## A new model of metalloantibiotic: Synthesis, structure and biological activity of a zinc(II) mononuclear complex carrying two enrofloxacin and sulfadiazine antibiotics

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## SINGLE-CRYSTAL X-RAY DIFFRACTION CHARACTERIZATIONS

 Table SI.1. Single-crystal X-ray diffraction data and crystal structure refinement results for complexes

 (1) and (3).

	(1)	(3)	
Formula	$C_{20}H_{23}N_9O_5S_2Zn_1$	$C_{48}H_{53}Cl_1F_2N_{10}O_{12}S_1Zn_1$	
Molecular weight (g.mol <sup>-1</sup> )	598.97	1132.90	
Temperature (K)	293	293	
Crystal system	Orthorhombic	Triclinic	
Space group	$Pna2_1$	P-1	
Flack parameter	0.11(2)	-	
Crystal shape	prismatic	needle	
Crystal color	colorless	colorless	
Crystal size (mm <sup>3</sup> )	0.11×0.11×0.11	0.09×0.35×0.50	
Density	1.591	1.540	
$\mu$ (mm <sup>-1</sup> )	1.201	0.683	
<b>a</b> (Å)	13.898(2)	12.6714(6)	
<b>b</b> (Å)	12.655(3)	14.1042(7)	
<b>c</b> (Å)	14.221(3)	15.0668(7)	
<b>α</b> (deg.)	90	68.569(5)	
<b>β</b> (deg.)	90	81.395(4)	
γ (deg.)	90	77.993(4)	
$V(Å^3)$	2501.0(9)	2443.7(2)	
Z	4	2	
No. refl. / unique refl. / R <sub>int</sub>	7690 / 4563 / 0.029	23105 / 11457 / 0.039	
$\mathbf{R}(\mathbf{F}) / \mathbf{R}_{\mathbf{w}}(\mathbf{F})$	0.0508 / 0.0660	0.0966 / 0.0938	
S	1.06	1.02	
No. refl. used	3458	6733	
No. refined parameters	335	677	
Electronic residue (e <sup>-</sup> .Å <sup>-3</sup> )	-0.82 / +0.32	-2.20/+ 1.41	
Absorption correction	Analytical	Analytical	

Table SI.2. Important bond lengths (Å), angles (°) and distances (Å) within structure of complex (1).

_	<b>1</b>	<u> </u>			/	
	Zn1-N7	2.185(5)	Zn1-N21	2.081(4)	Zn1-N27	2.738(5)
	Zn1-N41	2.049(5)	Zn1-O42	2.034(6)	Zn1-N1	2.413(5)
	N1-Zn1-N7	58.4(2)	N1-Zn1-N21	89.7(2)	N7-Zn1-N21	97.4(2)
	N1-Zn1-N27	143.3(2)	N7-Zn1-N27	124.9(2)	N21-Zn1-N27	54.0(2)
	N1-Zn1-N41	138.7(2)	N7-Zn1-N41	93.2(2)	N21-Zn1-N41	126.0(2)
	N27-Zn1-N41	76.7(2)	N1-Zn1-O42	87.0(2)	N7-Zn1-O42	136.6(2)
	N21-Zn1-O42	108.6(2)	N27-Zn1-O42	98.5(2)	N41-Zn1-O42	98.4(2)



Figure SI.1. Complex (1) with important labels.

**Table SI.3.** Important bond lengths (Å), angles (°) and distances (Å) within structure model of complex (2).

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Zn1-O2	2.118(2)	Zn1-O2	2.121(2)	Zn1-O10	2.044(2)
O2-Zn1-O2	84.47(9)	O2-Zn1-O10	85.51(8)	O2-Zn1-O10	95.96(8)
 O2-Zn1-O10	169.88(8)	O10-Zn1-O10	94.16(8)		



Figure SI.2. Structure model of complex (2) with important labels.

