

Supporting Information

A Novel Long-wavelength Fluorescent Probe for Selective Detection of Hydrogen Sulfide in Living Cells

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The pH stability of LC-OH

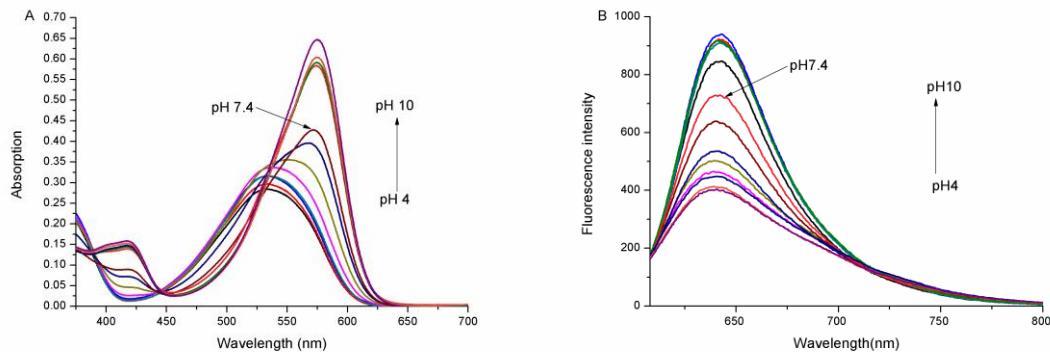


Figure S1. (A) UV absorption spectra of 10 μ M LC-OH in PBS (10 mM, pH = 4-10, 1% DMSO). (B) Fluorescence emission spectra of 10 μ M LC-OH in PBS (10 mM, pH = 4-10, 1% DMSO).

Mechanism study of LC-H₂S to H₂S

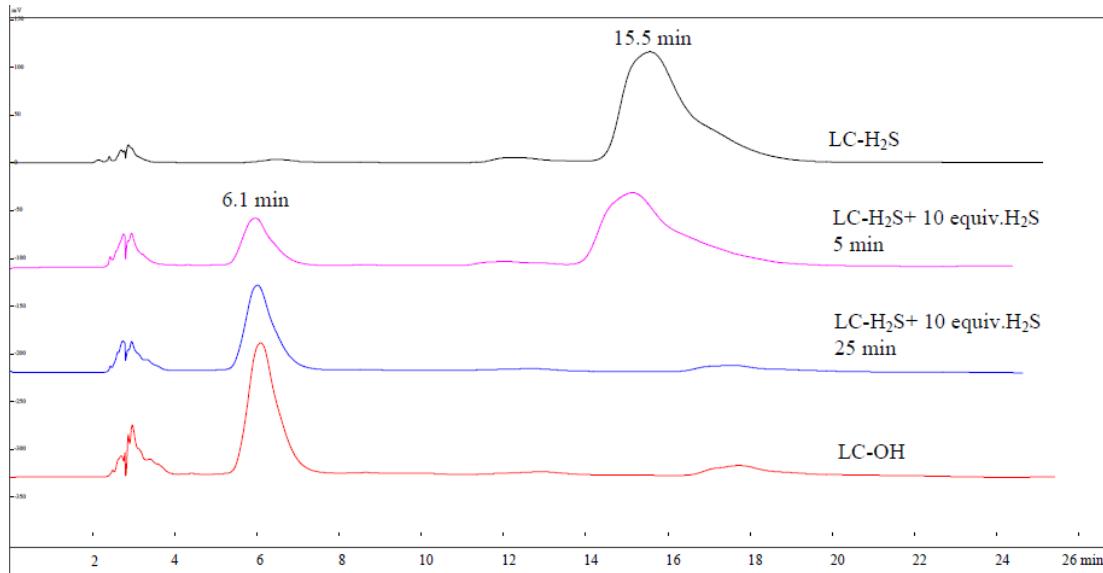
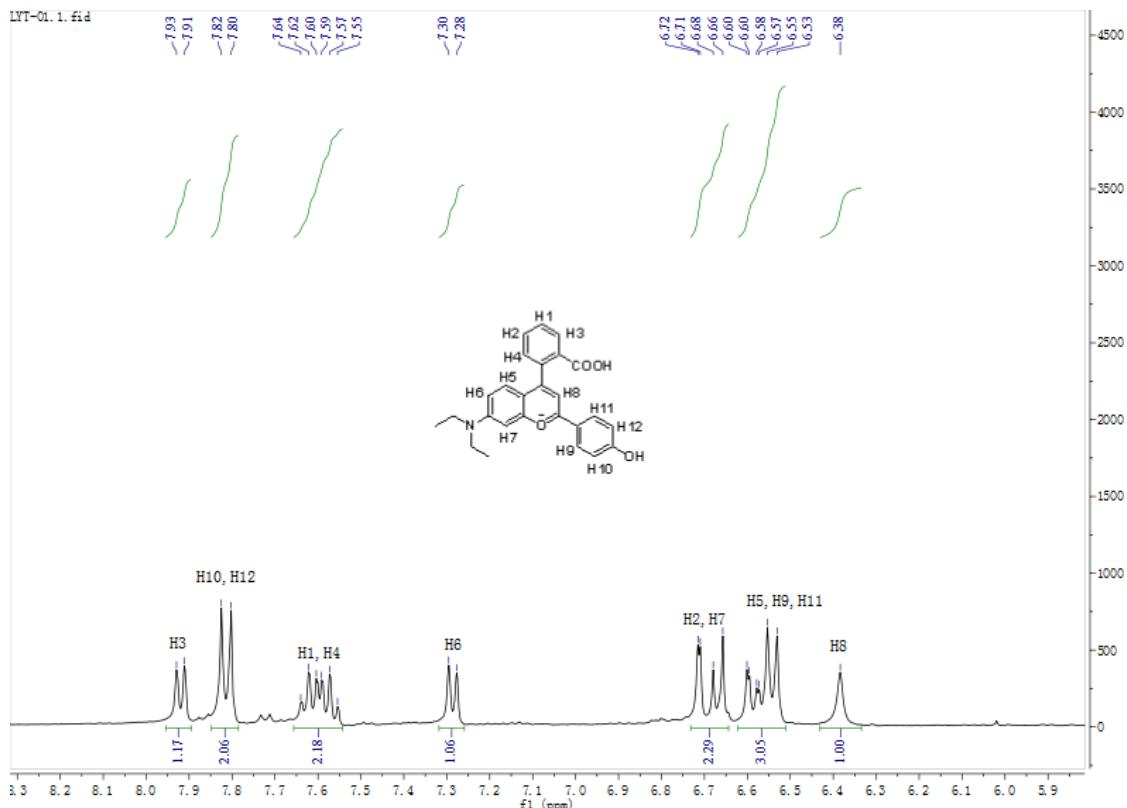


