Electronic Supplementary Material (ESI) for New Journal of Chemistry. This journal is © The Royal Society of Chemistry and the Centre National de la Recherche Scientifique 2019

New Journal of Chemistry

Electronic Supplementary Information (ESI)

Structural dependence of the optical properties of narrow band gap thiophenethiadiazologuinoxaline derivatives and their application in organic photovoltaic cells

Cristiana Costa^{a,b}, Joana Farinhas^a, João Avó^c, Jorge Morgado^{a,d}, Adelino M. Galvão^{b,*}, Ana Charas^{a,*}

1- Instituto de Telecomunicações, Instituto Superior Técnico, Av. Rovisco Pais 1, P-1049-001, Lisboa, Portugal.

2- Centro de Química Estrutural, Instituto Superior Técnico, Universidade de Lisboa, Portugal.

3- CQFM-IN and IBB-Institute for Bioengineering and Biosciences, Instituto Superior Técnico, University of Lisbon, Lisboa, Portugal.

4 - Department of Bioengineering, Instituto Superior Técnico, University of Lisbon, Lisboa, Portugal.

Contents

Figure S1. ¹ H NMR (400 MHz) of 1,2-bis(3,4-bis(2-(2-ethoxyethoxy)ethoxy)phenyl)ethane-1,2-dione (3)2
Figure S2. ¹³ C (100 MHz) and DEPT 135 NMR of 1,2-bis(3,4-bis(2-(2-ethoxyethoxy)ethoxy)phenyl) ethane-1,2-dione (3).2
Figure S3. ¹ H NMR (400 MHz) of 2,3-bis(3,4-bis(2-(2-ethoxyethoxy)ethoxy)phenyl)-5,8-dibromoquinoxaline (6)
Figure S4. ¹³ C (100 MHz) (top) and DEPT 135 NMR of 2,3-bis(3,4-bis(2-(2-ethoxyethoxy)ethoxy)phenyl)-5,8-
dibromoquinoxaline (6)
Figure S5. ¹ H NMR (400 MHz) of 6,7-bis(3,4-bis(2-(2-ethoxyethoxy)ethoxy)phenyl)-4,9-dibromo-[1,2,5] thiadiazolo[3,4-
g]quinoxaline (7)
Figure S6. ¹³ C (100 MHz) and DEPT135 NMR of 6,7-bis(3,4-bis(2-(2-ethoxyethoxy)ethoxy)phenyl)-4,9-dibromo-
[1,2,5]thiadiazolo[3,4-g]quinoxaline (7)
Figure S7. ¹ H NMR (400 MHz) of 1-bromo-4-[2-(2-ethoxyethoxy)ethoxy]benzene (8)
Figure S8. ¹ H NMR (400 MHz) of 1,2-bis(4-(2-(2-ethoxylethoxy)ethoxy)phenyl)ethane-1,2-dione (10)4
Figure S9. ¹³ C (100 MHz) and DEPT135 NMR of 1,2-bis(4-(2-(2-ethoxylethoxy)ethoxy)phenyl)ethane-1,2-dione (10)5
Figure S10. ¹ H NMR (400 MHz) of 4,9-dibromo-6,7-bis(4-(2-(2-ethoxylethoxy)ethoxy)phenyl)-[1,2,5]thiadiazolo[3,4-
g]quinoxaline (11)
Figure S11. ¹³ C (100 MHz) and DEPT 135 NMR of 4,9-dibromo-6,7-bis(4-(2-(2-ethoxylethoxy)ethoxy)phenyl)-
[1,2,5]thiadiazolo[3,4-g]quinoxaline (11)
Figure S12. ¹ H NMR (400 MHz) of TQT1
Figure S13. ¹³ C (100 MHz) and DEPT135 NMR of TQT1
Figure S14. ¹ H NMR (400 MHz) of TQT2
Figure S15. ¹³ C (100 MHz) and DEPT135 NMR of TQT2
Figure S16. ¹ H NMR (400 MHz) of PTQT
Figure S17. ¹ H NMR (400 MHz) of PQT
Figure S18. ¹³ C NMR (100 MHz) of PQT
Figure S19. Cyclic voltamogram of TQT1 obtained at the scan rate of 50 mV/s9
Figure S20. Cyclic voltamogram of TQT2 obtained at the scan rate of 50 mV/s
Figure S21. Cyclic voltamogram of PTQT obtained at the scan rate of 50 mV/s
Figure S22. Cyclic voltamogram of PQT obtained at the scan rate of 50 mV/S10
Figure S23. Stoke's shifts as a function of the solvent polarity parameter Et30 for TQT1, TQT2 and PQT10
Figure S24. Dependence of the HOMO level on the amount of LR exact Exchange for TQT1
Figure S25. UV-Visible absorption spectra of the pure compounds in the solvents used for preparing the active blends
and of active blends of the fabricated OSCs 11
Figure S26. J-V curves in a semilogarithmic representation for the best performing OSCs for each D:A pair in the dark. 11
Figure S27. J-V characteristics for the OSCs under AM1.5G (82 mW/cm ⁻) illumination and under dark
Figure S28. Phase AFM images of the active layers shown in Figure 6 of the main manuscript
Figure S29. AFM images of active layers of the fabricated cells

