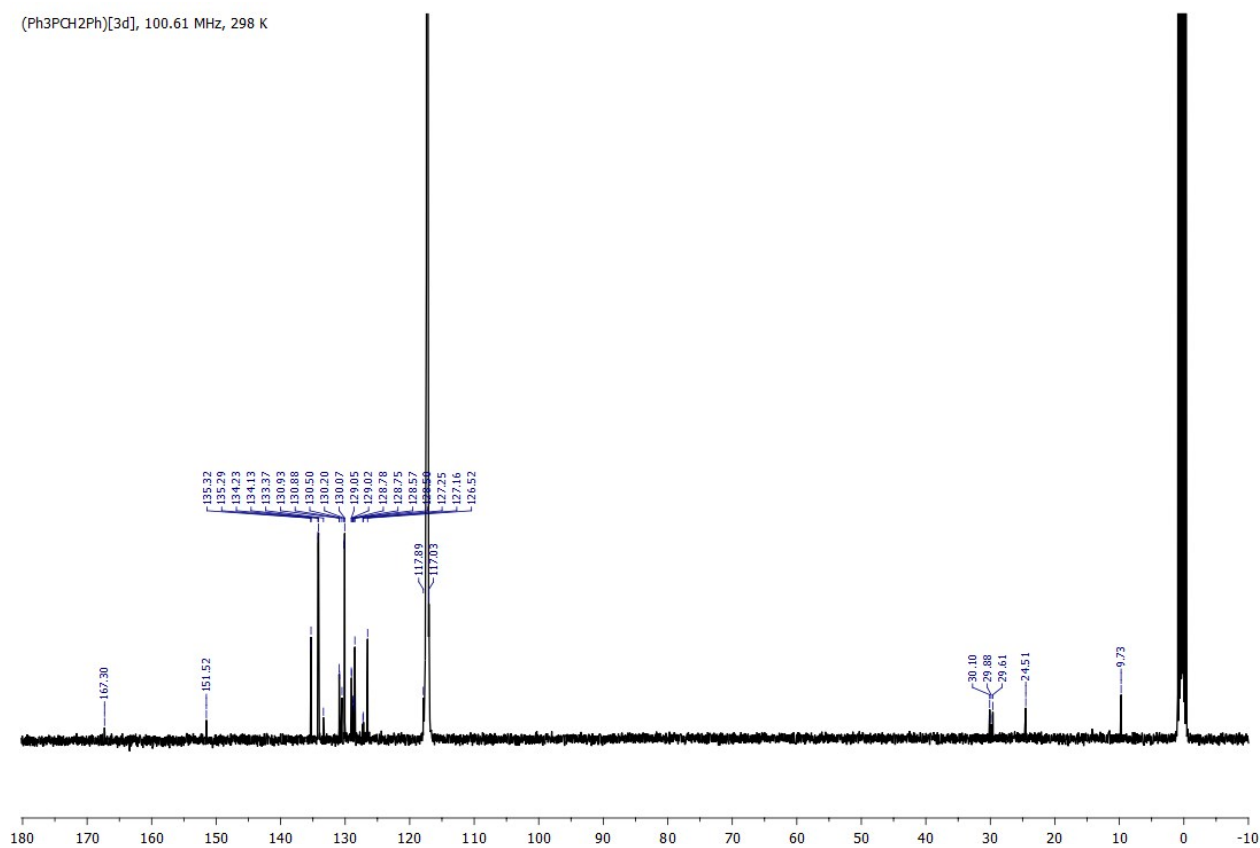
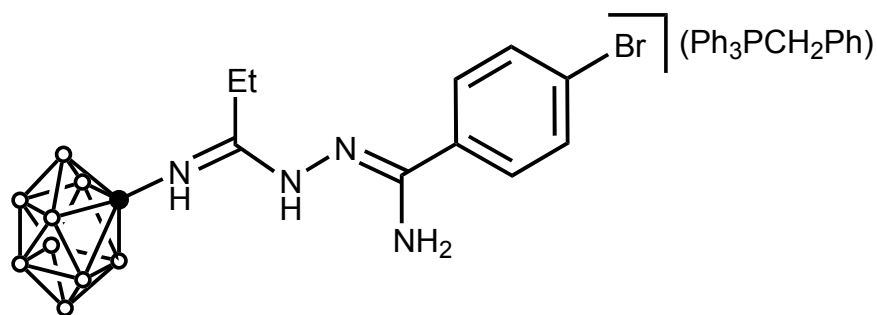


Figure 22S. ¹³C{¹H} NMR spectrum of (Ph₃PCH₂Ph)[3d].

(Ph₃PCH₂Ph)[3e]



Acquisition Parameter

Source Type	ESI	Ion Polarity	Negative	Set Nebulizer	1.0 Bar
Focus	Active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	3500 V	Set Dry Gas	4.0 l/min
Scan End	1200 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Source

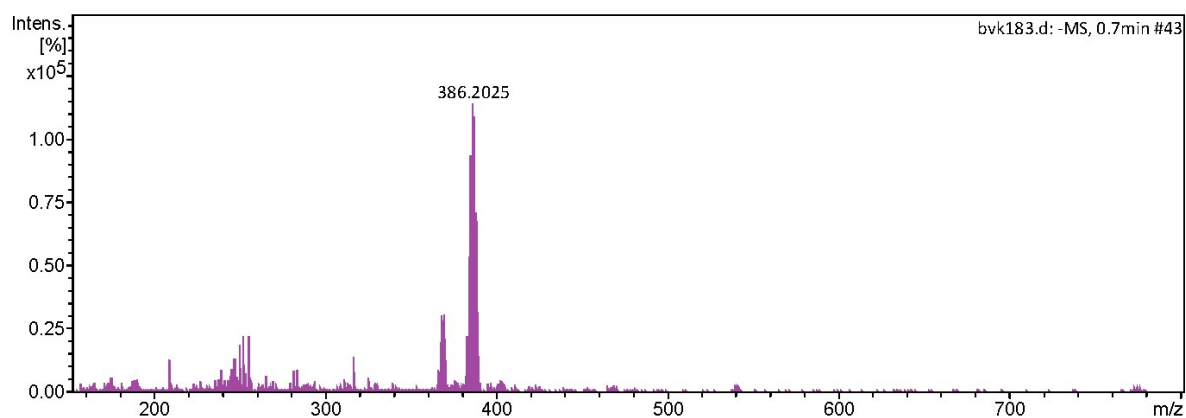


Figure 23S. HRESI-mass spectrum of (Ph₃PCH₂Ph)[3e].

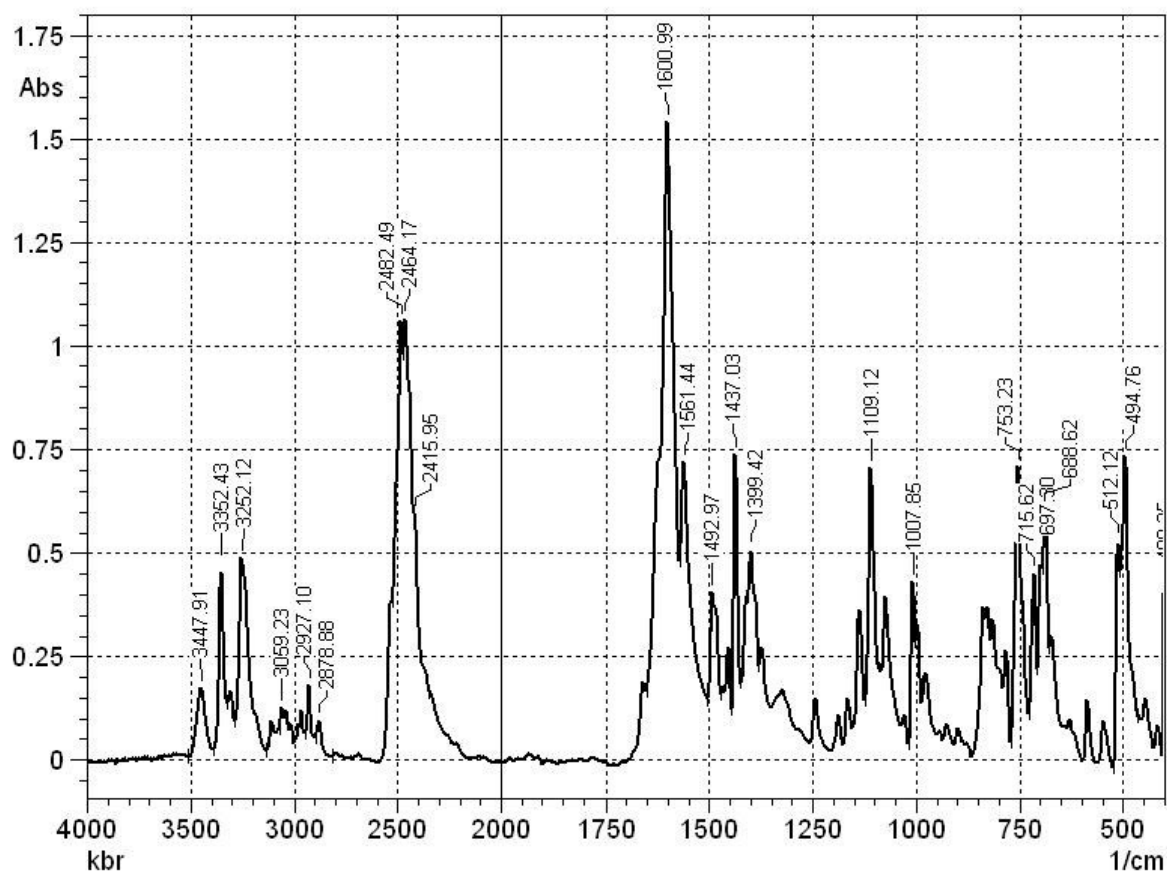


Figure 24S. IR spectrum of (Ph₃PCH₂Ph)[3e].

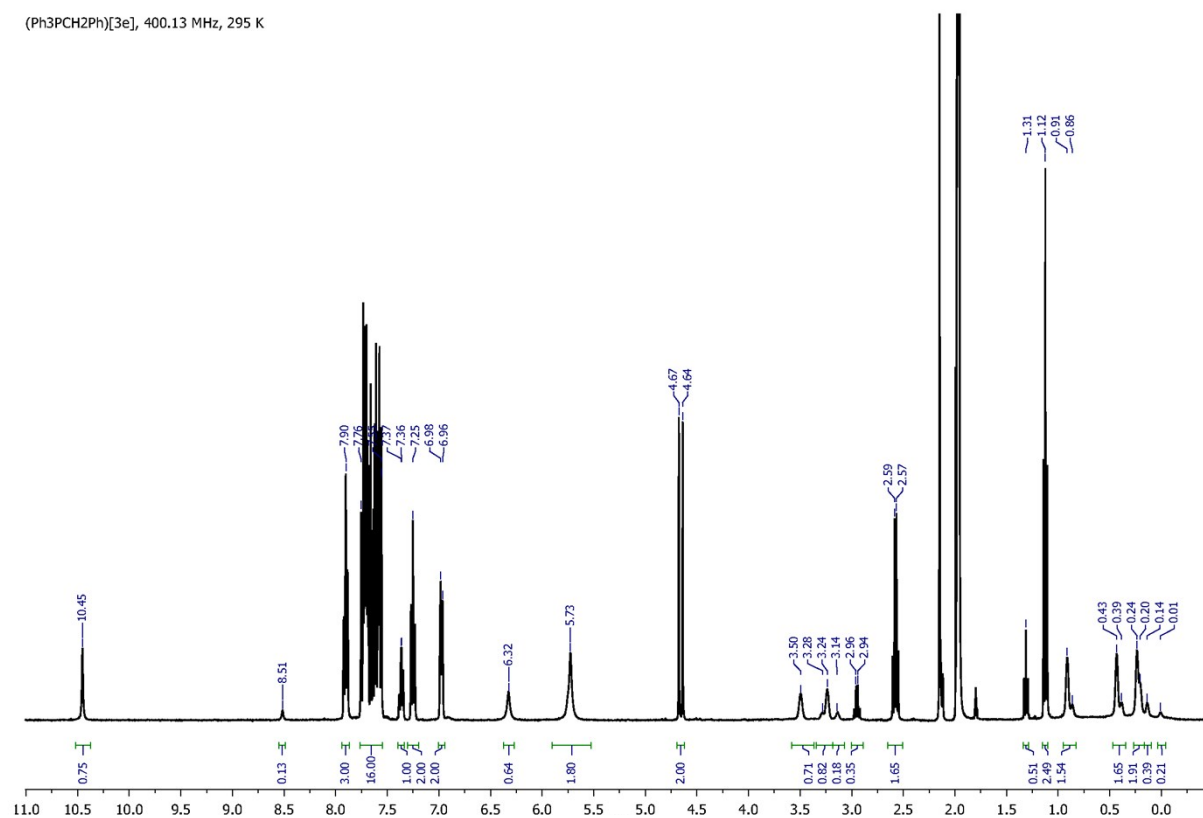


Figure 25S. ¹H{¹¹B} NMR spectrum of (Ph₃PCH₂Ph)[3e].

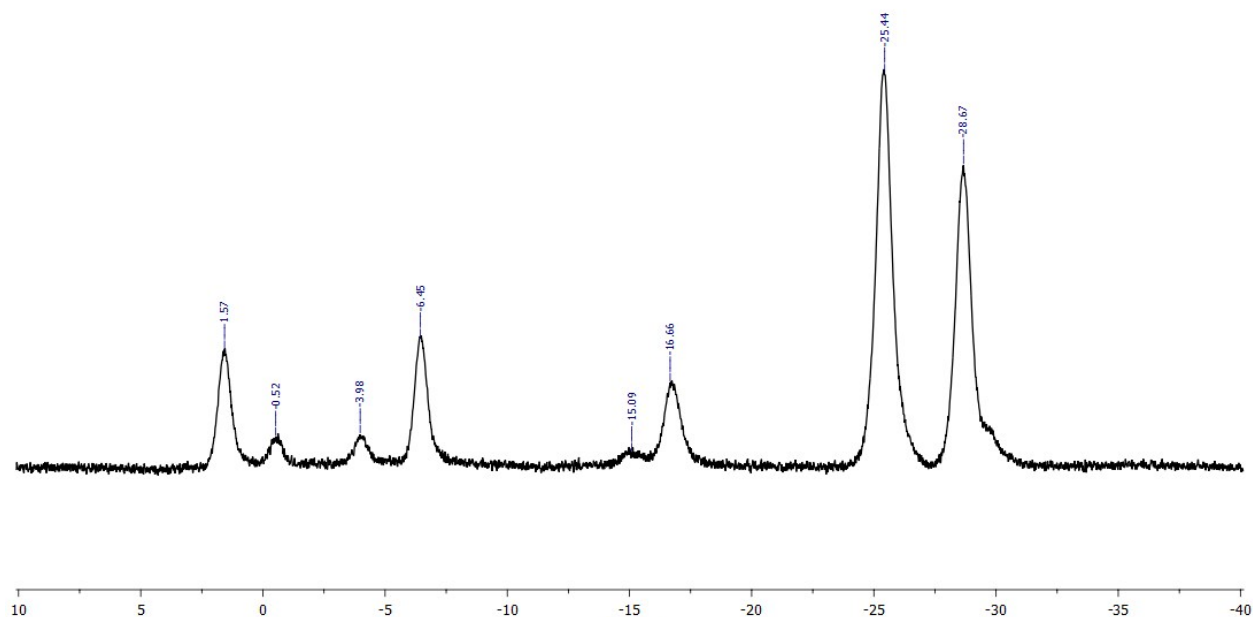
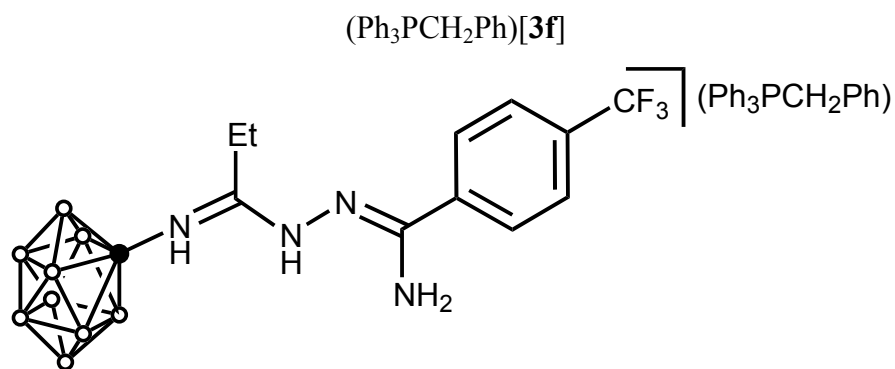


Figure 26S. ¹¹B{¹H} NMR spectrum of (Ph₃PCH₂Ph)[3e].



Acquisition Parameter

Source Type	ESI	Ion Polarity	Negative	Set Nebulizer	1.0 Bar
Focus	Active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	3500 V	Set Dry Gas	4.0 l/min
Scan End	1200 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Waste

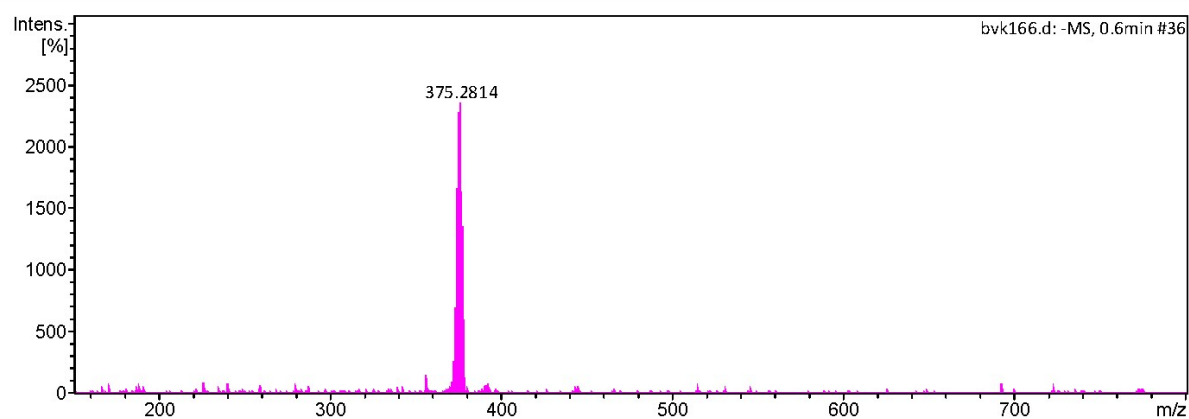


Figure 27S. HRESI[−]-mass spectrum of (Ph₃PCH₂Ph)[3f].

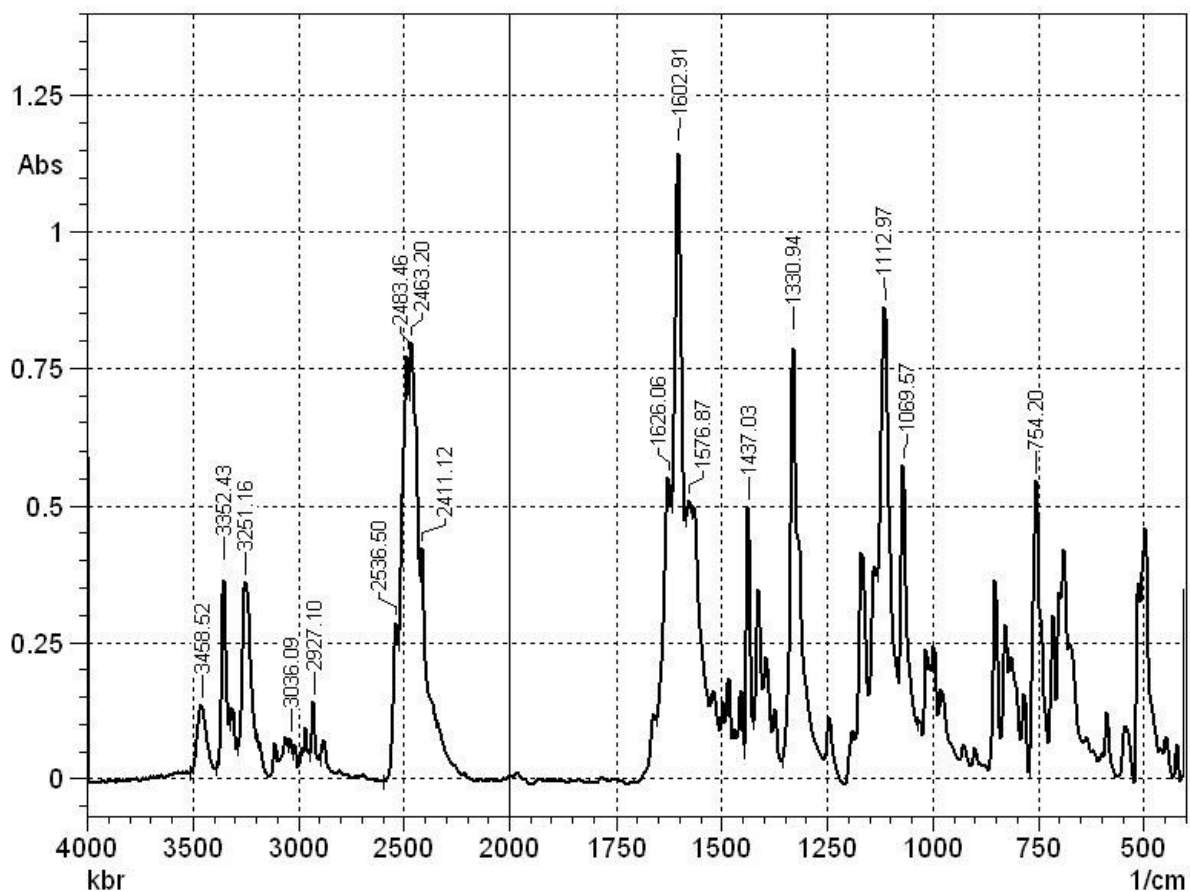


Figure 28S. IR spectrum of $(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3f}]$.

