

Supplementary Information.

New Journal of Chemistry

A molecular roundabout: triple cycle-arranged hydrogen bonds in light of experiment and theory

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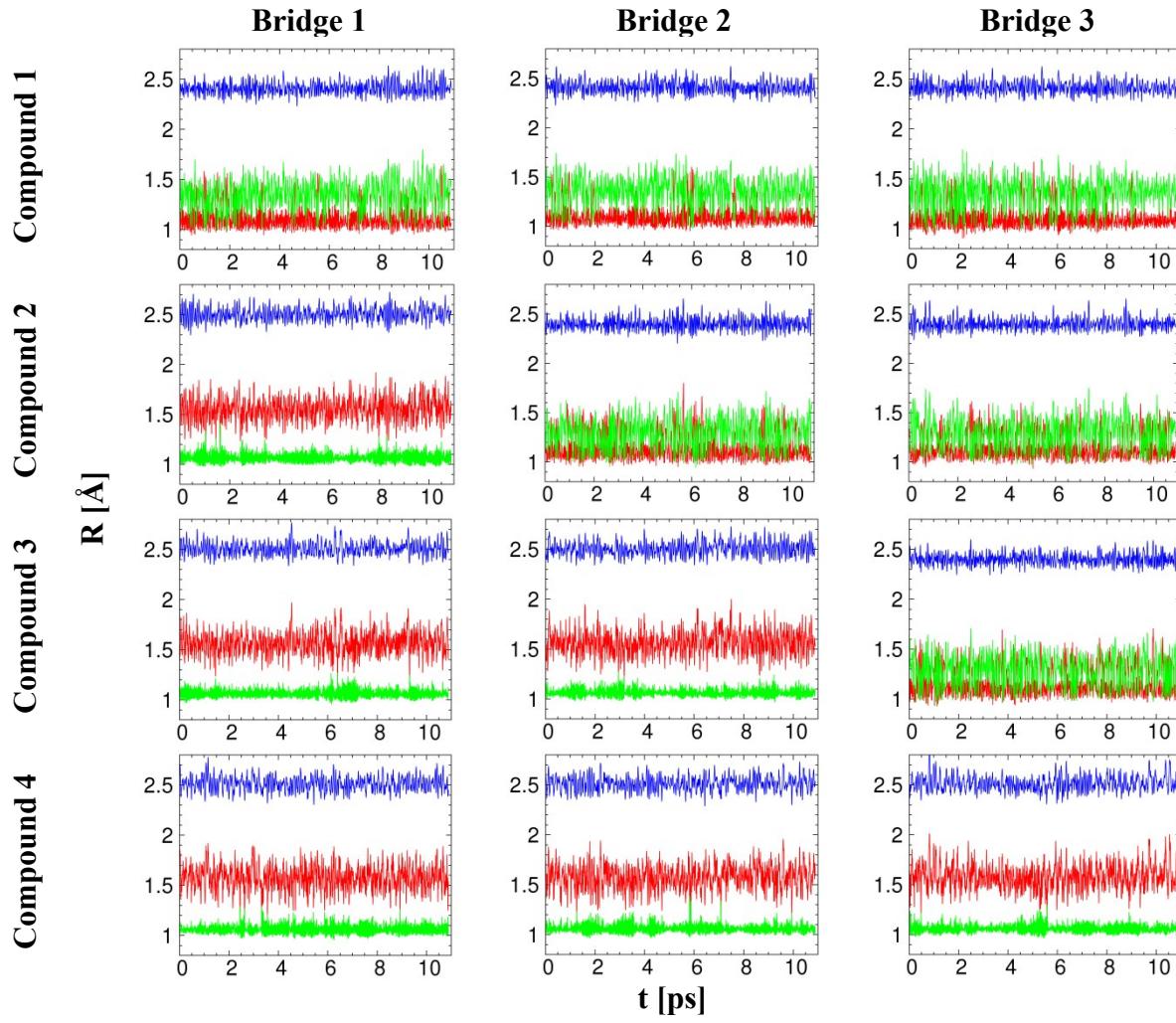


Fig. S1 Time evolution of the O···N, O-H and H-N interatomic distances in the three hydrogen bridges of the investigated compounds **1** – **4** obtained by gas-phase CPMD calculation. Red: hydroxyl oxygen – bridged hydrogen distance, green: acetyl oxygen or imine nitrogen – bridged hydrogen distance, blue: donor – acceptor distance.

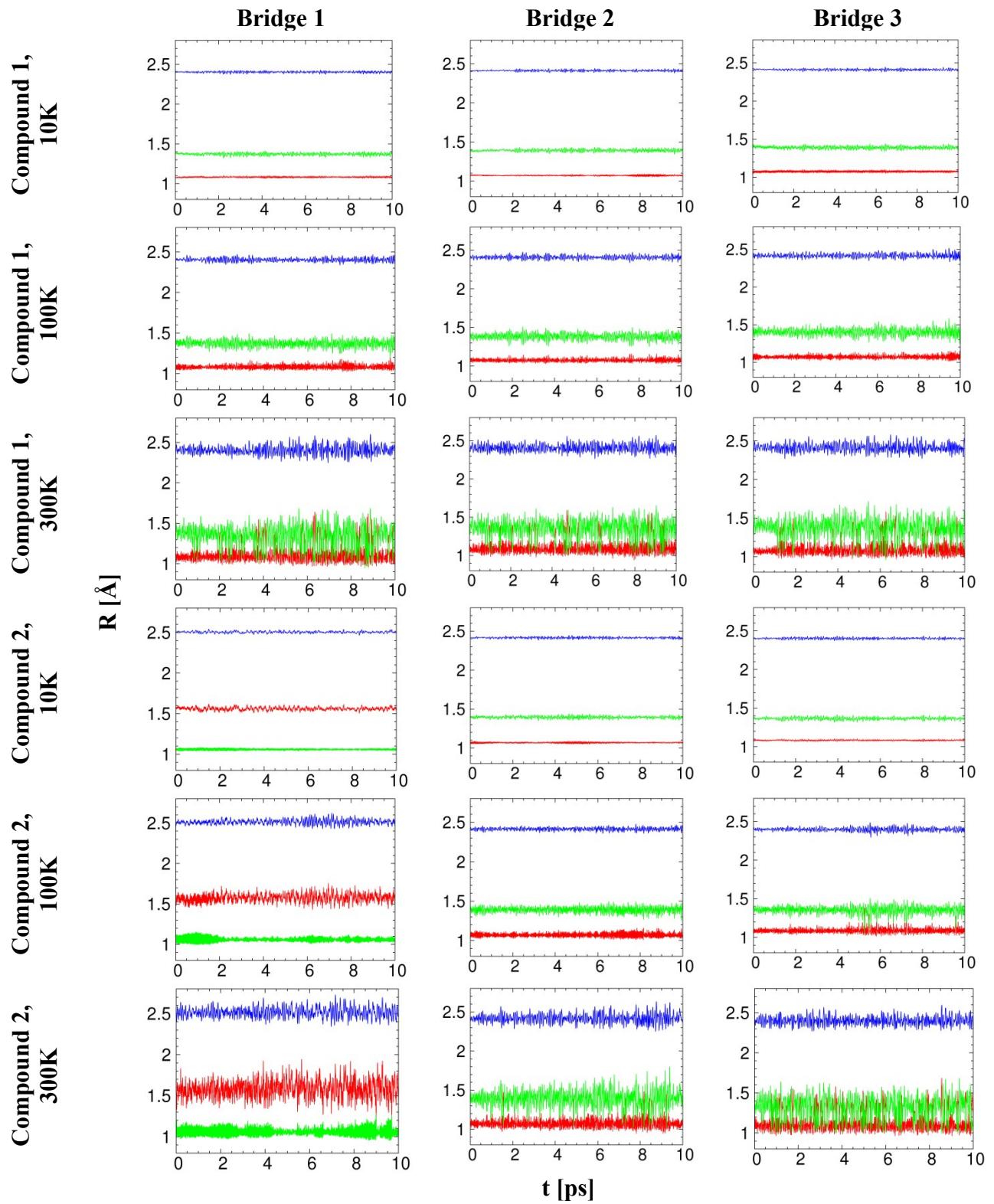
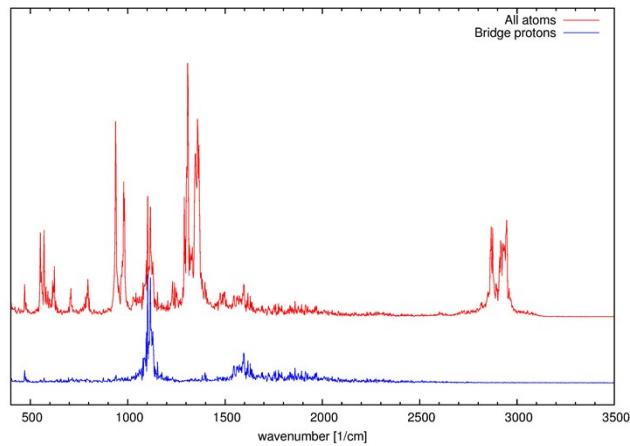
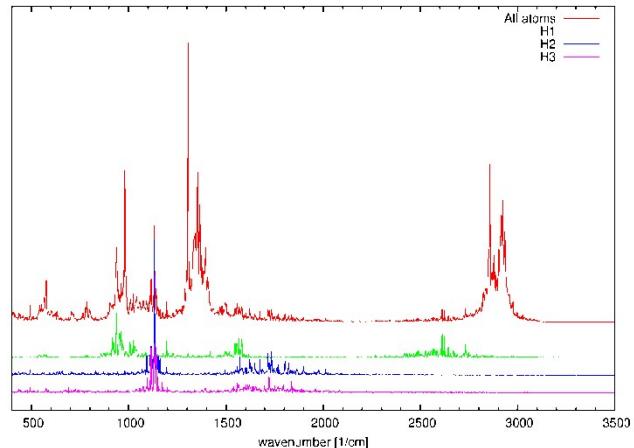


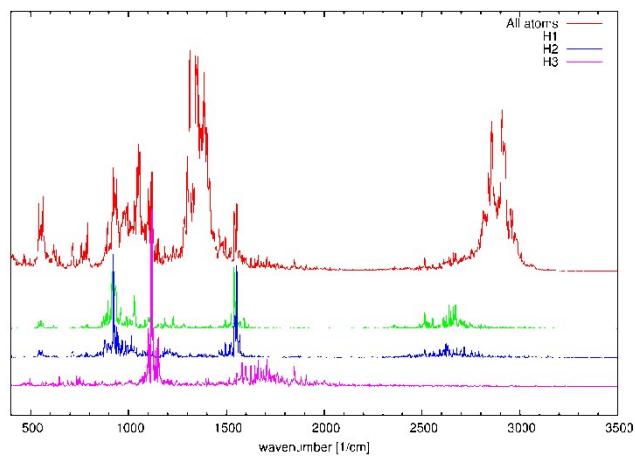
Fig. S2 Time evolution of interatomic distances in the three hydrogen bridges of the investigated compounds **1** and **2** for 10, 100 and 300 K (for which the X-ray structure was determined). Red: hydroxyl oxygen – proton distance, green: carbonyl oxygen or imine nitrogen – proton distance, blue: donor – acceptor distance. Results of the solid-state CPMD simulation.