Table S1 Performance parameters of the solar cells fabricated with TQT1, TQT2, PQTQ, and PQT under 82 mW.cm⁻²..12



Figure S1. ¹H NMR (400 MHz) spectrum of 1,2-bis(3,4-bis(2-(2-ethoxyethoxy)ethoxy)phenyl)ethane-1,2-dione (3) in CD_2CI_2 .



Figure S2. ¹³C (100 MHz) (top) and DEPT 135 (CH and CH3's appearing negative and CH2's appearing positive) NMR spectrum (bottom) of 1,2-bis(3,4-bis(2-(2-ethoxyethoxy)ethoxy)phenyl)ethane-1,2-dione (3) in CDCl₃.



Figure S3. ¹H NMR (400 MHz) spectrum of 2,3-bis(3,4-bis(2-(2-ethoxyethoxy)ethoxy)phenyl)-5,8-dibromoquinoxaline (6) in CDCl₃.



Figure S4. ¹³C (100 MHz) (top) and DEPT 135 (CH and CH3's appearing negative and CH2's appearing positive) (bottom) NMR spectra of 2,3-bis(3,4-bis(2-(2-ethoxyethoxy)ethoxy)phenyl)-5,8-dibromoquinoxaline (6) in CDCl₃.



Figure S5. ¹H NMR (400 MHz) spectrum of 6,7-bis(3,4-bis(2-(2-ethoxyethoxy)ethoxy)phenyl)-4,9-dibromo-[1,2,5]thiadiazolo[3,4-g]quinoxaline (7) in CD₂Cl₂.



Figure S6. ¹³C (100 MHz) (top) and DEPT 135 (CH and CH3's appearing negative and CH2's appearing positive) (bottom) NMR spectra of 6,7-bis(3,4-bis(2-(2-ethoxyethoxy)ethoxy)phenyl)-4,9-dibromo-[1,2,5]thiadiazolo[3,4-g]quinoxaline (7) in CD_2Cl_2 .



Figure S7. ¹H NMR (400 MHz) spectrum of 1-bromo-4-[2-(2-ethoxyethoxy)ethoxy]benzene (8) in CDCl₃.



Figure S8. ¹H NMR (400 MHz) spectrum of 1,2-bis(4-(2-(2-ethoxylethoxy)ethoxy)phenyl)ethane-1,2-dione (10) in CDCl₃.



Figure S9. ¹³C (100 MHz) (top) and DEPT 135 (CH and CH3's appearing positive and CH2's appearing negative) (bottom) NMR spectra of 1,2-bis(4-(2-(2-ethoxylethoxy)ethoxy)phenyl)ethane-1,2-dione (10) in CDCl₃.



