

Supplementary materials for

Fine-tuning effect of auxiliary ligand on two trigonal-bipyramid cobalt(II) complexes exhibiting field-Induced slow magnetic relaxation

Xiangyang Hou^a, Xiangyu Liu^{*b}, Xiao Wang^{*a}, Jijiang Wang^a, Long Tang^a, Ping Ju^a

^aDepartment of Chemistry and Chemical Engineering, Yan'an University Laboratory
of New Energy & New Function Materials, Yan'an Key Laboratory of Analytical
Technology and Detection, Yan'an University, Shaanxi 716000, China

^bState Key Laboratory of High-efficiency Utilization of Coal and Green Chemical
Engineering, College of Chemistry and Chemical Engineering, Ningxia University,
Yinchuan 750021, China

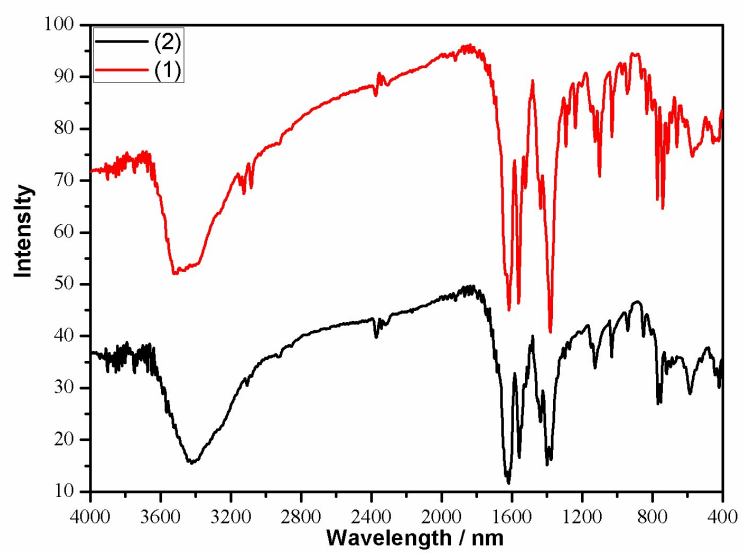
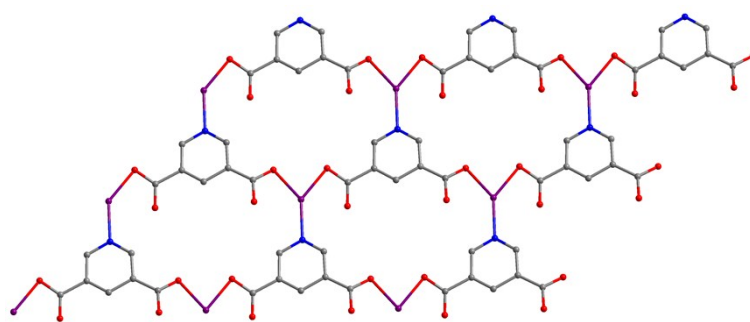
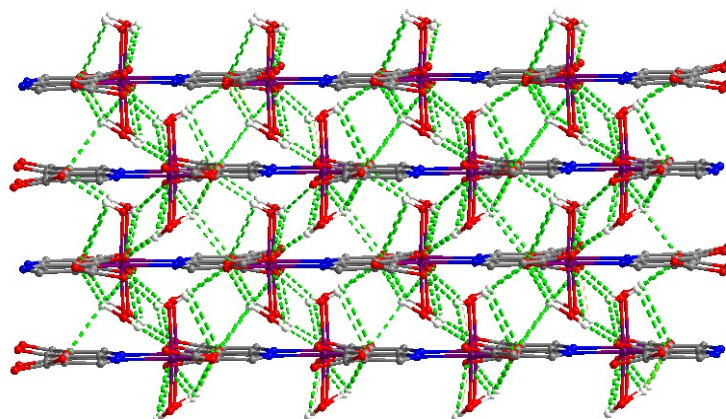