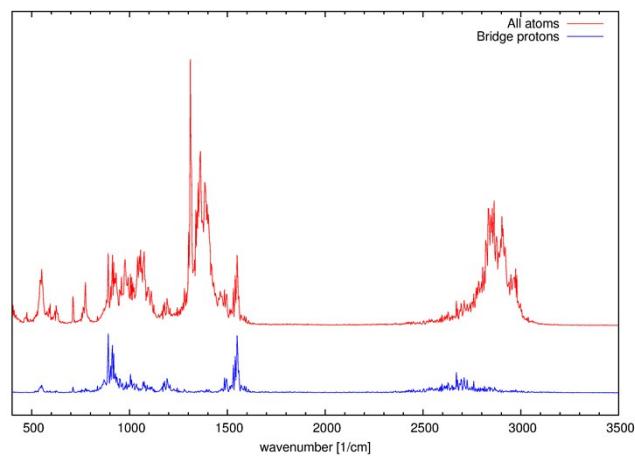
Compound 1



Compound 2



Compound 3



Compound 4

Fig. S3 Power spectra of atomic velocities – results of the CPMD runs for the studied compounds. Intensities (the y axis) are arbitrary and are not related to the IR absorption intensities. The spectra presented only for the bridged protons vibrational modes.

Compound 1

Power spectrum - arbitrary intensity

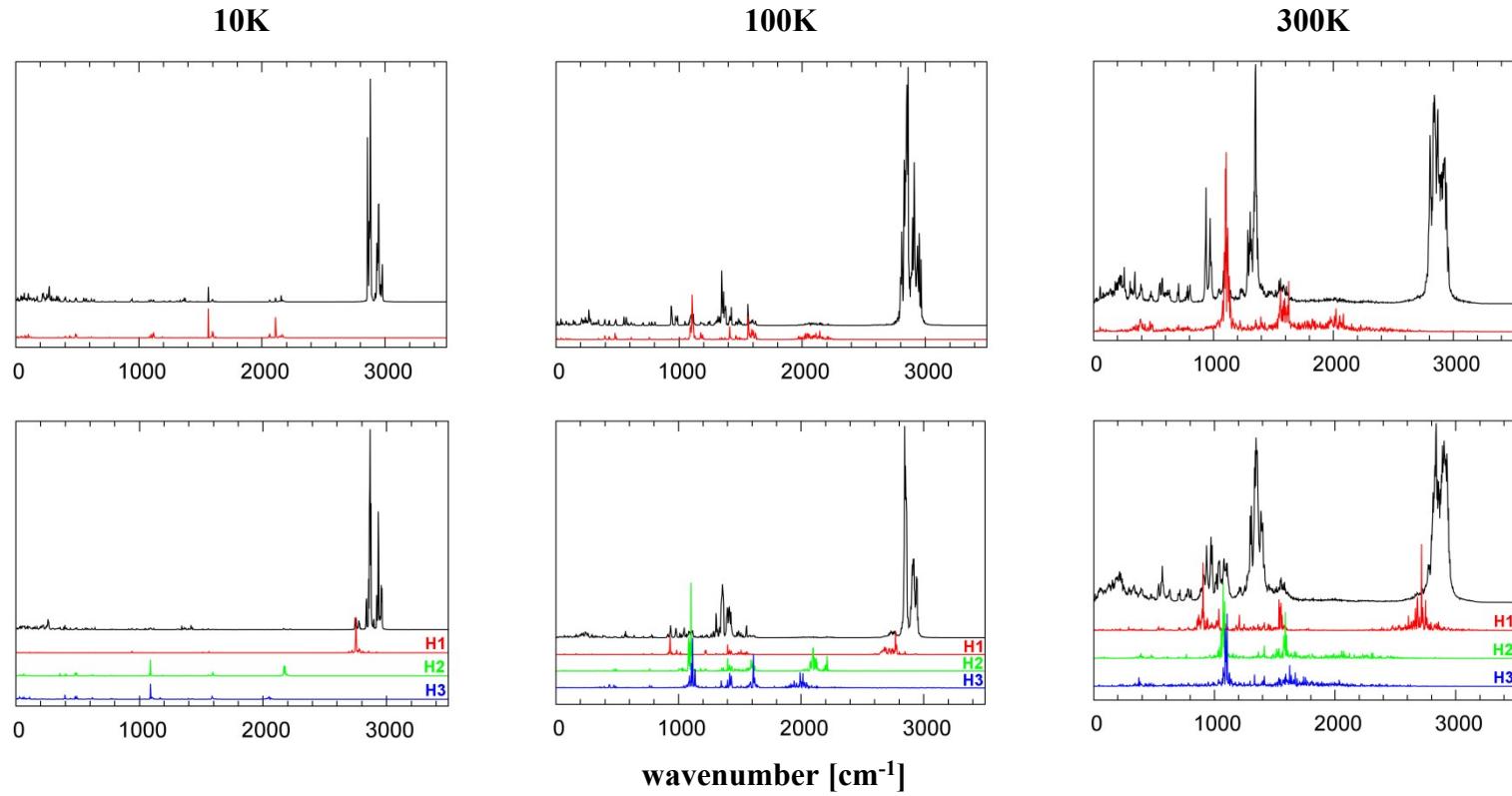


Fig. S4 Power spectra of the atomic velocity for the investigated compounds **1** and **2** - results of the solid-state CPMD simulation. For **1**, total power spectrum (black) and joint contribution of the bridge protons (red) are given. For **2**, total power spectrum (black) and individual contributions of the bridge protons are given.

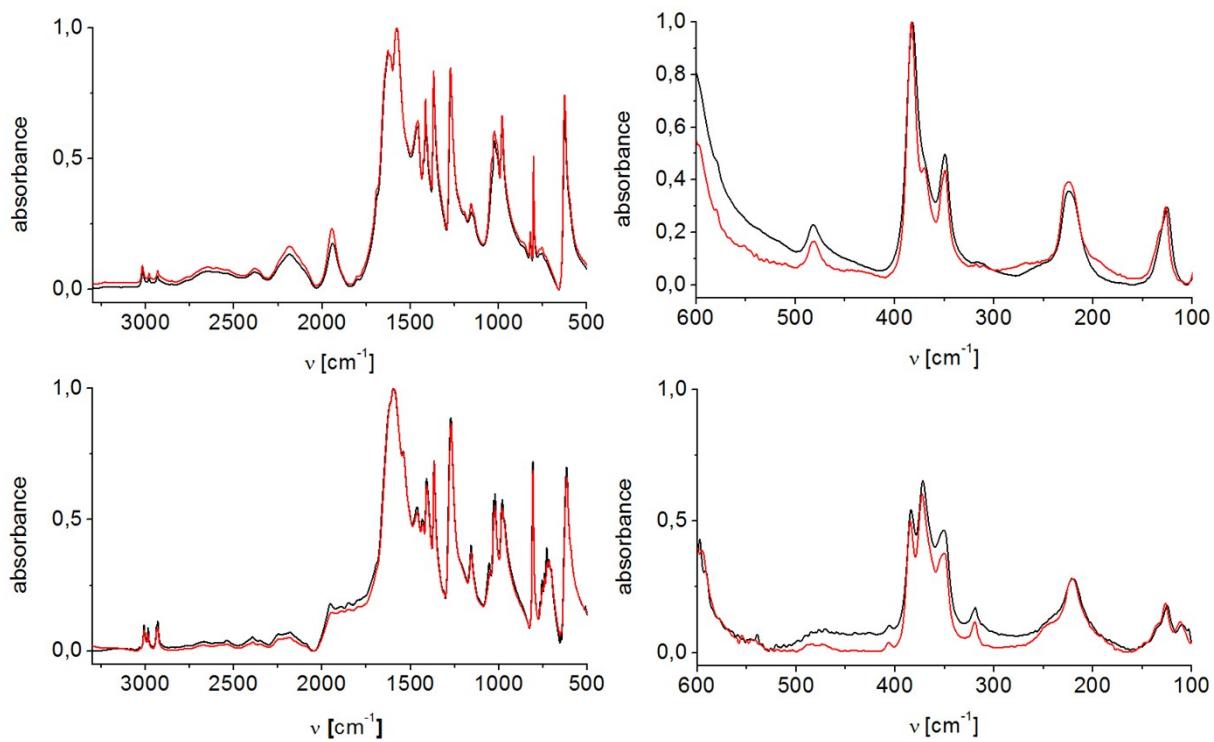


Fig. S5 Infrared spectra of 1,1',1''-(2,4,6-trihydroxybenzene-1,3,5-triyl)triethanone (**1**, upper spectra) and its deuteroderivative (**1D**, lower spectra) in 3300 - 500 cm^{-1} (left side) and 600 - 100 cm^{-1} (right side) regions measured as function of temperature - T = 300 K (black line) and 10 K (red line).

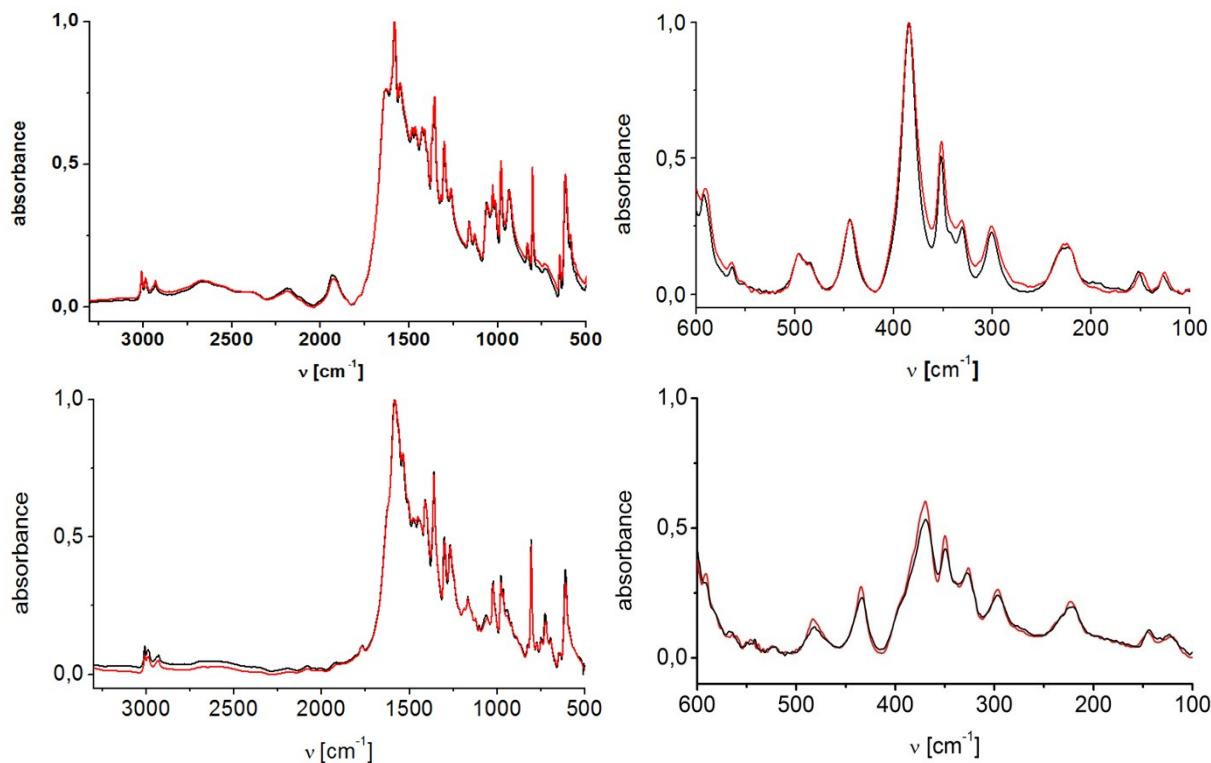


Fig. S6 Infrared spectra of (*1E*)-1-(3,5-diacetyl-2,4-dihydroxy-6-oxocyclohexa-2,4-dien-1-yl)-N-methylethaniminium (**2**, upper spectra) and its deuteroderivative (**2**, lower spectra) in $3300 - 500\text{ cm}^{-1}$ (left side) and $600 - 100\text{ cm}^{-1}$ (right side) regions measured as function of temperature - $T = 300\text{ K}$ (black line) and 10 K (red line).