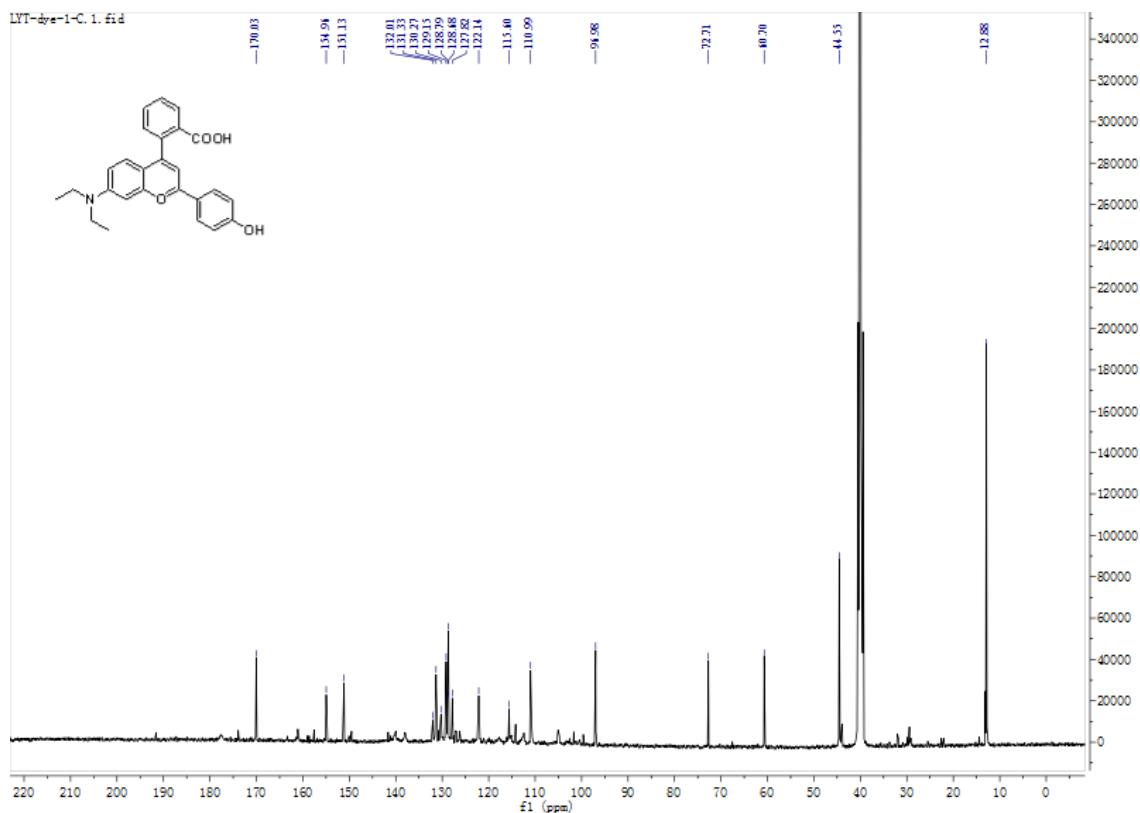
Figure S2. HPLC chromatograms: Black: **LC-H₂S**; Pink: **LC-H₂S** with 10 equiv. H₂S measured after 5 min mixing; Blue: **LC-H₂S** with 10 equiv. H₂S measured after 25 min mixing; Red: **LC-OH**. The concentration of both **LC-H₂S** and **LC-OH** are 10 μ M. Condition: column, 5 μ M C18, 250 \times 4.6 mm;

