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Band gap engineering of BiOI via Oxygen Vacancy induced by graphene for improved

photocatalysis

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Figure S1: Adsorption curves of MO by BiOI and rGO-BiOI nanocomposites samples.



Figure S2: The mineralization rate of rGO_{1/400}-BiOI for MO



Figure S3: (a) Zero-order kinetics (b) second-order kinetics of degradation for MO over BiOI and rGO-BiOI.

Table	S1.	Fitted	results	of MO	degradation	over BiOI an	1 rGO-BiOI

catalysts	BiOI	rGO _{1/200} -BiOI	rGO _{1/400} -BiOI	rGO _{1/600} -BiOI	rGO _{1/800} -BiOI	rGO _{1/1000} -BiOI
zero order R ²	0.984	0.959	0.916	0.975	0.963	0.997
slope (k ₀)	0.003	0.008	0.013	0.013	0.012	0.012
standard error	0.0001	0.0006	0.0017	0.0008	0.0010	0.0003
first order R ²	0.981	0.991	0.974	0.992	0.927	0.963
slope (k ₁)	0.002	0.007	0.026	0.015	0.016	0.012
standard error	0.0001	0.0003	0.0017	0.0006	0.0018	0.0010
second order \mathbb{R}^2	0.974	0.976	0.583	0.718	0.694	0.829
slope (k ₂)	0.001	0.006	0.090	0.028	0.026	0.015
standard error	0.00009	0.0003	0.0295	0.0070	0.0067	0.0027

 $(K_0=mg/L min, K_1=min^{-1}, K_2=L/mg min)$



Figure S4: Cycling experiment for degrading MO over rGO_{1/400}-BiOI



Figure S5: Photocatalytic activities of phenol over BiOI, rGO_{1/400}-BiOI and rGO_{1/600}-BiOI under visible light irradiation.



Figure S6: XRD patterns of rGO_{1/400}-BiOI before and after photoreaction.



Figure S7: Reactive species trapping experiments on the photocatalytic degradation of MO over pure

BiOI.