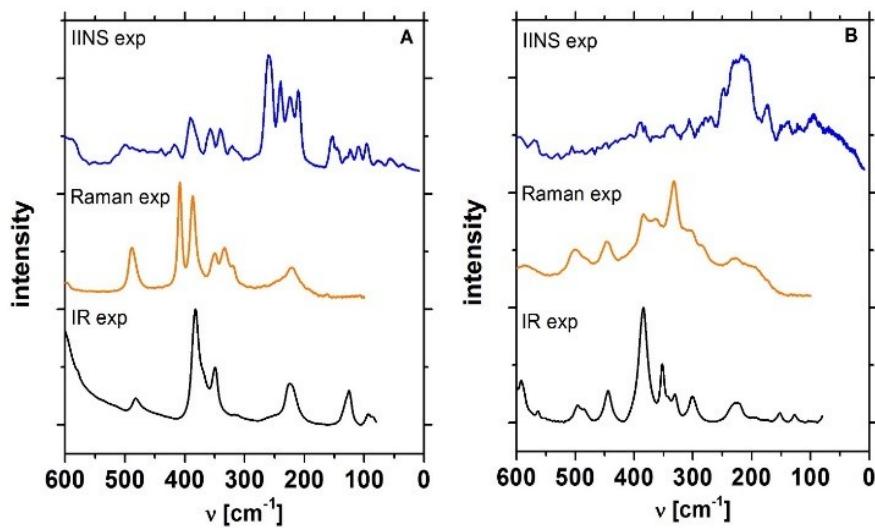


Fig. S7 Experimental IR (black colour), Raman (red colour) and INS (blue colour) spectra of 1,1',1''-(2,4,6-trihydroxybenzene-1,3,5-triyl)triethanone (**1**) (**A**) and (1*E*)-1-(3,5-diacetyl-2,4-dihydroxy-6-oxocyclohexa-2,4-dien-1-yl)-N-methylethaniminium (**2**) (**B**).

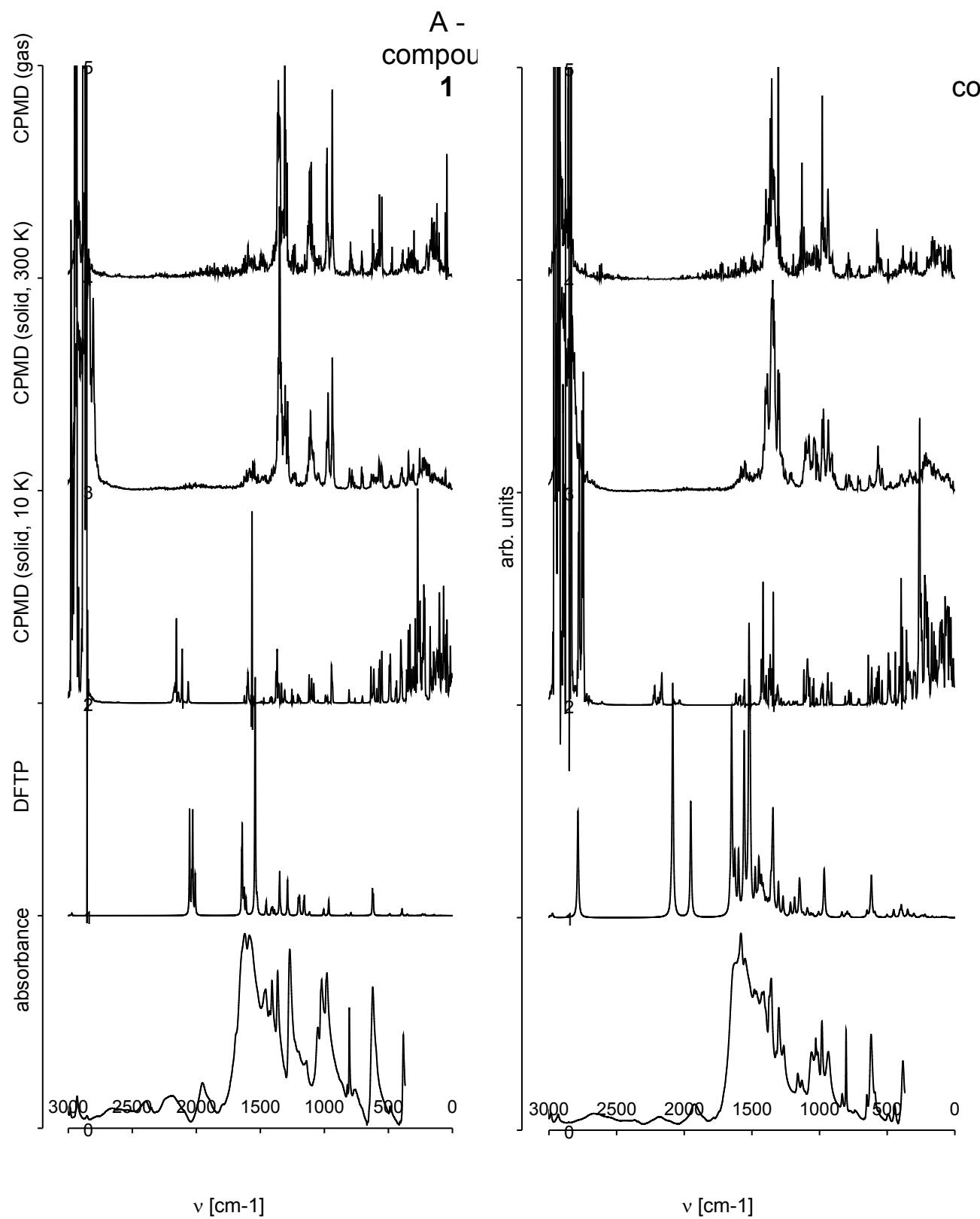


Fig. S8 Experimental and calculated DFTP IR spectra, as well as CPMD and CPMD-solid hydrogens vibrational spectra (300 K and 10 K) of 1,1',1''-(2,4,6-trihydroxybenzene-1,3,5-triyl)triethanone (**1**, panel A) and (1E)-1-(3,5-diacetyl-2,4-dihydroxy-6-oxocyclohexa-2,4-dien-1-yl)-N-methylethaniminium (**2**, panel B).

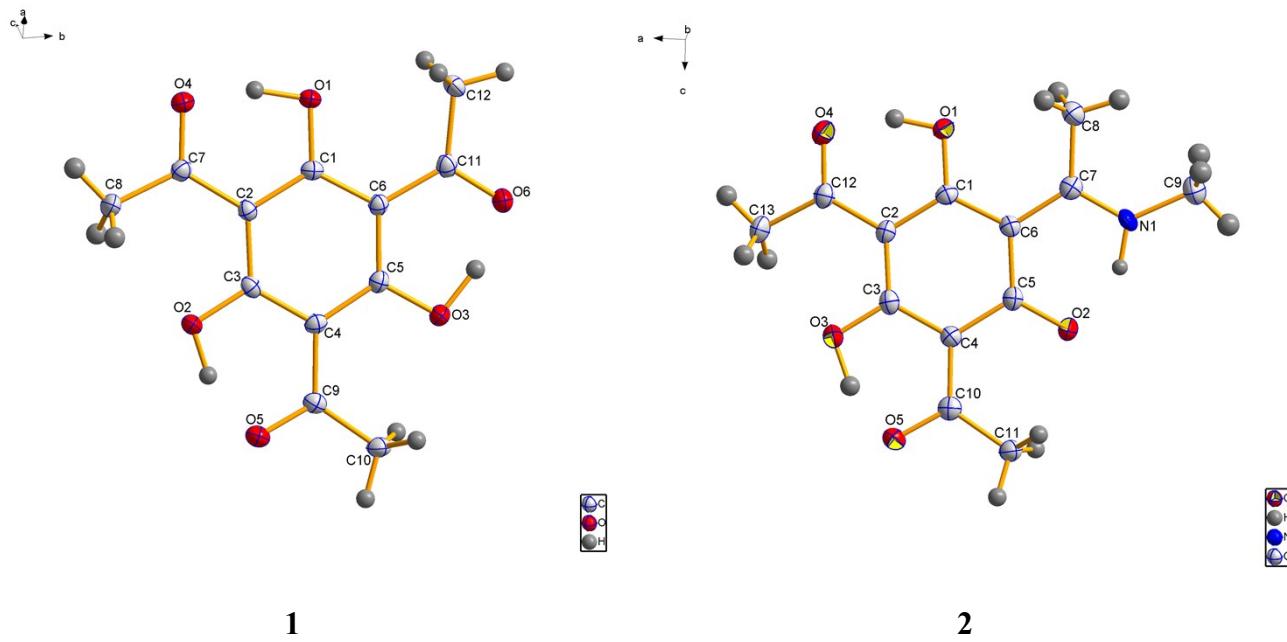


Fig. S9 The molecular structure and atom labelling scheme of $1,1',1''$ -(2,4,6-trihydroxybenzene-1,3,5-triyl)triethanone (**1**) and ($1E$)-1-(3,5-diacetyl-2,4-dihydroxy-6-oxocyclohexa-2,4-dien-1-yl)- N -methylethaniminium (**2**). Crystal structures were visualized using the DIAMOND program.⁶¹

Table S1. Definitions of internal coordinates used in potential energy distribution (PED) analysis for the assignment of the vibrational spectra of compounds **1** and **2**.