Fig. S1. The IR spectra of **1** and **2**

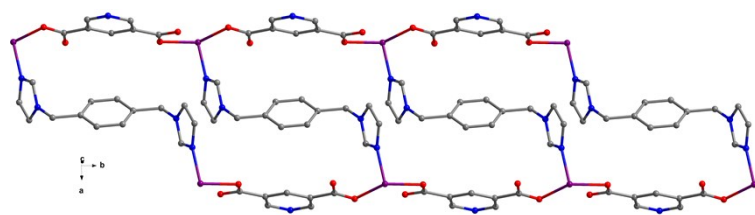


(a)

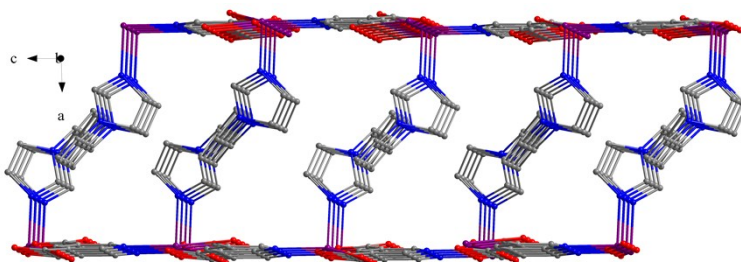


(b)

Fig. S2 (a) 2D network of complex **1** in ab-plane; (b) 3D network of **1**

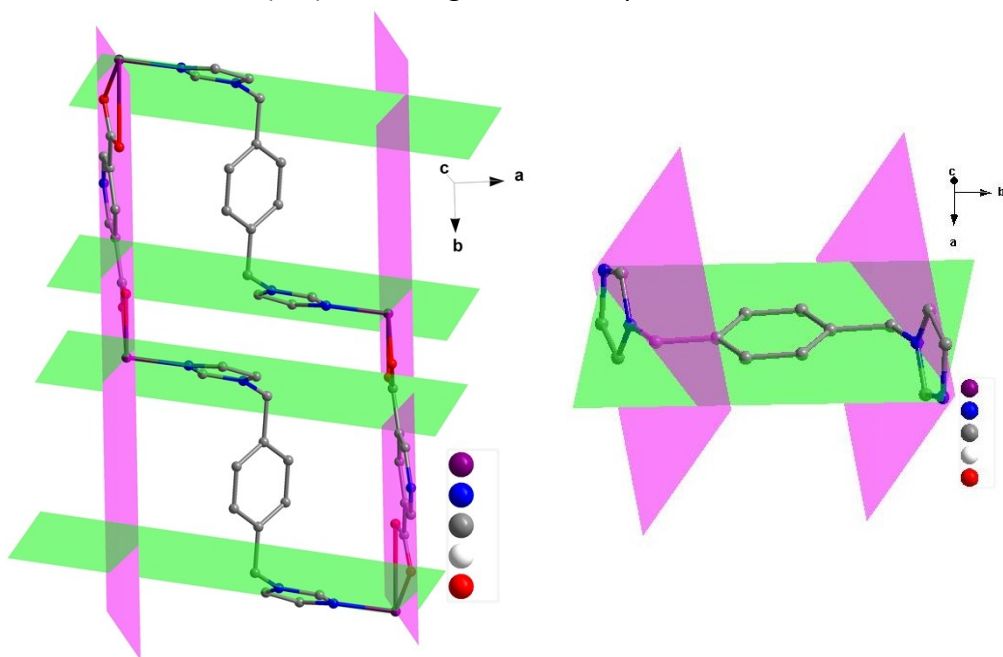


(a)



(b)

Fig. S3 (a) The 1D ladder chain structure of **2**; (b) the 1D double layer structure with (4,4) rhombic grid in the *ac* plane of **2**



(a)

(b)

