

Supporting Information

Ultrasensitive Point-of-Care Testing of Arsenic Based on Catalytic Reaction of Unmodified Gold Nanoparticles

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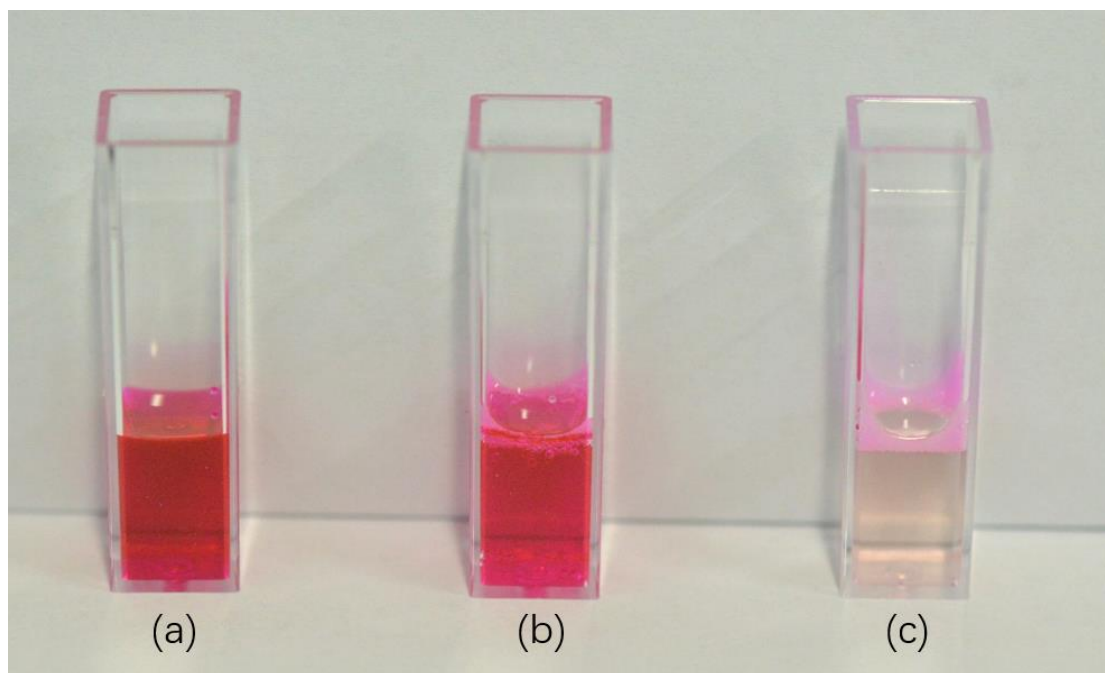


Figure S1. The contrast experiment to identify the catalytic activity of Au NPs. (a) Only Rhodamine B (1.0 mM). (b) Mixture of rhodamine B (1.0 mM) and NaBH₄ (10 mM). (c) Mixture of Au NPs (about 50 nM), rhodamine B (1.0 mM) and NaBH₄ (10 mM). The photo above was obtained in 5 minutes after mixture.

