

Supporting Information

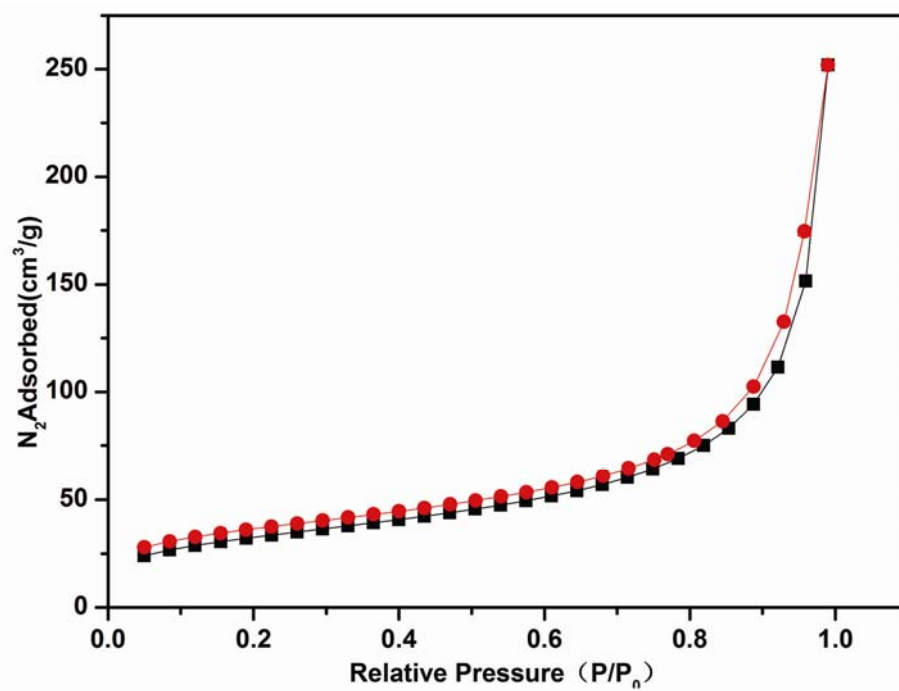
Porous organic polymer based on melamine and 5,5'-bis(bromomethyl)-2,2'-bipyridine: functionalization with lanthanide ions for chemical sensing, and highly efficient adsorption of methyl orange

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Materials	SBET (m ² /g)	Pore volume (cm ³ /g)	Pore size (nm)
Polymer	114	0.199	7

Fig. S1 N₂ adsorption-desorption isotherms for POP-1.

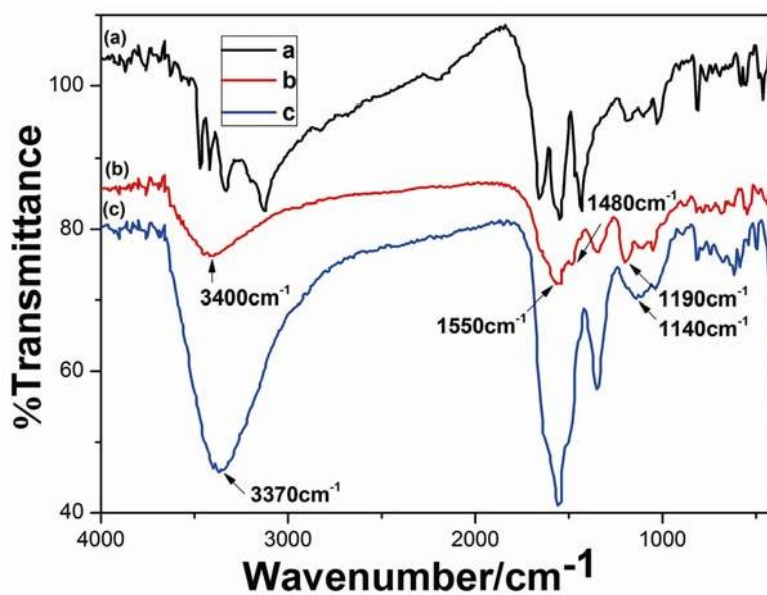


Fig. S2 1. Fourier transform infrared spectra (FT-IR) of (a) Melamine ; (b) POP-1; (c) Eu³⁺@POP-1.