eluent, MeOH/H₂O (v/v, 7/3); flow rate, 1.0 mL/min; temperature, 25 °C; detection wavelength, 570 nm; injection volume, 5 µL.

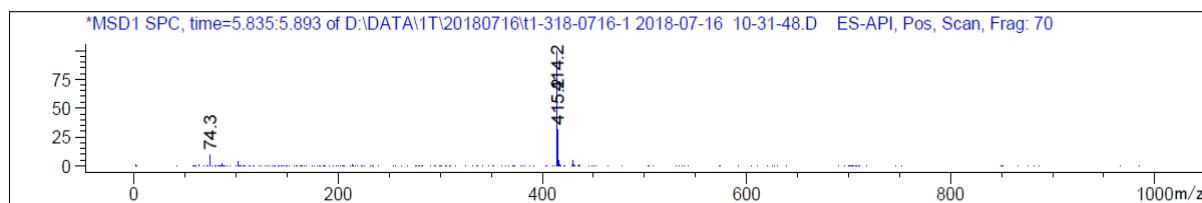
¹H-NMR Chart of LC-OH



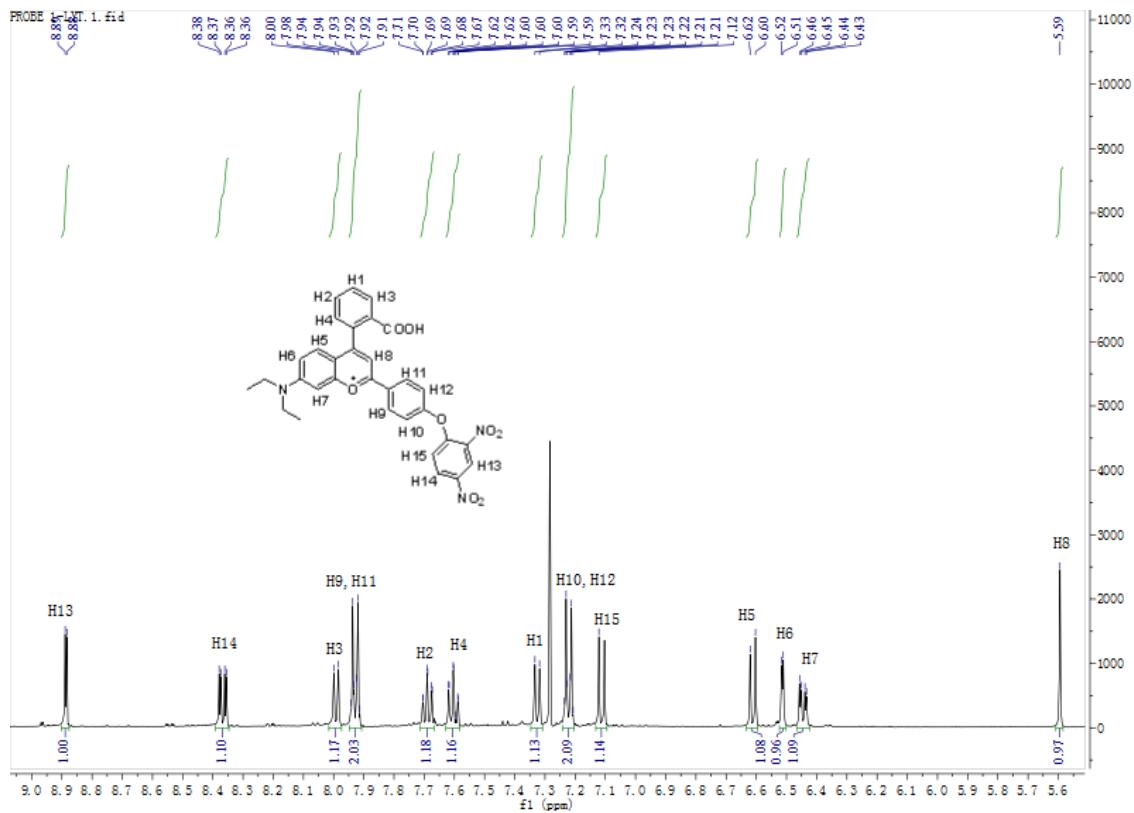
¹³C-NMR Chart of LC-OH



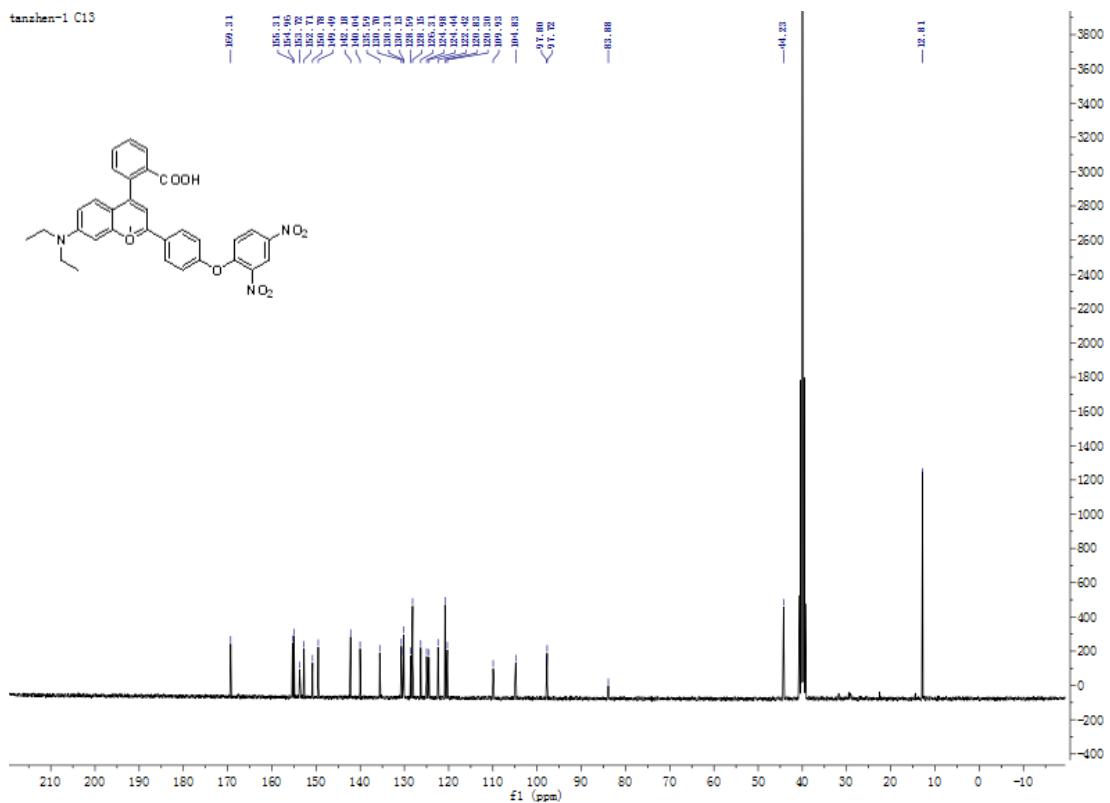
LCMS Chart of LC-OH



¹H-NMR Chart of LC-H₂S



¹³C-NMR Chart of LC-H₂S



LCMS Chart of LC-H₂S

