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Electronic Supporting Information

Heterogeneous Microwave-assisted Ullmann type Methodology for Synthesis of

Rigid-core Ionic Liquid Crystals.

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CCDC 1830443-1830446

ΘBr

Summary of Data CCDC 1830443

Compound Name: Formula: C22 H35 N2 1+,Br1 1-Unit Cell Parameters: a 20.4743(10) b 14.5386(6) c 7.5277(4) P21/c



Summary of Data CCDC 1830444

Compound Name: Formula: C21 H33 N2 1+,F6 P1 1-Unit Cell Parameters: a 8.3458(8) b 10.7404(11) c 26.252(3) P-1

Summary of Data CCDC 1830445

Compound Name: Formula: C22 H35 N2 1+,Br1 1-Unit Cell Parameters: a 19.1863(15) b 14.2196(12) c 8.1410(6) P21/c

ΘBI

Summary of Data CCDC 1830446

Compound Name: Formula: C33 H57 N2 O1 1+,Br1 1-Unit Cell Parameters: a 28.2048(7) b 14.7269(4) c 7.9430(2) P21/c

X-ray crystallography

The crystals were placed in oil, and a single crystal was selected, mounted on a glass fibre and placed in a low-temperature N_2 stream.

For compounds **1Br** and **3Br**, X-ray diffraction data collection was carried out on a Nonius Kappa-CCD diffractometer equipped with an Oxford Cryosystem liquid N₂ device, using Mo-K₀ radiation ($\lambda = 0.71073$ Å). The crystal-detector distance was 36mm. The cell parameters were determined (Denzo software) [1] from reflections taken from one set of 10 frames (1.0° steps in phi angle), each at 20s exposure. The structures were solved by Direct methods using the program SHELXS-97 [2]. The refinement and all further calculations were carried out using SHELXL-2014 [3]. The H-atoms were included in calculated positions and treated as riding atoms using SHELXL default parameters. The non-H atoms were refined anisotropically, using weighted full-matrix least-squares on F2. For **3Br**, a semi-empirical absorption correction was applied using the MULscanABS routine in PLATON [4]; transmission factors: $T_{min}/T_{max} = 0.58705/0.75162$.

For salts **2PF6** and **4Br**, X-ray diffraction data collection was carried out on a Bruker APEX II DUO Kappa-CCD diffractometer equipped with an Oxford Cryosystem liquid N₂ device, using Mo-Ka radiation ($\lambda = 0.71073$ Å). The crystal-detector distance was 38mm. The cell parameters were determined (APEX2 software) [5] from reflections taken from tree sets of 12 frames, each at 10s exposure. The structure was solved by Direct methods using the program SHELXS-97 [2]. The refinement and all further calculations were carried out using SHELXL-2014 [3]. The H-atoms were included in calculated positions and treated as riding atoms using SHELXL default parameters. The non-H atoms were refined anisotropically, using weighted full-matrix least-squares on F2. A semi-empirical absorption correction was applied using SADABS in APEX2 [5]; transmission factors for **2PF6**: $T_{min}/T_{max} = 0.6861/0.7458$; for **4Br**: $T_{min}/T_{max} = 0.4853/0.9119$.

- [1] "Kappa CCD Operation Manual", Nonius B. V., Ed. ; Delft: The Netherlands, 1997.
- [2] G. M. Sheldrick, Acta Cryst. 1990, A46, 467-473.
- [3] G. M. Sheldrick, Acta Cryst. 2015, C71, 3-8.
- [4] Spek, A.L. (2003), J.Appl.Cryst. 36, 7-13.
- [5] "M86-E01078 APEX2 User Manual", Bruker AXS Inc., Madison, USA, 2006.



Fig X1 : ORTEP views and a packing view of compound 2_PF₆. The ellipsoids enclose 90% of the electronic density. Selected weak interactions are represented with arrow (CH- π), dashed lines (π - π) or red lines (hydrogen bonds).



Fig X2 : ORTEP view and a packing view of compound 4_Br. The ellipsoids enclose 90% of the electronic density. Selected weak interactions (CH- π) are represented with arrows.



Fig X3 : ORTEP view and a packing view of compound 3_Br. The ellipsoids enclose 90% of the electronic density. Selected weak interactions (CH- π) are represented with arrows.



Fig X4 : ORTEP view and a packing view of compound 1_Br. The ellipsoids enclose 90% of the electronic density. Selected weak interactions (CH- π) are represented with arrows.

References for crystal structure views:

ORTEP views : E. Dowty, ATOMS, Shapesoftware, Kingsport, TN 37663 USA

Packing views : CHIMERA, J Comput Chem. 2004 Oct;25(13):1605-12