Fig.S4 The different types of planes for (4,4) rhombic grid (a) and 1,2-BIYB ligands (b)

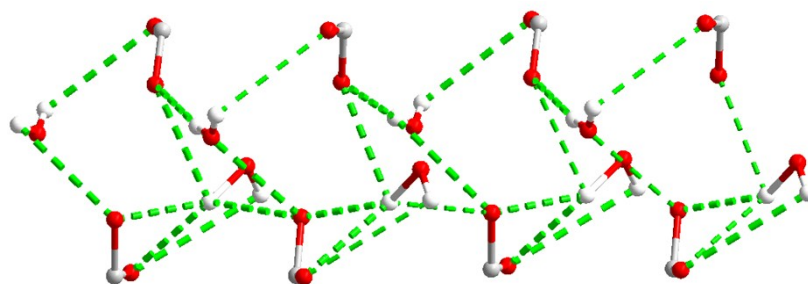


Fig.S5 The hydrogen bonding interactions in **2**

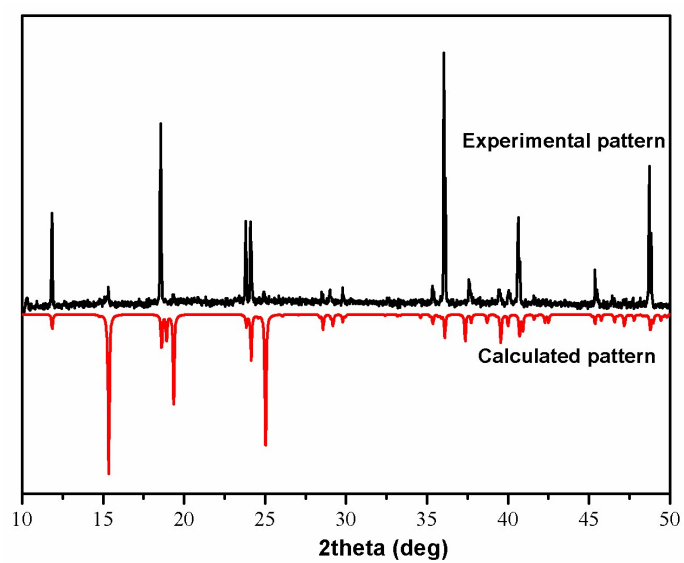
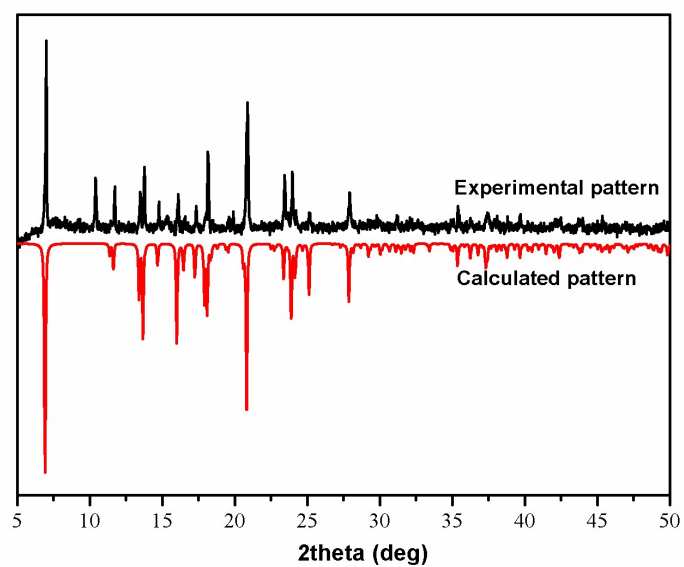


