

Electronic Supplementary Information (ESI)

Achieving highly practical capacitance of MnO₂ by using chain-like CoB alloy as support

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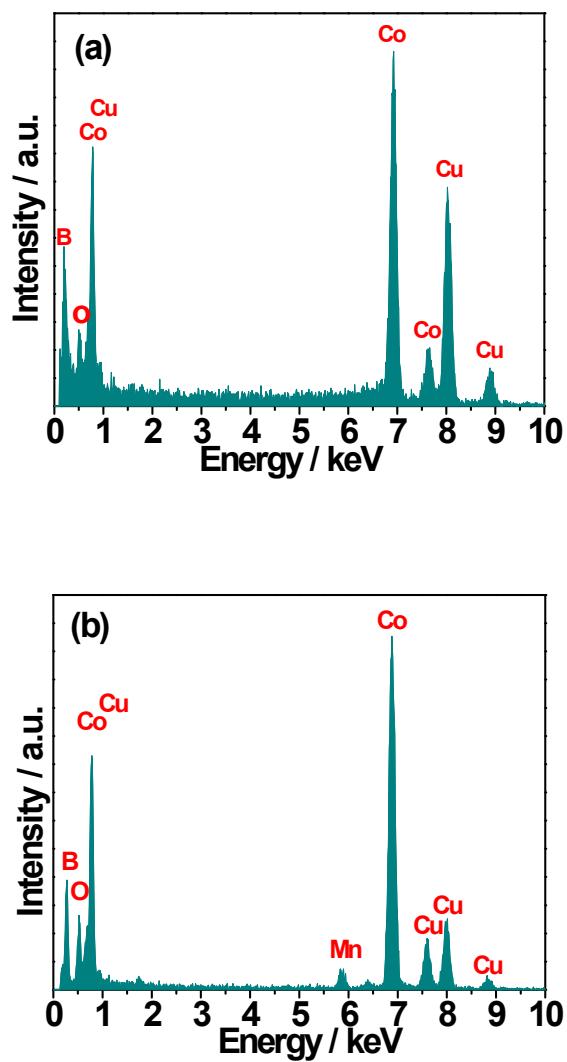


Figure S1. EDS patterns of (a)CoB and (b)CoB@MnO₂

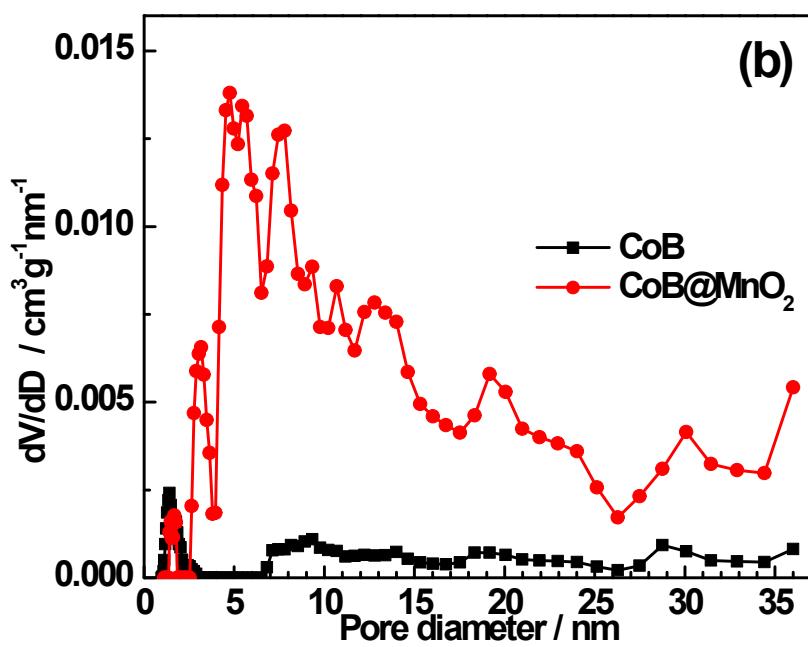
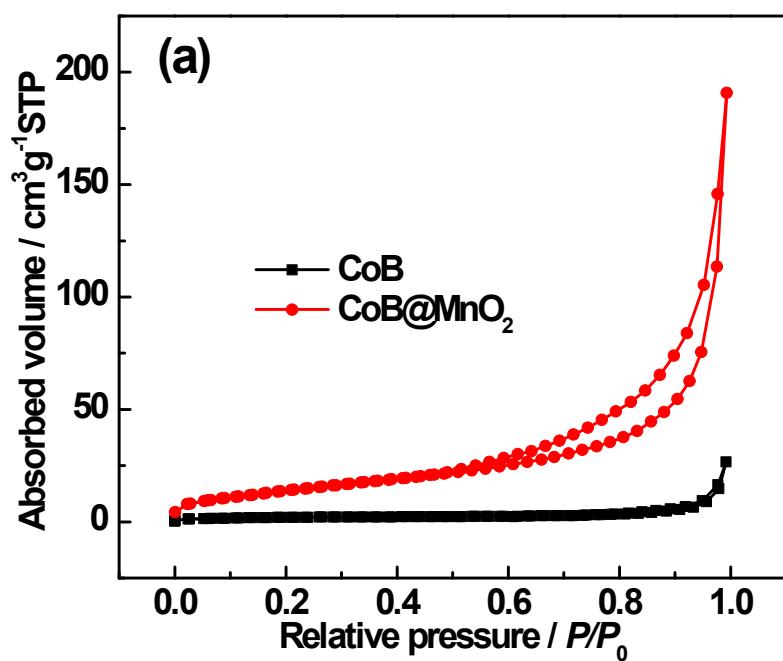


Figure S2. (a) N₂ isotherms and (b) pore size distribution of CoB and CoB@MnO₂.

