## **Supporting Information**

The syntheses, crystal structures, electrochemical and magnetic properties of trinuclear cyanide-bridged complexes [*cis*- $M^{II}(bpy)_2(CN)_2]_2Mn^{III}(salcy)(PF_6)$  (M = Fe, Ru, Os)

Ming Liu,<sup>a</sup> Jianmei Hu<sup>a</sup>, Meng Yu<sup>a</sup>, Weiwei Fan<sup>a</sup>, Wenwen Ding<sup>a</sup>, Yong Wang<sup>a</sup>\*

<sup>a</sup> Hubei Key Laboratory of Drug Synthesis and Optimization, Jingchu University of

Technology, Jingmen, Hubei, 448000, P. R. China

	1	2	3
Chemical formula	C <sub>64</sub> H <sub>52</sub> F <sub>6</sub> Fe <sub>2</sub> Mn	C <sub>69</sub> H <sub>66</sub> F <sub>12</sub> MnN <sub>17</sub>	C <sub>73</sub> H <sub>75</sub> F <sub>6</sub> MnN <sub>17</sub>
	N <sub>14</sub> O <sub>2</sub> P	$O_3P_2Ru_2$	O <sub>6</sub> Os <sub>2</sub> P
Formula weight	1360.80	1728.40	1866.81
Colour and Habit	Red prism	Yellow prism	Brown prism
Crystal Size / mm	0.36×0.29×0.10	0.26×0.10×0.07	0.20×0.10×0.05
<i>T /</i> K	123	123	123
Crystal system	monoclinic	monoclinic	monoclinic
Space group	$P2_1/n$	$P2_1/n$	$P2_1/n$
<i>a</i> / Å	15.566(12)	11.906(5)	11.835(5)
<i>b</i> / Å	23.082(17)	28.346(10)	28.365(10)
<i>c</i> / Å	22.534(18)	22.673(9)	22.093(9)
$\alpha$ / deg	90.00	90.00	90.00
$\beta$ / deg	104.679(18)	96.952(10)	98.794(7)
$\gamma/\deg$	90.00	90.00	90.00
V / Å <sup>3</sup>	7832(10)	7596(5)	7330(5)
Ζ	4	4	4
$D_{\rm calc}$ / g cm <sup>-3</sup>	1.154	1.511	1.692
$\mu/\mathrm{mm}^{-1}$	0.604	0.685	3.730
Completeness	96.6%	99.5%	99.8%

**Table S1** Detail of the Crystallographic Data Collection, Structural Determination and Refinement for Complexes 1-3

F(000)	2784	3496	3704
h, k, l, range	-18≤ <i>h</i> ≤18,	-10≤ <i>h</i> ≤14,	-14≤ <i>h</i> ≤12,
	-27≤k≤25,	-33≤k≤33,	<i>-</i> 33≤ <i>k</i> ≤33,
	-26≤ <i>l</i> ≤26	-26≤ <i>l</i> ≤26	-25≤ <i>l</i> ≤26
$\theta$ range (deg)	2.07-25.00	2.04-25.00	2.10-25.00
Reflections measured	13329	13308	12879
Independent	<u>8112</u>	10922	10329
reflections	0115	10822	
R <sub>int</sub>	0.0935	0.0544	0.0854
GOF	1.019	1.094	1.027
$R_1, \ _{\omega}R_2 \left[I > 2\sigma(I)\right]$	0.1110, 0.3087	0.0850, 0.2166	0.0813, 0.1804
$R_{1, \omega}R_{2}$ (all data)	0.1308, 0.3336	0.1012, 0.2297	0.0993, 0.1924