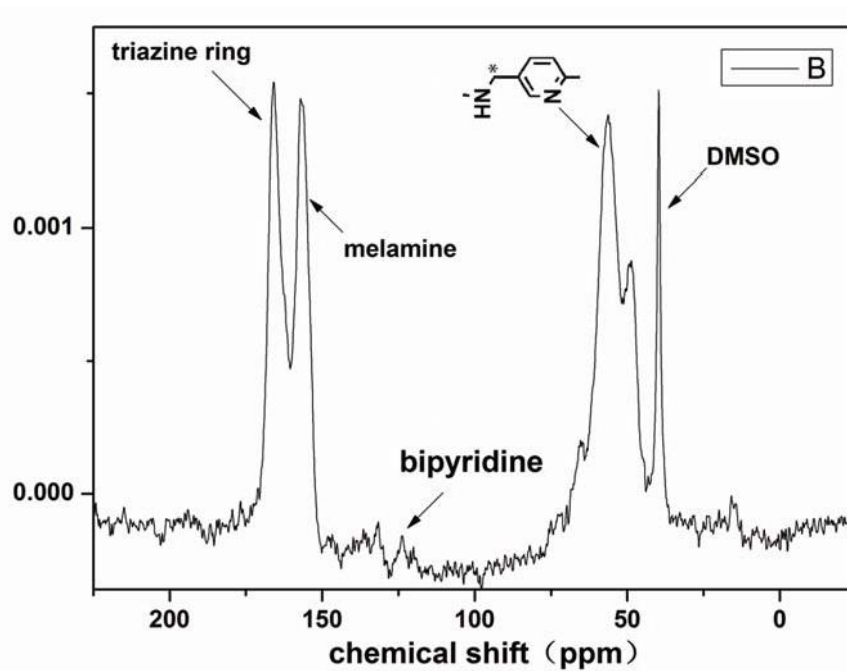


Fig.S3 Cross-polarization (CP) C^{13} MAS natural abundance NMR spectrum of POP-1

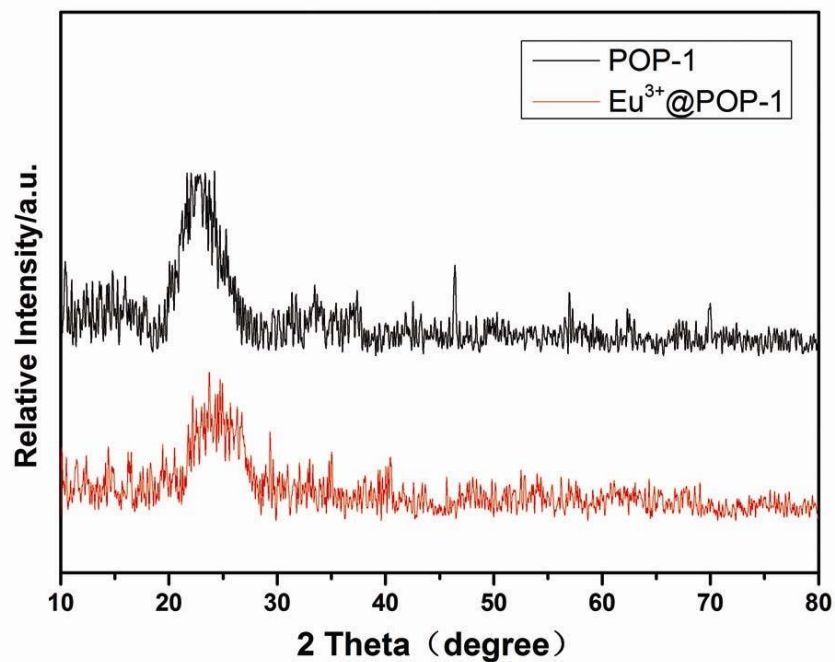


Fig. S4 Powder XRD patterns of POP-1 and $Eu^{3+}@POP-1$.