$(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3f}]$, 400.13 MHz, 295 K

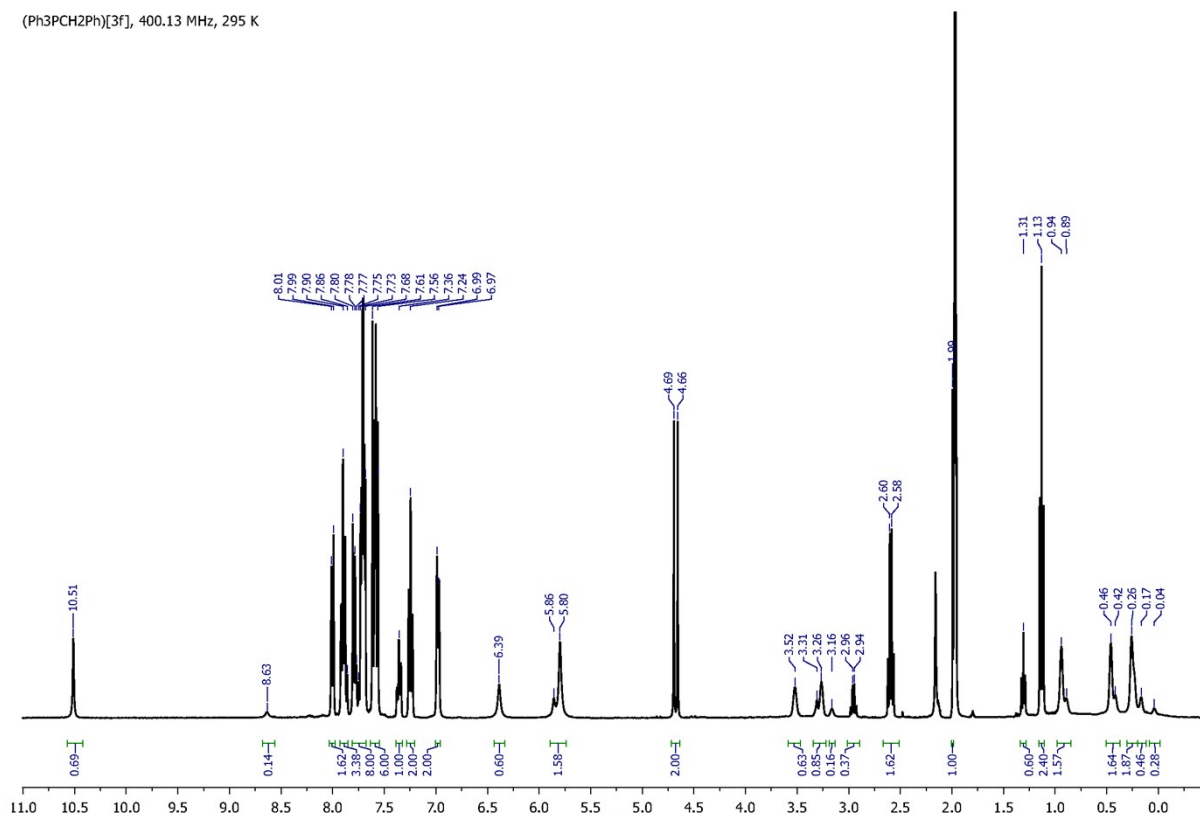


Figure 29S. $^1\text{H}\{^{11}\text{B}\}$ NMR spectrum of $(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3f}]$.

(Ph₃PCH₂Ph)[3f], 128.38 MHz, 298 K

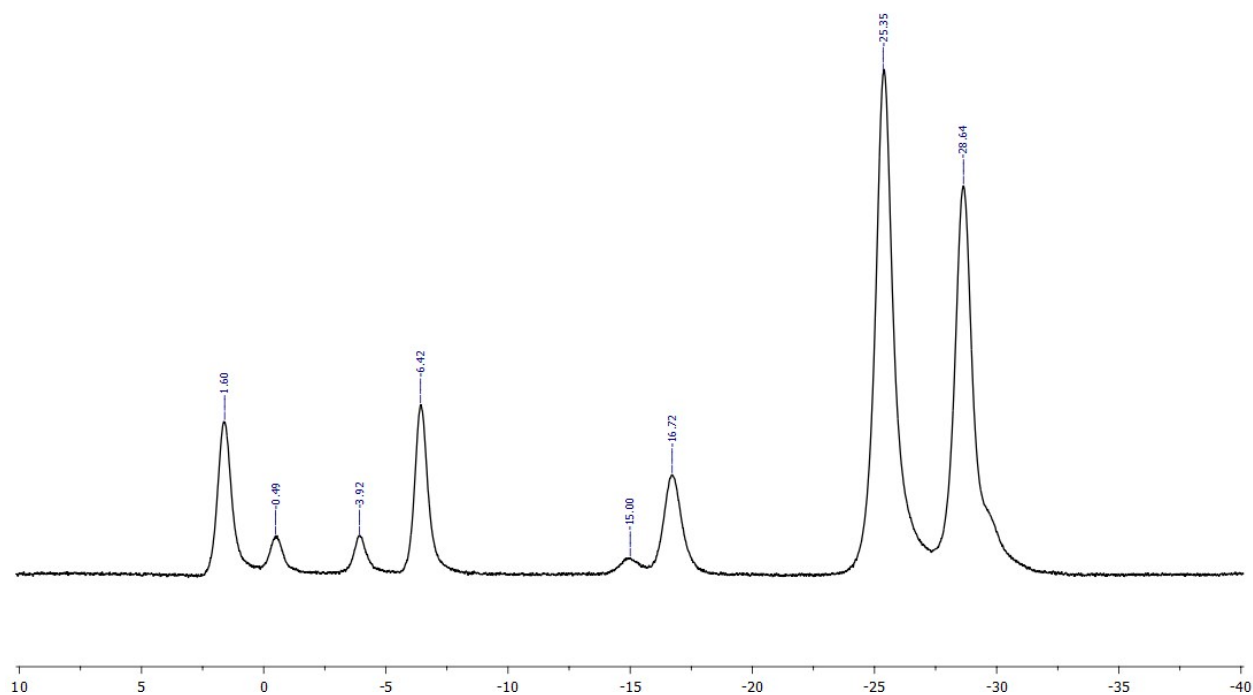
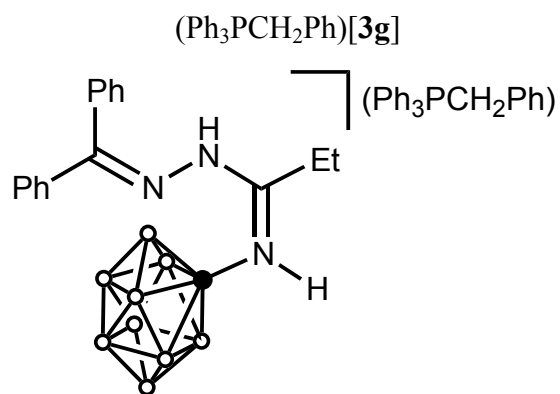


Figure 30S. $^{11}\text{B}\{^1\text{H}\}$ NMR spectrum of (Ph₃PCH₂Ph)[3f].



Acquisition Parameter

Source Type	ESI	Ion Polarity	Negative	Set Nebulizer	1.0 Bar
Focus	Active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	3500 V	Set Dry Gas	4.0 l/min
Scan End	1200 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Source

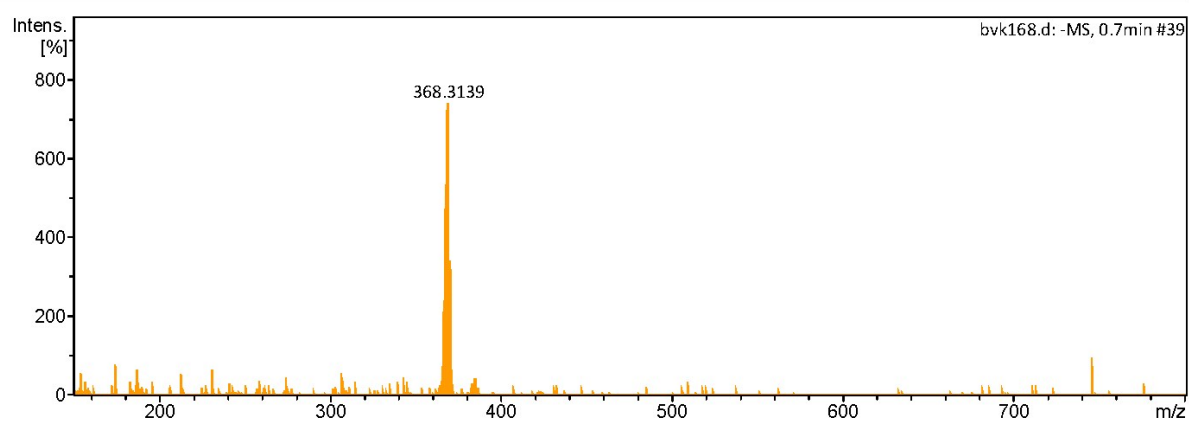


Figure 31S. HRESI⁻-mass spectrum of (Ph₃PCH₂Ph)[**3g**].

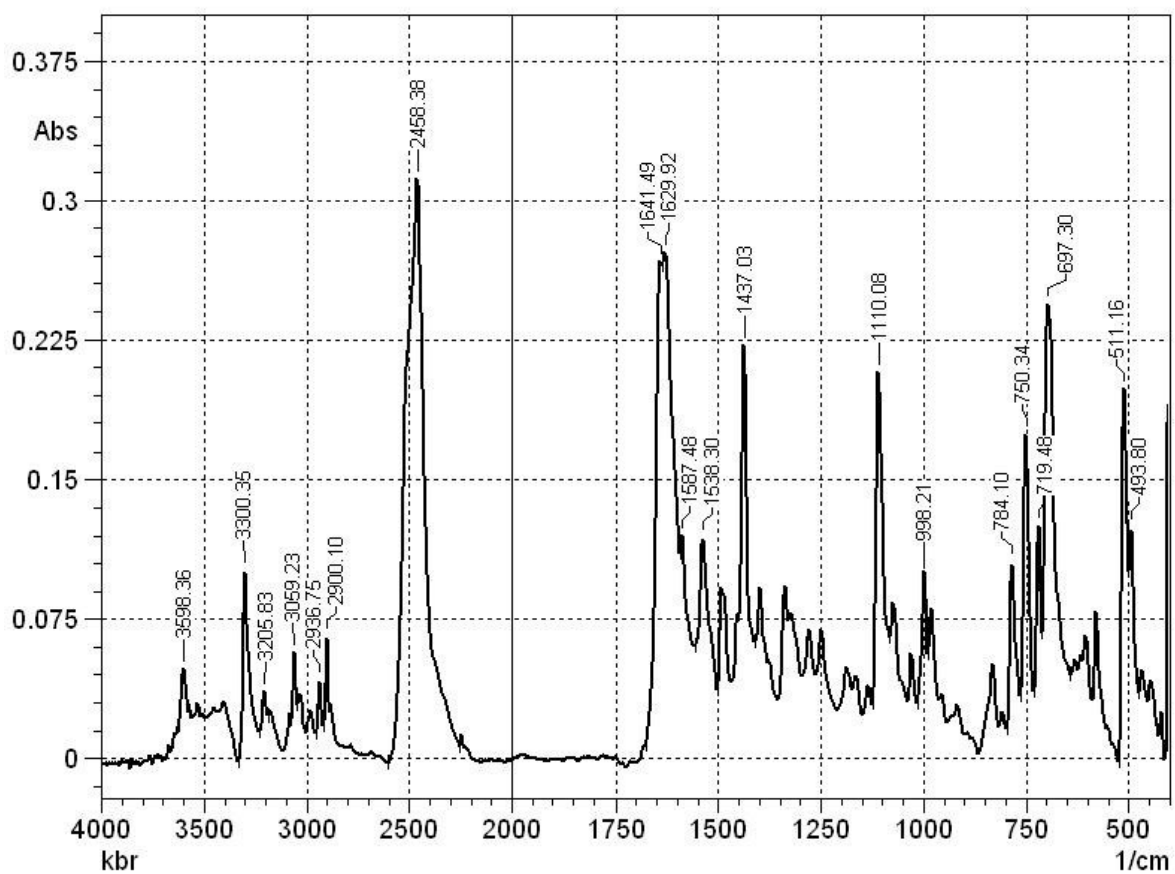


Figure 32S. IR spectrum of $(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3g}]$.

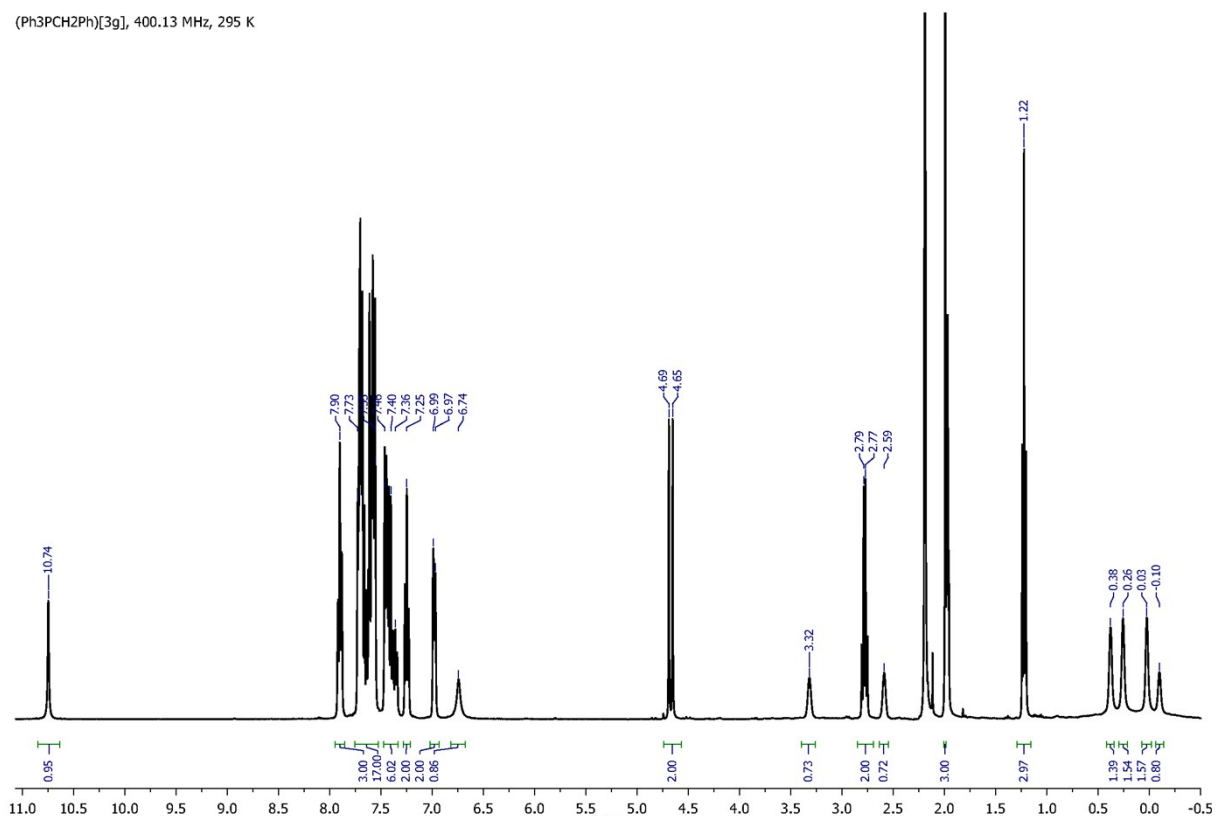


Figure 33S. $^1\text{H}\{^{11}\text{B}\}$ NMR spectrum of $(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{3g}]$.

(Ph₃PCH₂Ph)[3g], 128.38 MHz, 298 K

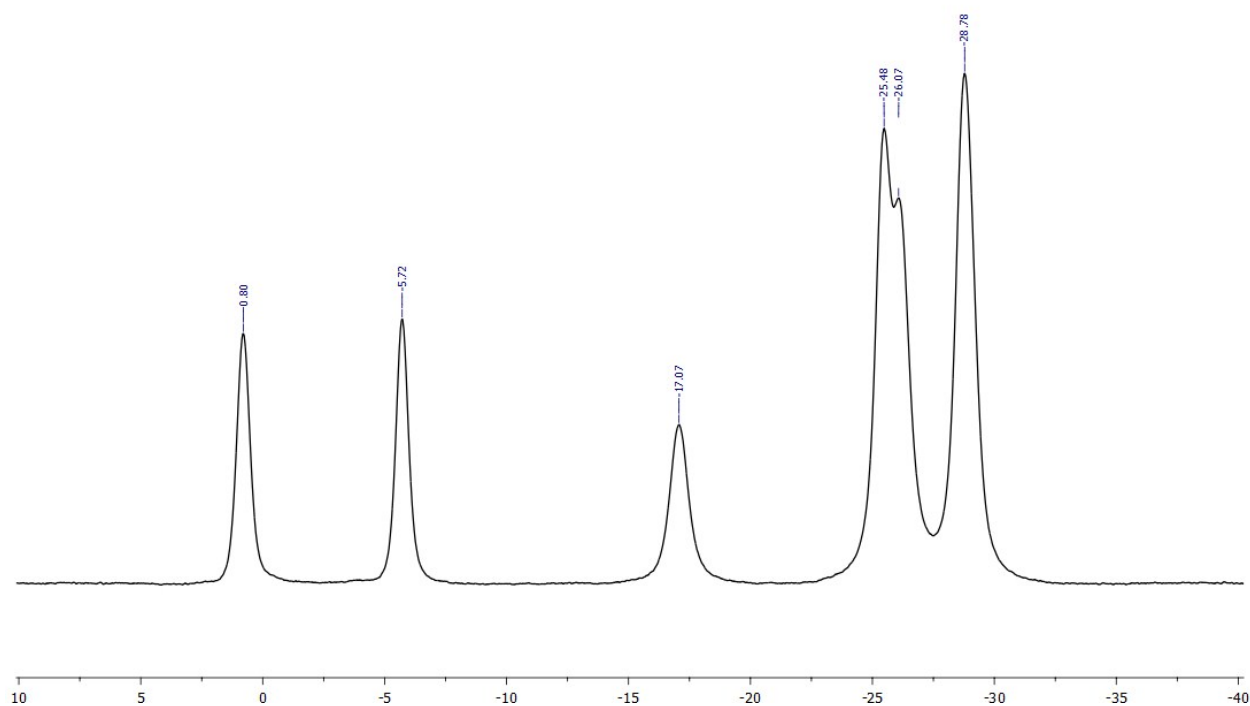


Figure 34S. ¹¹B{¹H} NMR spectrum of (Ph₃PCH₂Ph)[3g].

(Ph₃PCH₂Ph)[3g], 100.61 MHz, 298 K

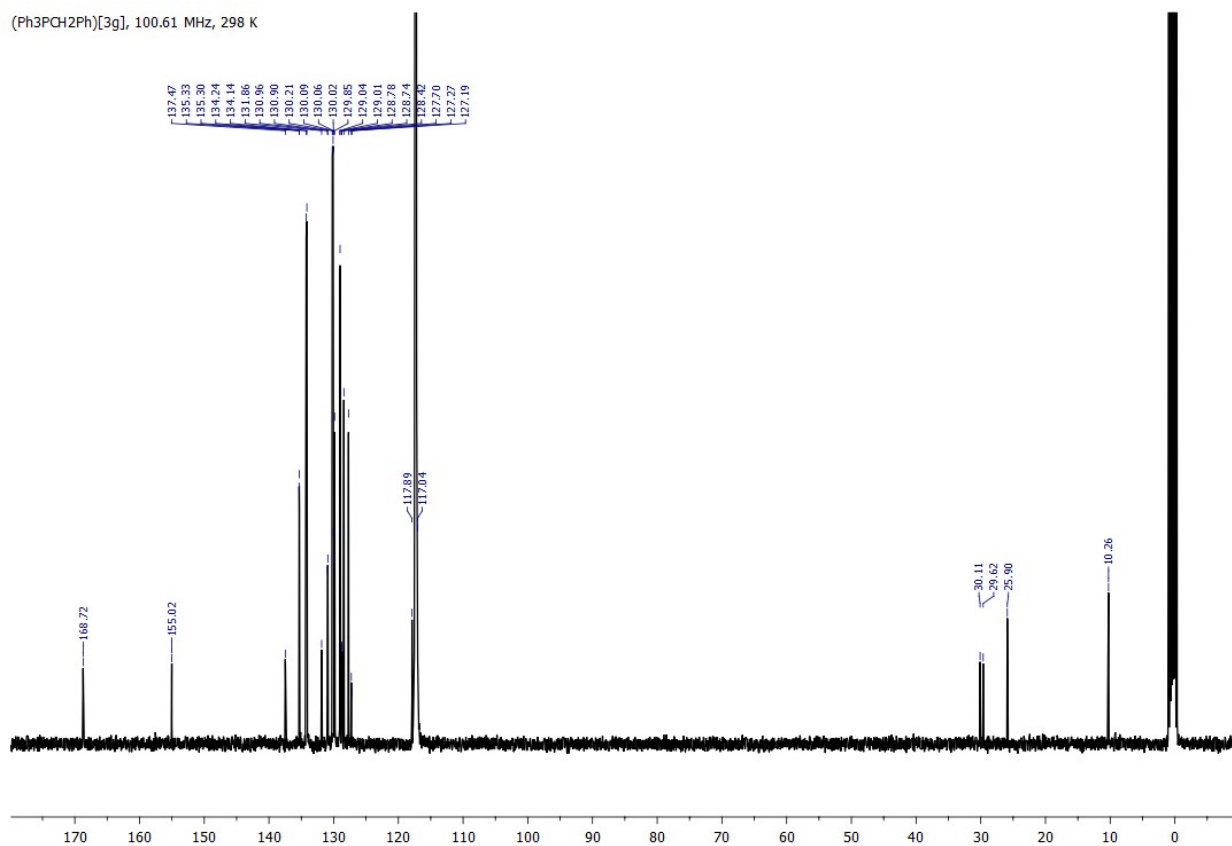
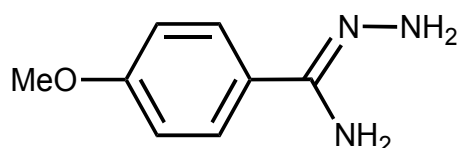


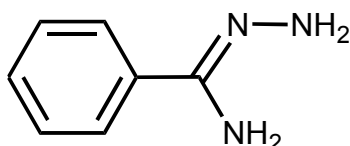
Figure 35S. ¹³C{¹H} NMR spectrum of (Ph₃PCH₂Ph)[3g].