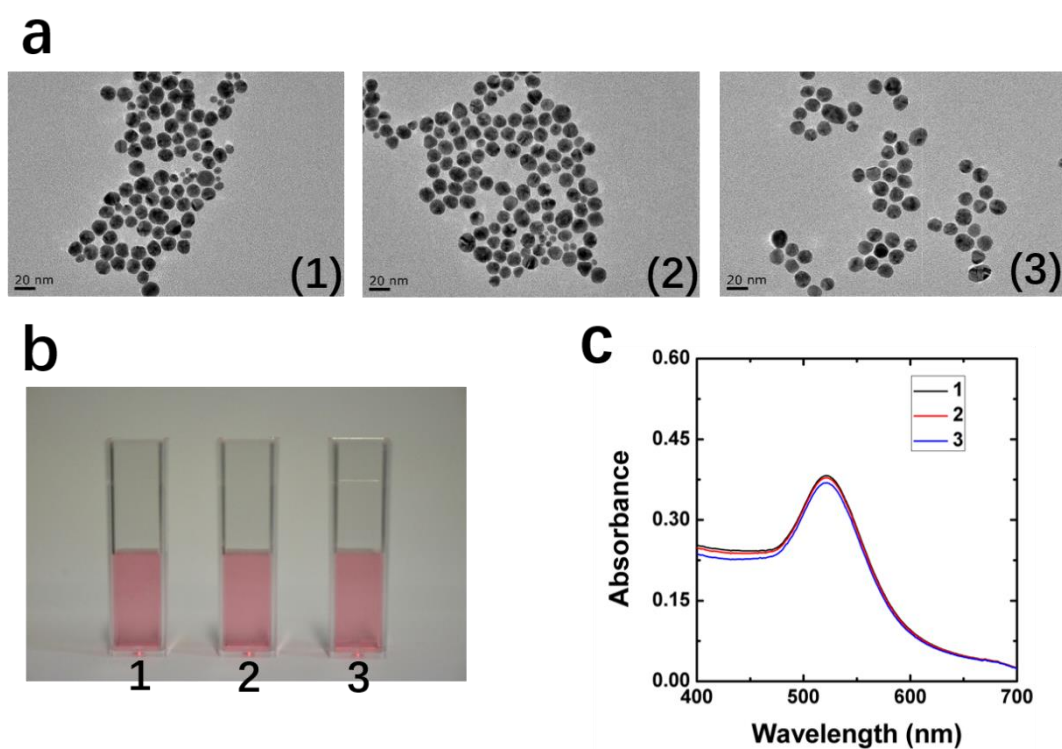


Figure S2. (a) TEM images of Au NPs solutions (1) bare Au NPs, (2) in the presence of AsO₂⁻, (3) in the presence of AsO₂⁻ and NaBH₄. (b) Visual observation and (c) UV-vis absorbance spectra of (1) bare Au NPs, (2) Au NPs + AsO₂⁻, (3) Au NPs + AsO₂⁻ + NaBH₄.