Fig. S6. PXRD patterns for compound 1 and 2

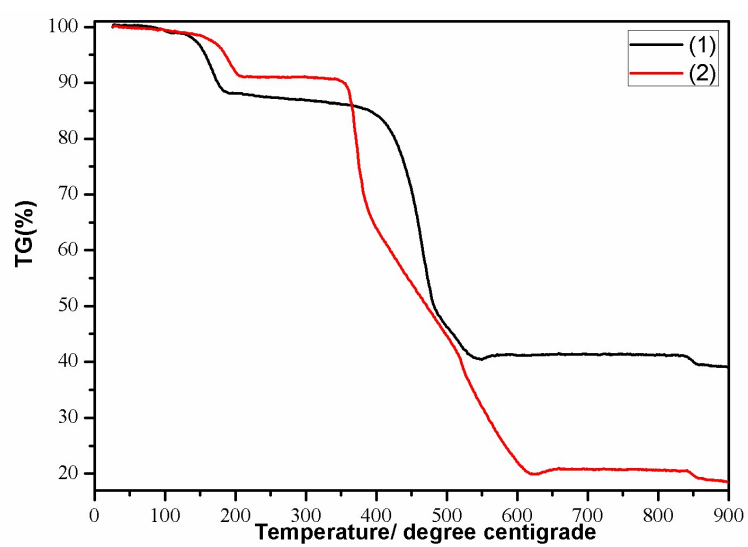


Fig.S7. The thermal stabilities of **1** and **2**

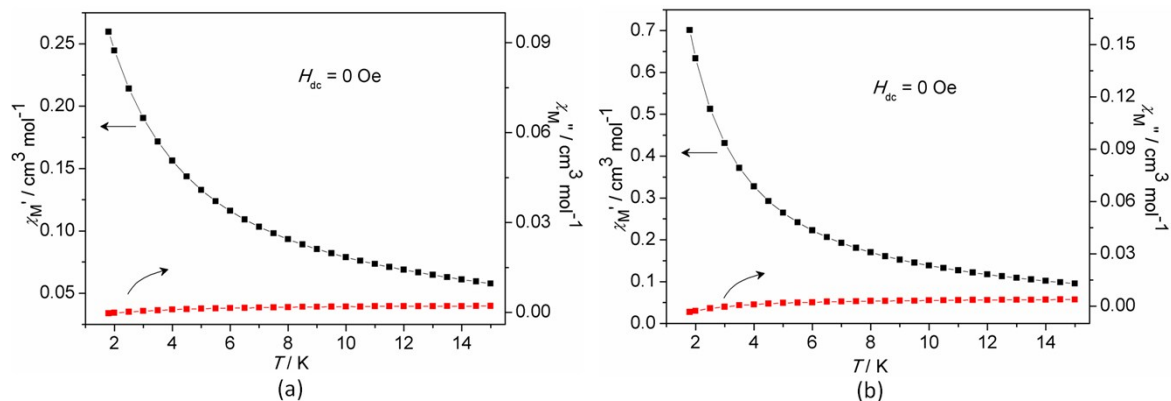


Fig. S8. Temperature dependence of the in-phase (χ') and out-of-phase (χ'') of the ac susceptibilities for **1** (a) and **2** (b) under zero dc field.