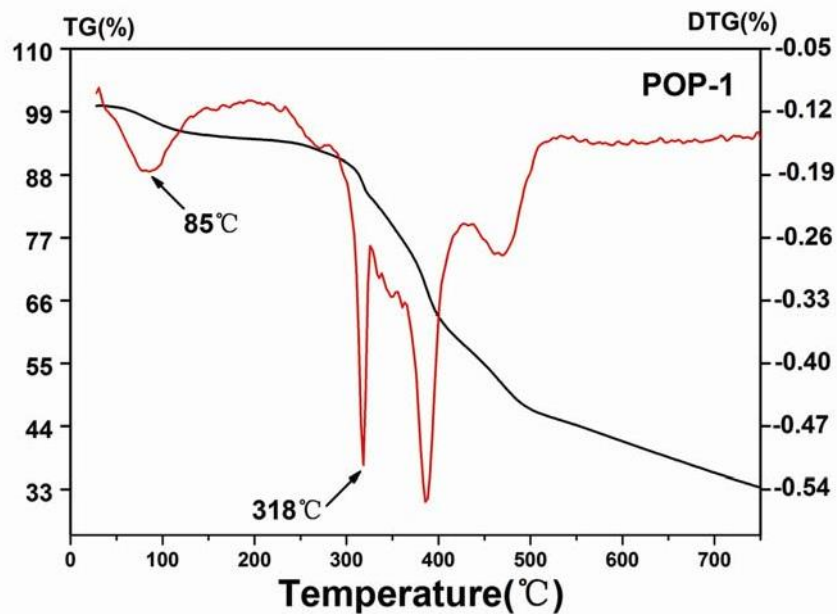


Fig. S5 Thermogravimetric analysis of POP-1.

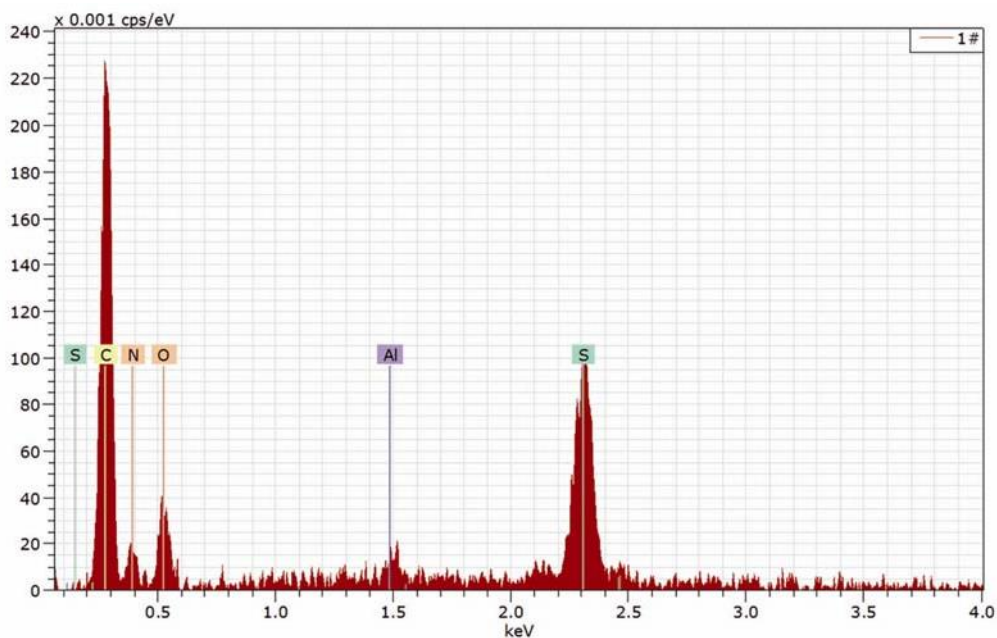


Fig. S6 Energy dispersive X-ray (EDX) analysis of POP-1.