v(CH) – stretching of the C-H bond
v(OH) – stretching of the O-H bond
v(OD) – stretching of the O-D bond
v(NH) – stretching of the N-H bond
v(ND) – stretching of the N-D bond
v(C=O) – stretching of the C=O bond
v(CO) – stretching of the C-O bond
v($C_{\text{alk}}C_{\text{alk}}$) – stretching of the $C_{\text{alk}}-C_{\text{alk}}$ bond
v($C_{\text{ar}}C_{\text{alk}}$) – stretching of the $C_{\text{ar}}-C_{\text{alk}}$ bond
v(C=N) – stretching of the C=N bond
v(CC) - ring stretching
v(NC) - stretching of the N- C_{alk} bond
α (CC) - in plane bending of the C=C=C chain in the aromatic ring
δ ($CC_{\text{im}}N$) - in plane bending of the $C-C_{\text{im}}=N$ chain
δ (CCH_3) – in plane bending of the $C-CH_3$ unit
δ (NCH_3) – in plane bending of the $N-CH_3$ unit
δ (CCC_{im}) - in plane bending of the $C=C-C_{\text{im}}$ chain
δ (CCC_{alk}) – in plane bending of the $C-C-C_{\text{alk}}$ chain
δ ($CC=O$) – in plane bending of the $C-C=O$ chain
δ (CCO) – in plane bending of the $C-C-O$ chain
δ (COH) - in plane bending of the $C-O-H$ chain
δ (COD) - in plane bending of the $C-O-D$ chain
δ (CNH) - in plane bending of the $C=N-H$ chain
δ (CND) - in plane bending of the $C-O-D$ chain
γ ($C=O$) – out of plane bending of the $C=O$ bond (the change in angle between the $C=O$ and $C-C$ bonds)
γ ($C=N$) – out of plane bending of the $C=N$ bond (the change in angle between the $C=N$ and $C-C$ bonds)
γ (COH) – out of plane bending of the $O-H$ bond (the change in angle between the $O-H$ and $C-O$ bonds)
γ (COD) – out of plane bending of the $O-D$ bond (the change in angle between the $O-D$ and $C-O$ bonds)
γ (CNH) – out of plane bending of the $N-H$ bond (the change in angle between the $O-H$ and $C-O$ bonds)
γ (CND) – out of plane bending of the $N-D$ bond (the change in angle between the $O-D$ and $C-O$ bonds)
γ (CO) – out of plane bending of the $C-O$ bond (the change in angle between the $C-O$ and $C=C$ bonds)
γ (CC_{alk}) – out of plane bending of the $C-C_{\text{alk}}$ (the change in angle between the $C-C_{\text{alk}}$ bond and plane defined by three carbon atoms in the ring)
τ (CC) – torsion angle in the ring (the change in the dihedral angle between two $C=C=C$ planes in the ring)
τ (CC_{alk}) – torsion angle in the ring (the change in the dihedral angle between $C=C=C$ and $C=C-C_{\text{alk}}$ planes)
τ (C=N) – torsion around the C=N bond (the change in the dihedral angle between $C-C=N$ and $C=C-C$ planes)
τ (CH_3) – torsion around the $C_{\text{alk}}-C_{\text{alk}}$ bond (the change in the dihedral angle between $C-C-H$ and $C-C-C$ planes)
τ (NC) – torsion around the N- C_{alk} bond (the change in the dihedral angle between $C=N-C$ and $C=C-C$ planes)

Table S2. Experimental (IR, Raman and IINS) and calculated (B3LYP/6-31+G(d,p)) vibrational frequencies (ν , cm⁻¹) with the assignments (PED, %) for 1,1',1''-(2,4,6-trihydroxybenzene-1,3,5-triyl)triethanone (**1**) and its OD derivative. Only internal coordinates contributing more than 5% to the normal coordinates are reported.

Compound 1														
IR		Raman		IINS		OH B3LYP/6-31+G(d,p), PED (%)						OD B3LYP/6-31+G(d,p), PED (%)		
OH	OD	OH	OD	OH	OD	ν	IR	R	PED (%)	ν	IR	PED (%)		
3033	3007	2983	2932	2857	2900-1000*	3030	3033	3166.5	11.0	83.0	v(CH) 99	3166.5	11.55	v(CH) 100
						3166.0	9.4	94.3	v(CH) 99	3165.9	8.47	v(CH) 100		
						3165.9	20.4	95.8	v(CH) 100	3165.9	20.04	v(CH) 100		
						3135.4	2.1	31.2	v(CH) 100	3135.4	2.13	v(CH) 100		
						3135.0	2.1	37.9	v(CH) 100	3135.0	2.09	v(CH) 100		
						3134.5	2.1	64.7	v(CH) 100	3134.5	2.07	v(CH) 100		
						3066.8	1.4	134.6	v(CH) 99	3066.8	1.51	v(CH) 99		
						3066.5	1.4	61.1	v(CH) 99	3066.5	1.47	v(CH) 99		
						3066.3	1.3	395.3	v(CH) 99	3066.3	1.39	v(CH) 99		
						2588.5	1355.7	27.4	v(OH) 94					
2700-900*	1692sh	1620	1582	1596wb	1537	2985	2929	2587.3	1354.9	27.3	v(OH) 97	1915.5	842.43	v(OD) 85
						2572.0	4.3	158.5	v(OH) 97	1914.7	834.63	v(OD) 88		
										1911.9	29.15	v(OD) 84		
						1683.2	594.6	40.3	$\delta(\text{COH})$ 44, v(C=O) 8, v(CC) 7					
						1682.0	609.3	40.5	$\delta(\text{COH})$ 41, v(C=O) 7, v(CC) 7, $\delta(\text{CCO})$ 5					
						1664.3	2.7	93.3	v(C=O) 44, $\delta(\text{COH})$ 29					
						1633.9	837.4	8.5	v(C=O) 26, v(CC) 17, $\delta(\text{CCC}_{\text{alk}})$ 16	1636.4	994.64	v(C=O) 30, $\gamma(\text{CC}_{\text{alk}})$ 21, v(CC) 11, $\delta(\text{CCH}_3)$ 5		
						1633.6	818.7	8.5	v(C=O) 28, v(CC) 11, $\delta(\text{CCC}_{\text{alk}})$ 10	1635.7	998.41	v(C=O) 29, $\delta(\text{CC=O})$ 13, v(CC) 7, $\gamma(\text{CC}_{\text{alk}})$ 6, $\delta(\text{CCH}_3)$ 5		
						1561.6	0.0	21.4	$\delta(\text{COH})$ 46, $\alpha(\text{CC})$ 9, v(CO) 24	1610.3	0.58	v(C=O) 48		
						1505.1	185.3	32.4	v(CC) 25, v(CO) 19, $\delta(\text{COH})$ 7, $\delta(\text{CCC}_{\text{alk}})$ 6	1563.6	300.10	v(CC) 45, $\gamma(\text{CO})$ 11, $\delta(\text{CCH}_3)$ 8, v(C=O) 6		
						1504.7	185.1	32.3	v(CC) 24, v(CO) 19, $\delta(\text{COH})$ 7, $\delta(\text{CCC}_{\text{alk}})$ 7	1563.3	299.68	v(CC) 40, $\gamma(\text{CO})$ 9, $\delta(\text{CCH}_3)$ 8, $\gamma(\text{C=O})$ 7, v(C=O) 7		
						1475.5	13.8	7.0	$\delta(\text{CCH}_3)$ 99	1478.1	0.00	$\delta(\text{CH}_3)$ 20, v(CO) 31		
						1475.3	11.1	10.9	$\delta(\text{CCH}_3)$ 99	1475.5	13.40	$\delta(\text{CH}_3)$ 99		
										1475.3	10.93	$\alpha(\text{CC})$ 93, $\tau(\text{CC})$ 6		

1461	1464	1460		1474.1	11.7	10.5	$\delta(\text{CCH}_3)$ 100	1474.1	11.52	$\delta(\text{CH}_3)$ 100
				1457.9	0.5	1.4	$\delta(\text{CCH}_3)$ 36, $\delta(\text{CC})$ 8	1462.7	23.19	$\delta(\text{CH}_3)$ 32, $\alpha(\text{CC})$ 31
				1451.0	55.2	14.9	$\delta(\text{CCH}_3)$ 64	1461.8	23.93	$\delta(\text{CH}_3)$ 47, $\alpha(\text{CC})$ 10
				1450.2	52.3	14.8	$\delta(\text{CCH}_3)$ 60	1451.4	0.15	$\delta(\text{CH}_3)$ 53, $\alpha(\text{CC})$ 24
1426	1426	1427	1427	1438.1	0.0	38.9	$\alpha(\text{CC})$ 7, $v(\text{C=O})$ 15	1415.1	189.78	$v(\text{CO})$ 18, $\alpha(\text{CC})$ 10, $\tau(\text{CC})$ 9, $v(\text{CC})$ 13, $v(\text{C=O})$ 12
1408	1408	1413	1411	1409.5	82.0	13.2	$\delta(\text{CCH}_3)$ 73, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 8	1414.1	186.30	$v(\text{CO})$ 17, $v(\text{CC})$ 13, $\delta(\text{CH}_3)$ 8, $v(\text{C=O})$ 8
				1407.8	81.7	13.2	$\delta(\text{CCH}_3)$ 73, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 5	1406.4	0.19	$\alpha(\text{CC})$ 62, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 8
			1386	1403.8	6.8	1.1	$\delta(\text{CCH}_3)$ 75	1405.0	1.78	$\tau(\text{CC})$ 57, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 8, $\delta(\text{CH}_3)$ 6
1365	1365	1370	1370	1346.7	0.4	1.4	$v(\text{CC})$ 50, $v(\text{C}_{\text{ar}}\text{C}_{\text{alk}})$ 18	1403.9	4.60	$\delta(\text{CH}_3)$ 65, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 9
				1343.0	197.0	91.1	$v(\text{C}_{\text{ar}}\text{C}_{\text{alk}})$ 24, $v(\text{CC})$ 18, $v(\text{C=O})$ 5, $\delta(\text{COH})$ 5	1356.7	0.01	$v(\text{CC})$ 75
			1327	1342.3	196.7	91.1	$v(\text{C}_{\text{ar}}\text{C}_{\text{alk}})$ 20, $v(\text{CC})$ 11, $v(\text{C=O})$ 6	1325.4	0.02	$v(\text{C}_{\text{ar}}\text{C}_{\text{alk}})$ 33, $v(\text{CC})$ 16, $v(\text{CO})$ 15
1309sh		1304	1307	1307.3	0.2	59.5	$v(\text{CC})$ 36	1304.8	278.14	$v(\text{C}_{\text{ar}}\text{C}_{\text{alk}})$ 31, $\delta(\text{COD})$ 9, $\delta(\text{CC=O})$ 9, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 7
1270	1273	1271	1271	1286.6	177.3	8.0	$v(\text{CO})$ 23, $v(\text{CC})$ 16, $\delta(\text{CC=O})$ 8, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 6, $v(\text{C}_{\text{ar}}\text{C}_{\text{alk}})$ 5	1304.5	278.09	$v(\text{C}_{\text{ar}}\text{C}_{\text{alk}})$ 29, $\delta(\text{CC=O})$ 12, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 7
1228sh				1286.2	175.6	8.1	$v(\text{CO})$ 26, $v(\text{CC})$ 10, $\delta(\text{CC=O})$ 7, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 7, $v(\text{C}_{\text{ar}}\text{C}_{\text{alk}})$ 6			
1201								1196.6	31.32	$v(\text{CO})$ 26, $\gamma(\text{COD})$ 10, $v(\text{CC})$ 9, $\delta(\text{CC=O})$ 7, $\gamma(\text{CCC}_{\text{alk}})$ 6
	1213	1184						1196.1	30.94	$\gamma(\text{COD})$ 30, $v(\text{CO})$ 25, $\gamma(\text{CCC}_{\text{alk}})$ 9, $v(\text{CC})$ 14
	1157		1160					1181.7	0.01	$\gamma(\text{COD})$ 48, $\delta(\text{CH}_3)$ 7
1141				1161.2	3.5	5.8	$v(\text{CC})$ 15, $v(\text{CO})$ 11, $\delta(\text{CCC}_{\text{alk}})$ 15, $\delta(\text{CCO})$ 6, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 5			
				1160.6	3.4	5.8	$v(\text{CC})$ 10, $v(\text{CO})$ 16, $\delta(\text{CCC}_{\text{alk}})$ 10, $\delta(\text{CCO})$ 7, $\delta(\text{CC=O})$ 5			
		1120		1103.3	0.0	18.1	$v(\text{CC})$ 21, $\delta(\text{CH}_3)$ 20, $\delta(\text{CC})$ 15, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 10	1091.9	0.07	$\delta(\text{CH}_3)$ 25, $\tau(\text{CC})$ 8, $v(\text{CC})$ 19
		1087		1100.5	203.0	0.0	$\gamma(\text{COH})$ 98	1080.4	53.29	$\gamma(\text{COD})$ 43, $\delta(\text{CH}_3)$ 6, $v(\text{CC})$ 5
1052	-		1072	1096.8	4.9	0.4	$\gamma(\text{COH})$ 98	1079.6	52.96	$\gamma(\text{COD})$ 46, $v(\text{CC})$ 6, $\delta(\text{CH}_3)$ 6
950	-			1093.7	16.8	0.4	$\gamma(\text{COH})$ 94	1050.0	6.46	$\tau(\text{CC})$ 62, $\delta(\text{COD})$ 18
	1058sh			1049.3	6.6	0.2	$\delta(\text{CCH}_3)$ 69, $\gamma(\text{C=O})$ 19	1049.3	2.63	$\delta(\text{CH}_3)$ 62, $\gamma(\text{C=O})$ 18
1020	1028	1033	1034	1048.7	10.8	0.2	$\delta(\text{CCH}_3)$ 69, $\gamma(\text{C=O})$ 19	1048.4	2.06	$\delta(\text{CH}_3)$ 70, $\gamma(\text{C=O})$ 19
			1015	1047.7	16.4	0.0	$\delta(\text{CCH}_3)$ 64, $\gamma(\text{C=O})$ 19			
				1000.2	5.0	24.8	$\delta(\text{CCH}_3)$ 32, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 12, $\delta(\text{CC=O})$ 5			
978	981	984	984	998.4	44.9	7.1	$\delta(\text{CCH}_3)$ 47, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 12, $v(\text{C}_{\text{ar}}\text{C}_{\text{alk}})$ 5	992.9	53.66	$\delta(\text{CH}_3)$ 32, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 29, $\tau(\text{CC})$ 10
864sh	967	867	977	997.3	45.3	7.1	$\delta(\text{CCH}_3)$ 48, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 13, $v(\text{C}_{\text{ar}}\text{C}_{\text{alk}})$ 6	991.9	51.70	$\tau(\text{CC})$ 21, $\delta(\text{CH}_3)$ 20, $v(\text{C}_{\text{alk}}\text{C}_{\text{alk}})$ 16