Table S1. A summary of our CoB-core@MnO₂-shell with the value obtained from some other conductive core@MnO₂-shell independent literatures.

Sample core-shell	The literature	The optimal window	The specific capacitance (F g ⁻¹)	^a Current density	^b Retention(%)/cycling number/current density	Electrolyte
CoB@MnO ₂	Our work	0.8 V	612.0	0.5 A g ⁻¹	97.7%/6000/2 A g ⁻¹	3 M LiOH
Au@MnO ₂	[18]	0.8 V	601	0.5 A g ⁻¹	85%/1000/1 A g ⁻¹	2 M Li ₂ SO ₄
Ni@MnO ₂	[45]	0.85 V	502	5 mV s ⁻¹	93%/500/5 mV s ⁻¹	0.1 M Na ₂ SO ₄
CNT@MnO ₂	[46]	1.0 V	110	5.3 A g ⁻¹	88%/2500/50 A g ⁻¹	1 M Na ₂ SO ₄
Petal-like C@MnO ₂	[8]	1.0 V	163	2 mV s ⁻¹	97.8%/5000/100 mV s ⁻¹	1 M Na ₂ SO ₄
C@MnO ₂	[11]	0.8 V	252	0.5 A g ⁻¹	74%/2000/5 A g ⁻¹	1 M Na ₂ SO ₄
CNF@MnO ₂	[47]	0.8 V	321	0.5 A g ⁻¹	100%/1000/5 A g ⁻¹	0.1 M Na ₂ SO ₄
PANI@MnO ₂	[17]	0.9 V	262	0.5 A g ⁻¹	93%/800/3A g ⁻¹	0.5 M Na ₂ SO ₄
PANI@MnO ₂	[48]	0.9 V	290	0.2 A g ⁻¹	96%/3000/2 A g ⁻¹	1 M KCl
PANI@MnO ₂	[49]	0.85 V	330	0.5 A g ⁻¹	94%/1000/1 A g ⁻¹	0.1 M Na ₂ SO ₄
PANI@MnO ₂	[50]	0.7 V	627	2 A g ⁻¹	55%/1000/10 A g ⁻¹	1 M H ₂ SO ₄

a: the specific capacitance and areal capacitance was obtained at the current density.

b: after cycling test, the retention of the specific capacitance.

c: the cycling test was carried out with the current density.

d: the value was calculated based on the data in the literature.

CNF: Carbon nanofiber, CNT: Carbon nanotube, PANI: Ployaniline

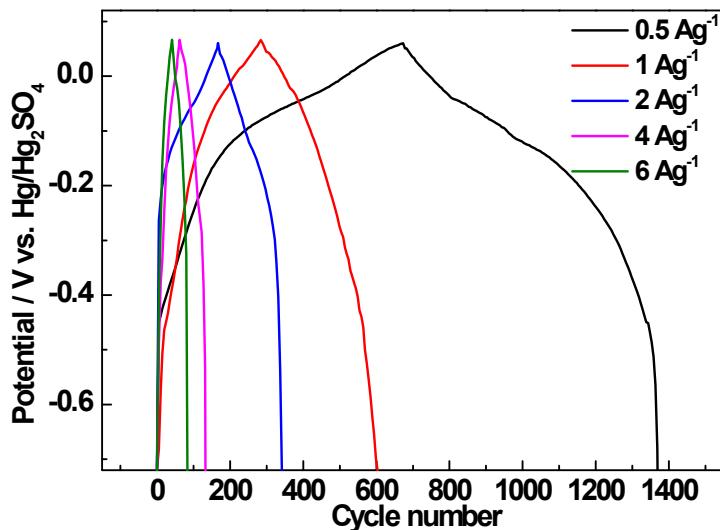


Figure S3. Galvanostatic charge-discharge curves of the CoB electrode at different

current densities of 0.5, 1, 2, 4, and 6 A g⁻¹ respectively.

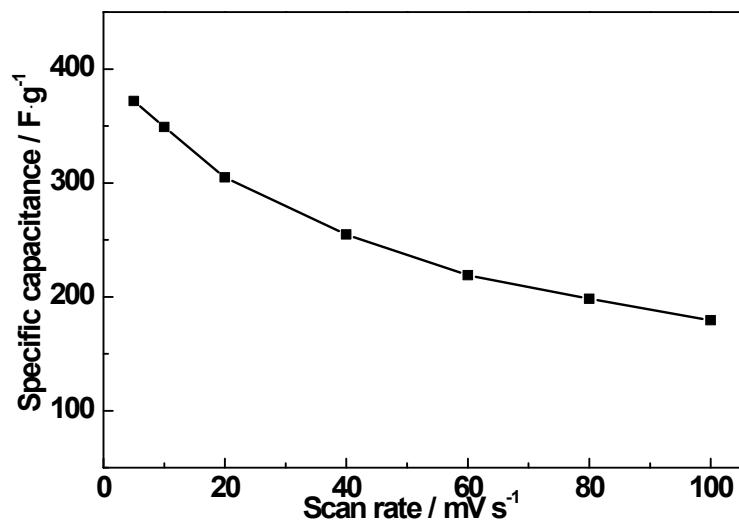


Figure S4. Specific capacitance of CoB@MnO₂ vs. scan rate.