Supplementary Information for

Electrochemical synthesis of some 2-aminobenzofuran-3carbonitrile and 2-aminobenzofuran-3-carboxylate derivatives: Product diversity by changing the applied current density †

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Control experiments: We initiated our studies by using **4BP** and **1b** in aqueous media, as model substrates to identify the optimized reaction conditions. As shown in Table below, when constant current electrolysis (CCE) of **4BP** and **1b** was performed in carbonate buffer (0.2M, pH= 9.0)/ ethanol (50:50) and at the current density 0.6 mA cm⁻² (**3b**) and 1.2 mA cm⁻² (**4b**), the desired products **3b** and **4b** were isolated on the best yield 90-95% yield (entry 4). Therefore, the following optimization experiments were carried out under this condition.

1	Yield (%) 3b				
Solvent	0.3 mA/cm ²	0.6 mA/cm 2	1.0 mA/cm 2	1.2 mA/cm 2	
Phosphate buffer solution (0.2 M, pH=5.0)	20	25	Trace	Trace	
Phosphate buffer solution (0.2 M, pH=7.0)	40	60	30	Trace	
Carbonate buffer solution (0.2 M, pH=9.0)	60	75	20	20	
Carbonate buffer (0.2M, pH=9.0) / Ethanol mixture (50:50)	80	95	30	15	
	Solvent Phosphate buffer solution (0.2 M, pH=5.0) Phosphate buffer solution (0.2 M, pH=7.0) Carbonate buffer solution (0.2 M, pH=9.0) Carbonate buffer (0.2M, pH=9.0) / Ethanol mixture (50:50)	Solvent0.3 mA/cm 2Phosphate buffer solution (0.2 M, pH=5.0)20Phosphate buffer solution (0.2 M, pH=7.0)40Carbonate buffer solution (0.2 M, pH=9.0)60Carbonate buffer (0.2 M, pH=9.0) / Ethanol mixture (50:50)80	Solvent 0.3 mA/cm 2 0.6 mA/cm 2 Phosphate buffer solution (0.2 M, pH=5.0) 20 25 Phosphate buffer solution (0.2 M, pH=7.0) 40 60 Carbonate buffer solution (0.2 M, pH=9.0) 60 75 Carbonate buffer (0.2M, pH=9.0) / Ethanol mixture (50:50) 80 95	Solvent 0.3 mA/cm 2 0.6 mA/cm 2 1.0 mA/cm 2 Phosphate buffer solution (0.2 M, pH=5.0) 20 25 Trace Phosphate buffer solution (0.2 M, pH=7.0) 40 60 30 Carbonate buffer solution (0.2 M, pH=9.0) 60 75 20 Carbonate buffer (0.2M, pH=9.0) / Ethanol mixture (50:50) 80 95 30	

Table S1. Optimization for the synthesis of 3b and 4b.

	1	Yield (%) 4b				
Entry	Solvent	0.6 mA/cm 2	1.0 mA/cm ²	1.2 mA/cm 2	1.5 mA/cm ²	
1	Phosphate buffer solution (0.2 M, pH=5.0)	Trace	Trace	30	20	
2	Phosphate buffer solution (0.2 M, pH=7.0)	Trace	30	40	30	
3	Carbonate buffer solution (0.2 M, pH=9.0)	10	60	70	50	
4	Carbonate buffer (0.2M, pH=9.0) / Ethanol mixture (50:50)	20	80	90	75	



FT-IR spectrum of 3a

MS spectrum of 3a

Abundance



m/z-->



¹H NMR spectrum of 3a

Expanded ¹H NMR spectrum of 3a



¹H NMR spectrum of 3a (with D₂O)





¹³C NMR spectrum of 3a









¹H NMR spectrum of 3b



Expanded ¹H NMR spectrum of 3b



¹H NMR spectrum of 3b (with D₂O)



¹³C NMR spectrum of 3b





FT-IR spectrum of 3c





¹H NMR spectrum of 3c



Expanded ¹H NMR spectrum of 3c



¹H NMR spectrum of 3c (with D₂O)



¹³C NMR spectrum of 3c





FT-IR spectrum of 4b





¹H NMR spectrum of 4b







¹H NMR spectrum of 4b (with D₂O)



¹³C NMR spectrum of 4b



Expanded ¹³C NMR spectrum of 4b





FT-IR spectrum of 4c





¹H NMR spectrum of 4c



Expanded ¹H NMR spectrum of 4c



¹H NMR spectrum of 4c (with D₂O)









Fig. S1. The cell designed for electrochemical synthesis