820		820		893										
804	807	805	808	826	844	833.8	5.7	1.8	v(C _{alk} C _{alk}) 25, v(C _{ar} C _{alk}) 21, α (CC) 8, δ (CCO) 7, δ (CCC _{alk}) 6	803.8	0.86	τ (CC _{alk}) 46, τ (CH ₃) 28, δ (CCC _{alk}) 8		
765		756		765		789.0	0.0	8.1	v(C _{alk} C _{alk}) 27, v(C _{ar} C _{alk}) 26, δ (CC) 8, δ (CCO) 7, δ (CCC _{alk}) 6	802.6	20.57	τ (CH ₃) 34, γ (CO) 21, τ (CC _{alk}) 19, δ (CCC _{alk}) 7		
									δ (CCO) 48, δ (CCC _{alk}) 36	799.3	128.29	γ (CO) 51, τ (CH ₃) 12, δ (CH ₃) 9, τ (CC _{alk}) 9, δ (CCC _{alk}) 9		
										649.6	0.02	δ (COD) 18, δ (CC=O) 35		
										622.1	103.16	δ (COD) 41, δ (CC=O) 24, δ (CH ₃) 8		
										594.1	0.00	δ (COD) 32, γ (C=O) 14, δ (CCC _{alk}) 12, τ (CC) 11, δ (CCO) 7		
										745.2	10.6	0.0 τ (CC) 50, γ (CO) 31, γ (CC _{alk}) 18		
										733.2	0.0	0.1 γ (CO) 47, τ (CC) 29, γ (CC _{alk}) 15		
708sh	702sh			714	696	-	733.2	0.0	0.1 γ (CO) 46, τ (CC) 29, γ (CC _{alk}) 16	724.7	0.01	δ (CCC _{alk}) 45, δ (CH ₃) 30, δ (CCO) 10, δ (CC=O) 5		
										649	647	661 669 649.7 0.0 16.3 δ (CC=O) 52		
622	615	612		635	629	625.7	107.2	1.0 δ (CC=O) 64, α (CC) 8	621.7	103.74	δ (CC=O) 65, δ (CH ₃) 8			
						625.3	107.8	1.0 δ (CC=O) 68, α (CC) 8	616.7	0.09	δ (CCH ₃) 41, v(CC) 21, v(CC) 21			
597sh				607	605	610	622.5	0.2	14.4 α (CC) 39, v(CC) 28, v(CO) 14	593.3	0.00	γ (C=O) 47, δ (CCC _{alk}) 18, δ (CH ₃) 15, δ (CC=O) 8		
						593			596.2	0.0	0.3 γ (C=O) 46, γ (CO) 15, δ (CCH ₃) 11, γ (CC _{alk}) 12			
						555			595.4	0.0	0.3 γ (C=O) 47, γ (CO) 18, δ (CCH ₃) 16, γ (CC _{alk}) 8			
						525			577.4	0.1	0.0 γ (C=O) 47, γ (CO) 17, τ (CC) 17, δ (CCH ₃) 11			
482		487		501		488.6	4.6	5.5 δ (CCO) 38, δ (CC=O) 18	474.6	4.38	γ (CO) 28, δ (COD) 14, δ (CC=O) 13, δ (CH ₃) 5			
						472			480			473.6 4.71 δ (CC=O) 28, γ (C=O) 19, γ (CO) 6, δ (CH ₃) 5		
						443			487.8	4.9	5.5 δ (CCO) 48, δ (CC=O) 19	412.7	0.00	δ (CC=O) 22, δ (COD) 11, γ (CO) 18, γ (C=O) 9
						436	428		422.4	0.0	3.5 δ (CCC _{alk}) 27, δ (CCO) 26, δ (CC=O) 22	392.1	13.97	δ (CH ₃) 20, γ (CO) 14, δ (CC=O) 14, γ (C=O) 9, v(OD) 6
						408	402		418			391.7 12.02 δ (CH ₃) 17, δ (COD) 17, γ (CO) 11, γ (C=O) 8, δ (CC=O) 8, v(OD) 6		
									403.5	10.3	2.7 δ (CC=O) 25, α (CC) 19, δ (CCO) 15	390.2	2.43	v(OD) 13, v(C _{ar} C _{alk}) 11, γ (CC _{alk}) 9, γ (CO) 8
									403.1	10.3	2.8 δ (CC=O) 32, α (CC) 18, δ (CCO) 17			

382		387		391	384	399.7	0.3	5.0	$\nu(C_{ar}C_{alk})$ 26, $\delta(CC=O)$ 18, $\alpha(CC)$ 6, $\nu(OH)$ 6	349.7	8.89	$\delta(CH_3)$ 22, $\gamma(CO)$ 6, $\delta(CC=O)$ 26, $\gamma(CC_{alk})$ 6
	373		373							349.5	8.69	$\delta(CH_3)$ 22, $\delta(COD)$ 20, $\delta(CC=O)$ 8, $\gamma(CO)$ 6, $\nu(C_{ar}C_{alk})$ 6, $\gamma(CC_{alk})$ 6
349	350	350	349	357	358	351.3	8.8	4.0	$\alpha(CC)$ 23, $\delta(CC=O)$ 20, $\delta(CCO)$ 6, $\delta(CCC_{alk})$ 6, $\nu(C_{ar}C_{alk})$ 5			
		333		341	348	351.0	8.6	4.0	$\alpha(CC)$ 23, $\delta(CC=O)$ 28, $\nu(C_{ar}C_{alk})$ 6, $\delta(CCC_{alk})$ 6	323.7	0.00	$\delta(CC=O)$ 34, $\delta(COD)$ 16
			333			324.2	0.0	9.6	$\delta(C_{alk}C=O)$ 50	309.4	0.00	$\delta(CCO)$ 31, $\delta(CC=O)$ 23, $\delta(CH_3)$ 28, $\tau(CC_{alk})$ 6
	323	320	320	323	328	310.2	0.0	0.2	$\gamma(CC_{alk})$ 54, $\tau(CC)$ 28, $\gamma(COH)$ 6	309.2	0.00	$\delta(CCO)$ 42, $\delta(CH_3)$ 28, $\delta(CC=O)$ 13, $\gamma(CO)$ 6
					311	310.0	0.0	0.2	$\gamma(CC_{alk})$ 55, $\tau(CC)$ 28, $\gamma(COH)$ 7	239.3	0.94	$\delta(CH_3)$ 32, $\tau(CC_{alk})$ 19, $\gamma(CC_{alk})$ 12, $\delta(CC=O)$ 6, $\delta(CCO)$ 5
				259	264	240.2	1.1	0.0	$\tau(CC)$ 33, $\tau(CH_3)$ 28, $\gamma(CC_{alk})$ 18	226.6	0.01	$\tau(CC_{alk})$ 39, $\gamma(CC_{alk})$ 38, $\delta(CH_3)$ 6, $\tau(CC)$ 6
				240	243	226.7	0.0	0.7	$\tau(CH_3)$ 76, $\delta(CCH_3)$ 11	219.8	0.02	$\tau(CC_{alk})$ 56, $\gamma(CC_{alk})$ 14, $\delta(CH_3)$ 7
				225	227	220.0	0.0	0.7	$\tau(CH_3)$ 70, $\delta(CCH_3)$ 7	218.1	12.24	$\gamma(CC_{alk})$ 71
				210	213	198.4	0.3	0.0	$\tau(CC)$ 35, $\tau(CH_3)$ 27, $\gamma(CC_{alk})$ 21	217.1	12.34	$\delta(CC=O)$ 46, $\gamma(CC_{alk})$ 27, $\gamma(CO)$ 5
223	223	221	221	154	153	218.3	12.4	1.9	$\delta(CCC_{alk})$ 71	197.5	0.36	$\delta(CH_3)$ 35, $\tau(CC_{alk})$ 19, $\delta(CC=O)$ 8, $\delta(CCO)$ 14, $\gamma(CC_{alk})$ 7
				145	148	217.3	12.5	1.9	$\delta(CCC_{alk})$ 71			
126	126			127	129	126.0	0.0	0.0	$\gamma(COH)$ 38, $\tau(CC_{alk})$ 33, $\tau(CC)$ 11	125.9	0.01	$\delta(CCO)$ 26, $\gamma(CO)$ 12, $\delta(CH_3)$ 11, $\tau(CH_3)$ 33
				124	123	124.6	0.0	0.0	$\gamma(COH)$ 38, $\tau(CC_{alk})$ 36, $\tau(CC)$ 11	124.6	0.00	$\tau(CC_{alk})$ 24, $\tau(CH_3)$ 37, $\gamma(CO)$ 14, $\delta(CH_3)$ 11
				111	111	110.9	4.0	0.0	$\gamma(COH)$ 40, $\tau(CC_{alk})$ 32, $\gamma(CC_{alk})$ 21	110.7	3.89	$\tau(CC_{alk})$ 14, $\gamma(CO)$ 14, $\tau(CH_3)$ 33, $\delta(CCO)$ 24, $\delta(CCO)$ 8
				95	94							
				79	77	58.2	3.6	0.0	$\tau(CC_{alk})$ 49, $\tau(CC)$ 29, $\tau(CH_3)$ 5	57.9	3.55	$\delta(CH_3)$ 30, $\delta(CCO)$ 23, $\tau(CH_3)$ 26, $\gamma(CC_{alk})$ 5
				61	57	48.4	0.1	0.2	$\tau(CC)$ 49, $\tau(CC_{alk})$ 30	47.9	0.12	$\delta(CH_3)$ 48, $\tau(CH_3)$ 17, $\delta(CCO)$ 4
				40	38	47.0	0.1	0.2	$\tau(CC_{alk})$ 37, $\tau(CC)$ 46, $\tau(CH_3)$ 6	46.5	0.06	$\tau(CH_3)$ 37, $\delta(CH_3)$ 47, $\tau(C_{ar}C_{alk})$ 6

* Sub-picks of stretching vibration mode $\nu(OH)$ observed in IR spectra: 2650, 2395, 2190, 1795 cm^{-1} and $\nu(OD)$: 2700, 2531, 2390 - 2340, 2250 – 2180 and 2000 - 1700 cm^{-1} . The $\nu(OH)$ bands are sensitive on deuteroreplacement in hydrogen bond.