Synthesis and Characterization of 2c–f

Dry gaseous HCl was bubbled through an ice-cooled vigorously stirred solution/suspension of corresponding benzonitrile (0.05 mol) in absolute MeOH (25 mL) for 3 h. The mixture was left stirring overnight and then poured into Et₂O (100 mL). Colorless crystals were collected and washed with two Et₂O 30-mL portions. The obtained iminoester hydrochloride was added to an ice-cooled stirred mixture of CH₂Cl₂ (80 mL) and saturated aqueous NaHCO₃ (80 mL). The organic layer was separated and aqueous phase was extracted with CH₂Cl₂ (50 mL). The combined organic phases were dried over CaCl₂ and evaporated to dryness. Hydrazine hydrate (0.035 mol) was added dropwisely to an ice-cooled stirred solution of the resulted iminoester (0.03 mol) in *i*PrOH (50 mL). The mixture was stirred for 1 h under gradual cooling to RT and then at RT overnight. The residue after evaporation of volatiles was treated with Et₂O (50 mL) and cooled in ice. Crystals were filtered off and dried in *vacuo*.

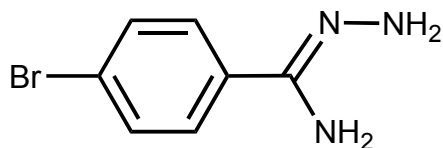


2c. Yield: 42%. Anal. Calcd for C₈H₁₁N₃O: C, 58.17; H, 6.71; N, 25.44. Found: C, 57.89; H, 6.75; N, 25.68. HRESI⁺-MS (*m/z*): 166.0969 ([M + H]⁺, calcd 166.09751). HRESI⁻-MS (*m/z*): 164.0813 ([M – H]⁻, calcd 164.0824). IR (KBr, selected bonds, cm⁻¹): 3396(m), 3269(m), 3176(m) ν(N–H); 2962(w) ν(C–H); 1642(s) ν(C=N); 1243(s) ν(C_{ar}–O–C). ¹H NMR (δ): 7.61 (d, *J* = 8.9 Hz, 2H, C₆H₄), 6.93 (d, *J* = 8.9 Hz, 2H, C₆H₄), 4.94 (s, br, 2H, NH₂), 4.39 (s, br, 2H, NH₂), 3.83 (s, 3H, CH₃). ¹³C{¹H} NMR (δ): 160.29 (NCN), 127.67, 126.88, 126.81, 113.49 (C₆H₄), 54.98 (OCH₃).

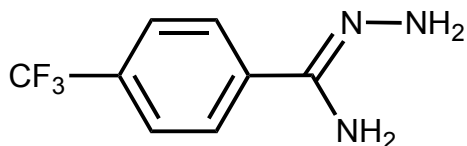


2d. Yield: 53%. Anal. Calcd for C₇H₉N₃: C, 62.20; H, 6.71; N, 31.09. Found: C, 62.03; H, 6.71; N, 31.17. HRESI⁺-MS (*m/z*): 136.0877 ([M + H]⁺, calcd 136.0869). HRESI⁻-MS (*m/z*): 134.0706 ([M – H]⁻, calcd 134.0718). IR (KBr, selected bonds, cm⁻¹): 3396(m), 3326(m), 3161(m) ν(N–H);

1640(s) $\nu(\text{C}=\text{N})$. ^1H NMR (δ): 7.72–7.67 (m, 2H, Ph), 7.41–7.36 (m, 3H, Ph), 4.93 (s, br, 2H, NH_2), 4.56 (s, br, 2H, NH_2). $^{13}\text{C}\{^1\text{H}\}$ NMR (δ): 148.87 (NCN), 135.85, 129.23, 123.75, 125.86 (Ph).



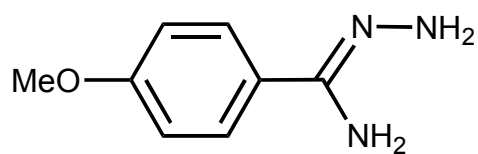
2e. Yield: 50%. Anal. Calcd for $\text{C}_7\text{H}_8\text{BrN}_3$: C, 39.28; H, 3.77; N, 19.63. Found: C, 38.99; H, 3.72; N, 19.85. HRESI⁺-MS: no characteristic peaks were detected. HRESI⁻-MS (m/z): 211.9821 ($[\text{M} - \text{H}]^-$, calcd 211.9823). IR (KBr, selected bonds, cm^{-1}): 3453(m), 3301(m), 3143(m) $\nu(\text{N}-\text{H})$, 1634(s) $\nu(\text{C}=\text{N})$. ^1H NMR (δ): 7.60 (d, $J = 8.6$ Hz, 2H, C_6H_4), 7.54 (d, $J = 8.6$ Hz, 2H, C_6H_4), 4.87 (s, br, 2H, NH_2), 4.58 (s, br, 2H, NH_2). $^{13}\text{C}\{^1\text{H}\}$ NMR (δ): 147.08 (NCN), 134.44, 131.17, 127.17, 122.08 (C_6H_4).



2f. Yield: 46%. HRESI⁺-MS (m/z): 204.0744 ($[\text{M} + \text{H}]^+$, 204.0749), HRESI⁻-MS (m/z): 202.0588 ($[\text{M} - \text{H}]^-$, 202.0592). IR (KBr, selected bonds, cm^{-1}): 3381(m), 3204(m) $\nu(\text{N}-\text{H})$, 1649(m) $\nu(\text{C}=\text{N})$, 1329(m), 1123(s), 1117(s) $\nu(\text{C}-\text{F})$. ^1H NMR (δ): 7.86 (d, $J = 8.2$ Hz, 2H, C_6H_4), 7.69 (d, $J = 8.2$ Hz, 2H, C_6H_4), 4.94 (s, br, 2H, NH_2), 4.69 (s, br, 2H, NH_2). $^{13}\text{C}\{^1\text{H}\}$ NMR (δ): 146.39 (NCN), 138.95 (C_{gem}), 129.55 (q, $J^2_{\text{CF}} = 32.1$ Hz, CCF_3), 125.78 (C_{ortho}), 125.04 (q, $J^3_{\text{CF}} = 3.9$ Hz, C_{meta}), 124.44 (q, $J^1_{\text{CF}} = 271.1$ Hz, CF_3).

Spectra of 2c-f

2c



Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	4500 V	Set Dry Gas	4.0 l/min
Scan End	1200 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Source

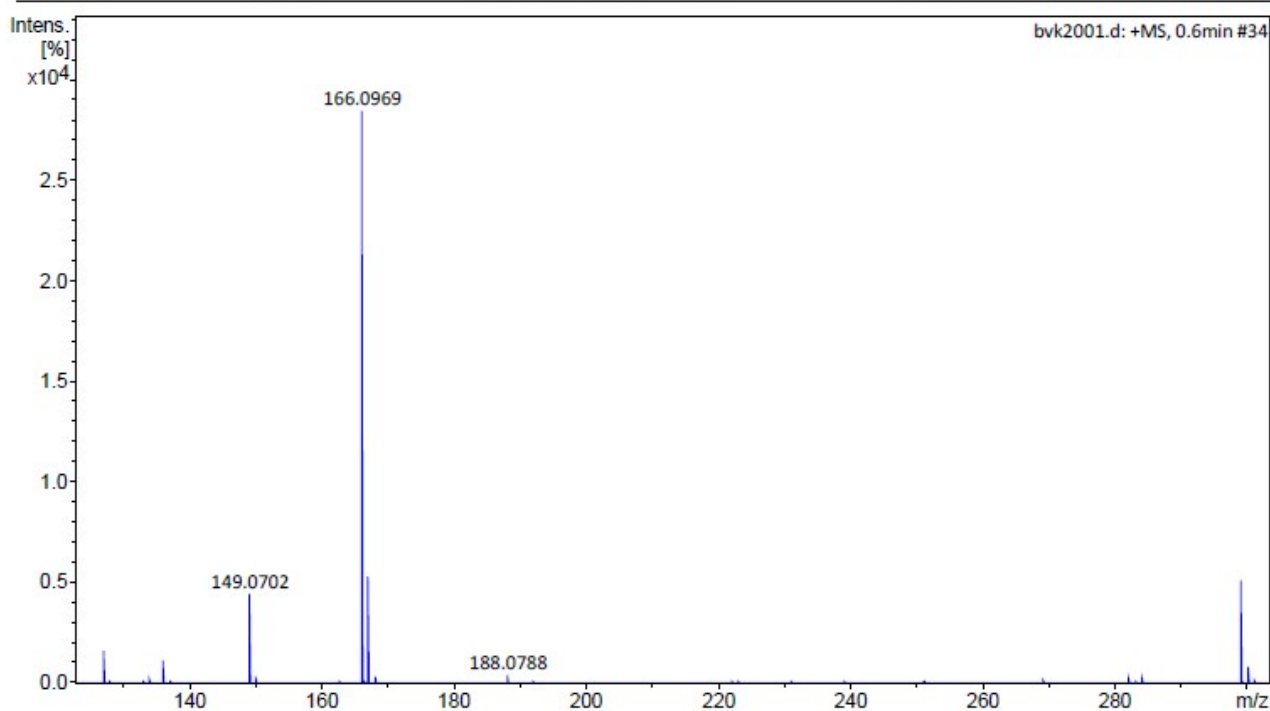


Figure 36S. HRESI⁺-mass spectrum of 2c.

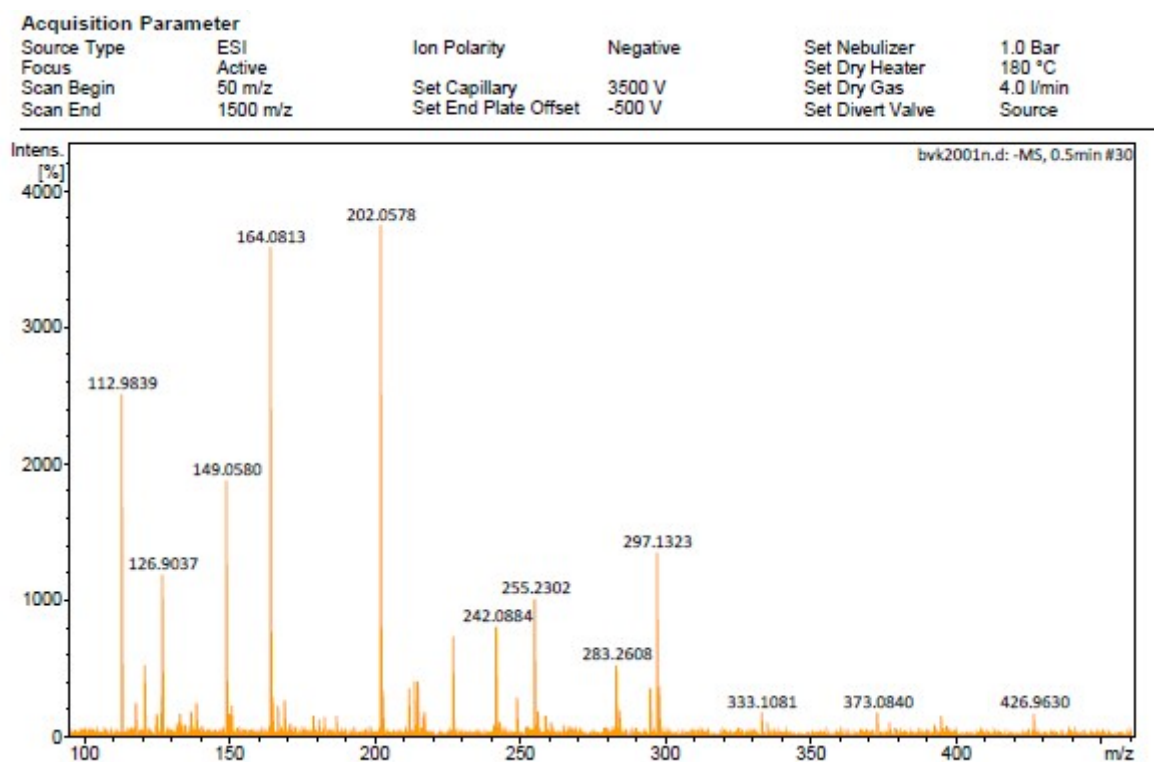


Figure 37S. HRESI⁻-mass spectrum of **2c**.

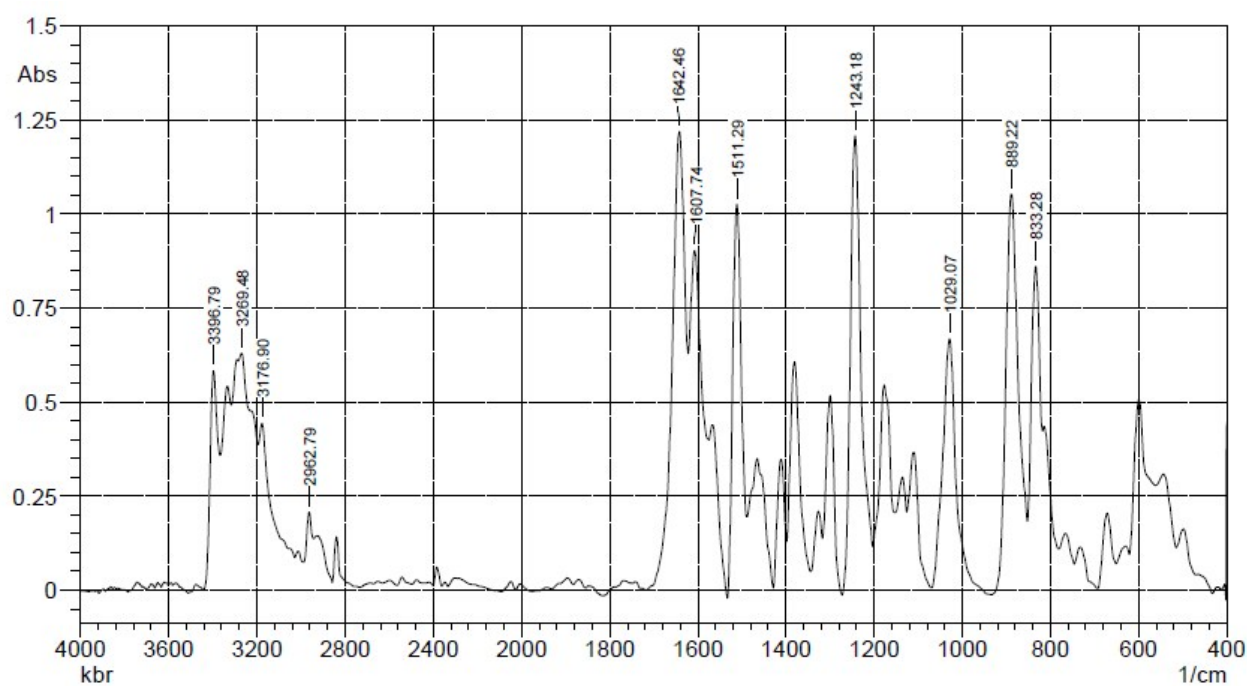


Figure 38S. IR spectrum of **2c**.

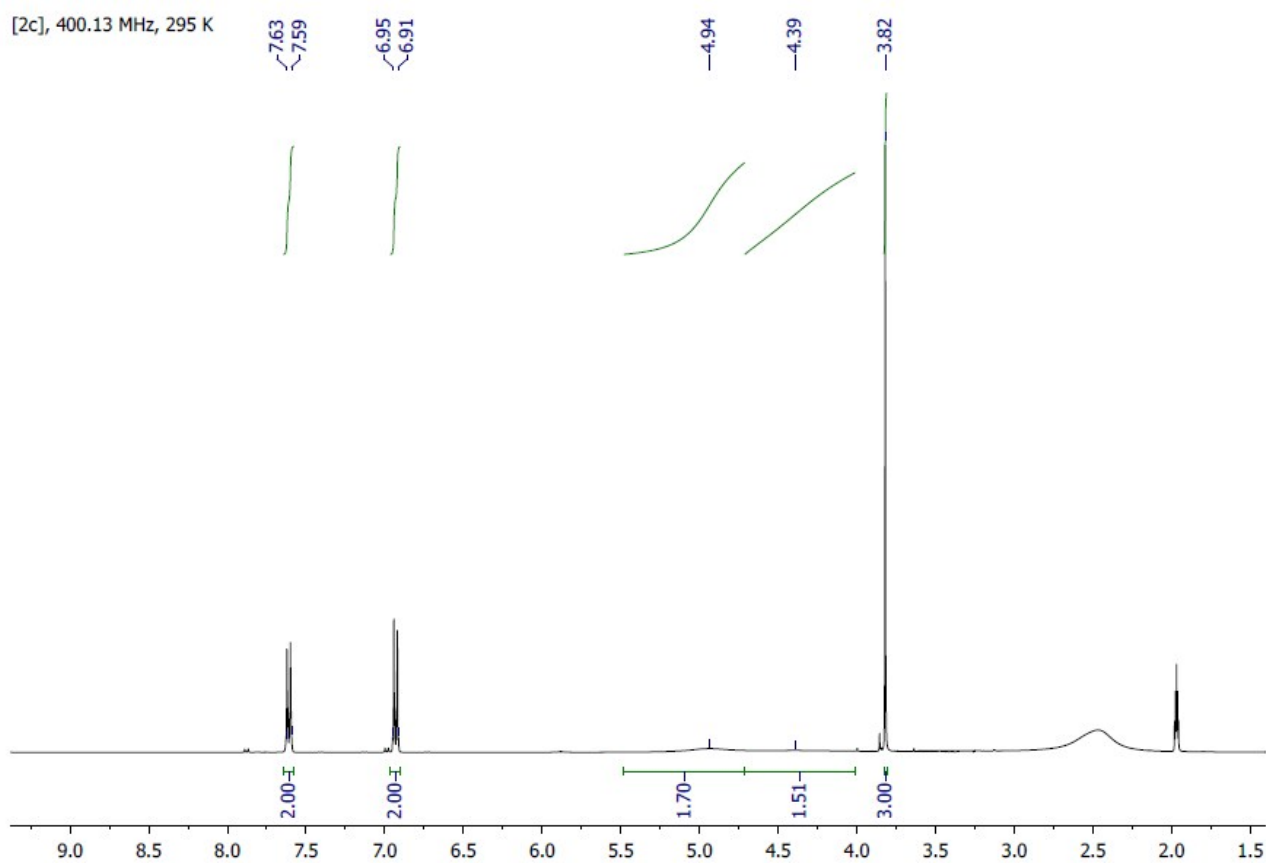


Figure 39S. ^1H NMR spectrum of **2c**.

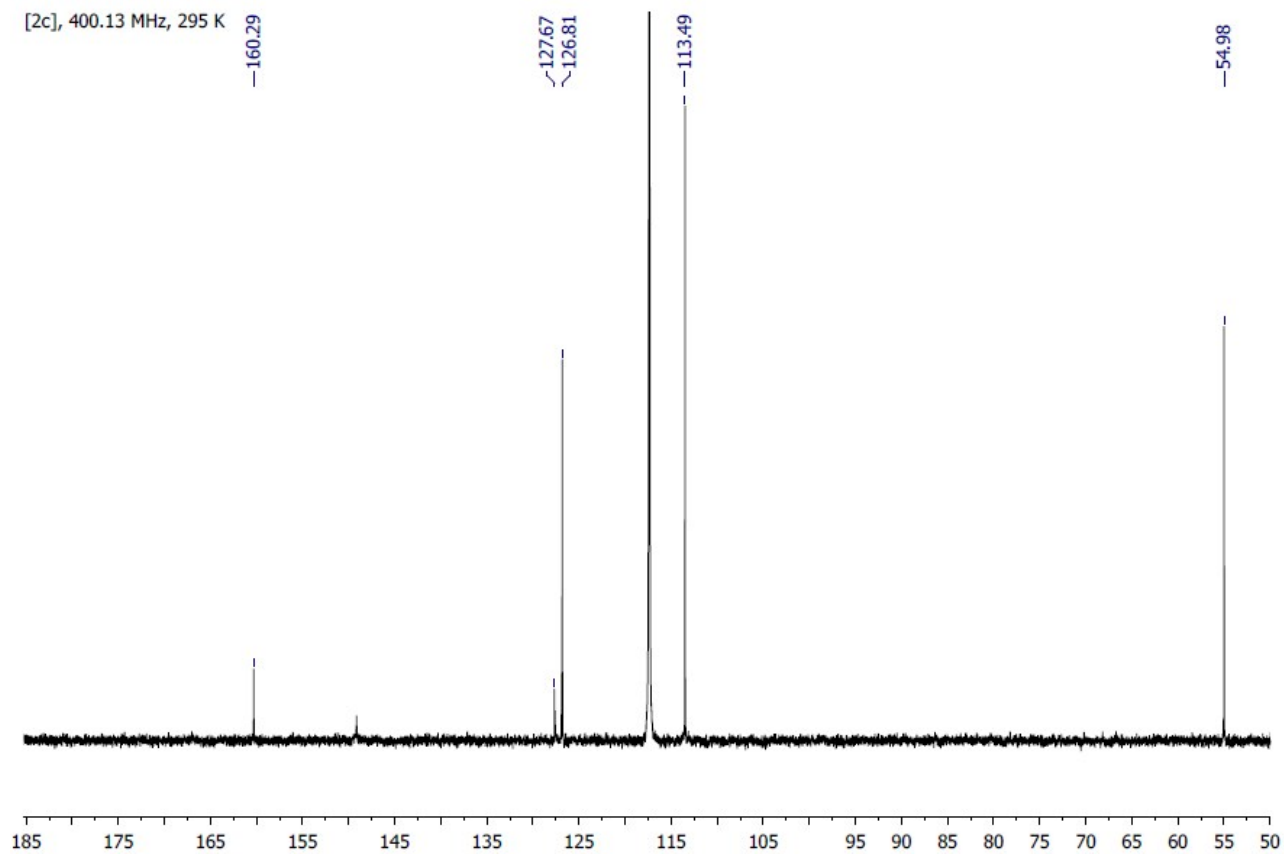


Figure 40S. ^{13}C NMR spectrum of **2c**.