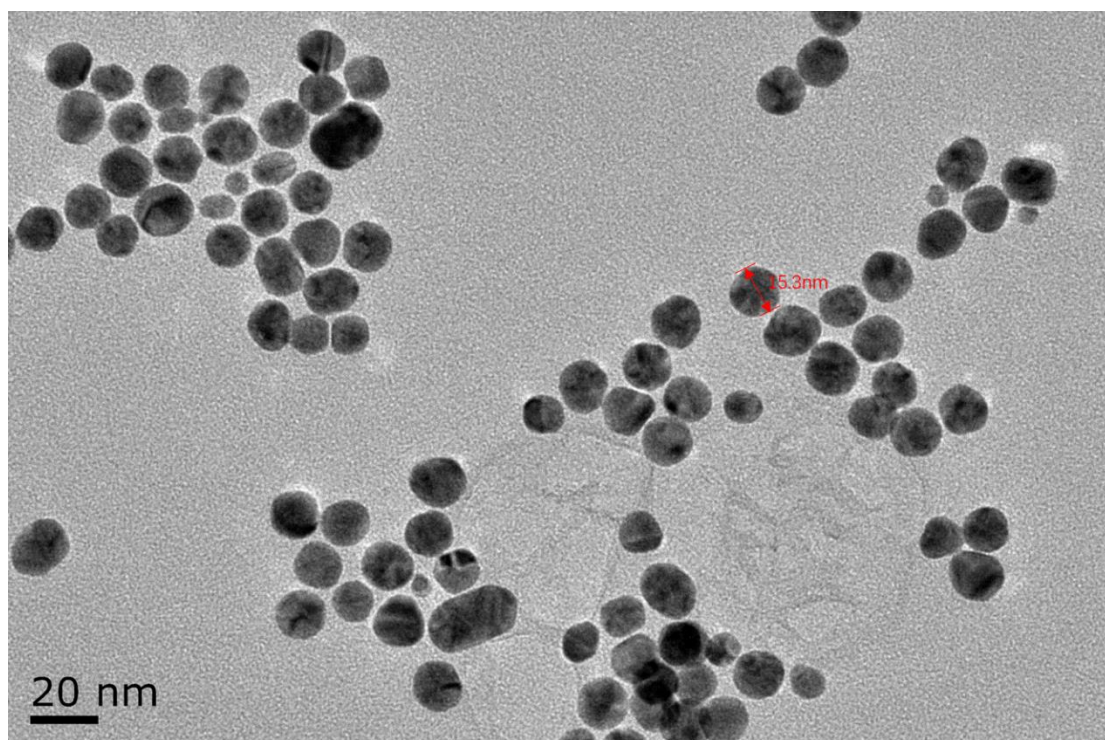


Figure S3. TEM image of bare Au NPs. The diameter of Au NPs is about 15.3 nm

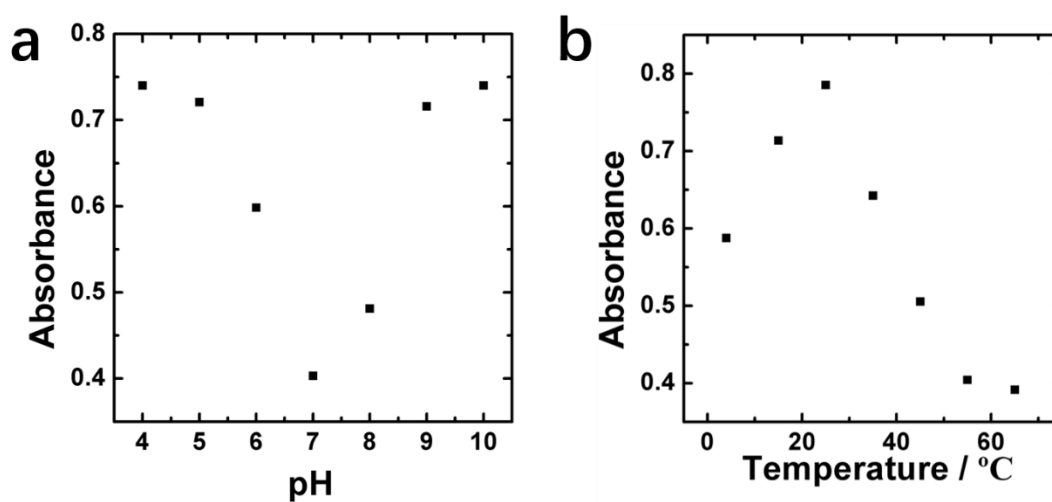


Figure S4. (a) Optimization experiment of pH using RhB-NaBH₄ solution with Au NPs (500 μ L 3.0 μ M Au NPs, 100 μ L 1.0 mM RhB and 50 μ L 10mM NaBH₄). (b) Optimization experiment of temperature using RhB-NaBH₄ solution with Au NPs (500 μ L 3.0 μ M Au NPs, 100 μ L 1.0 mM RhB and 50 μ L 10mM NaBH₄) and AsO₂⁻ (1.0 ppm, 500 μ L).