Table S3. Experimental and calculated (B3LYP/6-31+G(d,p)) vibrational frequencies (cm^{-1}) with the assignments (PED, %) for (1*E*)-1-(3,5-diacetyl-2,4-dihydroxy-6-oxocyclohexa-2,4-dien-1-yl)-N-methylethaniminium (**2**) and its OD/ND derivative. Only internal coordinates contributing more than 5% to the normal coordinates are reported.

Compound 2												
IR		Raman		IINS		DFT						
H	D	H	D	H	D	v	IR	R	OD B3LYP/6-31+G(d,p), PED (%)	v	IR	OD B3LYP/6-31+G(d,p), PED (%)
3015	3015	3010	3010	3010	3010	3175.5	21.2	76.7	v(CH) 98	3175.5	22.3	v(CH) 98
						3163.6	16.7	96.1	v(CH) 100	3163.6	16.4	v(CH) 100
						3159.0	19.0	91.9	v(CH) 100	3159.0	19.2	v(CH) 100
						3147.8	0.6	89.3	v(CH) 100	3147.8	0.6	v(CH) 100
						3147.7	9.9	41.3	v(CH) 98	3147.6	10.2	v(CH) 98
						3135.9	2.7	49.3	v(CH) 100	3135.9	2.7	v(CH) 100
						3133.4	3.5	42.5	v(CH) 100	3133.4	3.5	v(CH) 100
						3105.7	16.4	129.7	v(CH) 100	3105.6	16.2	v(CH) 100
						2986	2989	2989	3085.7	6.3	167.9	v(CH) 99
						2935	2937	2937	3066.8	3.1	230.7	v(CH) 99
2900- 1000*	2900- 1000*	2900- 1000*	2900- 1000*	2900- 1000*	2900- 1000*	3064.1	6.4	217.8	v(CH) 99	3064.1	6.6	v(CH) 99
						3043.7	46.3	426.9	v(CH) 99	3043.0	62.0	v(CH) 99
						2926.4	491.0	21.4	v(NH) 97			
						2439.8	915.4	48.0	v(OH) 93			
						2397.8	918.8	61.5	v(OH) 93			
										2165.2	280.5	v(ND) (93)
										1824.9	545.5	v(OD) 89, v(C=O) 13
										1801.9	678.1	v(OD) 89, v(C=O) 16
										1692.4	607.0	$\delta(\text{COH})$ 50, $\delta(\text{CCO})$ 6
										1626.0	866.4	v(C=N) 31, v(CO) 16, $\delta(\text{CC}_{\text{alk}})$ 9, $\delta(\text{CC}_{\text{alk}}\text{C}_{\text{alk}})$ 8, v(CC _{alk}) 7
1635	1635	1635	1635	1635	1635	1672.9	134.3	58.3	$\delta(\text{COH})$ 21, $\delta(\text{CNH})$ 19, v(C=N) 11, v(C=O) 10	1619.3	779.6	v(C=O) 22, $\delta(\text{CC}_{\text{alk}})$ 22, v(CC) 16, $\delta(\text{CC})$ 6
						1662.5	522.6	39.2	$\delta(\text{CNH})$ 22, $\delta(\text{COH})$ 28, v(C=O) 9, v(C=N) 8	1596.2	552.4	v(C=O) 24, $\delta(\text{CC}_{\text{alk}})$ 10, v(CO) 9
						1614sh	1615sh		v(CO) 17, $\delta(\text{CCC}_{\text{alk}})$ 10, v(C=N) 9, v(C=O) 13, $\alpha(\text{CC})$ 6	1570.8	278.67	v(CO) 16, v(CC) 16, v(C=N) 12, v(C=O) 5
						1586sh	1589		v(CO) 20, v(C=O) 19, $\delta(\text{CCC}_{\text{alk}})$ 7, v(CC) 6			
						1580	1580	1562	v(C=O) 17, v(CO) 8, $\delta(\text{COH})$ 8, v(CC) 6, $\delta(\text{CCC}_{\text{alk}})$ 6, $\delta(\text{CCC}_{\text{im}})$ 5	1540.6	134.87	v(CC) 25, v(CO) 13, $\delta(\text{CO})$ 11, v(C=O) 8, $\delta(\text{OD})$ 5
						1550	1536	1502	$\delta(\text{NCH}_3)$ 75	1512.5	76.65	$\delta(\text{CH}_3)$ 71, $\delta(\text{NCH}_3)$ 7
						1506			$\delta(\text{NCH}_3)$ 83, $\delta(\text{CH}_3)$ 12	1509.2	3.21	$\delta(\text{CH}_3)$ 88, $\delta(\text{NCH}_3)$ 7
						1485			$\delta(\text{CH}_3)$ 29, v(CC) 10, v(CO) 8	1495.0	550.33	v(CC) 15, $\delta(\text{CH}_3)$ 13
						1472	1475	1471	v(CC) 19, v(CO) 10, $\delta(\text{CH}_3)$ 15	1485.0	3.47	$\delta(\text{CH}_3)$ 69
						1475			$\delta(\text{CH}_3)$ 23, v(CC) 8, $\delta(\text{NCH}_3)$ 6, v(CO) 5, $\alpha(\text{CC})$ 5	1478.2	15.37	$\delta(\text{CH}_3)$ 94
									$\delta(\text{CH}_3)$ 81, $\delta(\text{NCH}_3)$ 13			

				1474.0	11.4	9.9	$\delta(\text{CH}_3)$ 93		1474.0	11.22	$\delta(\text{CH}_3)$ 100
				1468.5	10.5	8.9	$\delta(\text{CH}_3)$ 93		1468.5	10.45	$\delta(\text{CH}_3)$ 100
			1463	1467.0	168.2	57.8	$\delta(\text{CH}_3)$ 28, v(CC) 15, v(CO) 10		1463.3	122.74	$\delta(\text{CH}_3)$ 22, v(CC _{alk}) 14
		1452	1457						1455.7	2.62	$\delta(\text{CH}_3)$ 55
				1449.0	68.0	25.2	$\delta(\text{CH}_3)$ 61		1449.3	349.42	$\delta(\text{CH}_3)$ 57, v(N-C) 6
				1445.2	16.2	14.4	$\delta(\text{CH}_3)$ 59				
1424				1436.4	48.1	76.8	$\delta(\text{CH}_3)$ 39, v(C=N) 11		1436.4	25.61	$\delta(\text{CH}_3)$ 28, $\delta(\text{CC}_{\text{alk}})$ 5, v(C=O) 6
1413		1418sh	1415	1415.4	33.1	6.1	$\delta(\text{CH}_3)$ 80		1415.2	12.03	$\delta(\text{CH}_3)$ 70
1397sh	1407	1401		1408.0	72.1	12.1	$\delta(\text{CH}_3)$ 73, v(C _{alk} C _{alk}) 10		1406.5	72.58	$\delta(\text{CH}_3)$ 77, v(C _{alk} C _{alk}) 11
				1406.4	26.7	7.3	$\delta(\text{CH}_3)$ 65, v(C _{alk} C _{alk}) 9		1404.4	27.62	$\delta(\text{CH}_3)$ 68
1371		1372	1372						1392.5	36.39	v(CC) 23, v(CO) 21, v(C=O) 15, $\delta(\text{OD})$ 9
1358	1362			1354.2	101.0	35.5	v(C _{ar} C _{alk}) 21, v(CO) 12, $\delta(\text{CH}_3)$ 6, $\delta(\text{CC=O})$ 6, v(CC) 5		1368.8	24.54	v(CC) 17, v(CO) 13, v(CC _{alk}) 12, $\delta(\text{NCC}_{\text{alk}})$ 10, $\delta(\text{OD})$ 9, v(C _{alk} C _{alk}) 6
		1336	1336				v(C _{ar} C _{alk}) 20, v(CC) 26		1342.0	39.17	v(CC _{alk}) 30, v(CC) 16, v(CC) 7
1319		1319	1317	1323.0	136.2	102.9	v(CC) 29, $\delta(\text{CC=O})$ 6, v(C=O) 6, $\delta(\text{CCC}_{\text{alk}})$ 5, v(C _{ar} C _{alk}) 5		1330.2	160.93	v(CC) 25, v(CC _{alk}) 14, $\delta(\text{C=O})$ 10, v(CC) 9
1301	1301	1301	1301	1312.3	131.9	61.6	v(CC) 30, v(C _{ar} C _{alk}) 16, $\delta(\text{CC=O})$ 6		1293.2	129.43	v(CC) 21, $\delta(\text{C=O})$ 15, v(CC _{alk}) 8, $\delta(\text{CC}_{\text{alk}})$ 7, v(C _{alk} C _{alk}) 6
1263	1267	1280	1280	1283.0	10.2	12.1	v(CO) 22, v(C _{ar} C _{alk}) 12, v(CC) 13				
			1228	1266.0	29.3	41.6	v(CC) 27, v(CO) 13, v(C _{alk} C _{alk}) 6, $\delta(\text{CC=O})$ 6		1233.2	4.90	v(N-C) 14, $\delta(\text{ND})$ 13, v(CO) 9, v(CC _{alk}) 8, $\delta(\text{CC}_{\text{alk}})$ 7, $\delta(\text{CC}_{\text{alk}}\text{C}_{\text{alk}})$ 7,
									1199.5	50.65	v(CO) 20, $\delta(\text{OD})$ 10, $\delta(\text{CC}_{\text{alk}})$ 9, v(CC) 8, $\delta(\text{C=O})$ 5
1162			1188	1187.1	30.7	0.7	$\delta(\text{NCH}_3)$ 40, v(NC) 8, $\delta(\text{CNH})$ 7, v(CO) 5, $\delta(\text{CC}_{\text{im}}\text{N})$ 5		1181.7	39.23	$\delta(\text{OD})$ 20, v(CC) 15, $\delta(\text{CO})$ 6, v(CO) 6, $\delta(\text{CC}_{\text{alk}})$ 5
			1166	1163.4	3.1	28.9	$\delta(\text{CCC}_{\text{im}})$ 13, v(CC) 20, $\delta(\text{CCO})$ 7, $\delta(\text{CCC}_{\text{alk}})$ 7, v(C _{alk} C _{alk}) 6		1165.5	6.93	$\delta(\text{NCH}_3)$ 63
1128	1147sh	1139	1142	1151.8	4.3	2.9	$\delta(\text{NCH}_3)$ 15, $\delta(\text{CCC}_{\text{alk}})$ 10, v(CO) 9, v(CC) 8				
1104sh				1134.0	11.7	0.4	$\delta(\text{NCH}_3)$ 82, $\gamma(\text{COH})$ 6		1131.9	0.21	$\delta(\text{NCH}_3)$ 89, $\delta(\text{CH}_3)$ 5
1013				1130.6	142.3	0.3	$\gamma(\text{COH})$ 93, $\delta(\text{NCH}_3)$ 6				
			958	1128.1	0.5	0.5	$\gamma(\text{COH})$ 98				
			1168				$\delta(\text{OD})$		1105.4	25.94	$\delta(\text{CH}_3)$ 19, $\delta(\text{OD})$ 17, v(N-C) 11, v(CC) 8
			1099	1097.0	3.3	3.6	$\delta(\text{CH}_3)$ 12, $\alpha(\text{CC})$ 11, v(C _{alk} C _{alk}) 11, v(CC) 14, $\delta(\text{CC}_{\text{im}}\text{N})$ 6		1086.7	13.99	$\delta(\text{CH}_3)$ 20, $\delta(\text{OD})$ 14, $\delta(\text{CC})$ 8, v(CC) 9, v(C _{alk} C _{alk}) 5
1061	1076			1079.1	18.0	23.6	v(NC) 37, $\delta(\text{CH}_3)$ 11, v(C _{alk} C _{alk}) 7, v(CC) 7		1074.9	26.76	$\delta(\text{OD})$ 38, v(OD) 6
1029	1056	1050	1055	1057.9	0.5	0.2	$\delta(\text{CH}_3)$ 71, $\gamma(\text{C=N})$ 9, $\tau(\text{C=N})$ 7		1050.7	3.22	$\delta(\text{CH}_3)$ 78, $\gamma(\text{C=N})$ 13
	1029	1034	1033	1049.1	8.9	0.2	$\delta(\text{CH}_3)$ 73, $\gamma(\text{C=O})$ 18		1049.5	3.97	$\delta(\text{CH}_3)$ 63, $\gamma(\text{C=O})$ 18
				1048.1	3.4	0.3	$\delta(\text{CH}_3)$ 73, $\gamma(\text{C=O})$ 19		1048.2	2.15	$\delta(\text{CH}_3)$ 63, $\gamma(\text{C=O})$ 19
				1003.9	48.8	11.8	$\delta(\text{CH}_3)$ 48, v(C _{alk} C _{alk}) 10				
				1003.2	63.5	0.4	$\gamma(\text{CNH})$ 46, $\tau(\text{C=N})$ 27, $\delta(\text{CH}_3)$ 11,				

