## **Supporting Information**

## Mesoporous-Silica Induced Doped Carbon Nanotubes Growth From Metal-

## **Organic Frameworks**

Huang Zhou<sup>a</sup>, Daping He<sup>ab\*</sup>, Ibrahim Saana Amiinu<sup>a</sup>, Jinlong Yang<sup>a</sup>, Zhe Wang<sup>a</sup>, Jian Zhang<sup>a</sup>, Qirui Liang<sup>a</sup>, Shuai Yuan<sup>a</sup>, Jiawei Zhu<sup>a</sup>, Shichun Mu<sup>a\*</sup>

<sup>a</sup>State Key Laboratory of Advanced Technology for Materials Synthesis and Processing, Wuhan University of Technology, Wuhan 430070, China
E-mail: msc@whut.edu.cn, hedaping@whut.edu.cn
<sup>b</sup>Hubei Engineering Research Center of RF-Microwave Technology and Application, School of Science, Wuhan University of Technology, Wuhan 430070, China

\*Address correspondence to: msc@whut.edu.cn,hedaping@whut.edu.cn



Figure S1 FESEM images of as-prepared

ZIF-67 precursors.



Figure S2 TEM images (a, b) and particle-size distribution (c) of ZIF-67 precursors.



Figure S3 (a) TEM images,(b)HR-TEM image (c) N<sub>2</sub> adsorption-desorption

isotherms, and (d) pore-size distribution curves of as-prepared ZIF-67@mSiO<sub>2</sub>.



Figure S4 (a)  $N_2$  adsorption-desorption isotherms, and (b) pore-size distribution

curves of ZIF-67.



Figure S5 FESEM (a) and TEM (b) image of as-prepared



## of Co/N-CNTs.

**Figure S6** TEM image of (a) Co/N-CNTs (with Co nanoparticles partially removed by HF), HRTEM (b) and HAADF (c) image of a single CNT with Co removed and

(d-f) the corresponding EDS mapping images.



Figure S7 (a) X-Ray photoelectron spectroscopy (XPS) spectra of Co/N-CNTs obtained at different temperatures;(b) The total N content obtained from XPS;N1s XPS spectra of Co/N-CNTs obtained at different temperatures: (c)  $600^{\circ}$ C; (d)  $800^{\circ}$ C and (e)  $900^{\circ}$ C; (f) The percentage of pyridinic N of Co/N-CNTs obtained at different temperatures.



Figure S8 Co 2p XPS spectrum of Co/N-CNTs obtained at 700°C.



Figure S9 (a) Raman spectra of Co/N-CNTs synthesized at different pyrolysis temperatures (600-900 °C) and (b) the corresponding  $I_D/I_G$  value.



Figure S10 Raman spectra of Co NP/N-C synthesized at 700°C.



Figure S11  $N_2$  adsorption-desorption isotherms. The inset image shows the pore-size distribution calculated using the Barrett–Joyer–Halenda (BJH) method.



Figure S12 Stability of (a) Co/N-CNTs and (b) Pt/C symbolized by  $\Delta E_{1/2}$  before and

after the i-t test.



Figure S13 The i-t chronoamperometric responses for Co/N-CNTs and Pt/C in CO and O<sub>2</sub>-saturated solution ( $V_{CO}/V_{O2}=10\%$ ) during a constant potential at -0.35 V at a rotation rate of 1,600 rpm in 0.1 M KOH



Figure S14 Formation mechanic (a), Low-magnification (b) and (c) highmagnification FESEM images of Co NP/N-C synthesized at  $700^{\circ}$ C without mSiO<sub>2</sub> protection.



**Figure S15** TEM images (a,b) of ZIF-67@mSiO<sub>2</sub> after pyrolysis at different temperatures in argon for 0.5 h (mSiO<sub>2</sub> shell was removed by 1 M NaOH before characterization). And TEM image of ZIF-67 without mSiO<sub>2</sub> protection after pyrolysis at 580°Cin argon for 0.5h.



Figure S16 FESEM images of Co/N-CNTs synthesized at different pyrolysis

temperatures (600-900℃).



Figure S17 FESEM images of Co/N-CNTs-x (x indicates relative mass of mSiO<sub>2</sub>) obtained by pyrolyzed different mass of mSiO<sub>2</sub> coated ZIF-67 at 700°C after HF etching.

Catalyst	Eonset (V vs Ag/AgCl)	E <sub>1/2</sub> (V vs Ag/AgCl)	<b>J</b> (mA cm <sup>-</sup> <sup>2</sup> )	Ref.
Co/N-CNTs	-0.005	-0.154	~5.82	This work
Co,N-CNF	-0.082	-0.155	~5.71	<i>Adv. Mater.</i> <b>2015</b> , 28(8):1712- 1712.
P-Z8-Te- 1000	-0.161	-0.161	~5.70	J. Am. Chem. Soc 2014, 136, 14385- 14388.
Co@N- PGCS	-0.075	-0.151	~5.60	<i>Nanoscale</i> <b>2016</b> , 8, 13311-13320.
N-CNT aerogels	-0.055	-0.263	~4.80	small 2015, 11, 3903–3908
EDA-NCNT	/	-0.150	~4.91	J. Phys. Chem. C 2009, 113, 21008– 21013
Py-NCNT	/	-0.330	~1.57	J. Phys. Chem. C 2009, 113, 21008– 21013
Co3O4/rmG O	/	0.83 (V vs RHE )	/	<i>Nat. Mater.</i> <b>2011,</b> 10, 780-786.
NCNTFs	/	0.87 (V vs RHE )	~5.40	<i>Nat. Energy</i> <b>2016</b> , <i>1</i> , 15006
N-CNTs-550	/	0.86 (V vs RHE )	~5.20	J. Am. Chem. Soc. 2017, 139, 8212- 8221.
N-CNTs-435	/	0.81 (V vs RHE )	~5.20	J. Am. Chem. Soc. 2017, 139, 8212- 8221.

**Table S1** Comparison of ORR performance of all catalysts in this work with correlative literature values.