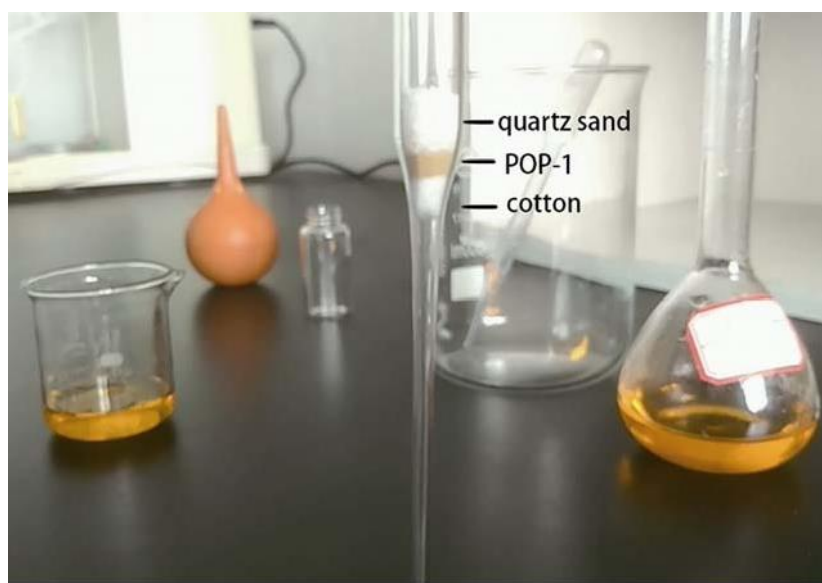


Fig. S7 The adsorption device for MO.

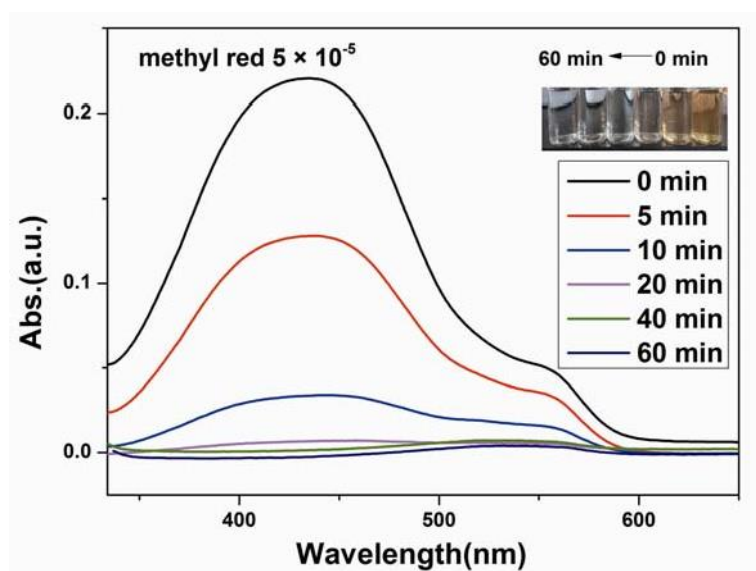


Fig. S8 UV-vis spectra of methyl red with POP-1 at given intervals, respectively. Inset: photographic images of dye solutions for different periods of time.

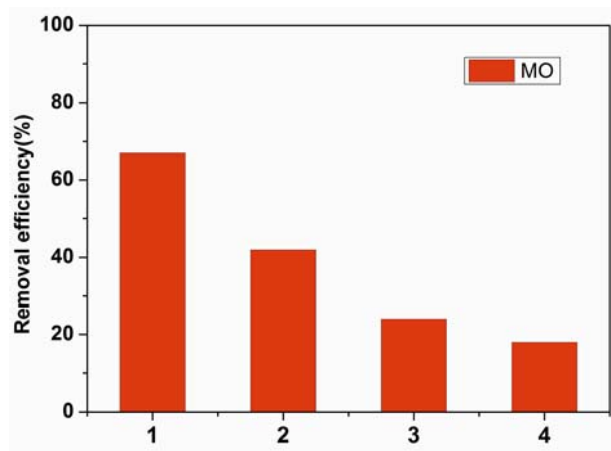


Fig. S9. The removal efficiency of as-synthesized adsorbent before and after four cycles.