2d

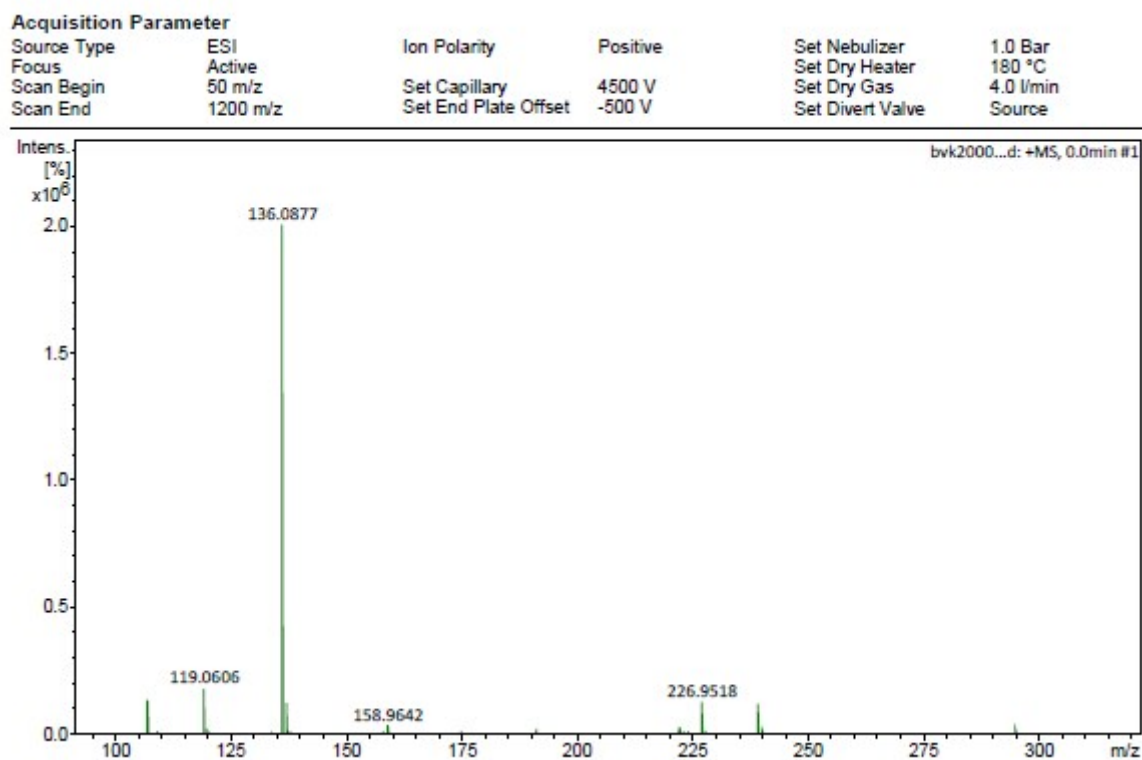
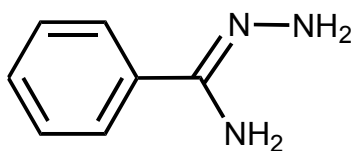


Figure 41S. HRESI⁺-mass spectrum of 2d.

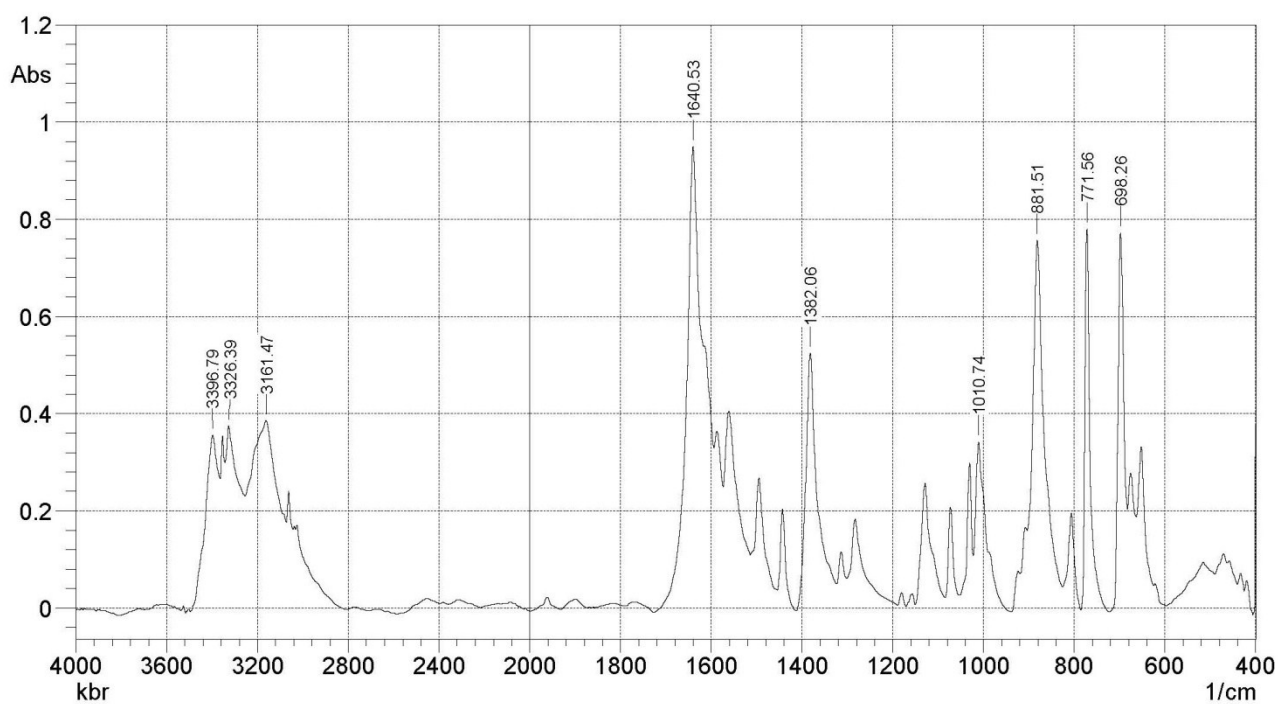


Figure 42S. IR spectrum of 2d .

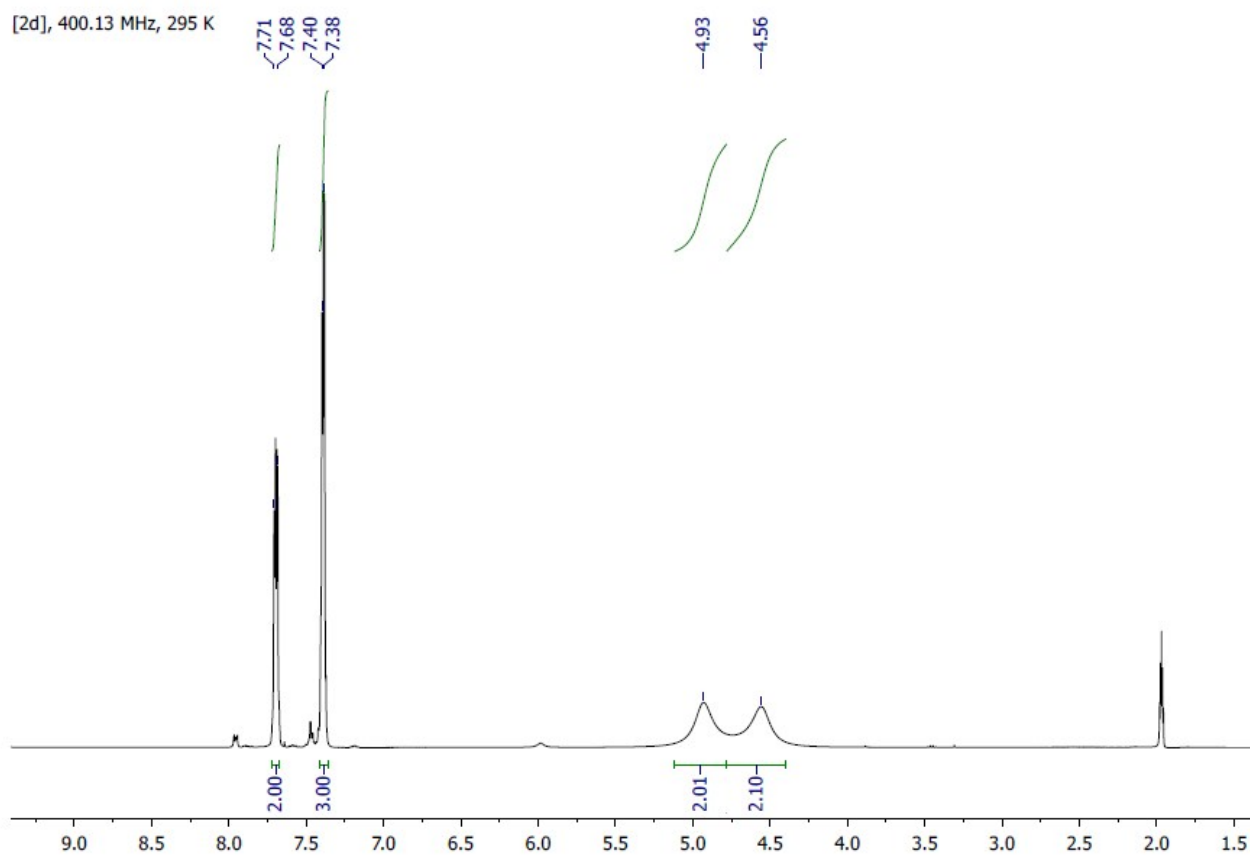


Figure 43S. ^1H NMR spectrum of **2d** .

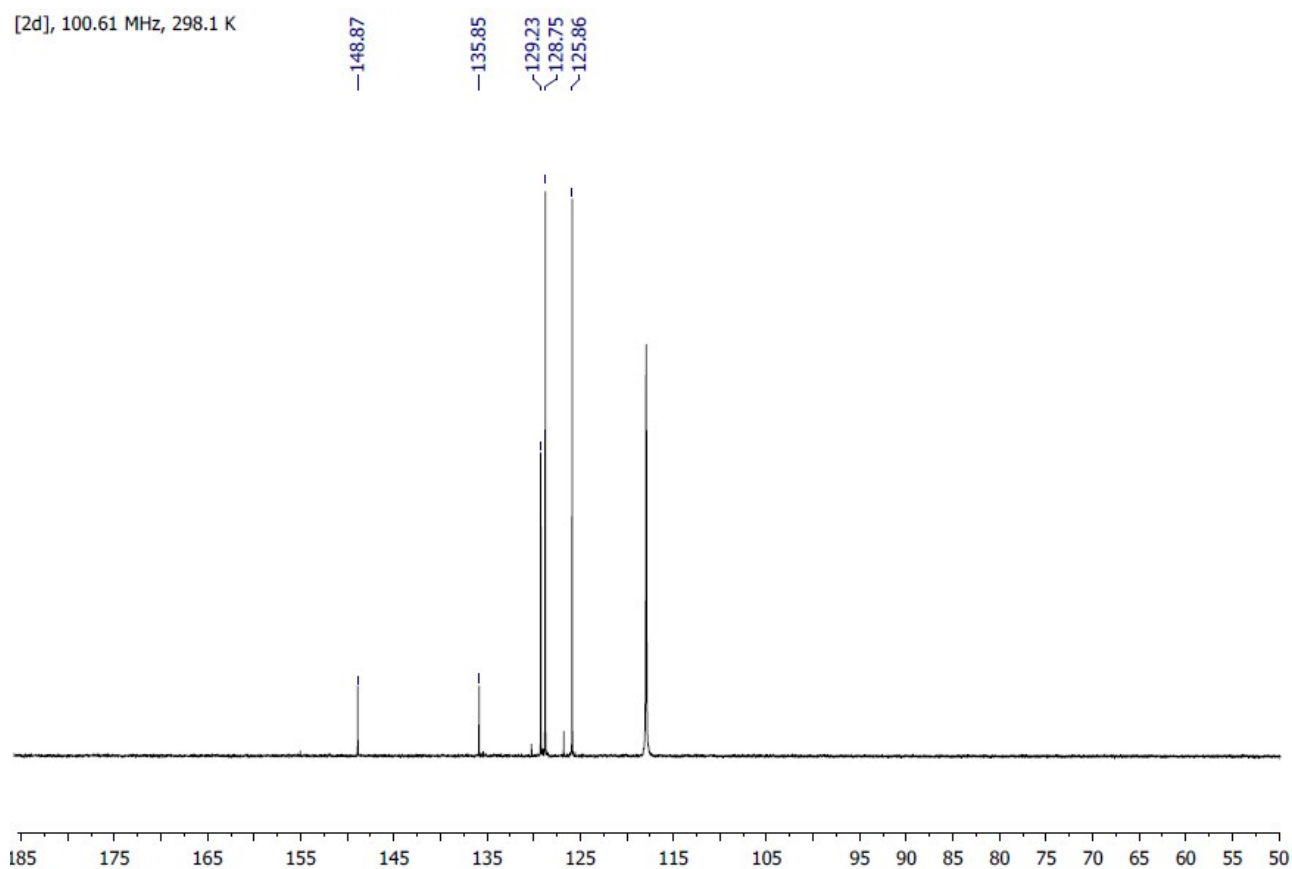
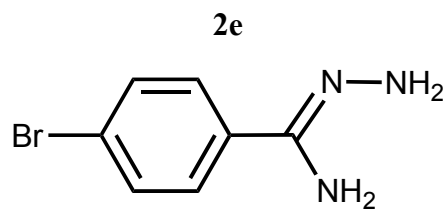


Figure 44S. ^{13}C NMR spectrum of **2d** .



Acquisition Parameter

Source Type	ESI	Ion Polarity	Negative	Set Nebulizer	1.0 Bar
Focus	Active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	3500 V	Set Dry Gas	4.0 l/min
Scan End	1500 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Source

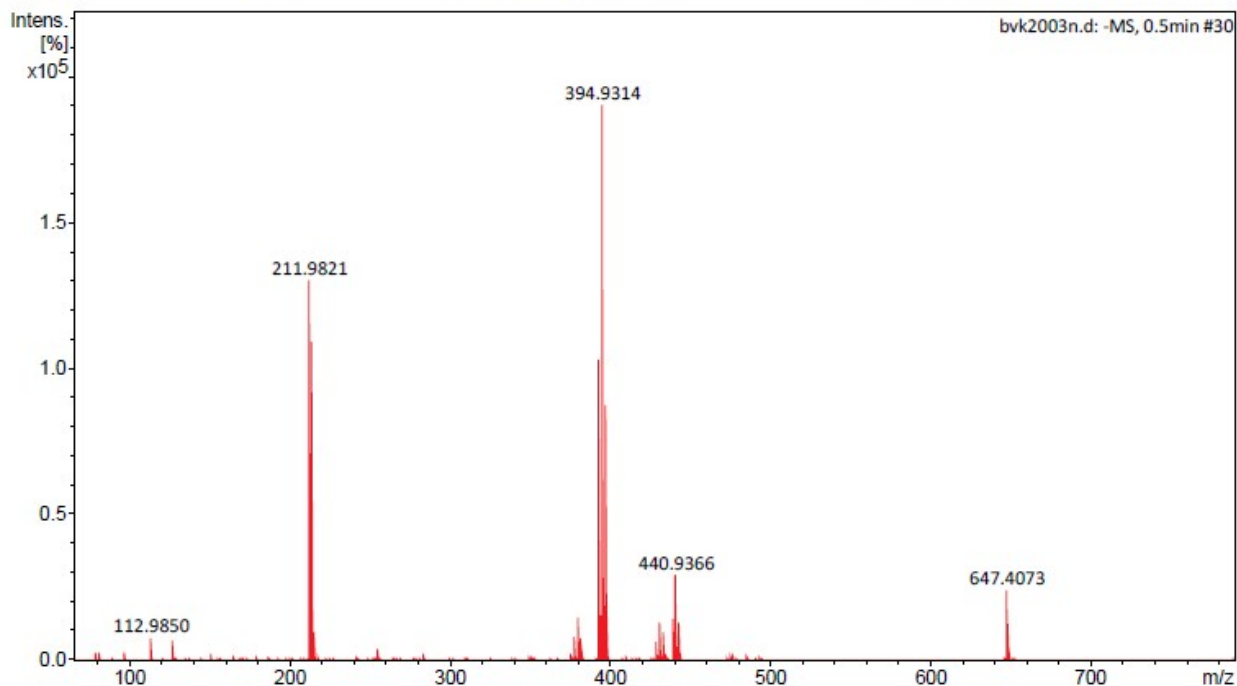


Figure 45S. HRESI⁻-mass spectrum of **2e** .

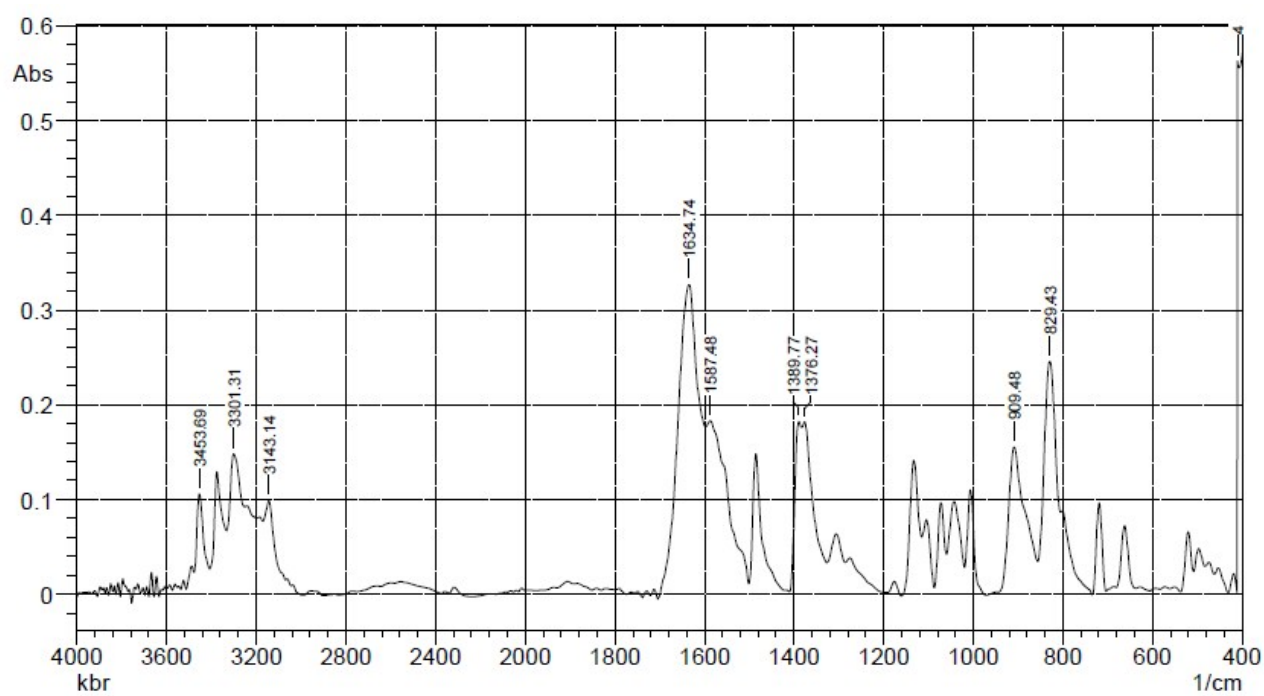


Figure 46S. IR spectrum of **2e** .

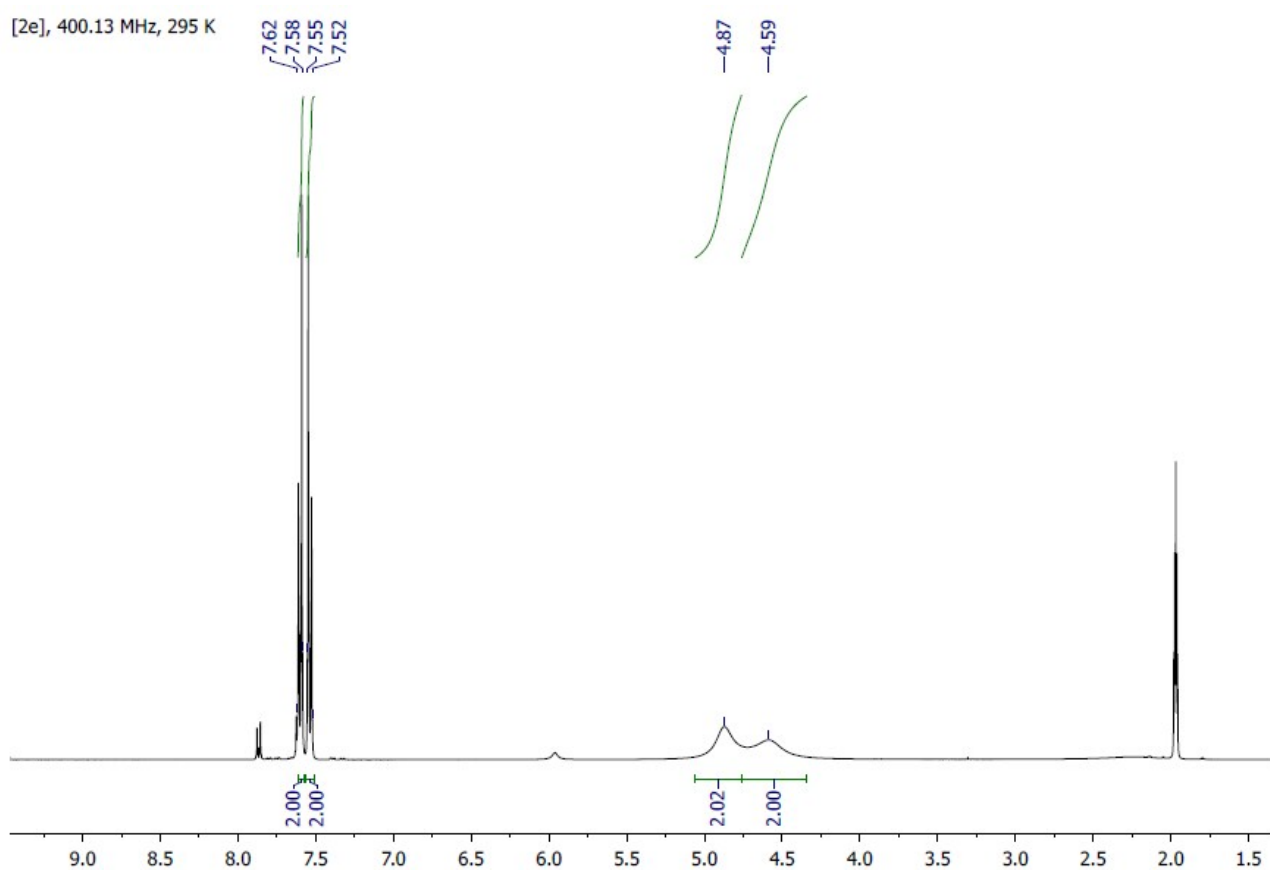


Figure 47S. ¹H NMR spectrum of **2e** .