981	975	982	976		997.4	29.9	12.9	$\gamma(C=N)$ 8 $\delta(CH_3)$ 51, $v(C_{alk}C_{alk})$ 15, $v(C_{ar}C_{alk})$ 6, $\delta(CC=O)$ 5, $v(C=O)$ 5 $\delta(CC=O)$ 5, $v(C=O)$ 5	997.2	55.13	$\delta(CH_3)$ 45, $v(C_{alk}C_{alk})$ 13, $\delta(OD)$ 5	
	969sh	953	967		965.7	2.3	13.2	$\delta(CH_3)$ 21, $v(NC)$ 12, $v(C_{alk}C_{alk})$ 14, $v(C=N)$ 6, $v(C_{ar}C_{alk})$ 6	992.3	25.36	$\delta(CH_3)$ 44, $v(C_{alk}C_{alk})$ 16, $v(CC_{alk})$ 5 $\delta(ND)$ 14, $\delta(CH_3)$ 14, $v(C_{alk}C_{alk})$ 20, $v(CC_{alk})$ 9, $v(C=N)$ 7 $v(N-C)$ 33, $\delta(ND)$ 28, $v(ND)$ 8, $\delta(CH_3)$ 7	
932								$\gamma(COH)$ $v(C_{alk}C_{alk})$ 25, $v(C_{ar}C_{alk})$ 24, $\alpha(CC)$ 8, $\delta(CCC_{im})$ 7	963.9	0.27		
831	945	945	951	873	838	842.6	17.4	2.5	$v(C_{alk}C_{alk})$ 23, $\delta(CCO)$ 14, $\delta(CCC_{alk})$ 11, $v(C_{ar}C_{alk})$ 9, $\alpha(CC)$ 5	956.6	3.31	
802	831	920	923	825		820.5	15.8	2.2	$\delta(CCO)$ 48, $\delta(CCC_{alk})$ 26, $\delta(CCC_{im})$ 8	839.8	22.08	$v(C_{alk}C_{alk})$ 43, $\delta(CC_{alk})$ 9, $\delta(CC)$ 8, $\delta(CO)$ 7
	805		885	751		798.0	2.4	5.6		813.0	11.91	$v(C_{alk}C_{alk})$ 26, $v(CC_{alk})$ 16, $\delta(CO)$ 11, $\delta(CC)$ 6, $\delta(CC_{alk})$ 10
770	775		750		779.4	8.0	0.1	$\tau(CC)$ 45, $\gamma(CO)$ 31, $\gamma(CC_{alk})$ 13	790.4	26.34	$\gamma(C-O)$ 28, $\gamma(C_{alk}C)$ 14	
	747								773.7	4.21	$\delta(CO)$ 45, $\delta(CC_{alk})$ 36	
	724				736.0	0.1	0.2	$\tau(CC)$ 28, $\gamma(CO)$ 49, $\gamma(CC_{alk})$ 11	752.1	8.96	$\tau(CC)$ 37, $\gamma(C-O)$ 31, $\gamma(C_{alk}C)$ 17, $\gamma(ND)$ 5	
736	731				744.7	1.9	0.2	$\gamma(CO)$ 47, $\gamma(CC_{alk})$ 18, $\tau(CC)$ 27	825.4	2.68	$\gamma(COD)$ 79, $\gamma(C-O)$ 7	
	695								824.1	90.43	$\gamma(COD)$ 77, $\gamma(C-O)$ 8, $\tau(CC)$ 5	
652	654	650		662	659.2	19.1	11.3	$\delta(CC=O)$ 32, $\delta(CC_{im}N)$ 16, $\alpha(CC)$ 8, $\delta(CNH)$ 7	731.8	1.98	$\gamma(C-O)$ 44, $\tau(CC)$ 25, $\gamma(C_{alk}C)$ 15	
622	647	618	613		628.3	98.8	4.2	$\delta(CC=O)$ 59, $\alpha(CC)$ 6	724.6	4.63	$\gamma(C-O)$ 50, $\tau(CC)$ 24, $\gamma(C_{alk}C)$ 5, $\gamma(ND)$ 5	
591	612			600	600	624.5	15.8	14.9	$\alpha(CC)$ 30, $\delta(CC=O)$ 11, $v(CC)$ 26, $v(CO)$ 7	655.8	22.13	$\delta(C=O)$ 35, $\delta(NC_{alk})$ 9, $\delta(CC)$ 7, $\delta(ND)$ 7
						604.1	14.4	1.6	$\delta(CC_{im}N)$ 34, $\delta(CNH)$ 13, $\delta(CC=O)$ 11, $\alpha(CC)$ 8	624.7	83.65	$\delta(C=O)$ 54, $\delta(CC)$ 14
	590sh	583	583			599.7	0.1	0.4	$\gamma(C=O)$ 47, $\gamma(CC_{alk})$ 10, $\gamma(CO)$ 10, $\delta(CH_3)$ 8	620.2	27.23	$\delta(CC)$ 25, $\delta(C=O)$ 24, $v(CC)$ 19, $v(CO)$ 7,
									597.8	0.12	$\gamma(C=O)$ 49, $\delta(CH_3)$ 16, $\delta(ND)$ 15, $\gamma(C_{alk}C)$ 15, $\gamma(C-O)$ 10	
566	547			572	-	593.0	0.4	0.5	$\gamma(C=O)$ 32, $\gamma(C=N)$ 18, $\gamma(CO)$ 17, $\delta(CH_3)$ 7, $\gamma(CC_{alk})$ 5	596.6	11.59	$\delta(NC_{alk})$ 27, $\delta(C_{alk}C_{alk})$ 7, $\delta(CC)$ 7, $\delta(C=O)$ 5
						576.3	0.2	0.1	$\gamma(C=N)$ 34, $\gamma(C=O)$ 24, $\delta(CH_3)$ 8, $\tau(CC)$ 7, $\gamma(CO)$ 6	588.6	0.32	$\gamma(C=O)$ 43, $\gamma(C-O)$ 17, $\delta(CH_3)$ 14, $\gamma(C=N)$ 7
497	485	502		499	503.5	8.4	10.3	$\delta(CCO)$ 35, $\delta(CC=O)$ 16, $\delta(CCC_{alk})$ 6	568.1	2.03	$\gamma(C=N)$ 46, $\delta(CH_3)$ 12, $\gamma(C=O)$ 8, $\gamma(C-O)$ 6	
			483		487.9	4.0	2.0	$\delta(CCO)$ 42, $v(CC)$ 13	490.8	11.73	$\delta(CO)$ 29, $\delta(C=O)$ 25, $\delta(CC_{alk})$ 7	
442	432	443	436		449.3	23.8	4.2	$\delta(CCC_{alk})$ 32, $\delta(CC=O)$ 21, $\delta(CCC_{im})$ 8, $\delta(CCO)$ 14	479.1	5.71	$\delta(CO)$ 40, $\delta(C_{alk}C_{alk})$ 12, $v(CC)$ 12, $v(OD)$ 5	
					411.2	11.2	7.6	$\alpha(CC)$ 16, $\delta(CC=O)$ 23, $v(C_{ar}C_{alk})$ 8,	442.9	17.68	$\delta(C=O)$ 31, $\delta(CC_{alk})$ 22, $\delta(CO)$ 12, $\delta(CC)$ 5	