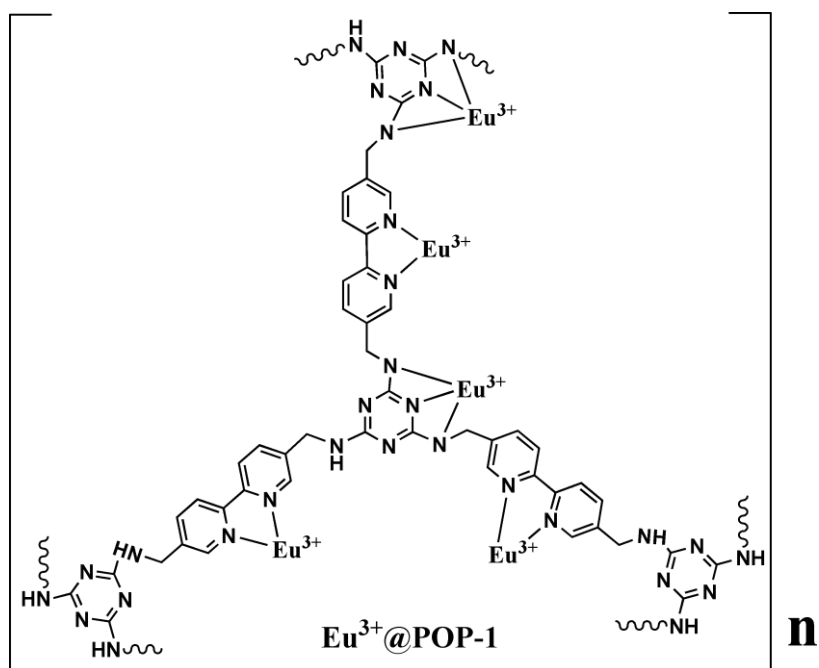


Fig. S10 Schematic diagram of $\text{Eu}^{3+}@POP-1$.