[2e], 400.13 MHz, 295 K

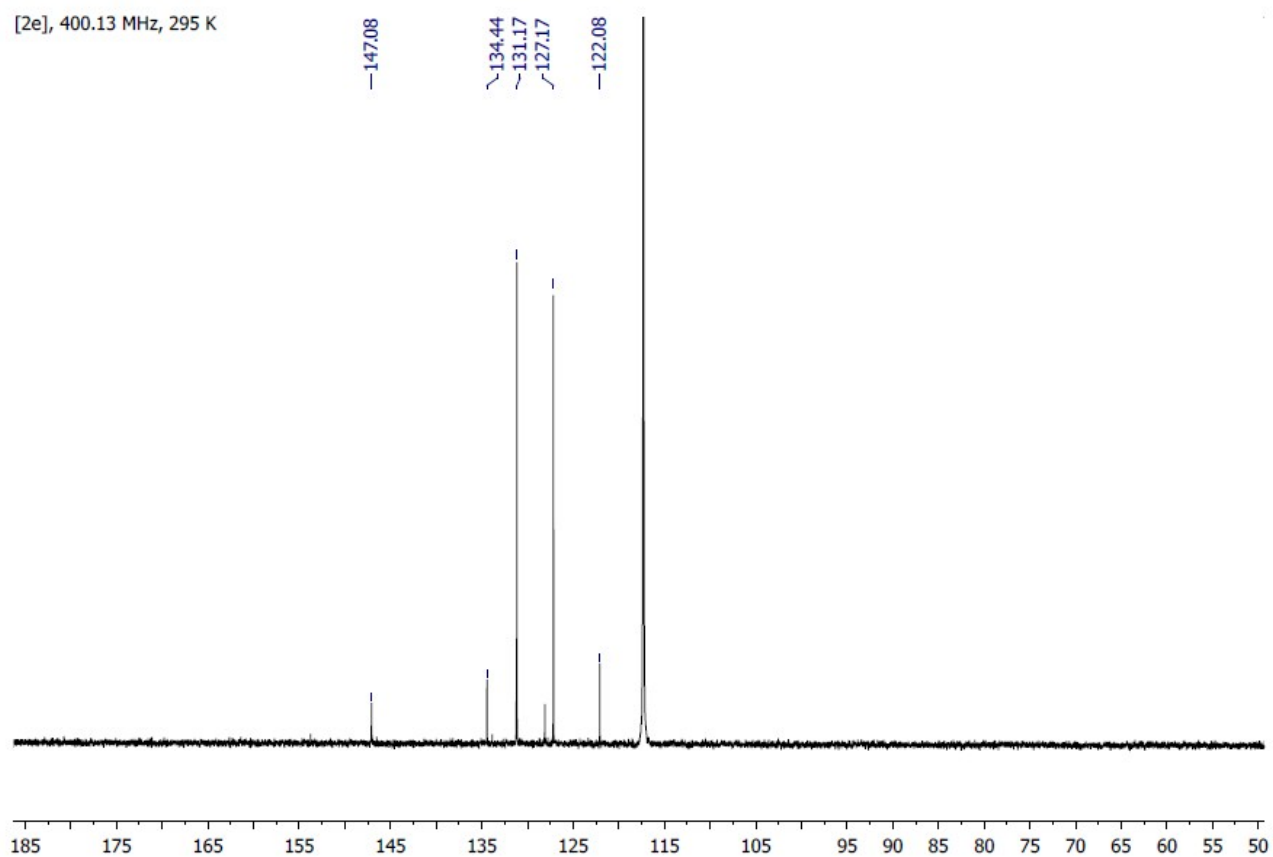
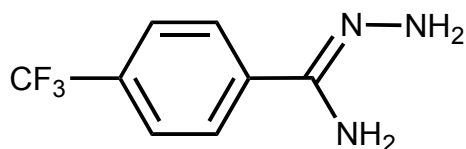


Figure 48S. ^{13}C NMR spectrum of **2e**.

2f



Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	1.0 Bar
Focus	Active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	4500 V	Set Dry Gas	4.0 l/min
Scan End	1500 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Source

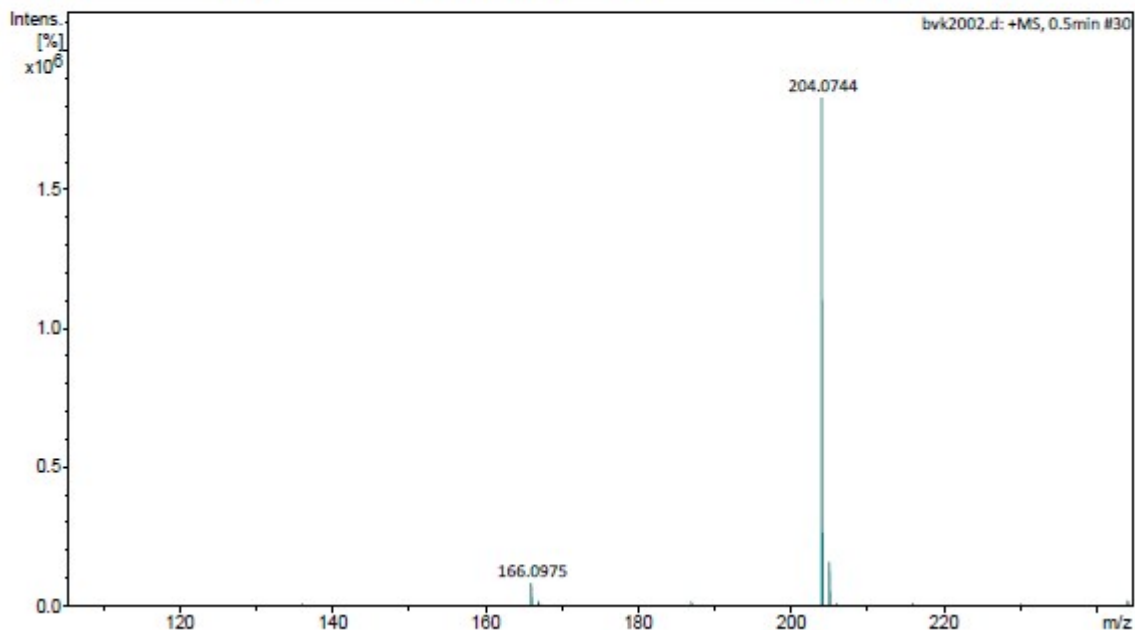


Figure 49S. HRESI⁺-mass spectrum of 2f .

Acquisition Parameter

Source Type	ESI	Ion Polarity	Negative	Set Nebulizer	1.0 Bar
Focus	Active			Set Dry Heater	180 °C
Scan Begin	50 m/z	Set Capillary	3500 V	Set Dry Gas	4.0 l/min
Scan End	1500 m/z	Set End Plate Offset	-500 V	Set Divert Valve	Source

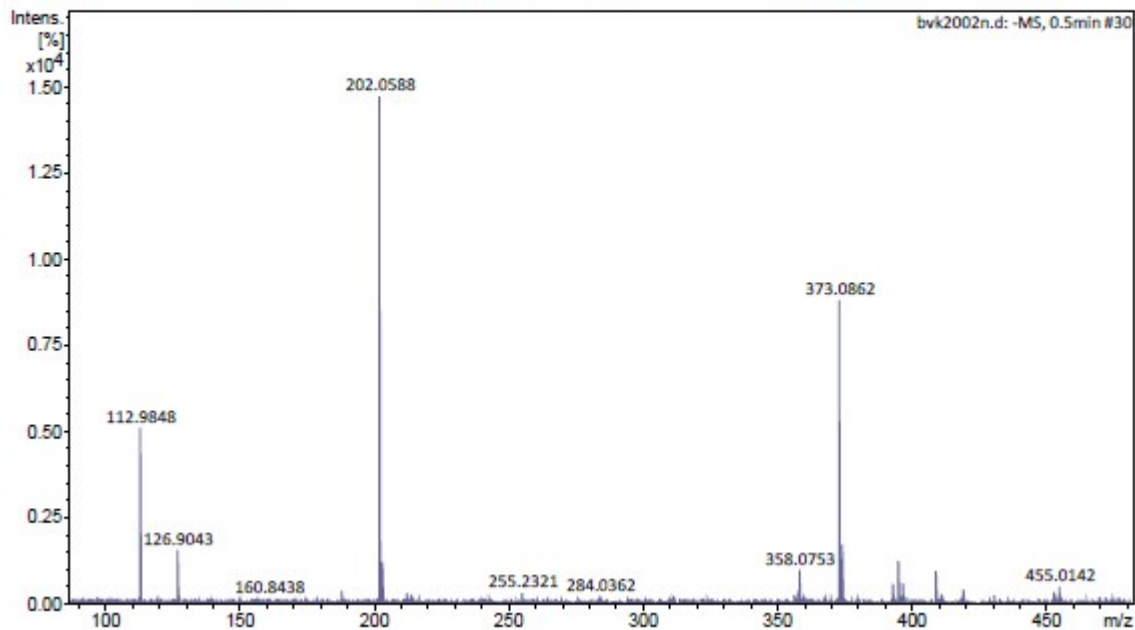


Figure 50S. HRESI⁻-mass spectrum of 2f .

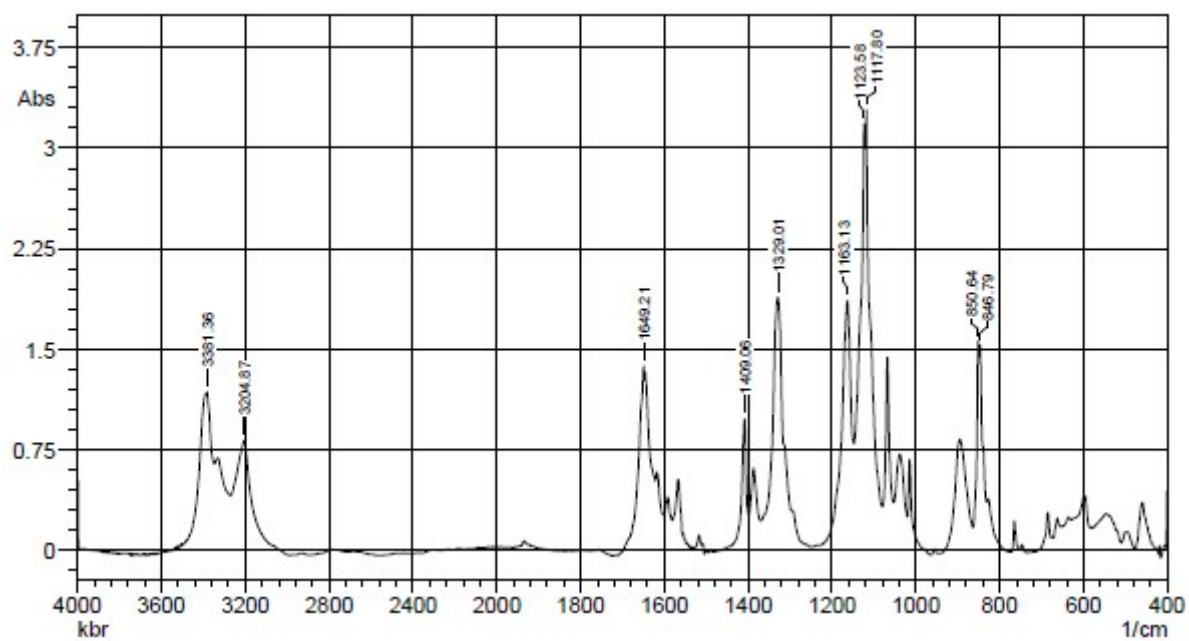


Figure 51S. IR spectrum of **2f**.

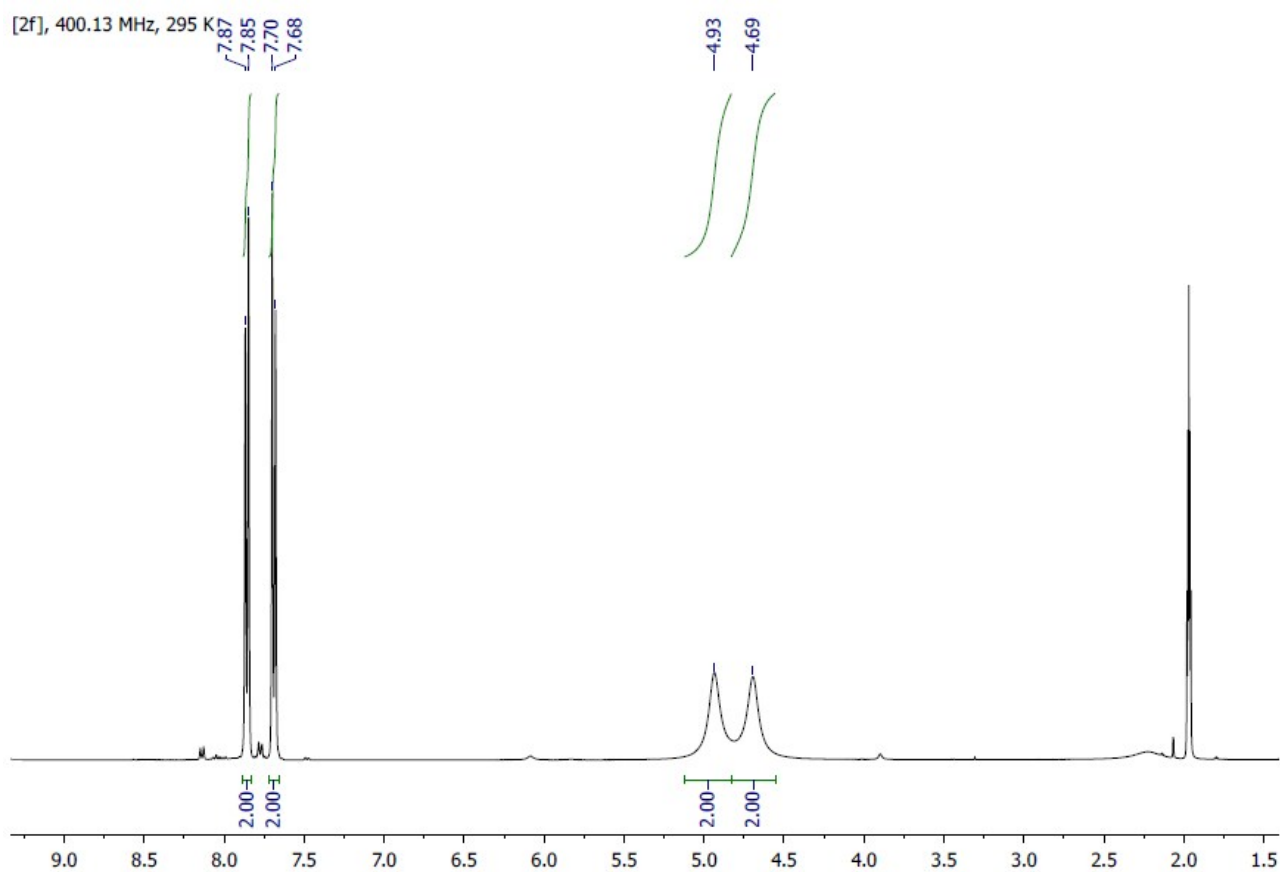


Figure 52S. ^1H NMR spectrum of **2f**.

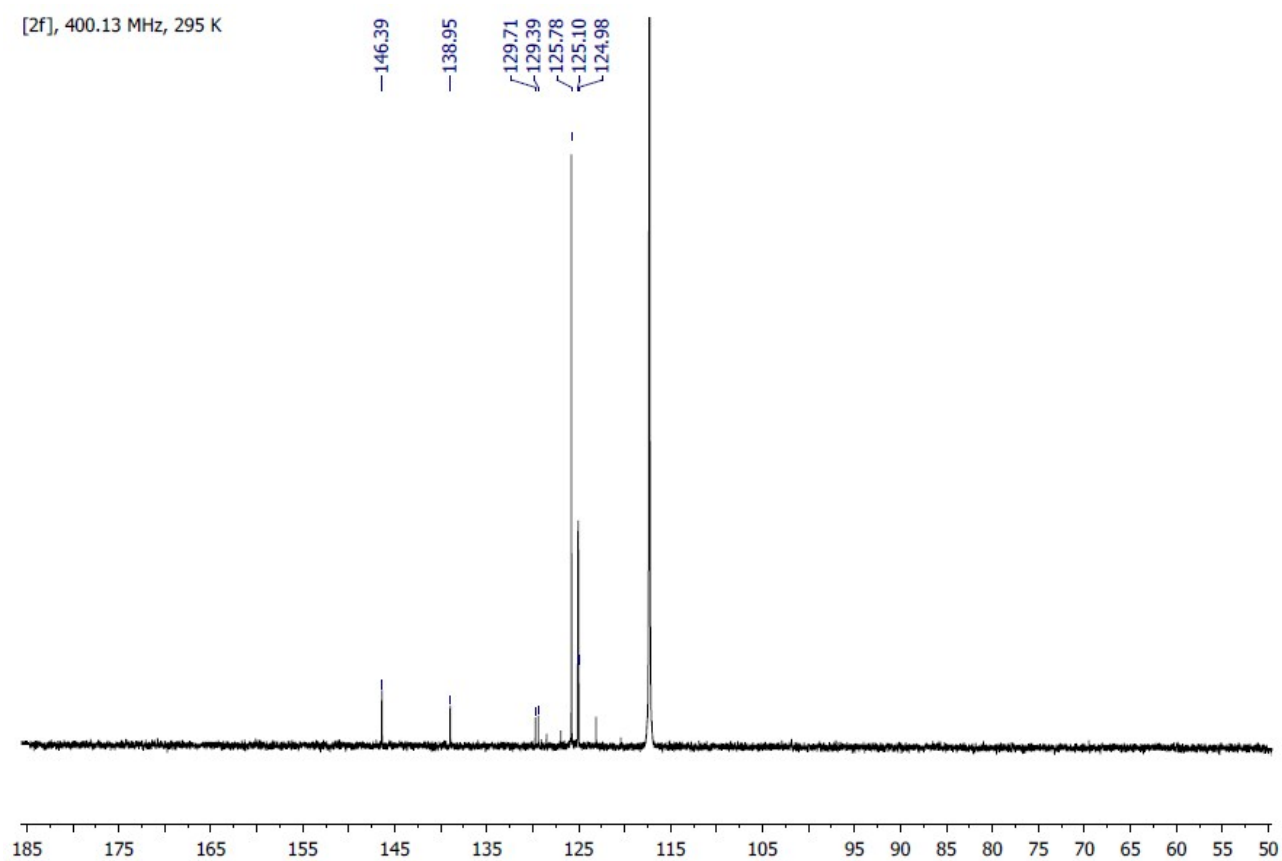


Figure 53S. ^{13}C NMR spectrum of **2f**.

IR identification of dimers of 2e

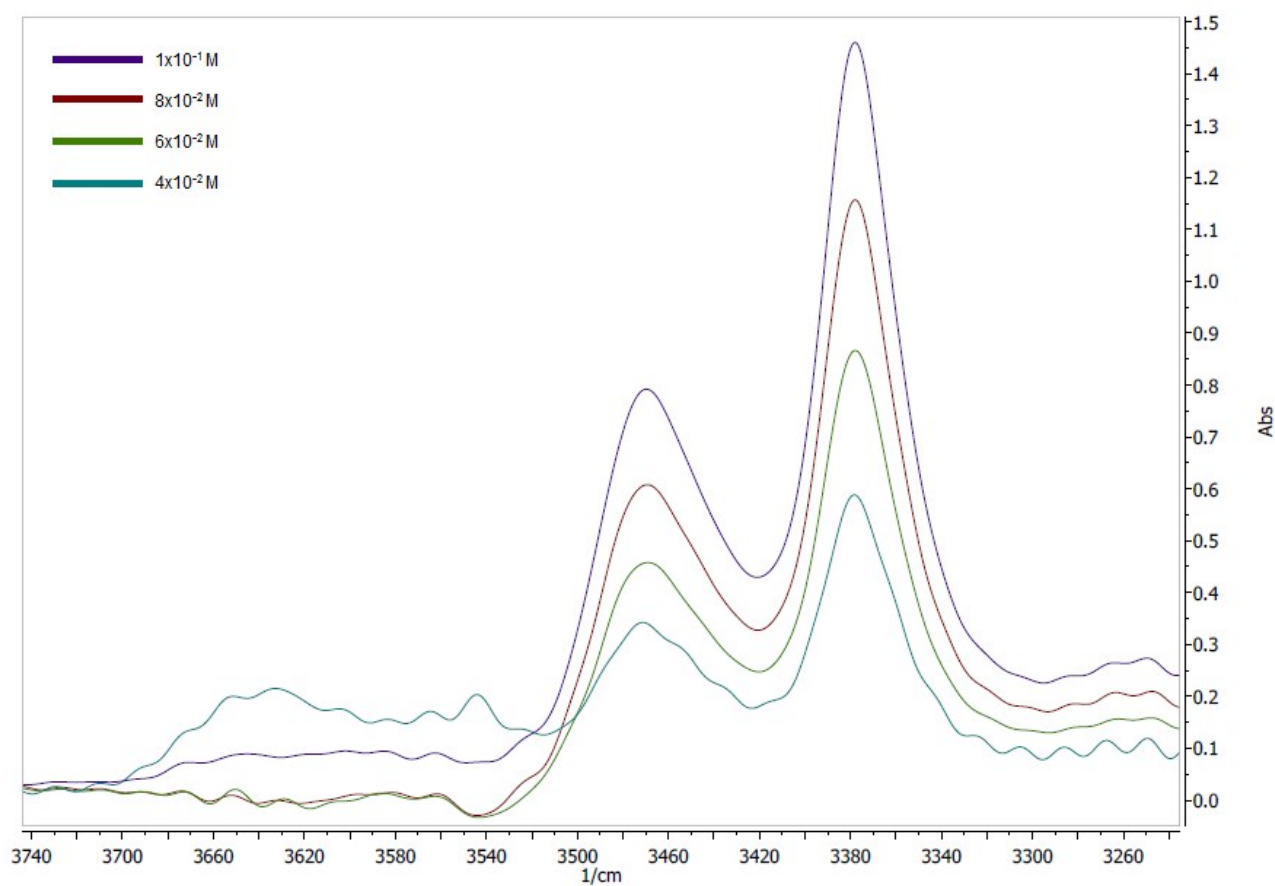


Figure 54S. Combined IR spectra of solution of **2e** in MeCN in the range of concentrations from 4×10^{-2} to 1×10^{-1} M.