**Table S2**. Selected Bond Length (Å) and Bond Angle (°) for Complexes 1-3 (M = Fe, 1; M = Ru,2; M = Os, 3)

	1	2	3
M1-C1	1.917(7)	1.984(7)	1.991(11)
M1-C2	1.925(9)	2.003(8)	1.992(13)
M2-C43	1.898(6)	1.994(7)	1.998(12)
M2-C44	1.915(7)	2.010(8)	1.991(14)
M1-N3	2.000(5)	2.126(6)	2.109(8)
M1-N4	1.974(6)	2.063(6)	2.058(10)
M1-N5	2.000(8)	2.108(6)	2.094(11)
M1-N6	1.970(7)	2.067(6)	2.051(10)
M2-N11	2.016(5)	2.118(5)	2.132(9)
M2-N12	1.976(5)	2.081(6)	2.054(11)
M2-N13	2.003(5)	2.092(6)	2.110(9)
M2-N14	1.987(6)	2.062(6)	2.078(10)
Mn1-N1	2.286(5)	2.278(6)	2.275(10)
Mn1-N9	2.264(5)	2.288(5)	2.280(10)
Mn1-O1	1.897(5)	1.900(5)	1.890(8)
Mn1-O2	1.902(5)	1.892(5)	1.904(9)
Mn1-N7	2.007(5)	1.992(6)	1.996(9)
Mn1-N8	1.985(5)	1.982(6)	1.982(10)
C1=N1	1.135(8)	1.166(9)	1.161(14)
C2≡N2	1.157(10)	1.168(10)	1.161(16)
C43≡N9	1.160(8)	1.161(8)	1.146(14)
C44=N10	1.173(8)	1.153(10)	1.173(17)
C1-M1-C2	88.6(3)	88.5(3)	89.3(5)
C43-M2-C44	89.6(3)	88.0(3)	90.4(5)
N1≡C1-M1	177.9(6)	175.9(6)	176.9(10)
N2≡C2-M1	179.0(7)	178.0(7)	175.7(10)
N9≡C43-M2	177.0(5)	176.3(6)	176.0(11)
N10=C44-M2	178.4(6)	177.0(6)	176.6(12)
C1≡N1-Mn1	141.6(5)	150.8(5)	149.2(9)

C43≡N9-Mn1	141.7(5)	149.3(5)	150.5(9)
O1-Mn1-O2	94.0(2)	93.0(2)	93.4(4)
N1-Mn1-N9	173.3(2)	170.4(2)	171.1(4)
M1…M2	9.95	7.94	7.93
Mn1…Mn2	12.77	12.15	11.71

**Table S3.** Data of Cyclic-Voltammetry, Cyanide Stretching Frequencies and ElectronicAbsorption Spectra for Complex 1-3 and Related Precursors.

Compound	<i>P</i> / V	$\nu_{\rm CN}$ (cm <sup>-1</sup> )	$\lambda_{\rm max},  {\rm nm}(\varepsilon,  {\rm dm}^3  {\rm mol}^{-1}  {\rm cm}^{-1})$
<i>cis</i> -Fe(bpy) <sub>2</sub> (CN) <sub>2</sub> ·3H <sub>2</sub> O	0.46	2069, 2079	382 (6252), 581 (6204)
<i>cis</i> -Ru(bpy) <sub>2</sub> (CN) <sub>2</sub> ·2H <sub>2</sub> O	0.86	2062,2078	346 (9428),483 (10385)
<i>cis</i> -Os(bpy) <sub>2</sub> (CN) <sub>2</sub> ·2H <sub>2</sub> O	0.47	2040, 2057	336         (11870), 372         (11569), 445           (10667), 496         (13074), 650         (3370)
[Mn(salcy)](ClO <sub>4</sub> )·2H <sub>2</sub> O		—	406 (3915), 475 (1387)
1	-0.41, 0.75	2090, 2098	382 (10397), 569 (8962), 610 (8023)
2	-0.42, 0.77	2051, 2071	335 (17276), 473 (13322), 493 (13120)
3	-0.13, 0.35	2031, 2059	341         (19103), 379         (17118), 407           (15893), 510         (14401), 672         (3647)



**Figure S1.** Cyclic voltammogram of complex **1** in CH<sub>3</sub>CN containing 0.1 M (Bu<sub>4</sub>N)(PF<sub>6</sub>) electrolyte. Scan rate was 100 mV/s.



**Figure S2.** Cyclic voltammogram of complex **2** in  $CH_3CN$  containing 0.1 M ( $Bu_4N$ )( $PF_6$ ) electrolyte. Scan rate was 100 mV/s.



**Figure S3.** Cyclic voltammogram of complex **3** in CH<sub>3</sub>CN containing 0.1 M (Bu<sub>4</sub>N)(PF<sub>6</sub>) electrolyte. Scan rate was 100 mV/s.



**Figure S4.** Temperature dependence of the in-phase and out-of-phase ac susceptibility for **2** in zero dc and 3 Oe ac applied field