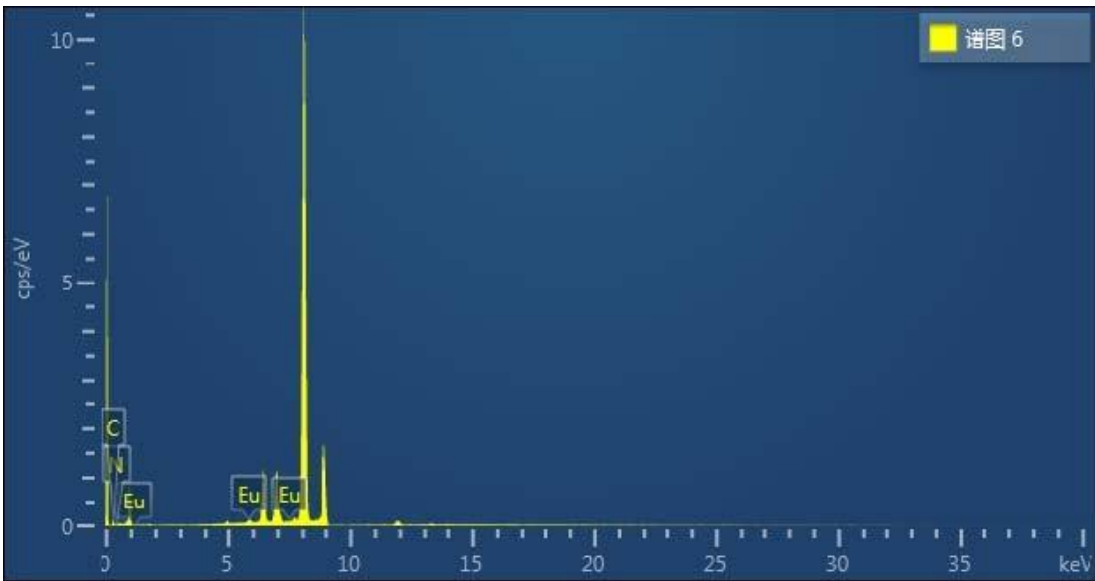


Fig. S11 Energy dispersive X-ray (EDX) analysis of Eu^{3+} @ARPOP-1

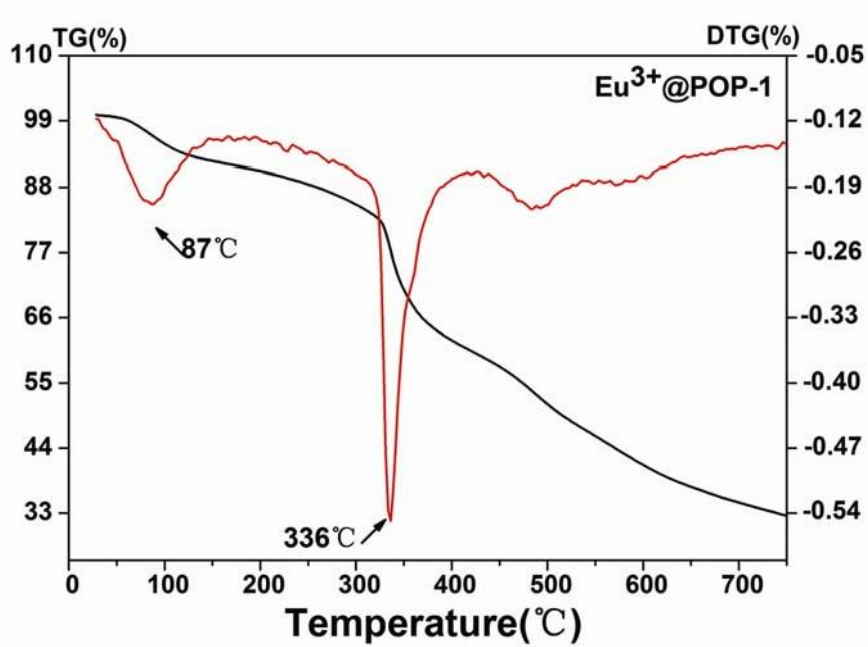


Fig. S12 Thermogravimetric analysis of Eu^{3+} @POP-1.

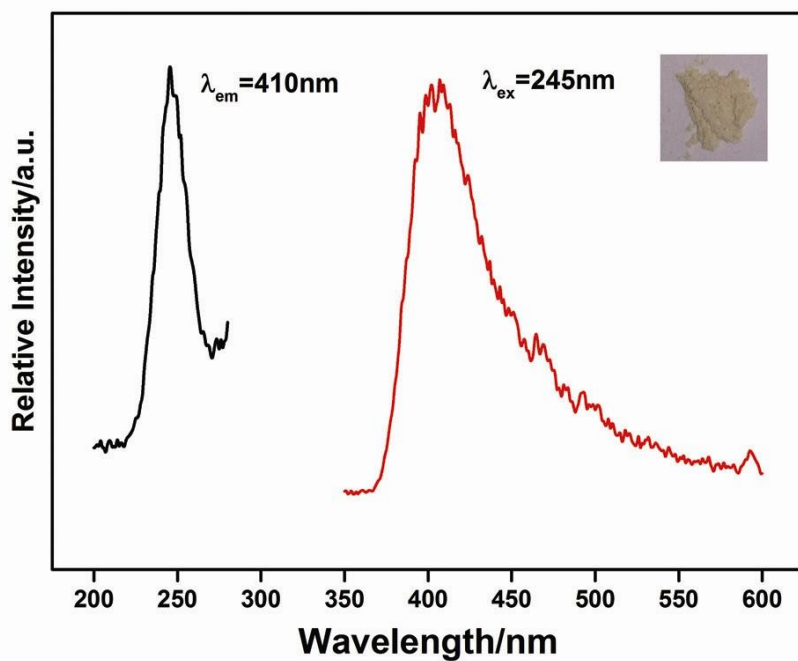


Fig. S13 Excitation spectrum and emission spectrum of POP-1.