Theoretical studies

Theoretical calculations of relative energies for different isomers of (Ph₃PCH₂Ph)[3a–g] in acetonitrile solution. Compounds (Ph₃PCH₂Ph)[3a–g] can exist as two isomers exhibiting different *E/Z* configurations of the oxime C=N bond. In order to determine the relative stability for these isomers, we carried out geometry optimization procedure in acetonitrile solution for model structures based on the experimental X-ray data. The most stable isomers in all cases are those for which crystal structures have been experimentally obtained (**Table 2S**). The calculated total electronic energies, enthalpies, Gibbs free energies and entropies for equilibrium structures of all model species in acetonitrile solution are presented in **Table 3S**.

Table 1S. Relative stability of different isomers of (Ph₃PCH₂Ph)[**3a–g**] in terms of total electronic energies (ΔE_s), enthalpies and Gibbs free energies (ΔH_s and ΔG_s) (in kcal/mol) in acetonitrile solution (M06-2X/6-311++G** level of theory).

Relative stability (<i>E</i>)→(<i>Z</i>)	ΔE_s	ΔH_s	ΔG_s
(Ph ₃ PCH ₂ Ph)[3a]	1.1	1.3	0.7
(Ph ₃ PCH ₂ Ph)[3b]	−3.6	−3.6	−3.6
(Ph ₃ PCH ₂ Ph)[3c]	−4.8	−4.7	−4.9
(Ph ₃ PCH ₂ Ph)[3d]	6.0	6.0	5.9
(Ph ₃ PCH ₂ Ph)[3e]	7.4	7.1	5.4
(Ph ₃ PCH ₂ Ph)[3f]	5.7	5.8	5.6
(Ph ₃ PCH ₂ Ph)[3g]	−4.9	−4.9	−4.0

Table 2S. Calculated total electronic energies, enthalpies, Gibbs free energies (in Hartree) and entropies (in cal/mol•K) in acetonitrile solution (E_s , H_s , G_s , S_s) for different isomers of (Ph₃PCH₂Ph)[**3a–g**] (M06-2X/6-311++G** level of theory).

Compound	E_s	H_s	G_s	S_s
(<i>E</i>)-(Ph ₃ PCH ₂ Ph)[3a]	-538.164079	-537.881633	-537.937543	117.67
(<i>Z</i>)-(Ph ₃ PCH ₂ Ph)[3a]	-538.162312	-537.879541	-537.936376	119.62
(<i>E</i>)-(Ph ₃ PCH ₂ Ph)[3b]	-690.809101	-690.485867	-690.550606	136.26
(<i>Z</i>)-(Ph ₃ PCH ₂ Ph)[3b]	-690.814806	-690.491620	-690.556328	136.19
(<i>E</i>)-(Ph ₃ PCH ₂ Ph)[3c]	-977.158046	-976.730726	-976.808215	163.09
(<i>Z</i>)-(Ph ₃ PCH ₂ Ph)[3c]	-977.165741	-976.738171	-976.816095	164.00
(<i>E</i>)-(Ph ₃ PCH ₂ Ph)[3d]	-862.647585	-862.255796	-862.328569	153.16
(<i>Z</i>)-(Ph ₃ PCH ₂ Ph)[3d]	-862.637993	-862.246272	-862.319136	153.35
(<i>E</i>)-(Ph ₃ PCH ₂ Ph)[3e]	-3436.225092	-3435.841783	-3435.917974	160.36
(<i>Z</i>)-(Ph ₃ PCH ₂ Ph)[3e]	-3436.213333	-3435.830542	-3435.909420	166.01
(<i>E</i>)-(Ph ₃ PCH ₂ Ph)[3f]	-1199.703100	-1199.302922	-1199.383826	170.28
(<i>Z</i>)-(Ph ₃ PCH ₂ Ph)[3f]	-1199.694083	-1199.293632	-1199.374829	170.89
(<i>E</i>)-(Ph ₃ PCH ₂ Ph)[3g]	-1038.294371	-1037.835007	-1037.915564	169.55
(<i>Z</i>)-(Ph ₃ PCH ₂ Ph)[3g]	-1038.302251	-1037.842771	-1037.921943	166.63

Table 3S. Cartesian atomic coordinates for equilibrium structures of all model species in acetonitrile solution (unless otherwise specified). Nuclear charges of elements are shown in the second column.

Model structure	Charge	X	Y	Z
<i>(E)</i> -(Ph ₃ PCH ₂ Ph)[3f]				
	7	2.316320	0.416696	-0.146469
	1	1.423140	-0.072604	-0.138200
	7	-0.138489	1.461809	0.097787
	7	0.978030	2.275253	-0.020446
	1	0.920173	3.256040	0.244277
	7	-1.387857	3.225377	-0.844374
	1	-0.570743	3.725307	-1.169731
	1	-2.237308	3.436111	-1.349922
	6	-3.484455	-1.045488	-0.149952
	1	-3.401642	-2.124430	-0.219318
	6	-2.466096	1.139983	-0.195718
	6	2.194608	1.712346	-0.047159
	6	3.775967	2.745665	1.573137
	1	4.668163	3.368394	1.656341
	1	4.001356	1.763384	1.991524
	1	2.979785	3.202797	2.164983
	6	3.364795	2.633555	0.099003
	1	3.086631	3.614567	-0.291807
	1	4.189960	2.248352	-0.497830
	6	-1.242143	1.979291	-0.332685
	6	-4.724870	-0.445860	0.062834
	6	-2.356180	-0.251435	-0.274062

	1	-1.387205	-0.706199	-0.439263
	6	-3.712788	1.727965	0.013794
	1	-3.810050	2.804343	0.096764
	6	-4.847240	0.934314	0.145540
	1	-5.813050	1.393671	0.316302
	6	-5.937110	-1.321759	0.156743
	5	4.374954	-2.878647	1.158990
	1	4.246887	-3.733288	1.975192
	5	4.794968	-2.966382	-0.481967
	1	5.020720	-4.025042	-0.986202
	5	3.563732	-0.460165	-0.245914
	5	3.253359	-2.201569	0.083344
	1	2.110611	-2.551175	0.068526
	5	3.989756	-1.706543	-1.492225
	1	3.445013	-2.044872	-2.498988
	5	4.765925	-0.201869	-1.405123
	1	4.867137	0.661164	-2.216011
	5	4.184887	-1.200452	1.281049
	1	3.851978	-0.684708	2.305353
	5	5.729438	-1.957803	0.716636
	1	6.777198	-2.129141	1.263209
	5	5.308127	-0.276368	0.206137
	1	5.921904	0.634787	0.673402
	5	5.742289	-1.529813	-1.038749
	1	6.746222	-1.715119	-1.657552
	9	-6.283007	-1.830103	-1.044232
	9	-5.740319	-2.378696	0.965571

	9	-7.014119	-0.672276	0.621297
(Z)-(Ph ₃ PCH ₂ Ph)[3f]				
	7	3.490801	-0.345440	-0.519040
	7	0.948695	-1.524121	-0.033987
	7	2.174514	-2.096398	0.281872
	7	0.223570	-2.428300	2.010349
	6	-3.021638	-0.713858	-1.098361
	6	-1.336603	-1.264065	0.537386
	6	3.332117	-1.584009	-0.135233
	6	4.576700	-3.206257	-1.575521
	6	4.490912	-2.542336	-0.196968
	6	0.031915	-1.759106	0.848197
	6	-3.848424	-0.259994	-0.071248
	6	-1.766692	-1.213051	-0.791983
	6	-2.173162	-0.808410	1.555525
	6	-3.432955	-0.302516	1.252568
	6	-5.210817	0.251903	-0.430190
	5	2.532233	2.917094	1.803573
	5	2.846275	3.703608	0.332112
	5	2.730708	0.984803	-0.366484
	5	3.592160	2.112522	0.760453
	5	3.373270	2.570490	-0.970208
	5	2.074470	1.766781	-1.713278
	5	1.954671	1.443262	1.187668
	5	1.202004	3.038516	0.758857
	5	1.042593	1.642661	-0.372217
	5	1.686907	3.235415	-0.971950

	9	-5.161208	1.219471	-1.364626
	9	-5.990773	-0.721810	-0.944091
	9	-5.872606	0.759354	0.619009
	1	4.440718	-0.180300	-0.840303
	1	2.180586	-3.081474	0.541177
	1	1.153228	-2.483408	2.405691
	1	-0.541295	-2.519745	2.661764
	1	-3.357621	-0.679117	-2.128868
	1	5.404469	-3.916798	-1.590690
	1	4.747132	-2.462545	-2.356500
	1	3.655872	-3.747033	-1.804883
	1	4.360430	-3.301621	0.577287
	1	5.408052	-1.993871	0.023312
	1	-1.112528	-1.566119	-1.579151
	1	-1.844857	-0.815827	2.588680
	1	-4.076183	0.058747	2.045123
	1	2.673496	3.282607	2.926122
	1	3.283846	4.815062	0.327367
	1	4.690730	1.790324	1.102646
	1	4.347781	2.837882	-1.606214
	1	1.951843	1.374276	-2.828616
	1	1.631677	0.555479	1.921606
	1	0.195623	3.565877	1.127832
	1	-0.032458	1.142385	-0.480849
	1	1.173105	4.106341	-1.607171
<i>(E)</i> -(Ph ₃ PCH ₂ Ph)[3d]				
	7	0.735509	1.824747	-0.001646

	1	1.023407	2.772376	-0.236230
	7	-1.010270	0.341578	0.108770
	1	-0.265842	-0.349559	0.033502
	7	1.622037	0.774730	-0.211430
	6	3.812632	-0.107563	0.070411
	6	-0.577989	1.573169	0.055936
	6	5.173680	0.198047	0.000149
	1	5.513279	1.225305	0.073475
	7	3.209649	2.087088	0.927067
	1	2.515004	2.727965	1.286944
	1	4.067921	2.046776	1.457943
	6	-1.492562	2.757738	0.013667
	1	-0.967894	3.615328	0.439609
	1	-2.361842	2.545609	0.633636
	5	-2.572087	-1.958871	-0.147221
	1	-1.551617	-2.579871	-0.190503
	6	5.688839	-2.137187	-0.313267
	1	6.417873	-2.926008	-0.460033
	5	-3.863117	-2.299980	-1.191798
	1	-3.979288	-3.129162	-2.035549
	6	3.394992	-1.434964	-0.055566
	1	2.339016	-1.670656	0.004074
	6	2.804284	0.967429	0.281656
	6	4.331193	-2.444476	-0.242890
	1	4.000664	-3.473025	-0.330206
	6	6.106987	-0.815696	-0.194074
	1	7.160767	-0.569566	-0.255412

	5	-3.490684	0.293433	1.458881
	1	-3.349969	1.124091	2.297286
	5	-3.106612	-1.357203	1.471140
	1	-2.626631	-1.856911	2.443349
	5	-2.430677	-0.206679	0.240201
	5	-4.091372	0.414134	-0.128831
	1	-4.480159	1.463860	-0.543448
	5	-3.269329	-0.715915	-1.276986
	1	-2.857228	-0.259897	-2.301023
	6	-1.923852	3.061976	-1.426830
	1	-1.065170	3.349398	-2.037724
	1	-2.637440	3.887443	-1.426512
	1	-2.401995	2.194377	-1.884656
	5	-4.233268	-2.343530	0.462486
	1	-4.695032	-3.332395	0.947774
	5	-4.775956	-0.740754	1.099459
	1	-5.772610	-0.697874	1.755665
	5	-4.931839	-1.091822	-0.666373
	1	-6.008221	-0.979512	-1.171228
(Z)-(Ph ₃ PCH ₂ Ph)[3d]				
	7	0.351075	2.174555	-0.396052
	7	2.228684	0.973428	0.301649
	7	-0.581285	1.241194	0.028808
	6	-2.794856	0.407402	-0.134799
	6	1.643128	2.077898	-0.074827
	6	-3.714047	-0.155017	-1.023711
	7	-1.940215	2.095705	-1.687663

	6	2.422063	3.364836	-0.123171
	6	-4.806638	-1.282522	0.807501
	6	-2.889444	0.118538	1.229412
	6	-1.696224	1.281425	-0.625147
	6	-3.891859	-0.721830	1.697001
	6	-4.714559	-0.999202	-0.551885
	5	1.523052	-1.404480	1.655267
	5	3.042325	-1.781861	0.991109
	5	1.938321	-0.539932	0.264173
	5	0.560448	-1.718647	0.294284
	6	2.291745	4.134196	1.195736
	5	1.670723	-2.965328	1.020496
	1	0.014982	3.109123	-0.623227
	1	3.206169	1.138197	0.526203
	1	-3.638878	0.038393	-2.088108
	1	-1.165768	2.396259	-2.265432
	1	-2.828031	2.021939	-2.161908
	1	2.043301	3.970086	-0.950315
	1	3.467209	3.133048	-0.331722
	1	-5.587860	-1.939073	1.173448
	1	-2.175925	0.557865	1.916403
	1	-3.961211	-0.938063	2.756860
	1	-5.417734	-1.438977	-1.249819
	1	1.236050	-1.004131	2.737425
	1	4.027094	-1.669111	1.657487
	1	-0.622801	-1.596212	0.341599
	1	1.246407	4.366141	1.411144

	1	2.844003	5.072887	1.130934
	1	2.697530	3.553653	2.026592
	1	1.448159	-3.911577	1.714057
	5	3.160909	-1.391023	-0.765501
	5	2.465215	-2.568894	-1.760016
	5	1.410609	-1.335126	-1.261232
	5	2.967721	-3.107790	-0.230244
	5	1.213755	-3.056560	-0.725708
	1	4.102973	-0.746497	-1.118072
	1	2.760174	-2.938187	-2.850980
	1	0.835873	-0.659091	-2.063028
	1	3.748343	-4.007198	-0.139175
	1	0.450881	-3.909790	-1.067368
(Z)-(Ph ₃ PCH ₂ Ph)[3c]				
	7	2.736943	1.477472	-0.409373
	1	3.237327	2.350987	-0.530917
	8	-6.973009	0.019621	0.230690
	7	-0.718033	0.680304	-0.051067
	7	0.657679	0.548099	-0.187400
	1	1.110200	-0.327662	0.074858
	6	-1.384778	-0.385341	-0.357908
	6	-5.639095	-0.157906	0.103831
	6	-2.860200	-0.320441	-0.202636
	6	-4.892105	0.989723	-0.194980
	1	-5.409526	1.934823	-0.312103
	7	-0.839643	-1.567158	-0.758895
	1	0.101923	-1.574125	-1.131864

	1	-1.455481	-2.237366	-1.197965
	6	-3.521402	0.907573	-0.341193
	1	-2.949691	1.798128	-0.574359
	6	-3.610419	-1.456111	0.092154
	1	-3.124611	-2.416240	0.228630
	6	1.430227	1.613096	-0.383643
	6	-4.993090	-1.385724	0.250311
	1	-5.544228	-2.285116	0.490490
	6	0.800036	2.964866	-0.533055
	1	-0.091586	2.872332	-1.153457
	1	1.512440	3.612180	-1.046258
	5	4.720175	-0.628479	-1.336920
	1	5.061394	-0.394286	-2.455742
	5	4.920145	-1.744141	1.252501
	1	5.034607	-2.623189	2.051883
	5	5.321315	0.454371	-0.015350
	1	5.755365	1.520575	-0.335102
	5	3.558926	0.207869	-0.228988
	5	4.294649	-0.049739	1.400881
	1	3.832423	0.577266	2.305837
	5	3.260794	-1.340451	0.671091
	1	2.319114	-1.729387	1.295809
	5	3.167207	-1.234243	-1.026286
	1	2.269123	-1.483941	-1.765706
	5	5.949350	-0.396905	1.309593
	1	6.840992	-0.128757	2.048108
	5	5.949963	-1.237809	-0.164966

	1	6.972360	-1.670208	-0.603967
	6	0.435080	3.554087	0.834894
	1	-0.022617	4.535253	0.698751
	1	1.326086	3.674961	1.454388
	1	-0.272894	2.911312	1.359861
	5	4.432046	-2.182484	-0.433647
	1	4.521836	-3.324645	-0.767565
	6	-7.766998	-1.119318	0.542698
	1	-8.792638	-0.761113	0.601655
	1	-7.473512	-1.548291	1.504544
	1	-7.689350	-1.878458	-0.240565
<i>(E)</i> -(Ph ₃ PCH ₂ Ph)[3c]				
	7	2.437138	0.733674	-0.446004
	8	-7.179215	-1.204952	0.209333
	7	-1.090957	0.337720	-0.248713
	7	0.158639	0.948252	-0.328197
	6	-2.065215	1.151256	0.002641
	6	-5.988965	-0.569450	0.135620
	6	-3.430378	0.567719	0.050307
	6	-4.860818	-1.341451	0.445167
	7	-1.943597	2.497595	0.163074
	6	-3.600658	-0.779538	0.398910
	6	-4.556904	1.327342	-0.255469
	6	1.275060	0.236345	-0.111194
	6	-5.833978	0.770489	-0.219314
	6	1.152584	-1.119115	0.509660
	5	5.300981	0.533121	-1.299399

	5	3.852564	0.168566	-0.281547
	5	4.571653	-1.411533	0.228708
	5	4.470550	-0.943012	-1.400951
	6	0.965316	-2.193318	-0.569057
	5	6.038795	-1.048168	-0.786603
	6	-8.351754	-0.457616	-0.094257
	1	2.404463	1.650694	-0.887092
	1	0.240935	1.801376	-0.880941
	1	-4.999053	-2.379435	0.724584
	1	-1.034585	2.877504	0.393294
	1	-2.715208	2.985487	0.595707
	1	-2.733313	-1.381683	0.642888
	1	-4.459043	2.365866	-0.552967
	1	-6.686399	1.386193	-0.473687
	1	0.299820	-1.113902	1.187865
	1	2.055340	-1.315120	1.085637
	1	5.436959	1.209663	-2.273414
	1	4.078006	-2.436084	0.591520
	1	3.933555	-1.459782	-2.326426
	1	0.874687	-3.172095	-0.095089
	1	1.821791	-2.215592	-1.245236
	1	0.060876	-2.004804	-1.150546
	1	6.838904	-1.763209	-1.309767
	1	-9.184736	-1.145689	0.033325
	1	-8.329634	-0.097378	-1.126223
	1	-8.466736	0.386535	0.590917
	5	6.106480	-0.867505	1.010767

	5	5.070876	1.366276	0.287167
	5	4.556489	-0.000483	1.368673
	5	6.067572	0.752661	1.512457
	5	6.623723	0.506476	-0.071003
	1	6.726982	-1.674387	1.635356
	1	4.749253	2.516560	0.259403
	1	3.789645	-0.040708	2.283539
	1	6.578542	1.293182	2.439794
	1	7.699960	0.907762	-0.397651
<i>(E)</i> -(Ph ₃ PCH ₂ Ph)[3a]				
	7	0.951316	-0.620600	0.074446
	1	1.047501	-1.594283	0.357862
	7	3.223485	-0.805078	-0.173940
	1	4.101879	-0.375769	-0.442043
	6	2.099633	-0.090851	-0.256886
	7	3.197153	-2.125936	0.281647
	1	3.597768	-2.728677	-0.431976
	1	3.765180	-2.199172	1.121874
	6	2.259815	1.334530	-0.684966
	1	1.343511	1.660358	-1.172553
	1	3.073627	1.386366	-1.411339
	5	-0.473533	-0.073418	0.043570
	5	-1.571651	0.525864	1.363553
	1	-1.379378	0.754395	2.519517
	5	-1.645012	-1.196772	0.831601
	1	-1.178417	-2.018381	1.563200
	5	-2.605396	1.334823	0.108460

	1	-3.315333	2.293134	0.167551
	5	-2.989103	-1.544586	-0.141934
	1	-3.638175	-2.536353	-0.235511
	5	-1.605495	-1.048919	-0.982663
	1	-1.104006	-1.737789	-1.819926
	5	-1.508617	0.737618	-1.211206
	1	-1.259868	1.161868	-2.299399
	6	2.581746	2.220823	0.525552
	1	2.603003	3.267555	0.217717
	1	3.557661	1.964395	0.943162
	1	1.827840	2.108563	1.306540
	5	-0.928919	1.546911	0.169677
	1	-0.293885	2.544840	0.265579
	5	-3.112806	-0.049836	-0.933698
	1	-3.971560	0.133283	-1.743533
	5	-3.154817	-0.202628	0.881994
	1	-4.051447	-0.155343	1.669228
(Z)-(Ph ₃ PCH ₂ Ph)[3a]				
	7	-0.791627	-0.632159	-0.665496
	7	-1.976704	1.133184	0.188878
	6	-1.942518	-0.065840	-0.390722
	7	-3.134126	1.860456	0.491911
	6	-3.220091	-0.802667	-0.661079
	5	0.616700	-0.128418	-0.369853
	5	2.024639	0.190273	-1.460473
	5	2.710608	1.320445	-0.208850
	5	1.291278	1.000700	0.879313

	6	-3.688807	-1.557010	0.590399
	5	1.117861	1.439685	-0.758063
	5	1.911571	-1.359678	-0.531484
	5	2.944560	-1.406683	0.812097
	5	1.399211	-0.786889	1.119323
	5	2.887954	0.237550	1.228314
	5	3.403740	-0.337428	-0.423387
	1	-0.878608	-1.564072	-1.056746
	1	-1.104093	1.602847	0.402308
	1	-3.711990	1.339767	1.145714
	1	-3.668395	2.031053	-0.355680
	1	-3.031753	-1.509145	-1.470173
	1	-3.985014	-0.102776	-0.998801
	1	2.145751	0.128024	-2.645732
	1	3.441820	2.259208	-0.301621
	1	0.766293	1.652305	1.731383
	1	-4.602380	-2.107600	0.362234
	1	-3.897561	-0.879446	1.420586
	1	-2.927851	-2.270128	0.913813
	1	0.512169	2.334776	-1.252932
	1	1.636082	-2.347427	-1.143985
	1	3.528194	-2.315068	1.308828
	1	0.677192	-1.275477	1.935669
	1	3.507650	0.640067	2.165948
	1	4.477589	-0.441958	-0.934133
(Z)-(Ph ₃ PCH ₂ Ph)[3b]				
	8	2.976320	-1.504446	-1.146087

