

Supporting Information

Theoretical insight into the catalytic activities of the oxygen reduction reaction on the transition metal-N₄ doped graphene

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Table S1. The free energy changes of ΔG_x ($x=1-4$) in ORR on M-N₄-C

	ΔG_1	ΔG_2	ΔG_3	ΔG_4
Cr	-3.33	-1.56	-0.07	0.04
Mn	-1.21	-2.37	-0.79	-0.55
Fe	-1.88	-2.16	-0.74	-0.14
Co	-0.83	-1.39	-1.59	-1.11
Ni	0.003	-0.95	-1.91	-2.06
Ru	-1.91	-2.36	-0.62	-0.03
Rh	-0.79	-1.42	-1.57	-1.14
Pd	0.30	-0.63	-2.16	-2.43
Os	-2.40	-2.61	0.007	0.08
Ir	-0.78	-1.82	-1.16	-1.16
Pt	0.12	-0.66	-1.90	-2.48

Table S2. Bader charge on metal, nitrogen and the ten nearest carbon atoms (C-10) surrounding the M-N₄ moiety. NC is the sum of the MN₄ and C-10.

	Metal	N ₄	C-10	NC
Cr	1.25	-4.80	3.37	-0.18
Mn	1.28	-5.09	3.42	-0.39
Fe	1.00	-4.99	3.42	-0.57
Co	0.84	-4.94	3.44	-0.66
Ni	0.82	-4.93	3.46	-0.65
Mo	1.27	-5.17	3.55	-0.35
Tc	1.24	-5.03	3.35	-0.44
Ru	0.90	-4.89	2.97	-1.02
Rh	0.61	-4.84	3.10	-1.13
Pd	0.70	-4.80	3.06	-1.04
Os	0.93	-4.71	2.73	-1.05
Ir	0.73	-4.67	2.88	-1.06
Pt	0.72	-4.75	2.99	-1.04

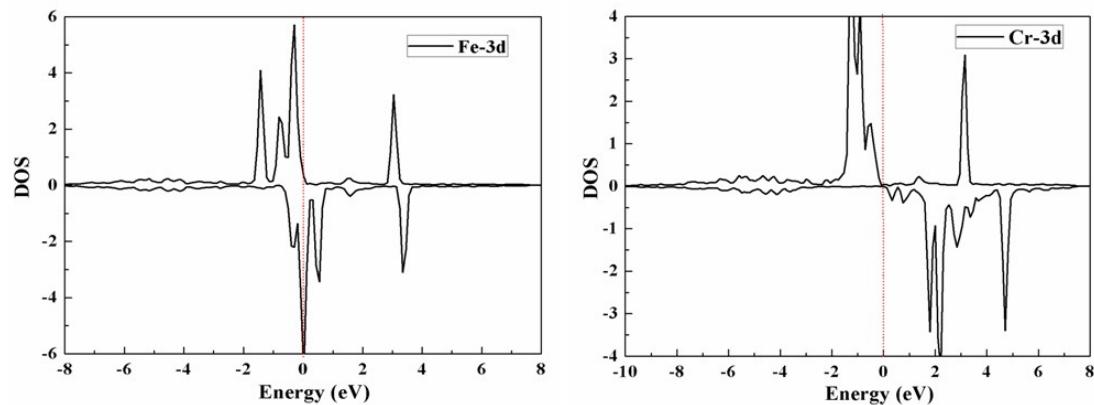


Fig. S1. Density of states for Fe-N₄-C and Cr-N₄-C. The vertical red line is the Fermi Level.