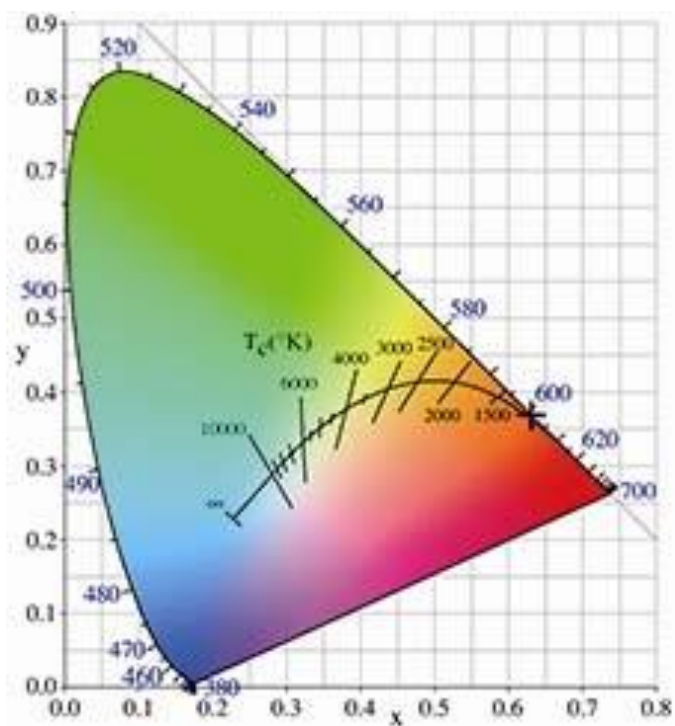


Fig. S14 CIE chromaticity diagram of $\text{Eu}^{3+}@POP-1$.

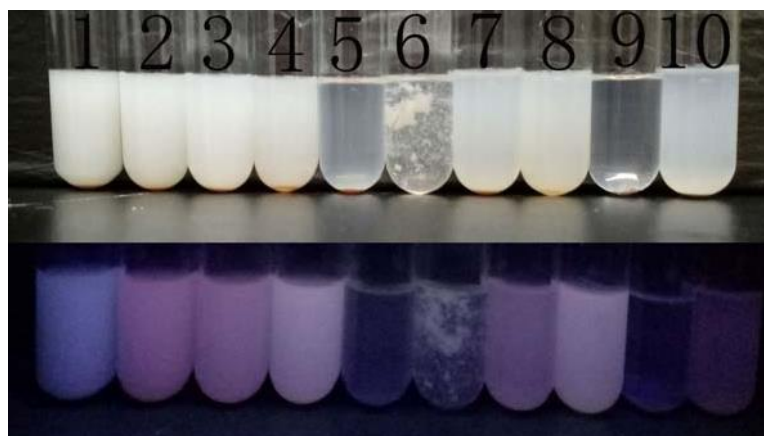


Fig. S15 Eu^{3+} @POP-1 introduced into various pure solvents under UV light
(1- CHCl_3 , 2- CHCl_2 , 3-Ethyl acetate, 4-DMF, 5-Toluene, 6-Methanol, 7-THF, 8- CCl_4 , 9-Diethyl ether, 10-Acetone).

Adsorbents	Uptake capacity (mg g ⁻¹)	Ref
5% GO/MIL-100(Fe)	1189	1
ZnAl-LDH-LDO	1015	2
POP-1	738.6	This work
MZIF-67	738	3
Mesoporous Y-Fe ₂ O ₃ /SiO ₂ nanocomposites	476	4
Fe ₃ O ₄ /Al ₂ O ₃ /chitosan	416	5
Activated carbon	200	6

Table. S1 Comparison of MO adsorption capacity for adsorbents.

References

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