	7	2.687985	-0.359536	0.783086
	1	2.994506	-0.190012	1.733129
	7	1.371979	-0.075714	0.506029
	1	0.686742	-0.799579	0.711745
	6	4.768200	-1.551802	0.452996
	1	4.745992	-2.616878	0.695005
	1	5.055724	-0.990303	1.341433
	1	5.503074	-1.397035	-0.337066
	7	-0.203925	1.058415	-0.700522
	1	-0.375203	1.883900	-1.265727
	6	1.036389	0.928236	-0.318481
	5	-3.574736	-0.589297	1.255827
	1	-4.110642	-1.006548	2.237536
	6	2.075489	1.931837	-0.714338
	1	1.681076	2.505423	-1.553577
	1	2.968551	1.405413	-1.055029
	5	-2.461814	0.816853	0.993872
	1	-1.982796	1.632392	1.722635
	6	3.412269	-1.163126	-0.065148
	5	-4.096456	0.864112	0.552810
	1	-4.962975	1.613536	0.868505
	5	-3.004152	0.922746	-0.741377
	1	-2.995672	1.832970	-1.514669
	6	2.415375	2.862239	0.456003
	1	3.148980	3.600638	0.129607
	1	2.837192	2.307617	1.295940
	1	1.523886	3.390820	0.799392

	5	-1.413676	0.199563	-0.338874
	5	-4.118091	-0.481052	-0.482266
	1	-5.131819	-0.802891	-1.024091
	5	-1.383135	-1.487404	-0.451925
	1	-0.499068	-2.216076	-0.768660
	5	-1.799720	-0.863487	1.077837
	1	-1.163627	-1.171743	2.039873
	5	-2.571790	-0.713868	-1.382969
	1	-2.615937	-0.885868	-2.562358
	5	-2.972994	-1.780907	0.035716
	1	-3.374336	-2.902957	0.095491
(E)-(Ph₃PCH₂Ph)[3b]				
	8	3.645460	-1.770736	-0.426027
	7	3.413560	0.234902	0.595466
	7	2.180375	-0.075836	1.107633
	6	5.506401	-0.255755	-0.514193
	7	-0.088852	-0.269975	0.905965
	6	1.040122	0.157098	0.423086
	5	-3.723283	-0.066109	-1.308314
	6	1.106010	0.976422	-0.825729
	5	-2.071721	-0.790700	-1.128709
	6	4.123295	-0.696074	-0.123807
	5	-3.485778	-1.707249	-0.951815
	5	-2.565171	-1.523917	0.458815
	6	1.148005	2.471813	-0.486732
	5	-1.529551	-0.053708	0.420750
	5	-4.218891	-0.805364	0.282893

	5	-2.353078	1.368862	0.828389
	5	-2.331333	0.980126	-0.825168
	5	-3.030606	-0.068114	1.425317
	5	-3.854373	0.962155	0.173131
	1	3.856015	1.075260	0.946609
	1	2.161047	-0.674078	1.928905
	1	5.584524	-0.289776	-1.602052
	1	6.223670	-0.966617	-0.100144
	1	5.744426	0.747725	-0.163424
	1	-0.018918	-0.805198	1.770348
	1	-4.348643	0.254466	-2.273222
	1	0.230406	0.746827	-1.428412
	1	1.994388	0.691567	-1.390137
	1	-1.225288	-1.099930	-1.912200
	1	-3.847033	-2.686037	-1.520970
	1	-2.143279	-2.466275	1.059630
	1	1.166594	3.051634	-1.410573
	1	2.038687	2.721427	0.094280
	1	0.264875	2.763142	0.085243
	1	-5.281199	-1.135089	0.717199
	1	-1.960314	2.329708	1.406899
	1	-1.898294	1.757795	-1.620030
	1	-3.200955	-0.206765	2.598401
	1	-4.764054	1.731533	0.246218
<i>(E)</i> -(Ph ₃ PCH ₂ Ph)[3e]				
	35	6.213914	-1.481000	0.148921
	7	0.057069	1.492364	0.153940

	7	-2.377148	0.412084	-0.133998
	1	-1.477502	-0.063975	-0.098363
	7	-1.068825	2.290401	0.011354
	1	-1.030125	3.273641	0.269757
	7	1.286213	3.247210	-0.828384
	1	0.460395	3.737062	-1.146842
	1	2.123642	3.452233	-1.355746
	6	1.155455	2.009014	-0.293180
	6	4.653639	-0.396653	0.044736
	6	-2.275873	1.710585	-0.042614
	6	2.384704	1.179087	-0.161422
	6	-3.462678	2.616180	0.062212
	1	-4.266404	2.208856	-0.549065
	1	-3.189154	3.594848	-0.337474
	6	2.280080	-0.212884	-0.197145
	1	1.308528	-0.677057	-0.317274
	6	3.413174	-1.008202	-0.097825
	1	3.327496	-2.086673	-0.139811
	6	3.639650	1.769981	-0.010288
	1	3.742279	2.848192	0.042258
	5	-3.609595	-0.483926	-0.246611
	6	-3.912017	2.745729	1.523565
	1	-3.137620	3.224671	2.126916
	1	-4.132215	1.767037	1.953041
	1	-4.815601	3.355396	1.574886
	5	-3.279108	-2.217849	0.104591
	1	-2.131289	-2.550791	0.109967

	5	-4.242383	-1.217954	1.278499
	1	-3.931524	-0.686519	2.301760
	5	-4.406094	-2.899889	1.171584
	1	-4.277487	-3.744100	1.998496
	5	-3.999365	-1.749569	-1.486119
	1	-3.435271	-2.090545	-2.481218
	6	4.782346	0.984340	0.096552
	1	5.752584	1.449618	0.218387
	5	-5.760587	-1.594113	-1.059823
	1	-6.752073	-1.801449	-1.691530
	5	-4.801066	-3.010532	-0.474219
	1	-5.003716	-4.077732	-0.970176
	5	-4.798274	-0.255723	-1.425984
	1	-4.900719	0.596831	-2.247678
	5	-5.363313	-0.321532	0.177813
	1	-5.997527	0.585004	0.625903
	5	-5.767285	-2.003382	0.700336
	1	-6.820372	-2.184673	1.233336
(Z)-(Ph ₃ PCH ₂ Ph)[3e]				
	35	5.605097	0.368314	-0.517684
	7	-0.869727	-1.667513	0.016789
	7	-3.425056	-0.529462	-0.610679
	7	-2.113226	-2.192774	0.378917
	7	-0.331163	-1.810039	2.299435
	6	-0.043451	-1.557181	1.008896
	6	3.861057	-0.221764	-0.034297
	6	-3.254753	-1.738611	-0.132151

	6	1.330748	-1.100643	0.667249
	6	-4.398218	-2.716037	-0.176100
	6	1.878283	-1.447397	-0.568677
	6	3.147489	-1.012845	-0.926426
	6	2.064732	-0.302336	1.545625
	5	-2.737070	0.838047	-0.420478
	6	-4.473534	-3.391846	-1.549886
	5	-3.988704	2.149206	-0.411483
	5	-3.161552	1.667645	1.135542
	5	-3.831276	3.206983	0.902014
	5	-2.614729	2.236643	-1.574172
	6	3.335719	0.143284	1.198042
	5	-1.307490	2.952353	-0.537306
	5	-2.976676	3.643901	-0.496189
	5	-1.262039	1.346556	-1.063923
	5	-1.437198	1.551997	0.616408
	5	-2.143515	3.158947	1.051252
	1	-4.371797	-0.415330	-0.960666
	1	-2.117966	-3.154100	0.713083
	1	-1.289967	-1.930895	2.593997
	1	0.372876	-1.719757	3.015676
	1	-5.324319	-2.180552	0.038795
	1	-4.254247	-3.466995	0.603398
	1	1.309549	-2.067033	-1.250815
	1	3.570785	-1.292068	-1.883188
	1	1.648006	0.004408	2.498789
	1	-3.543225	-3.917872	-1.775229

	1	-4.656813	-2.657477	-2.336864
	1	-5.289048	-4.116498	-1.559323
	1	-5.059915	1.854923	-0.854302
	1	-3.515902	0.971558	2.040348
	1	-4.656602	3.792368	1.526345
	1	-2.863576	2.333712	-2.737833
	1	3.898281	0.769240	1.879435
	1	-0.395907	3.679075	-0.796078
	1	-3.163900	4.707478	-1.005738
	1	-0.444404	0.759535	-1.695508
	1	-0.649006	1.069757	1.371014
	1	-1.594286	3.794912	1.900399
(Z)-(Ph ₃ PCH ₂ Ph)[3g]				
	6	-1.275277	-1.324799	0.035002
	6	-0.572645	4.024087	0.847593
	1	0.222936	4.163094	1.582676
	1	-1.301112	3.319198	1.251453
	1	-1.068535	4.983664	0.693610
	6	-0.696193	-3.243740	1.378555
	1	-0.751763	-3.781677	2.318031
	6	0.001649	3.524809	-0.483771
	1	-0.795403	3.384743	-1.214233
	1	0.712756	4.249037	-0.883235
	7	0.030140	1.089338	-0.156977
	1	0.524688	0.214933	0.023114
	6	-5.497516	1.300601	0.133069
	1	-6.006762	2.232867	0.349601

	6	-0.547523	-1.854905	-1.031925
	1	-0.488152	-1.307797	-1.967391
	6	0.728239	2.227511	-0.298955
	7	2.029940	2.189066	-0.225951
	1	2.477682	3.091281	-0.351427
	6	-3.441024	0.050860	-0.111843
	7	-1.332221	1.111344	-0.220483
	6	-1.355725	-2.027236	1.239576
	1	-1.927761	-1.615990	2.064839
	6	-6.235816	0.154267	-0.162202
	1	-7.319022	0.194889	-0.180023
	6	-5.577520	-1.042012	-0.426951
	1	-6.144463	-1.936828	-0.657283
	6	-1.956850	-0.003387	-0.105310
	6	-4.186667	-1.096957	-0.396904
	1	-3.681564	-2.033111	-0.606523
	6	-4.110734	1.251034	0.159505
	1	-3.536673	2.138735	0.398533
	6	0.105388	-3.076451	-0.890945
	1	0.674426	-3.481040	-1.720472
	6	0.034579	-3.768571	0.314333
	1	0.549008	-4.716369	0.426176
	5	4.634827	0.966931	0.681370
	1	5.375678	1.841068	1.012653
	5	4.456323	-0.855852	-1.600173
	1	4.826876	-1.310354	-2.633683
	5	4.344857	-0.714660	1.316215

	1	4.841999	-1.327048	2.211439
	5	2.907366	-0.396957	-1.090143
	1	1.962776	-0.480504	-1.815786
	5	4.316609	0.753242	-1.088101
	1	4.592801	1.665652	-1.806929
	5	2.642661	-0.668369	0.686285
	1	1.639789	-1.222481	1.027410
	5	3.899569	-1.618073	-0.189806
	1	3.834928	-2.809441	-0.134725
	5	5.312628	-0.462475	-0.189450
	1	6.491458	-0.642548	-0.136099
	5	3.298295	0.568007	1.649506
	1	2.903114	1.050232	2.660665
	5	2.946285	0.995581	0.048003
(Z)-(Ph ₃ PCH ₂ Ph)[3g]				
	6	2.629522	1.392085	-0.050358
	6	-1.077319	-2.208695	1.481689
	6	2.271797	3.510311	1.053689
	6	-1.133392	-1.948726	-0.030135
	7	0.201797	0.083427	-0.306326
	6	4.729170	-3.090274	-0.452885
	6	3.343404	2.042015	-1.059822
	6	-1.019734	-0.482803	-0.296987
	7	-2.068410	0.282144	-0.441443
	6	3.661462	-0.958405	-0.042527
	7	1.321553	-0.681236	-0.175850
	6	2.094157	2.130674	1.008033

	6	5.921333	-2.602950	0.082178
	6	5.983431	-1.295135	0.552165
	6	2.457179	-0.089723	-0.093857
	6	4.861569	-0.473416	0.486765
	6	3.607191	-2.275104	-0.516674
	6	3.507452	3.422245	-1.016848
	6	2.974733	4.156544	0.040393
	5	-4.897716	0.341719	-1.525790
	5	-5.889868	-0.994206	-0.791930
	5	-4.541507	-1.378997	0.367411
	5	-4.301496	-1.231548	-1.312223
	5	-3.562219	-0.046255	-0.363899
	1	-1.881660	-1.677445	1.995039
	1	-0.120820	-1.886135	1.898976
	1	-1.195823	-3.276643	1.670869
	1	1.862102	4.077868	1.881266
	1	-0.314819	-2.462129	-0.532324
	1	-2.076966	-2.300016	-0.440189
	1	0.283610	1.092725	-0.431988
	1	4.677354	-4.105652	-0.829003
	1	3.768055	1.465323	-1.874957
	1	-1.860681	1.273932	-0.543341
	1	1.549682	1.622124	1.797753
	1	6.797771	-3.239589	0.125811
	1	6.906541	-0.909622	0.969801
	1	4.919395	0.543151	0.859032
	1	2.684472	-2.649879	-0.943800

	1	4.054664	3.924295	-1.806486
	1	3.110480	5.231488	0.075678
	1	-4.891880	0.845598	-2.607526
	1	-6.765646	-1.670369	-1.239714
	1	-4.225669	-2.383940	0.928700
	1	-3.823799	-1.988957	-2.093169
	5	-5.725276	1.175620	1.153160
	5	-4.351762	0.189320	1.245177
	5	-4.595939	1.400346	-0.090763
	5	-6.003565	-0.483752	0.937391
	5	-6.251177	0.734246	-0.398160
	1	-6.178197	1.946354	1.936357
	1	-3.635890	0.202103	2.201027
	1	-4.089258	2.463946	-0.291401
	1	-6.770814	-1.061735	1.646816
	1	-7.235893	1.228525	-0.858670
(Z)-(Ph ₃ PCH ₂ Ph)[3g] (experimental X-ray geometry)				
	6	5.958932	1.627777	3.511440
	6	5.486494	6.977228	3.266705
	1	5.659189	7.205621	2.329590
	1	6.174747	6.355080	3.583047
	1	5.507877	7.792130	3.810456
	6	7.124657	0.400488	1.782183
	1	7.849948	0.326315	1.173593
	6	4.113760	6.322452	3.387042
	1	3.957843	6.050329	4.326187
	1	3.412726	6.975883	3.136350

	7	4.649030	3.979382	2.848163
	1	4.680480	3.292972	2.298456
	6	6.341660	3.446303	8.091432
	1	5.914806	3.916787	8.798566
	6	5.011612	0.608522	3.579121
	1	4.291604	0.670101	4.194886
	6	4.018145	5.118428	2.495273
	7	3.382666	5.168576	1.363101
	1	3.008046	5.941873	1.172240
	6	6.419630	2.777128	5.767663
	7	5.240935	3.918068	4.088220
	6	7.014243	1.519010	2.603293
	1	7.660615	2.213278	2.546170
	6	7.499091	2.710612	8.340634
	1	7.864992	2.685077	9.216836
	6	8.116273	2.016202	7.312285
	1	8.914094	1.529526	7.481488
	6	5.840524	2.821575	4.402667
	6	7.570096	2.028492	6.030537
	1	7.980274	1.527477	5.334638
	6	5.813046	3.490627	6.809278
	1	5.035653	4.008431	6.636826
	6	5.119532	-0.500518	2.743934
	1	4.467412	-1.189631	2.784408
	6	6.177276	-0.600871	1.853384
	1	6.250938	-1.361789	1.290005
	5	2.822292	4.187240	-1.458667

	1	2.729208	5.069707	-2.142789
	5	1.071732	2.165551	-0.250691
	1	0.083045	1.674040	-0.063620
	5	3.587751	2.539872	-1.701237
	1	4.106199	2.092995	-2.588132
	5	2.318190	2.576668	0.824358
	1	2.170156	2.342588	1.909966
	5	1.501192	3.813900	-0.262874
	1	0.700074	4.568480	-0.052791
	5	3.977873	2.438247	0.078916
	1	4.807267	1.921334	0.626729
	5	2.592653	1.482763	-0.608861
	1	2.663078	0.367321	-0.682227
	5	1.774752	2.727447	-1.702996
	1	1.191985	2.608779	-2.651752
	5	4.372415	3.795341	-0.876066
	1	5.361530	4.287499	-1.062596
	5	3.196833	4.071839	0.317019
(Z)-(Ph ₃ PCH ₂ Ph)[3g] (gas phase optimized geometry)				
	6	-1.234163	-1.347099	0.027444
	6	-0.649914	4.025118	0.866582
	1	0.136947	4.166470	1.609913
	1	-1.365866	3.302256	1.258961
	1	-1.161237	4.977967	0.717519
	6	-0.576727	-3.266835	1.329135
	1	-0.617259	-3.831275	2.253325
	6	-0.058235	3.527040	-0.457681

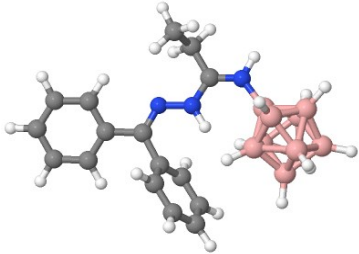
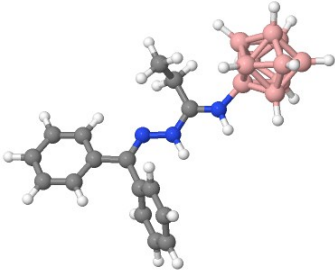
	1	-0.853265	3.359898	-1.185463
	1	0.630830	4.267191	-0.869407
	7	0.023906	1.095533	-0.085231
	1	0.561675	0.239461	0.097948
	6	-5.497308	1.241736	0.039809
	1	-6.015095	2.181939	0.192132
	6	-0.426900	-1.789392	-1.020351
	1	-0.341921	-1.192683	-1.921215
	6	0.704869	2.249421	-0.262658
	7	1.999537	2.223833	-0.210432
	1	2.459841	3.112682	-0.359817
	6	-3.423818	0.001519	-0.109642
	7	-1.327966	1.089551	-0.165682
	6	-1.317924	-2.099276	1.201029
	1	-1.940435	-1.750096	2.018175
	6	-6.224707	0.073975	-0.186930
	1	-7.307807	0.102504	-0.214777
	6	-5.550733	-1.126863	-0.371778
	1	-6.105569	-2.040747	-0.550121
	6	-1.939854	-0.038316	-0.084710
	6	-4.159977	-1.165338	-0.330922
	1	-3.639125	-2.104176	-0.478001
	6	-4.111861	1.206796	0.080159
	1	-3.541528	2.109749	0.260759
	6	0.323444	-2.951322	-0.884635
	1	1.004350	-3.244586	-1.674298
	6	0.251361	-3.687004	0.291420

	1	0.869412	-4.568335	0.413061
	5	4.598438	0.970602	0.671913
	1	5.351555	1.839805	0.990310
	5	4.345425	-0.839844	-1.622651
	1	4.686556	-1.294181	-2.666221
	5	4.264704	-0.707060	1.303486
	1	4.744215	-1.333002	2.197340
	5	2.813155	-0.348854	-1.095948
	1	1.861768	-0.386618	-1.822893
	5	4.255705	0.772621	-1.103211
	1	4.531060	1.687906	-1.824165
	5	2.561054	-0.619815	0.689590
	1	1.536760	-1.129797	1.046720
	5	3.778705	-1.596806	-0.207253
	1	3.673926	-2.786536	-0.155438
	5	5.226008	-0.475402	-0.214645
	1	6.398958	-0.684739	-0.172856
	5	3.253944	0.604275	1.655223
	1	2.860718	1.102205	2.661669
	5	2.916235	1.022154	0.053222
(E)-(Ph ₃ PCH ₂ Ph)[3g] (gas phase optimized geometry)				
	6	2.624126	1.398742	-0.048752
	6	-1.134874	-2.346716	1.210788
	6	2.247261	3.497840	1.090864
	6	-1.162856	-1.956800	-0.274714
	7	0.213158	0.095037	-0.382751
	6	4.682613	-3.134917	-0.243324

	6	3.278090	2.084094	-1.074392
	6	-1.038264	-0.476737	-0.416610
	7	-2.066872	0.293098	-0.507730
	6	3.656699	-0.955456	-0.024636
	7	1.319311	-0.658247	-0.251265
	6	2.110633	2.114378	1.035549
	6	5.903965	-2.614267	0.182893
	6	5.999531	-1.266266	0.504116
	6	2.460130	-0.084638	-0.116353
	6	4.884331	-0.439957	0.400096
	6	3.569713	-2.315669	-0.348646
	6	3.411910	3.466759	-1.018362
	6	2.896648	4.174841	0.063432
	5	-4.925566	0.524347	-1.419264
	5	-5.903205	-0.876013	-0.788063
	5	-4.500394	-1.432511	0.233176
	5	-4.355005	-1.086561	-1.434972
	5	-3.567114	-0.031507	-0.389670
	1	-1.950873	-1.849684	1.737983
	1	-0.179319	-2.078364	1.668438
	1	-1.276785	-3.424246	1.304773
	1	1.841760	4.044374	1.933856
	1	-0.342744	-2.431552	-0.814386
	1	-2.117095	-2.242759	-0.717259
	1	0.290063	1.108404	-0.387695
	1	4.600516	-4.185080	-0.498394
	1	3.680349	1.526674	-1.913534

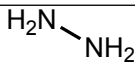
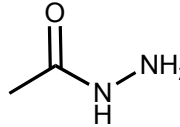
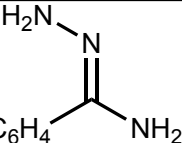
	1	-1.894861	1.295697	-0.550231
	1	1.602686	1.582382	1.833839
	1	6.772730	-3.257379	0.260917
	1	6.943072	-0.852889	0.841267
	1	4.966833	0.608806	0.661037
	1	2.618725	-2.713503	-0.680818
	1	3.914980	3.992040	-1.821802
	1	2.997586	5.252946	0.104045
	1	-4.941396	1.158128	-2.430068
	1	-6.816445	-1.478031	-1.262637
	1	-4.155242	-2.497847	0.653688
	1	-3.917571	-1.745863	-2.326073
	5	-5.559923	1.044903	1.395685
	5	-4.210456	0.025002	1.285836
	5	-4.509249	1.399468	0.116223
	5	-5.895183	-0.576024	1.002536
	5	-6.195237	0.801454	-0.163786
	1	-5.934801	1.723207	2.296697
	1	-3.420085	-0.080565	2.181429
	1	-3.955926	2.461563	0.005823
	1	-6.624453	-1.224499	1.688511
	1	-7.192165	1.364639	-0.497378

Table 4S. Calculated total energies, enthalpies, Gibbs free energies (in Hartree) and entropies (in cal/mol•K) (E, H, G, S) for two isomers of (Ph₃PCH₂Ph)[**3g**] (M06-2X/6-311++G** level of theory, optimization in gas phase).