Table SI.4. Important bond leng	ths (Å), angles (°)	and distances (Å)	) within structure of com	plex (3).
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-	Zn1-O2	2.124(4)	Zn1-O10	1.983(4)	Zn1-O29	2.006(5)
	Zn1-O50	2.052(4)	Zn1-N55	2.045(5)		
	O2-Zn1-O10	87.2(2)	O2-Zn1-O29	90.4(2)	O10-Zn1-O29	107.6(2)
	O2-Zn1-O50	175.3(2)	O10-Zn1-O50	89.1(2)	O29-Zn1-O50	88.0(2)
	O2-Zn1-N55	85.4(2)	O10-Zn1-N55	145.0(2)	O29-Zn1-N55	106.6(2)
	O50-Zn1-N55	99.3(2)				



Figure SI.3. Complex (3) with important label

**POWDER DIFFRACTION CHARACTERIZATIONS** 



**Figure SI-4**. Powder diffraction pattern highlighting sample purity of complex (1). Red line : observed pattern, blue line: calculated pattern.



**Figure SI-5**. Powder diffraction pattern highlighting sample purity of complex (2). Red line : observed pattern, blue line: calculated pattern.



**Figure SI-6**. Powder diffraction pattern highlighting sample purity of complex (3). Red line : observed pattern, blue line: calculated pattern.

Table SI–5 : <sup>1</sup>H NMR data (600 MHz, DMSO-d<sub>6</sub>)

Sulfadiazine :  $\delta$  = 5.93 (s, 2H),  $\delta$  = 6.56 (d, J = 8.7 Hz, 2H),  $\delta$  = 6.99 (t, J = 5 Hz, 1H),  $\delta$  = 7.61 (d, J = 8.7 Hz, 2H),  $\delta$  = 8.47 (d, J = 5 Hz, 2H),  $\delta$  = 11.15 (s, 1H)

Enrofloxacin :  $\delta = 1.05$  (t, J = 7.2 Hz, 3H),  $\delta = 1.18$  (m, 2H),  $\delta = 1.31$  (m, 2H),  $\delta = 2.41$  (q, J = 7.2 Hz, 2H),  $\delta = 2.58$  (m, 4H),  $\delta = 3.34$  (m, 4H),  $\delta = 3.82$  (m, 1H),  $\delta = 7.56$  (d, J = 7.4 Hz, 1H),  $\delta = 7.89$  (d, J = 13.5 Hz, 1H),  $\delta = 8.65$  (s, 1H) ,  $\delta = 15.16$  (s, 1H)

(2):  $\delta = 1.06$  (t, J = 7.2 Hz, 3H),  $\delta = 1.12$  (s, 2H),  $\delta = 1.31$  (m, 2H),  $\delta = 2.67$  (s, 4H),  $\delta = 3.35$  (s, 4H),  $\delta = 3.79$  (m, 1H),  $\delta = 7.53$  (d, J = 5 Hz, 1H),  $\delta = 7.89$  (d, J = 13.5 Hz, 1H),  $\delta = 8.67$  (s, 1H),  $\delta = 8.81$  (s, 1H)

(3) :  $\delta = 1.05$  (t, J = 7.2 Hz, 6H),  $\delta = 1.13$  (m, 2H),  $\delta = 1.19$  (m, 2H),  $\delta = 1.24$  (m, 2H),  $\delta = 1.33$  (m, 4H),  $\delta = 3.33$  (m, 8H),  $\delta = 3.82$  (m, 2H),  $\delta = 5.92$  (s, 2.7H),  $\delta = 6.56$  (d, J = 8.7 Hz, 3.4H),  $\delta = 6.99$  (s, 1.8H),  $\delta = 7.54$  (d, J = 7.2 Hz, 1H),  $\delta = 7.56$  (d, J = 7.2 Hz, 1H),  $\delta = 7.61$  (d, J = 8.5 Hz, 3H),  $\delta = 7.88$  (d, J = 13.5 Hz, 1H),  $\delta = 7.91$  (d, J = 13.5 Hz, 1H),  $\delta = 8.47$  (d, J = 4.5 Hz, 3H), 8.67 (s, 1H),  $\delta = 8.87$  (s, 1H),  $\delta = 11.15$  (s, 1H),  $\delta = 15.16$  (s, 1H)

<sup>19</sup>F NMR (400 MHz, DMSO-d<sub>6</sub>) Enrofloxacin :  $\delta = -123.7$  ppm (2) :  $\delta = -123.7$  ppm (2F),  $\delta = -124.7$  ppm (1F) (3) :  $\delta = -123.7$  ppm (1F),  $\delta = -124.6$  ppm (1F)



**Figure SI-7:** IR spectra of Enrofloxacin (**Erx**), (2) and (3) in a solid state, (A) comparison of (2) with the ligand Erx ; (B) comparison of (3) with **Erx**.