

Selective Liquid Phase Oxidation of Ethyl Benzene to Acetophenone by Palladium Nanoparticles Immobilized on g-C₃N₄-rGO Composite as Recyclable Catalyst

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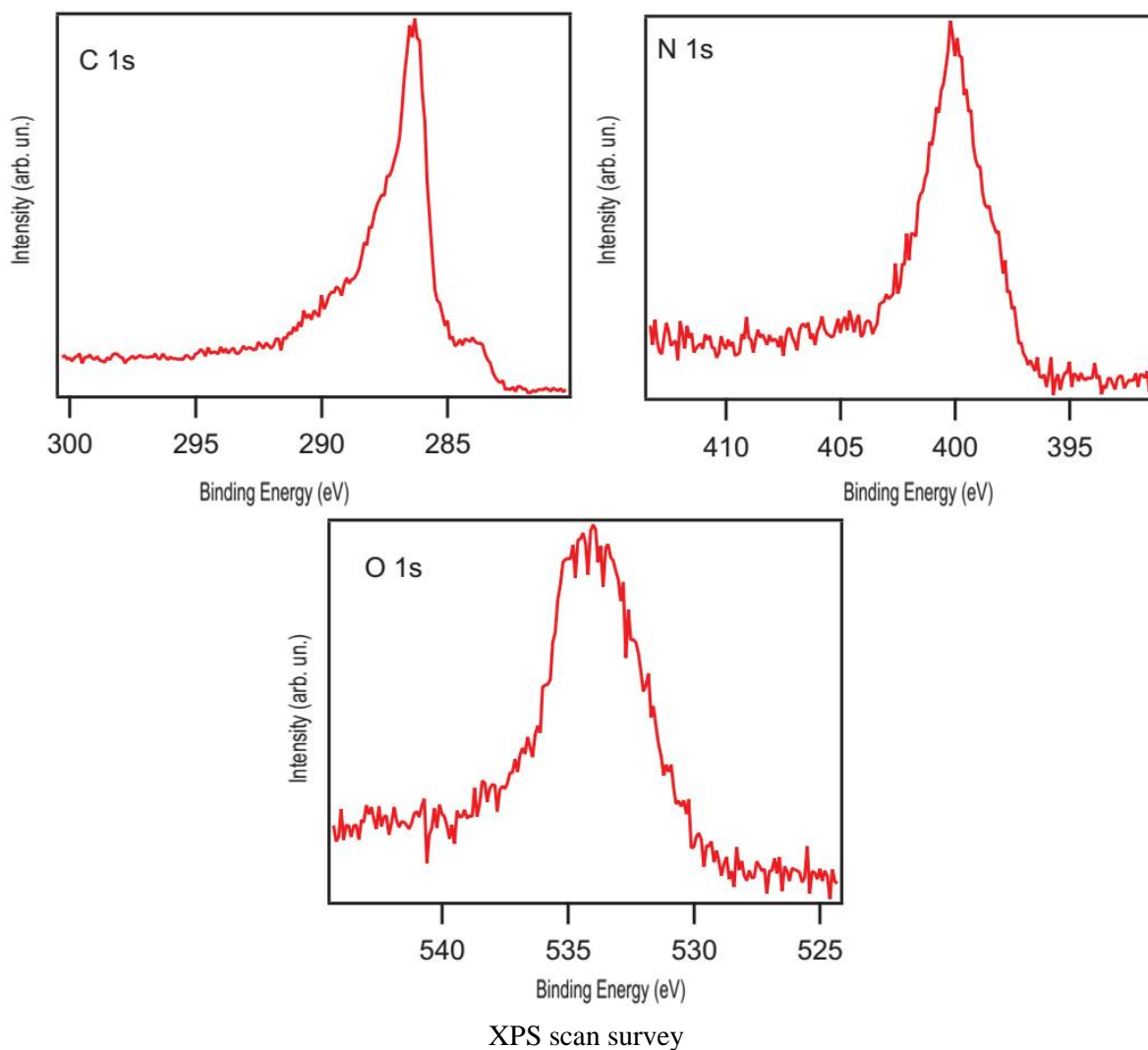


Figure S1. The XPS spectra carbon (1s), nitrogen (1s) and oxygen (1s)