Isomer	E	H	G	S
 (Z)-(Ph ₃ PCH ₂ Ph)[3g]	-1038.186703	-1037.727663	-1037.806072	165.03
 (E)-(Ph ₃ PCH ₂ Ph)[3g]	-1038.167566	-1037.709305	-1037.792369	174.82

Kinetic measurements

Table 5S. Observed pseudo first-order Rate constants for the six nucleophiles, **2a–c**; **2e–2g**; vary [Nu(H)]; Other details in Column 1.

Nucleophile	Conc, M	k_{obs} (Standard Deviation) (s ⁻¹)			
		35°C	25°C	15°C	5°C
 Chemical Formula: H ₄ N ₂ Exact Mass: 32.04 Wavelength 335 nm (Ph ₃ PCH ₂ Ph)[1] = 5 × 10 ⁻⁴ M 2a	0.003	-	0.000647 (0.000014)	-	-
	0.004	-	0.0009890 (0.000218)	-	-
	0.005	-	0.001465 (0.000036)	-	-
	0.006	-	0.002015 (0.000056)	-	-
	0.0075	-	0.003145 (0.000081)	-	-
	0.01	-	0.005343 (0.000167)	-	-
 Chemical Formula: C ₂ H ₆ N ₂ O Exact Mass: 74.05 Wavelength 335 nm (Ph ₃ PCH ₂ Ph)[1] = 5 × 10 ⁻⁴ M 2b	0.05	-	0.000489 (0.000014)	-	-
	0.06	-	0.000749 (0.000024)	0.000163 (0.000003)	-
	0.075	-	0.001088 (0.000031)	0.000224 (0.000007)	-
	0.09	-	0.001630 (0.000116)	0.000577 (0.000016)	-
	0.1	-	0.001930 (0.000032)	0.000718 (0.000012)	-
 p -MeOC ₆ H ₄ Chemical Formula: C ₈ H ₁₁ N ₃ O Exact Mass: 165.09 Wavelength 350 nm (Ph ₃ PCH ₂ Ph)[1]	0.00025	-	0.000715 (0.000008)	-	-
	0.0005	0.002340 (0.00002)	0.001532 (0.000026)	-	-
	0.001	0.006729 (0.000014)	0.004486 (0.000008)	0.004847 (0.000007)	-
	0.0015	0.016398 (0.000079)	0.011214 (0.000021)	0.009860 (0.000030)	-
	0.002	0.025030	0.019112	0.017223	-

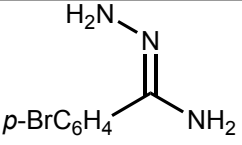
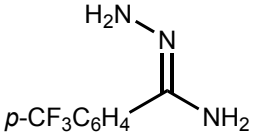
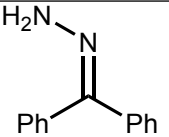
$= 5 \times 10^{-5} \text{ M}$ 2c		(0.000152)	(0.000056)	(0.000091)	
	0.0025	0.034918 (0.000373)	-	-	-
 Chemical Formula: C ₇ H ₈ BrN ₃ Exact Mass: 212.99 Wavelength 360 nm (Ph ₃ PCH ₂ Ph)[1] $= 5 \times 10^{-5} \text{ M}$ 2e	0.00025	-	0.000196 (0.000001)	-	-
	0.0005	-	0.000470 (0.000004)	-	-
	0.00075	-	0.001380 (0.000016)	-	-
	0.001	0.0021812 (0.000002)	0.001711 (0.000001)	0.001579 (0.000011)	0.001426 (0.000005)
	0.002	0.0069203 (0.000012)	0.006142 (0.000007)	0.005544 (0.000077)	0.004065 (0.000012)
	0.003	0.018409 (0.000065)	0.013626 (0.000037)	0.012456 (0.000029)	0.011402 (0.000045)
	0.004	0.030796 (0.000241)	0.022034 (0.000107)	0.020261 (0.000114)	0.017673 (0.000099)
 Chemical Formula: C ₈ H ₈ F ₃ N ₃ Exact Mass: 203.07 Wavelength 360 nm (Ph ₃ PCH ₂ Ph)[1] $= 5 \times 10^{-5} \text{ M}$ 2f	0.0005	-	0.000277 (0.000001)	-	-
	0.00075	-	0.000609 (0.000005)	-	-
	0.001	-	0.000836 (0.000001)	0.000849 (0.000009)	0.000748 (0.000001)
	0.002	-	0.004892 (0.000011)	0.003516 (0.000006)	0.002702 (0.000006)
	0.003	-	0.011421 (0.000084)	0.007606 (0.000037)	0.006429 (0.000023)
	0.004	-	-	0.013889 (0.000104)	0.009976 (0.000054)
 Chemical Formula: C ₁₃ H ₁₂ N ₂ Exact Mass: 196.10 Wavelength 370 nm (Ph ₃ PCH ₂ Ph)[1] $= 5 \times 10^{-5} \text{ M}$ 2g	0.025	-	0.0000290 (0.000002)	-	-
	0.05	-	0.000121 (0.000002)	0.000140 (0.000104)	-
	0.075	-	0.000319 (0.000005)	0.000293 (0.000005)	-
	0.09	-	0.000449 (0.0000011)	-	-
	0.1	-	0.000539 (0.00017)	0.000505 (0.00002)	-

Table 6S. Data used to construct **Fig. 5**: *k*_{obs} vs *k*_{obs}_Calc for **2a–2c**; **2e–2g** at 25 °C (details in **Table 5S**).

	2g		2b		2a		2f		2e		2c	
L	KOBS1	KOBS1_CALC	KOBS2	KOBS2_CALC	KOBS3	KOBS3_CALC	KOBS4	KOBS4_CALC	KOBS5	KOBS5_CALC	KOBS6	KOBS6_CALC
(M)	(s ⁻¹)	(s ⁻¹)	(s ⁻¹)	(s ⁻¹)	(s ⁻¹)	(s ⁻¹)	(s ⁻¹)	(s ⁻¹)	(s ⁻¹)	(s ⁻¹)	(s ⁻¹)	(s ⁻¹)
0	0	0	0	0		0	0	0	0	0	0	0
0.00025		3.42E-09		1.23E-08		3.41E-06		7.86E-05	0.000196	8.86E-05		0.000301
0.0005							0.000278	0.000314	0.000471	0.000354	0.001532	0.001204
0.00075								0.000707	0.00138	0.000797		0.002709
0.001							0.000836	0.001257	0.001711	0.001418	0.004486	0.004816
0.0015								0.002828		0.00319	0.011215	0.010836
0.002							0.004892	0.005028	0.006143	0.00567	0.019112	0.019264
0.0025								0.007856		0.00886		0.030099
0.003					0.000648	0.000491	0.011421	0.011313	0.013627	0.012759		0.043343
0.004					0.000989	0.000874		0.020112	0.022034	0.022682		
0.005					0.001465	0.001365		0.031425		0.03544		
0.006					0.002015	0.001966						
0.0075					0.003145	0.003071						
0.01		5.47E-06		1.97E-05	0.005343	0.00546						
0.025	2.9E-05	3.42E-05		0.000123		0.034127						
0.05	0.000121	0.000137	0.00049	0.000491								
0.06		0.000197	0.000749	0.000708								
0.075	0.00032	0.000308	0.001089	0.001106								
0.09	0.00045	0.000443	0.00163	0.001592								
0.1	0.000539	0.000547	0.00193	0.001966								
0.115		0.000723		0.0026								

Table 7S. Observed pseudo first-order Rate constants for three nucleophiles [Fig. 6], **2c**; **2e** and **2f**; (Vary [B]).

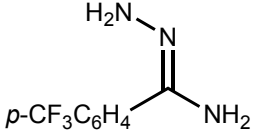
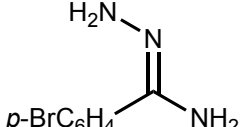
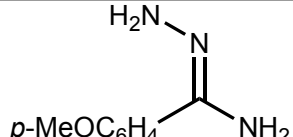
Nucleophile, [Nu] = 2.5×10^{-4} M	Conc [B], M	k_{obs} (Standard Deviation) (s^{-1})		
		35 °C	25 °C	15 °C
 Chemical Formula: $\text{C}_8\text{H}_8\text{F}_3\text{N}_3$ Exact Mass: 203.07 2f	0.003	-	0.000797 (0.000013)	-
	0.004	-	0.000893 (0.000015)	-
	0.005	-	0.001041 (0.000029)	-
	0.01	-	0.001561 (0.000039)	-
 Chemical Formula: $\text{C}_7\text{H}_8\text{BrN}_3$ Exact Mass: 212.99 2e	0.003	0.001677 (0.000038)	0.001145 (0.000025)	0.000923 (0.000019)
	0.004	0.002125 (0.000053)	0.001385 (0.000036)	0.001098 (0.000027)
	0.005	0.002477 (0.000069)	0.001599 (0.000042)	0.001208 (0.000033)
	0.01	0.004047 (0.000162)	0.002531 (0.000096)	0.001401 (0.000049)
 Chemical Formula: $\text{C}_8\text{H}_{11}\text{N}_3\text{O}$ Exact Mass: 165.09 2c	0.003	-	0.003773 (0.000227)	-
	0.004	-	0.004592 (0.000317)	-
	0.005	-	0.004851 (0.000371)	-

Table 8S. Data from Global fit illustrated in **Fig. 7**: k_{obs} vs $k_{\text{obs_Calc}}$ for 2e; 5–35 °C (Details in **Table 5S**).

T	L	KOBS	KOBS_CALC
(°C)	(M)	(s⁻¹)	(s⁻¹)
35	0		0
35	0.00025		0.000104
35	0.0005		0.000415
35	0.00063		0.000658
35	0.00075		0.000933
35	0.00088		0.001285
35	0.001	0.00218	0.001659
35	0.0015		0.003732
35	0.002	0.00692	0.006635
35	0.0025		0.010368
35	0.003	0.016031	0.014929
35	0.0033		0.018065
35	0.00365		0.0221
35	0.004	0.026675	0.026541
35	0.0042		0.029262
35	0.0046		0.035101
30	0.00063		0.00062
30	0.00075		0.000879
30	0.00088		0.001209
30	0.001	0.001695	0.001562
30	0.0015		0.003514
30	0.002	0.006864	0.006247
30	0.0025		0.009761
30	0.003		0.014056
30	0.0033		0.017008
30	0.00365		0.020807
30	0.004	0.024498	0.024989
30	0.0042		0.027551

30	0.0046		0.033048
30	0.0048		0.035984
25	0.00025	0.0002	9.17E-05
25	0.0005	0.00047	0.000367
25	0.00063		0.000583
25	0.00075	0.00138	0.000826
25	0.00088		0.001137
25	0.001	0.00171	0.001468
25	0.0015		0.003303
25	0.002	0.00614	0.005872
25	0.0025		0.009175
25	0.003	0.01363	0.013211
25	0.0033		0.015986
25	0.00365		0.019556
25	0.004	0.02203	0.023487
25	0.0042		0.025894
25	0.0046		0.031061
20	0.00063		0.000547
20	0.00075		0.000775
20	0.00088		0.001066
20	0.001	0.001658	0.001377
20	0.0015		0.003099
20	0.002	0.005914	0.005509
20	0.0025		0.008607
20	0.003	0.013174	0.012394
20	0.0033		0.014997
20	0.00365		0.018347
20	0.004	0.021534	0.022035
20	0.0042		0.024293
20	0.0046		0.029141
20	0.0048		0.03173
15	0.00025		8.06E-05
15	0.0005		0.000322

15	0.00063		0.000512
15	0.00075		0.000725
15	0.00088		0.000999
15	0.001	0.00158	0.00129
15	0.0015		0.002901
15	0.002	0.00554	0.005158
15	0.0025		0.00806
15	0.003	0.01246	0.011606
15	0.0033		0.014043
15	0.00365		0.01718
15	0.004	0.02026	0.020632
15	0.0042		0.022747
15	0.0046		0.027286
15	0.0048		0.029711
5	0.00063		0.000446
5	0.00075		0.000632
5	0.00088		0.00087
5	0.001	0.00143	0.001124
5	0.0015		0.002528
5	0.002	0.00407	0.004495
5	0.0025		0.007023
5	0.003	0.0114	0.010113
5	0.0033		0.012237
5	0.00365		0.014971
5	0.004	0.01767	0.017979
5	0.0042		0.019822
5	0.0046		0.023778

Table 9S. Reverse Rate constants for Nucleophiles **2a–2c**; **2e–2f** at 25 °C in MeCN; other details in Caption **Table 5S**.

Nucleophile	Fig. 5 Eq. 5b ^{a)}	Fig. 5 Eq. 4a ^{b)}	Fig. 6 Eq. 6ab ^{c)}
	k_r	k_r	
2a	0.00017(18)	0.00017(18)	--
2b	0.000013(196)	-0.000004(196)	--
2c	0.00009(17)	0.00009(17)	0.0023(3)
2e	0.00034(12)	0.00033(12)	0.00059(11)
2f	-0.00017(16)	-0.00017(16)	0.00047(12)
2g	0.000004(171)	-0.000007(171)	--

^{a)} Eq. 5b; Data in **Fig. 5**. ^{b)} Eq. 4a; Data in **Fig. 5**. ^{c)} Eq. 6ab; Data in **Fig. 6**

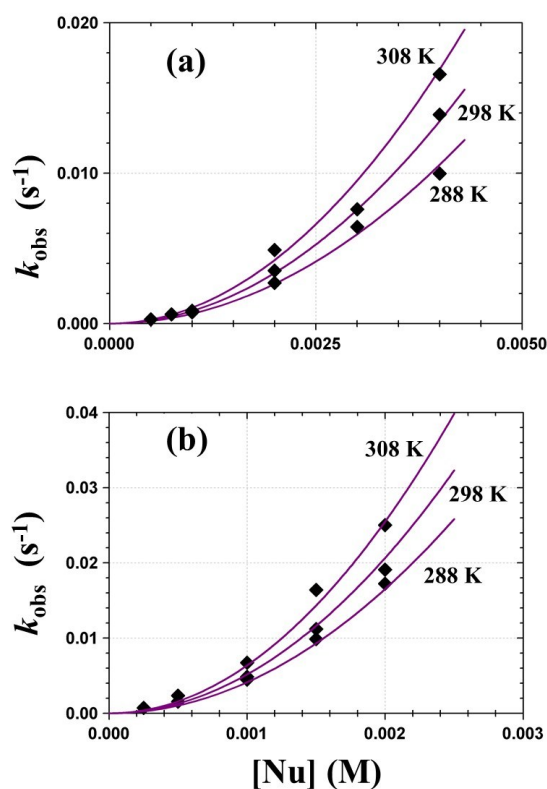


Figure 55S. Plots of k_{obs} vs. $[\text{Nu(H)}]$ for the nucleophilic addition of (a) **2f** and (b) **2c** to (Ph₃PCH₂Ph)[**1**] at 25 °C in MeCN. Details in Experimental.

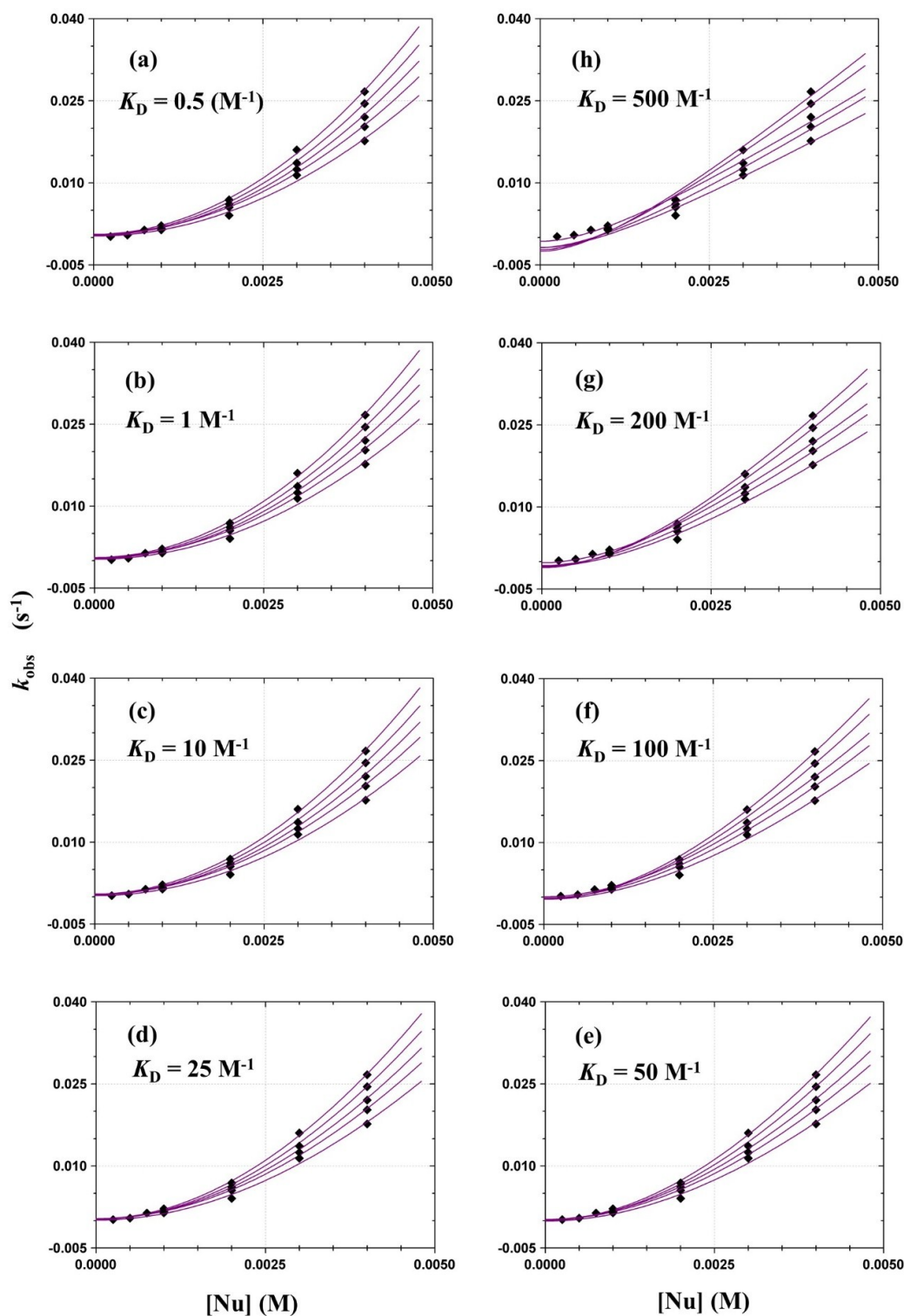


Figure 56S. Plots of k_{obs} vs. $[\text{Nu}]$ in MeCN for the nucleophilic addition of **2e** to $(\text{Ph}_3\text{PCH}_2\text{Ph})[\mathbf{1}]$ as a function of temperature to evaluate the magnitude and influence of K_D in Eq. 4a. A value of $K_D=1 \text{ M}^{-1}$ was selected for fits.

Literature

1. W. J. Geary, *Coord. Chem. Rev.*, 1971, **7**, 81–122.
2. (a) D. S. Bolotin, V. K. Burianova, A. S. Novikov, M. Y. Demakova, C. Pretorius, P. P. Mokolokolo, A. Roodt, N. A. Bokach, V. V. Suslonov, A. P. Zhdanov, K. Y. Zhizhin, N. T. Kuznetsov and V. Y. Kukushkin, *Organometallics*, 2016, **35**, 3612–3623; (b) D. S. Bolotin, M. Y. Demakova, E. A. Daines, M. S. Avdontseva, A. P. Zhdanov, K. Y. Zhizhin and N. T. Kuznetsov, *Russ. J. Gen. Chem.*, 2017, **87**, 37–43; (c) A. L. Mindich, N. A. Bokach, M. L. Kuznetsov, M. Haukka, A. P. Zhdanov, K. Y. Zhizhin, S. A. Miltsov, N. T. Kuznetsov and V. Y. Kukushkin, *ChemPlusChem*, 2012, **77**, 1075–1086.
3. L. J. Bellamy, *The infra-red spectra of complex molecules*, London: Methuen and co. Ltd; New York: John Wiley and sons, Inc., 1958.
4. I. B. Sivaev, N. A. Votnova, V. I. Bragin, Z. A. Starikova, L. V. Goeva, V. I. Bregadze and S. Sjoberg, *J. Organomet. Chem.*, 2002, **657**, 163–170.
5. (a) A. P. Zhdanov, M. V. Lisovsky, L. V. Goeva, G. A. Razgonyaeva, I. N. Polyakova, K. Y. Zhizhin and N. T. Kuznetsov, *Russ. Chem. Bull.*, 2009, **58**, 1694–1700; (b) A. P. Zhdanov, I. N. Polyakova, G. A. Razgonyaeva, K. Y. Zhizhin and N. T. Kuznetsov, *Russ. J. Inorg. Chem.*, 2011, **56**, 847–855.
6. F. H. Allen, O. Kennard, D. G. Watson, L. Brammer, A. G. Orpen and R. Taylor, *J. Chem. Soc., Perkin Trans.*, 1987, **12**, S1–S19.
7. A. Fischer and D. Wiebelhaus, *Z. Krist. – New Cryst. St.*, 1997, **212**, 335–336.