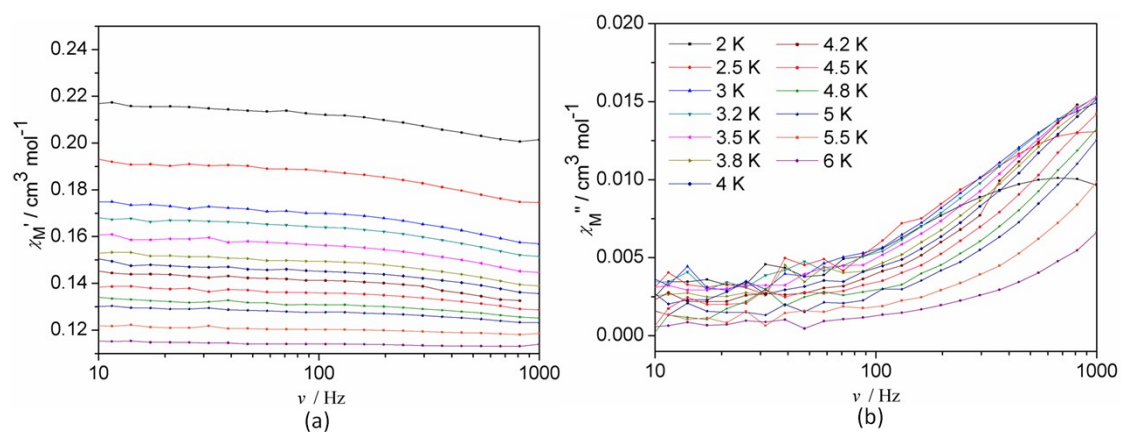


Fig. S9 Frequency dependence of χ' and χ'' susceptibilities for **1** at applied DC fields of 3000 Oe.

Table S1 Crystal data and structure refinement summary for **2**

Compound	2		
<i>T</i> /K	293(2)	<i>r</i> (°)	90.00
Empirical formula	C ₁₄ H ₁₄ N ₃ O ₆ Co	<i>V</i> / Å ³	1514.7(5)
Formula weight	379.21	<i>D_c</i> / Mg•m ⁻³	1.663
Crystal system	monoclinic	<i>F</i> (000)	776
Space group	P2(1)/c	Reflections collected	8955
<i>a</i> / Å	12.804(2)	Data/restraints/parameter	3638/229/
		<i>s</i>	4
<i>b</i> / Å	9.8015(17)	<i>Z</i>	4
<i>c</i> / Å	12.078(2)	final <i>R</i>	0.0372
<i>α</i> (°)	90.00	<i>w_R</i> indices(all data)	0.0768
<i>β</i> (°)	92.057(3)	GOF on <i>F</i> ²	1.03

Table S2 Selected bond lengths (Å) and angles (°) for **2**

Compound	2				
Co1-O1#1	2.0464(17)	Co1-O3#2	2.0630(18)	Co1-N1	2.077(2)
Co1-O5	2.102(2)	Co1- N2	2.1349(19)	O1#1-Co1- N1	90.72(8)
O3#2-Co1-N1	98.19(9)	O1#1-Co1-O3	130.71(7)	O1#1-Co1-O5	84.78(8)
O3#2-Co1-O5	87.58(9)	N1-Co1-O5	174.16(9)	O1#1-Co1-N2	135.84(7)
O3#2-Co1-N2	91.91(7)	N1-Co1-N2	94.24(8)	O5-Co1-N2	86.46(8)

Symmetry codes: #1: x, 1/2-y, 1/2+z; #2: x, 3/2-y, -1/2+z;

Table S3 The geometrical parameters of all hydrogen bonds for **2**

D-H...A	d (D-H)/Å	d (H...A)/ Å	d (D-H...A)/Å	∠DHA/°
O(5)-H(5)A...O(1)#1	0.85(2)	2.01(2)	2.824(3)	160(2)
O(5)-H(5)B...O(3)#2	0.86(3)	1.92(3)	2.762(3)	168(3)
O(6)-H(6)A...O(3)#3	0.87(4)	2.06(4)	2.858(3)	153(4)
O(6)-H(6)A...O(1)	0.86(4)	2.07(4)	2.923(4)	168(4)

Symmetry code: #1: -1+x, y, 1+z; #2: 1-x, y, -z; #3: x, 1+y, z.