							v(OH) 7, δ (CCC _{im}) 7				
					403.6	13.9	2.3	α (CC) 18, δ (CC=O) 38, δ (CCO) 17, 403.5	13.96	δ (C=O) 24, δ (CC) 12, δ (ND) 9, v(CC _{alk}) 9, v(OD) 9, δ (CC _{alk}) 6	
		383	372		390.4	2.1	6.9	v(OH) 6		δ (C=O) 31, δ (CO) 23, v(OD) 10, δ (CC _{alk}) 9, δ (CC) 8	
385	384sh			389 386			δ (CC _{imN}) 47, δ (CCO) 19, δ (CNH) 6, α (CC) 5	392.1	13.25		
	373				349.9	17.4	2.7	v _o (OO) and v _o (ON)		δ (CC _{alk} C _{alk}) 38, δ (CO) 22, δ (CC) 11, v(OD) 6	
		363		339 347	360.9	7.4	4.2	α (CC) 26, δ (CC=O) 16, δ (CNH) 6, δ (CCC _{im}) 6, v(CC) 6, v(C _{ar} C _{alk}) 6	384.6	8.35	δ (CC) 24, v(CC _{alk}) 13, v(CC) 7, δ (C=O) 6
351	350	332	331		332.5	0.0	0.2	α (CC) 24, v(C _{ar} C _{alk}) 12, v(CC) 7, δ (CC=O) 17, δ (CCO) 6	359.6	7.25	γ (CC _{alk}) 40, τ (CC) 21, τ (CN) 11, γ (COH) 8, γ (CNH) 7
331	329			305 302	327.5	2.4	20.6	γ (CC _{alk}) 40, τ (CC) 21, τ (CN) 11, γ (COH) 8, γ (CNH) 7	347.8	15.92	δ (CC) 25, δ (C=O) 17, δ (ND) 8, δ (CC _{alk}) 7, v(CC _{alk}) 6, v(CC) 5
							δ (CC=O) 26, v(C _{ar} C _{alk}) 12, δ (CCC _{im}) 7, v(CC) 6	330.8	0.02	γ (C _{alk} C) 42, τ (CC) 21, τ (C=N) 12, τ (COD) 8, γ (ND) 6	
								326.0	2.05	δ (C=O) 25, v(CC _{alk}) 7, v(CC) 6, δ (ND) 6, δ (CC _{alk}) 6	
302	297	303	298		303.7	0.0	0.3	γ (CC _{alk}) 58, τ (CC) 22, γ (COH) 8	303.2	0.05	γ (C _{alk} C) 58, τ (CC) 22, τ (COD) 8
		283		278 275	296.7	14.5	4.0	δ (CC _{imN}) 35, δ (CCC _{alk}) 26, δ (CCC _{im}) 13, δ (CNH) 9, δ (CC=O) 7	293.7	14.59	δ (CC _{alk}) 27, δ (NC _{alk}) 22, δ (C _{alk} C _{alk}) 13, δ (ND) 9, δ (C=O) 6
227	222	229	223	248 252	243.3	0.8	0.3	τ (CH ₃) 43, τ (CC) 24, γ (CC _{alk}) 9	242.9	0.75	τ (C _{alk} C _{alk}) 45, τ (CC) 23, γ (C _{alk} C) 9
				217	225.5	0.9	0.9	τ (CC) 22, τ (CH ₃) 34, γ (CC _{alk}) 19	224.8	0.80	τ (C _{alk} C _{alk}) 37, τ (CC) 21, γ (C _{alk} C) 18
					218.8	12.7	4.1	δ (CCC _{alk}) 72, δ (CCO) 5	218.6	12.67	δ (CC _{alk}) 78
					213.9	0.0	0.8	τ (CH ₃) 72, δ (CH ₃) 5	213.8	0.01	τ (C _{alk} C _{alk}) 75, δ (CH ₃) 6
				173	197.2	0.0	0.1	τ (CC) 21, τ (CH ₃) 26, γ (CC _{alk}) 26, τ (C=N) 6	196.3	0.00	τ (C _{alk} C _{alk}) 24, γ (C _{alk} C) 26, τ (CC) 22, τ (C=N) 7
150	144				181.3	0.9	0.9	γ (CNH) 30, τ (CH ₃) 23, τ (NC) 17, τ (CC) 7	179.9	0.77	γ (ND) 29, τ (N-C) 20, τ (C=N) 18, τ (C _{alk} C _{alk}) 6, τ (CC) 6
					167.0	0.4	0.3	τ (NC) 32, γ (CNH) 26, γ (COH) 10, τ (C=N) 9, τ (CC) 12	166.7	0.44	τ (N-C) 29, γ (ND) 28, τ (CC) 12, τ (C=N) 10, τ (COD) 10
					166.5	1.2	1.6	δ (CCC _{im}) 34, δ (CCC _{alk}) 14, δ (CC _{imN}) 11, δ (CCO) 6 δ (CNH) 6	165.3	1.17	δ (CC _{alk}) 46, δ (NC _{alk}) 11, δ (ND) 9, δ (CO) 6
				136	127.4	0.3	0.0	γ (COH) 36, τ (CH ₃) 25, τ (CC _{alk}) 13, τ (CC) 12	127.4	0.28	τ (C=O) 37, τ (COD) 36, τ (CC) 12
124	125			122	123.0	2.6	0.0	γ (COH) 39, τ (CC _{alk}) 31, τ (CC) 5	123.0	2.63	τ (COD) 39, τ (C=O) 31, τ (CC) 5
		93			73.8	0.4	0.0	τ (CC _{alk}) 37, τ (C=N) 22, τ (CH ₃) 11, τ (CC) 7	73.6	0.46	τ (C=N) 47, τ (C=O) 11, τ (C _{alk} C _{alk}) 11, τ (CC) 5
					55.5	3.6	0.0	τ (CC) 28, τ (CC _{alk}) 42, τ (CH ₃) 12	55.2	3.50	τ (C=O) 43, τ (CC) 28, τ (C _{alk} C _{alk}) 12
					42.6	0.7	0.1	τ (CC) 55, τ (CC _{alk}) 28	42.3	0.63	τ (CC) 55, τ (C=O) 28
					31.6	1.0	0.1	τ (CC _{alk}) 33, τ (CC) 44, τ (CH ₃) 15	31.4	1.02	τ (CC) 43, τ (C=N) 33, τ (C _{alk} C _{alk}) 12

* Sub-picks of stretching vibration mode v(OH): 1924, 1744 and v(OD): 2078, 1803. These bands are sensitive on deuteronsubstitution in hydrogen bonding. **Assignments made on base of experimental deuteronsubstitution spectrum.

Table S4. Selected normal modes of compound **1** calculated by DFPT method (frequencies in cm^{-1}).

$v(\text{OH}) = 2082$	$v(\text{OH}) = 2074$	$v(\text{OH}) = 2063$
$\delta(\text{OH}) = 1639$	$\delta(\text{OH}) = 1664$	$\delta(\text{OH}) = 1665$
$\delta(\text{OH}) = 1568$	$\delta(\text{OH}) = 1566$	$\delta(\text{OH}) = 1536$
$\delta(\text{OH}) = 1477$	$\delta(\text{OH}) = 1467$	
$\gamma(\text{OH}) = 1170$	$\gamma(\text{OH}) = 1156$	$\gamma(\text{OH}) = 1153$
$v_\sigma = 492$	$v_\sigma = 491$	$v_\sigma = 425$
$v_\beta = 146$	$v_\beta = 135$	$v_\beta = 118$

Table S5. Selected normal modes of compound **2** calculated by DFTP method (frequencies in cm^{-1}).

$\nu(\text{NH}) = 2798$	$\nu(\text{OH}) = 2133$	$\nu(\text{OH}) = 1958$	
$\delta(\text{OH}) = 1656$	$\delta(\text{OH}) = 1632$	$\delta(\text{OH}) = 1630$	
$\delta(\text{OH}) = 1617$	$\delta(\text{NH}) = 1562$	$\delta(\text{NH}) = 1263$	
$\gamma(\text{OH}) = 1181$	$\gamma(\text{OH}) = 1132$	$\gamma(\text{NH}) = 1021$	$\gamma(\text{NH}) = 979$
780	779	617, 616, 615, 614	
610, 609	591 - 590	554	
$\nu_{\sigma} = 500, 499, 498, 497$	$\nu_{\sigma} = 485$	$\nu_{\sigma} = 483-482$	$\nu_{\sigma} = 454, 450$

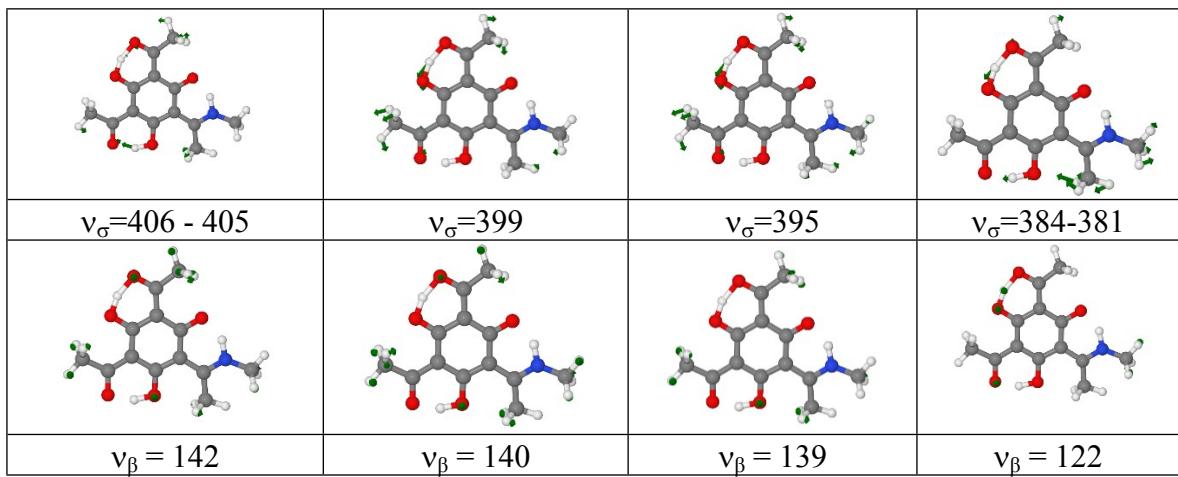


Table S6. Crystal data and structure refinement for 1,1',1''-(2,4,6-trihydroxybenzene-1,3,5-triyl)triethanone (**1**) and (1*E*)-1-(3,5-diacetyl-2,4-dihydroxy-6-oxocyclohexa-2,4-dien-1-yl)-N-methylethaniminium (**2**).

Identification code	1	2
Empirical formula	C ₁₂ H ₁₂ O ₆	C ₁₃ H ₁₅ N O ₅
Formula weight	252.22	265.26
Temperature	100(2) K	100(2) K
Wavelength	0.71073 Å	0.71073 Å
Crystal system, space group	Monoclinic, P2 ₁ /c	Orthorhombic, Pbca
Unit cell dimensions	a = 9.298(3) Å b = 16.631(4) Å β = 106.90(4) ° c = 7.232(3) Å	a = 17.6428(7) Å b = 7.0164(3) Å c = 19.1027(9) Å
Volume	1070.0(6) Å ³	2364.70(18) Å ³
Z, Calculated density	4, 1.566 mg/m ³	8, 1.490 mg/m ³
Absorption coefficient	0.127 mm ⁻¹	0.115 mm ⁻¹
F(000)	528	1120
Crystal size	0.15 x 0.13 x 0.11 mm	0.25 x 0.10 x 0.08 mm
Theta range for data collection	3.19 to 26.07 °	3.14 to 28.67 °
Limiting indices	-11<=h<=11, -20<=k<=19, -8<=l<=8	-22<=h<=19, -9<=k<=9, -21<=l<=25
Reflections collected / unique	11575/2111 [R(int) = 0.0371]	11591/2870 [R(int) = 0.0620]
Absorption correction	Analytical	Analytical
Max. and min. transmission	0.911 and 0.798	0.678 and 0.567
Completeness to theta = 25.00	99.8 %	99.6%
Refinement method	Full-matrix least-squares on F ²	Full-matrix least-squares on F ²
Data / restraints / parameters	2111/0/166	2870/0/176
Goodness-of-fit on F ²	0.94	0.90
Final R indices [I>2sigma(I)]	R1 = 0.0349, wR2 = 0.0886	R1 = 0.0492, wR2 = 0.0983
R indices (all data)	R1 = 0.0517, wR2 = 0.0941	R1 = 0.1082, wR2 = 0.1089
Largest diff. peak and hole	0.339 and -0.266 e.Å ⁻³	0.320 and -0.246 e.Å ⁻³