## **Supplementary Information**

to the article

**Title:** Self-assembly of cucurbiturils and cyclodextrins to supramolecular millstones with naphthalene derivatives capable of translocations in the host cavities

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**Fig. S1** <sup>1</sup>H NMR spectrum of compound **1** (500.13 MHz, DMSO- $d_6$ , 25 °C).



**Fig. S2** <sup>1</sup>H NMR spectrum of compound **2** (500.13 MHz, DMSO- $d_6$ , 26 °C).



**Fig. S3** <sup>1</sup>H NMR spectrum of compound **3** (500.13 MHz, DMSO- $d_6$ , 26 °C).



**Fig. S4** <sup>1</sup>H NMR spectrum of compound **4** (500.13 MHz, DMSO- $d_6$ , 26 °C).



**Fig. S5** <sup>1</sup>H NMR spectrum of compound **5** (500.13 MHz, DMSO- $d_6$ , 28 °C).

3-C, 5-C, 3'-C



**Fig. S6** <sup>13</sup>C NMR spectrum of compound **1** (125.76 MHz, DMSO- $d_6$ , 25 °C).





**Fig. S7** <sup>13</sup>C NMR spectrum of compound **2** (125.76 MHz, DMSO- $d_6$ , 26 °C).



**Fig. S8** <sup>13</sup>C NMR spectrum of compound **3** (125.76 MHz, DMSO- $d_6$ , 25 °C).



**Fig. S9** <sup>13</sup>C NMR spectrum of compound 4 (125.76 MHz, DMSO- $d_6$ , 25 °C).



**Fig. S10** <sup>13</sup>C NMR spectrum of compound **5** (125.76 MHz, DMSO- $d_6$ , 26 °C).



**Fig. S11** <sup>1</sup>H NMR spectrum of complex (1)<sub>2</sub>@CB[8]·6.5H<sub>2</sub>O, which was obtained by crystallization ( $C_{\text{complex}} = 3 \times 10^{-4} \text{ M}$ ), D<sub>2</sub>O, 25 °C.



**Fig. S12** <sup>1</sup>H NMR spectrum of complex 2(**5**)@ $\gamma$ -CD·6H<sub>2</sub>O, which was obtained by crystallization ( $C_{\text{complex}} = 3 \times 10^{-4} \text{ M}$ ), D<sub>2</sub>O, 25 °C.



**Fig. S13** <sup>1</sup>H NMR spectrum of complex (5)<sub>2</sub>@CB[8]·5H<sub>2</sub>O, which was obtained by crystallization (sat.,  $C_{\text{complex}} < 1 \times 10^{-4}$  M), D<sub>2</sub>O, 25 °C.



regions) of (a, c) compound 4 and (b, d) a 1:6.0 mixture of compound 4 regions) of (a, c) compound 4 and (b, d) a 1:7.4 mixture of compound and  $\beta$ -CD ( $C_4 = 5.0 \times 10^{-4}$  M), D<sub>2</sub>O–MeCN- $d_3$  (10:1, v/v), 25 °C.

Fig. S14 <sup>1</sup>H NMR spectra ((a, b) aromatic and (c, d) aliphatic proton Fig. S15 <sup>1</sup>H NMR spectra ((a, b) aromatic and (c, d) aliphatic proton **4** and  $\gamma$ -CD ( $C_4 = 4.3 \times 10^{-4}$  M), D<sub>2</sub>O–MeCN- $d_3$  (10:1, v/v), 25 °C.



Fig. S16 <sup>1</sup>H NMR spectra ((a, b) aromatic and (c, d) aliphatic proton Fig. S17 <sup>1</sup>H NMR spectra ((a, b) aromatic and (c, d) aliphatic proton regions) of (a, c) compound 5 and (b, d) a 1:5.8 mixture of compound 5 regions) of (a, c) compound 5 and (b, d) a 1:6.1 mixture of compound and  $\beta$ -CD ( $C_5 = 5.2 \times 10^{-4}$  M), D<sub>2</sub>O–MeCN- $d_3$  (10:1, v/v), 25 °C.

**5** and  $\gamma$ -CD ( $C_5 = 5.2 \times 10^{-4}$  M), D<sub>2</sub>O–MeCN- $d_3$  (10:1, v/v), 25 °C.



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6.4×10<sup>-4</sup> M), D<sub>2</sub>O–MeCN-d<sub>3</sub> (10:1, v/v), 25 °C.



**Fig. S20** <sup>1</sup>H NMR spectrum of a 1:1.9 mixture of compound **5** and CB[7] ( $C_5 = 6.4 \times 10^{-4}$  M), D<sub>2</sub>O–MeCN- $d_3$  (10:1, v/v), 25 °C.