Table S4. Selected magneto-structural data for some 2D Co²⁺ compounds

L.p	Compound	Intralayer Co... Co distances/ Å	Magnetic properties	Re f
1	[Co(dmpzm)($\mu_{1,5}$ -dca) ₂] _n	7.4900 8.7527	isolated cations ma	1
2	[Co ₂ ($\mu_{1,5}$ -dca) ₄ (4-cypy) ₄] _n	7.745 8.037 8.349 8.404	AF (<i>J</i> = -0.80 cm ⁻¹)	2
3	[Co($\mu_{1,5}$ -dca) ₂ (3-cypy) ₂] _n	8.194	AF (<i>J</i> = -1.18 cm ⁻¹)	2
4	[Co(btrm) ₂ ($\mu_{1,5}$ -dca)]ClO ₄	8.539	AF(<i>J</i> not given)	3
5	[Co($\mu_{1,5}$ -dca) ₂ (dmdpy)] _n	8.102	AF (<i>J</i> = -0.08 cm ⁻¹)	4
6	[Co(azbbpy)(4,4'-bipy) _{0.5} (DMF)(NCS) ₂]·MeOH	13.92	magnetically isolated cations	5
7	[Co(azbbpy)(bpe) _{0.5} (DMF)(NCS) ₂]·0.25H ₂ O	13.67	magnetically isolated cations	5
8	[Co(atz) ₂ (dca) ₂] _n	8.041	magnetically isolated cations	6
9	[Co(ppad) ₂] _n	8.302	magnetically isolated cations	7
10	[Co(tbta)N ₃]ClO ₄ ·3CH ₃ CN	8.186(1)	AF (<i>J</i> = -0.48 cm ⁻¹)	8
11	[{ArNdCMe} ₂ (NPh)]Co(NCS) ₂	8.67		9
12	[{ArNdCMe} ₂ (NPh)]Co(NCS) ₂	9.96		9

Table S5. The best results fitted for **2** under 1500 Oe dc field by a generalized Debye model

<i>T</i> (K)	χ_T	χ_S	α
2.0	0.54	0.19	0.32
2.2	0.50	0.17	0.31
2.5	0.46	0.17	0.28
2.8	0.42	0.15	0.27
3.0	0.40	0.15	0.24
3.2	0.38	0.15	0.21
3.5	0.35	0.14	0.18
3.8	0.33	0.12	0.19
4.0	0.31	0.12	0.15

References

- (1)Q. Y. Li, W. H. Zhang, H. X. Li, X. Y. Tang, J. P. Lang, Y. Zhang, X. Y. Wang, S. Gao, *Chin. J. Chem.* 2006, **24**, 1716.
- (2)M. Du, Q. Wang, Y. Wang, X. J. Zhao, J. J. Ribas, *Solid State Chem.* 2006, **179**, 3926.
- (3)X. Y. Chen, P. Cheng, S. P. Yan, D. Z. Liao, Z. H. Jiang, *Z. Anorg. Allg. Chem.* 2005, **631**, 3104.
- (4)B. L. Livia, C. C. Charlane, P. G. Guilherme, G. F. V. Maria, D. Renata, C. M. Flavia , *POLYHEDRON*, 2013, **50**, 16.
- (5)A. E. Ion, S. Nica, A. M. Madalan, S. Shova, J. Vallejo, M. Julve, F. Lloret, M. Andruh, *Inorg. Chem.* 2015, **54**, 16.
- (6)P. G. Joanna, K. Tomasz , M. Barbara , V. Julia, L. Francesc, J. Miguel, *Dalton.Trans.* 2015, **44**, 2989.
- (7)X. Y. Liu, S. Lin, I. H. Zhou, P. P. Cen, X. Y. Jin, G. Xie, S. P. Chen, Q. L. Hu, *Inorg. Chem.* 2015, **54**, 8884.
- (8)S. David, G. M. Sommer, A. Mihail, D. Serhiy, H. Stephan, M. Franc, N. Frank, S. Biprajit, *J. Am. Chem. Soc.* 2015, **137**, 1993.
- [9]T. Jurca, A. Farghal, P. H. Lin, I. Korobkov, M. Murugesu, D. S. Richeson, *J. Am. Chem. Soc.* 2011, **133**, 15814.