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Supporting Information

Facile microwave and deposition method synthesis of magnetic biomass carbon

based Bi₂O₃ photocatalyst and mechanism insight

Wei Ma^{a*}, Na Wang^a, Yao Lu^{ab}, Ziyang Lu^c, Xu Tang^d, Songtian Li^{a*}

- a. School of Chemistry and Environmental Engineering, Pingdingshan University,
 Pingdingshan 467099, PR China
- b. College of Chemistry and Molecular Engineering, Zhengzhou University, Zhengzhou, 450000, PR China
- c. School of the Environment and Safety Engineering, Jiangsu University, Jiangsu, Zhenjiang, 212013, PR China
- Institute for Advanced Materials, School of Materials Science, Jiangsu University, Zhenjiang 212013, P.R. China

*Corresponding authors:

Tel: 86-375-2657731 Fax: 86-375-265773

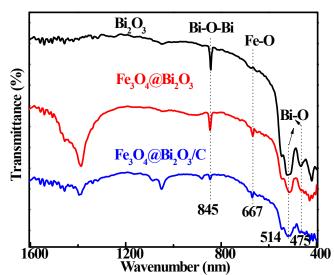
E-mail address: maweihenan@126.com, lisongtian@126.com

Photocatalytic degradation and trapping experiments

The photocatalytic activities of the as-prepared photocatalysts are measured by degradation of tetracycline (100 mL, 20 mg L⁻¹) under visible-light (250 W Xenon lamp covered with a UV filter λ > 420 nm) irradiation. Before irradiation start, 0.05 g photocatalyst is suspended in tetracycline solution to reach adsorption/desorption equilibration in dark. During illumination process, 4 mL suspension was withdrawn in 20 min interval. The trapping experiments are the same as photocatalytic experiment, additional add triethanolamine (TEOA, 1 mM), isopropanol (IPA, 1 mM) and benzoquinone (BQ, 1 mM) before the photocatalytic degradation process. The absorbance of tetracycline solution is monitored by UV-vis spectrophotometer. The degradation efficiency of Dr is calculated using C₀-C/C₀, here, C₀ and C are the initial and final concentrations of tetracycline, respectively.

Photo-electrochemical measurements

The photo-electrochemical performance of the photocatalyst is investigated by the photocurrent response and electrochemical impedance spectroscopy (EIS) on a CHI 852C electrochemical station. Briefly, 0.005 g photocatalyst is dispersed in 1.5 mL ethanol and 1.0 mL ethanol glycol, the dispersion mixture is dipcoated onto FTO substrates (1.0 cm²) and used as corresponding working electrodes. A Pt electrode is used as the counter electrode and an Ag/AgCl electrode in saturated KCl solution is employed as the reference electrode.



FT-IR spectra analysis

Fig. S1 FT-IR spectra of Bi₂O₃, Fe₃O₄@Bi₂O₃, and Fe₃O₄@Bi₂O₃/C.