**Fig. S21** <sup>1</sup>H NMR spectra of (*a*) compound **2** and (*b*) a 1:1.9 mixture of compound **2** and CB[7] ( $C_2 = 5.7 \times 10^{-4}$  M), D<sub>2</sub>O–MeCN- $d_3$  (10:1, v/v), 25 °C.



**Fig. S22** NOESY spectrum of an equimolar mixture of compound **3** and  $\beta$ -CD ( $C_3 = C_{CD} = 6 \times 10^{-3}$  M), D<sub>2</sub>O, 25 °C.



**Fig. S23** Absorption spectrum of compound **1** ( $C = 2 \times 10^{-5}$  M), water, ambient temperature, 1-cm quartz cell. **Fig. S24** Fluorescence spectrum of compound **1** ( $C = 1 \times 10^{-6}$  M), water, ambient temperature. The fluorescence was excited by light at 356 nm.



Fig. S25 (*a*) Absorption and (*b*) fluorescence spectra of compound 2 ( $C_2 = 2 \times 10^{-5}$  M for absorption and  $C_2 = 1 \times 10^{-5}$  M for fluorescence) and respective evaluated spectra of complex 2@ $\beta$ -CD, water, ambient temperature. The fluorescence was excited by light at 313 nm.



Fig. S26 (*a*) Absorption and (*b*) fluorescence spectra of compound 4 ( $C_4 = 2 \times 10^{-5}$  M for absorption and  $C_4 = 1 \times 10^{-5}$  M for fluorescence) and respective evaluated spectra of complex 4@ $\beta$ -CD, water, ambient temperature. The fluorescence was excited by light at 319 nm.

(*l* 



**Fig. S27** (*a*) Absorption and (*b*) fluorescence spectra of compound **5** ( $C_5 = 2 \times 10^{-5}$  M for absorption and  $C_5 = 1 \times 10^{-5}$  M for fluorescence) and respective evaluated spectra of complex **5**@ $\beta$ -CD, water, ambient temperature. The fluorescence was excited by light at 311 nm.



**Fig. S28** (*a*) Absorption and (*b*) fluorescence spectra of compound **2** ( $C_2 = 2 \times 10^{-5}$  M for absorption and  $C_2 = 1 \times 10^{-5}$  M for fluorescence) and respective evaluated spectra of complex **2**@ $\gamma$ -CD, water, ambient temperature. The fluorescence was excited by light at 332 nm.



**Fig. S29** (*a*) Absorption and (*b*) fluorescence spectra of compound 4 ( $C_4 = 2 \times 10^{-5}$  M for absorption and  $C_4 = 1 \times 10^{-5}$  M for fluorescence) and respective evaluated spectra of complex 4@ $\gamma$ -CD, water, ambient temperature. The fluorescence was excited by light at 319 nm.



**Fig. S30** (*a*) Absorption and (*b*) fluorescence spectra of compound **5** ( $C_5 = 2 \times 10^{-5}$  M for absorption and  $C_5 = 1 \times 10^{-5}$  M for fluorescence) and respective evaluated spectra of complex **5**@ $\gamma$ -CD, water, ambient temperature. The fluorescence was excited by light at 321 nm.



**Fig. S31** (*a*) Absorption and (*b*) fluorescence spectra of compound **2** ( $C_2 = 2 \times 10^{-5}$  M for absorption and  $C_2 = 1 \times 10^{-6}$  M for fluorescence) and respective evaluated spectra of complexes **2**@CB[7] and **2**@(CB[7])<sub>2</sub>, water, ambient temperature. The fluorescence was excited by light at 356 nm.



**Fig. S32** (*a*) Absorption and (*b*) fluorescence spectra of compound 4 ( $C_4 = 2 \times 10^{-5}$  M for absorption and  $C_4 = 1 \times 10^{-6}$  M for fluorescence) and respective evaluated spectra of complexes 4@CB[7] and 4@(CB[7])<sub>2</sub>, water, ambient temperature. The fluorescence was excited by light at 356 nm.



**Fig. S33** (*a*) Absorption and (*b*) fluorescence spectra of compound **5** ( $C_5 = 2 \times 10^{-5}$  M for absorption and  $C_5 = 1 \times 10^{-6}$  M for fluorescence) and respective evaluated spectra of complexes **5**@CB[7] and **5**@(CB[7])<sub>2</sub>, water, ambient temperature. The fluorescence was excited by light at 359 nm.



**Fig. S34** (*a*) Absorption and (*b*) fluorescence spectra of compound **5** ( $C_5 = 2 \times 10^{-5}$  M for absorption and  $C_5 = 1 \times 10^{-6}$  M for fluorescence) and respective evaluated spectra of complexes **5**@CB[8] and (**5**)<sub>2</sub>@CB[8] (per molecule of naphthylpyridine derivative), water, ambient temperature. The fluorescence was excited by light at 333 nm.