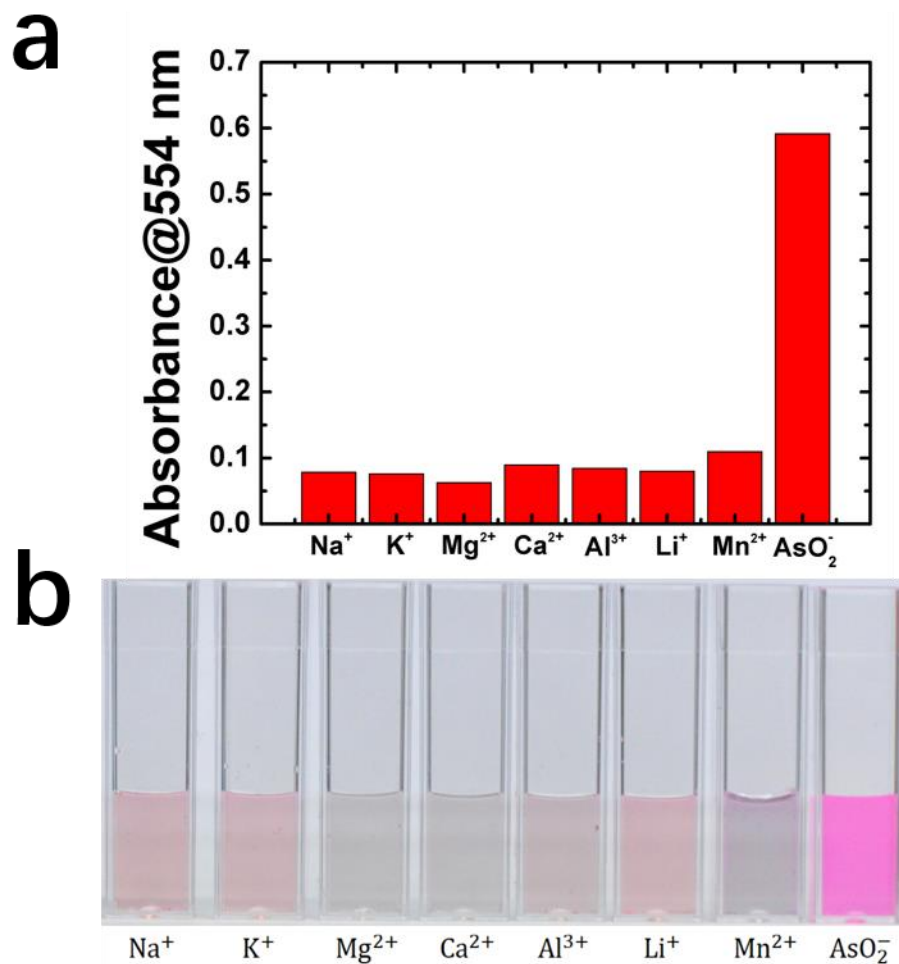


Figure S5.(a)UV-vis absorbance spectra of the RhB solution at a concentration of 50 μM in the presence of Na^+ , K^+ , Mg^{2+} , Ca^{2+} , Al^{3+} , Li^+ , Mn^{2+} and AsO_2^- (1.0 ppm for AsO_2^- and 10 ppm for other ions).(b) Photographs of Au NP-catalyzed RhB- NaBH_4 solutions in the presence of AsO_2^- and other cations.

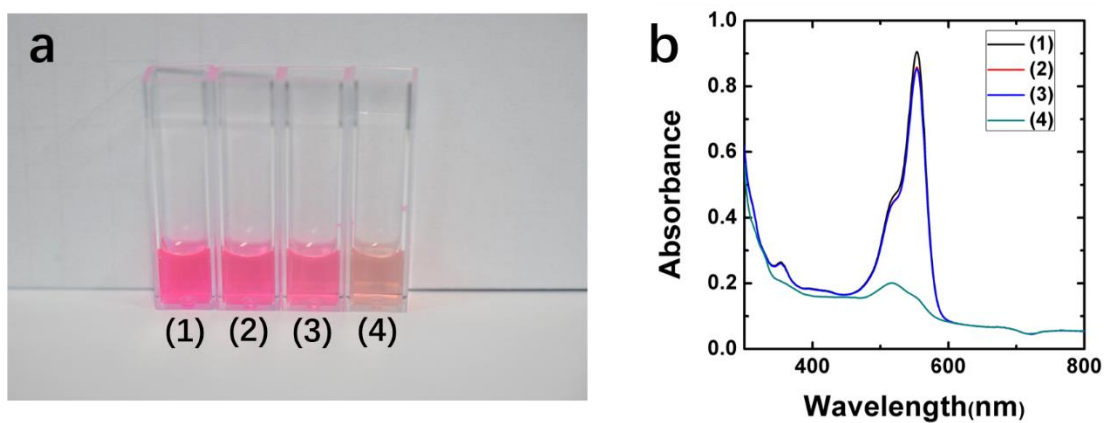


Figure S6. Visual Observation and UV-vis spectra of (1)100 ppb AsO_4^{3-} , (2)50 ppb AsO_4^{3-} and 50 ppb AsO_2^- , (3) 100 ppb AsO_2^- and (4)blank control.