Figure S10. ¹H NMR (400 MHz) spectrum of 4,9-dibromo-6,7-bis(4-(2-(2-ethoxylethoxy)ethoxy)phenyl)-[1,2,5]thiadiazolo[3,4-g]quinoxaline (11) in CDCl₃.



Figure S11. ¹³C (100 MHz) (top) and DEPT 135 (CH and CH3's appearing negative and CH2's appearing positive) (bottom) NMR spectra of 4,9-dibromo-6,7-bis(4-(2-(2-ethoxylethoxy)ethoxy)phenyl)-[1,2,5]thiadiazolo[3,4-g]quinoxaline **(11)** in CDCl₃.



Figure S12. ¹H NMR (400 MHz) spectrum of TQT1 in CD₂Cl₂.



Figure S13. ¹³C (100 MHz) (top) and DEPT 135 (CH and CH3's appearing negative and CH2's appearing positive) (bottom) NMR spectra of TQT1 in CDCl₃.

Hn



Figure S14. ¹H NMR (400 MHz) spectrum of TQT2 in CD₂Cl₂.



Figure S15. ¹³C (100 MHz) (top) and DEPT 135 (CH and CH3's appearing negative and CH2's appearing positive) (bottom) NMR spectrum of TQT2 in CD₂Cl₂.



Figure S16. ¹H NMR (400 MHz) spectrum of PTQT in CD₂Cl₂.



Figure S18. ¹³C NMR (100 MHz) spectrum of PQT in CD₂Cl₂.



Figure S19. Cyclic voltammogram of TQT1 obtained at the scan rate of 50 mV/s.



Figure S20. Cyclic voltammogram of TQT2 obtained at the scan rate of 50 mV/s.



Figure S21. Cyclic voltamogram of PTQT obtained at the scan rate of 50 mV/s.



Figure S22. Cyclic voltammogram of PQT obtained at the scan rate of 50 mV/s



Figure S23. Stoke's shifts as a function of the solvent polarity parameter Et30 for TQT1, TQT2 and PQT.



Figure **S24.** Dependence of the HOMO level on the amount of LR exact Exchange for **TQT1**



Figure **S25.** Top: UV-Visible absorption spectra (300-900 nm) of the pure compounds (TQT1, TQT2, PTQT, PQT) in the solvents used for preparing the active blends (chlorobenzene or chloroform) and (bottom) of active blends of the fabricated OSCs.



Figure S26. J-V curves in a semilogarithmic representation for the best performing OSCs for each D:A pair in the dark.

Active layer	Ratio	J _{sc}	V _{oc} (V)	FF	PCE (%)	Thickn
	(D:A)	(mA/cm ²)			best/ave ^a	(nm)
TQT1:PC ₇₀ BM	1:1	-1.20	0.81	0.23	0.29/0.24	90
	1:2	-2.16	0.88	0.21	0.51/0.46	85
	1:3	-2.93	0.88	0.27	0.88/0.70	90
TQT2:PC ₇₀ BM	1:1	-2.02	0.64	0.28	0.42/0.36	120
	1:2	-3.05	0.64	0.29	0.66/0.55	100
	1:3	-3.07	0.65	0.30	0.71/0.63	80
	1:4	-2.28	0.64	0.27	0.49/0.45	105
PTQT:PC ₇₀ BM	1:1	-0.33	0.24	0.21	0.02/0.02	150
PQT:PC ₆₀ BM	1:2	-1.17	0.82	0.48	0.61/0.54	95

Table S1. Performance parameters of the solar cells fabricated with TQT1, TQT2, PQTQ, and PQT (under 82 mW.cm⁻²).

^a Best values followed by the averages calculated from at least 8 devices.



Figure S27. *J-V* characteristics for the OSCs under AM1.5G (82 mW/cm²) illumination (left) and in dark conditions (semilogarithmic scale) (right).



Figure S28. Phase AFM images of the active layers shown in Figure 6 of the main manuscript.



Figure S29. AFM images of active layers of the fabricated cells.