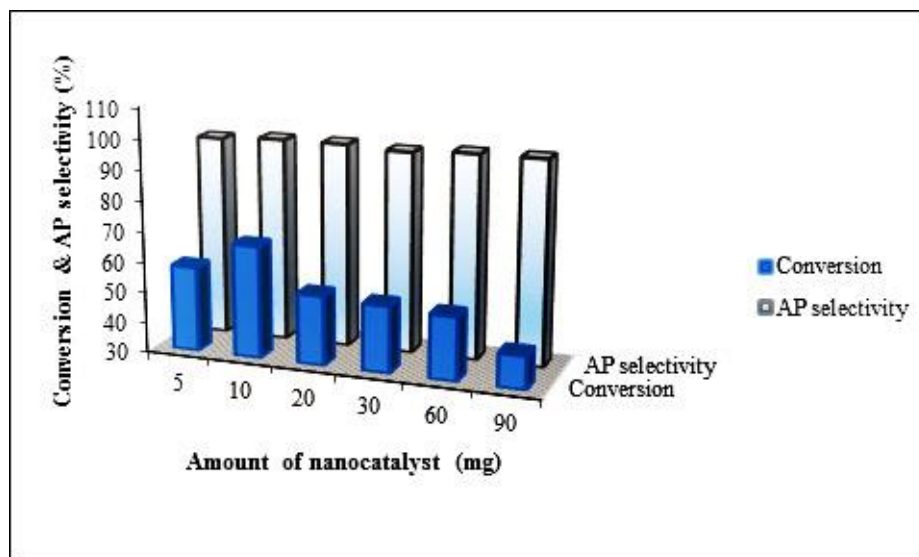


Figure S2. The effect of amount of catalyst on the oxidation of EB
 Reaction conditions: EB: TBHP (mol ratio) = 1:4, 36 h, temperature = reflux, 10 mL CH₃CN, 10 mg Pd/g-C₃N₄-rGO

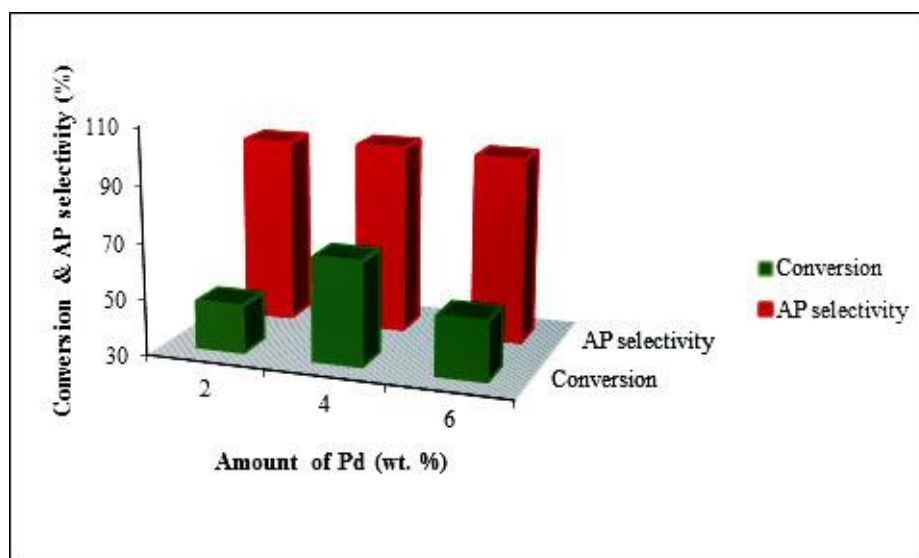
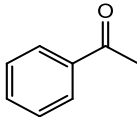
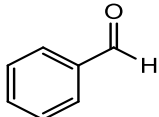


Figure S3. The effect of Pd loading on the oxidation of EB
 Reaction conditions: EB: TBHP (mol ratio) = 1:4, 24 h, temperature = reflux, 10 mL CH₃CN, 10 mg x%Pd/g-C₃N₄-rGO

Table S1. The activity of recycled Pd/gC₃N₄-rGO in the oxidation of EB

Entry	Cycle number	Conversion (%)	Selectivity (%)	
				
1	1	67	97	3
2	2	67	96	4
3	3	66	96	4
4	4	65	96	4
5	5	63	96	4

Reaction conditions: EB: TBHP (mol ratio) = 1:4, 24 h, temperature = reflux, 10 mL CH₃CN, 10 mg Pd/g-C₃N₄-rGO

Table S2. An overview of the research on oxidation of EB

Catalyst	Oxidant	Reaction time (h)	T(°C)	Conversion (%)	Selectivity (%) ^a			Reference
					AP	BZ	Others	
Si/Al-pr-NH-et-N methyl-2-pyridylketone-Mn	TBHP	8	120	91	98.1	1.8	-	92
Mn(TMCP) ^b	TBHP	6	150	40.8	96.6	2.2	1.2	93
Pd-nanoparticles	TBHP	24	80	92.3	93.5	0.6	5.9	72
Si/Al-pr-NH-et-N methyl-2-pyridylketone-Mn	TBHP	2	80	33	74	13	12	55
Si/Al-pr-NH-et-N methyl-2-pyridylketone-Mn	TBHP	24	80	67	93	27	4.3	55
Ag/SiO ₂	TBHP	12	120	38	88	-	12	52
Co-N-C/CeO ₂ ^c	O ₂	5	120	33.1	74.8	2.5	22.7	58
Co-N-C/g-C ₃ N ₄ ^d	O ₂	5	120	28	78.1	1.9	20	19
Pd/g-C ₃ N ₄ -rGO	TBHP	24	80	67	97	3	-	Present work

^a AP: Acetophenone, BZ: Benzaldehyde

^b 5, 10, 15, 20-tetrakis (4-methoxy carbonyl phenyl) porphyrin)

^c Cobalt(II) 5-(4-carboxyphenyl)-10,15,20-triphenylporphyrin

^d Cobalt(II) 5,10,15,20-tetraphenylporphyrin