

Supporting information

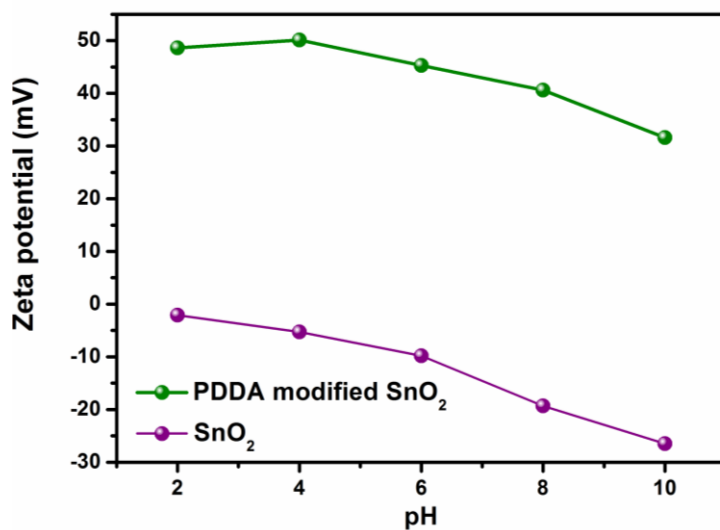


Fig. S1 Zeta potential of SnO₂ and PDDA modified SnO₂ in solution at different pH values

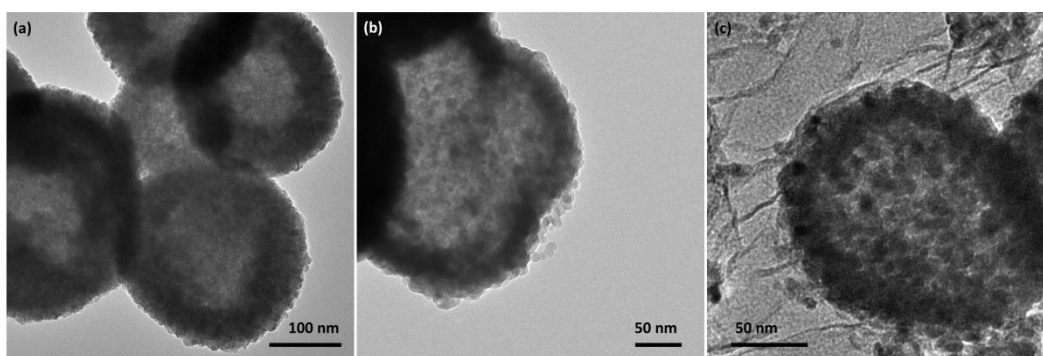


Fig. S2 TEM images of (a) SnO₂, (b) SnO₂/C and (c) SnO_{2-x}/N-rGO

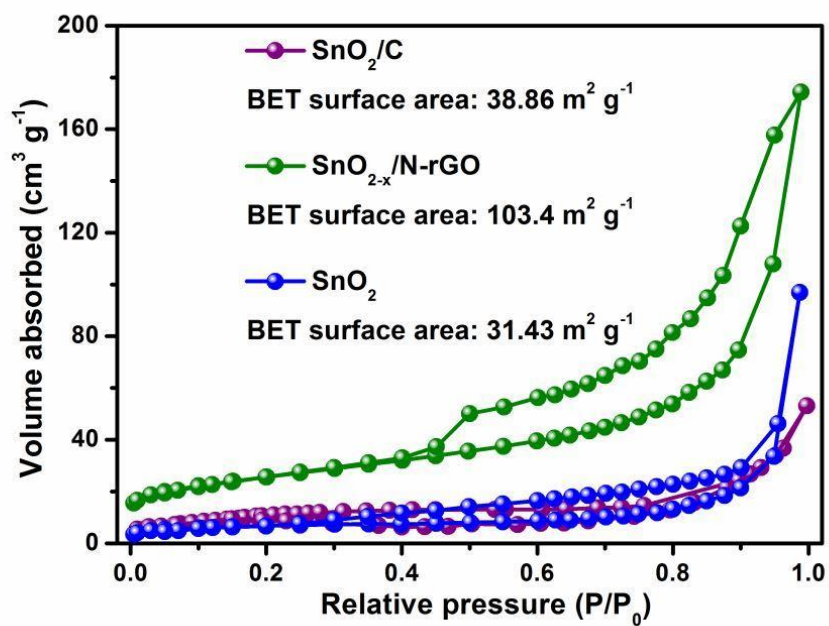


Fig. S3 N₂ adsorption/desorption isotherms of the SnO₂, SnO₂/C and SnO_{2-x}/N-rGO

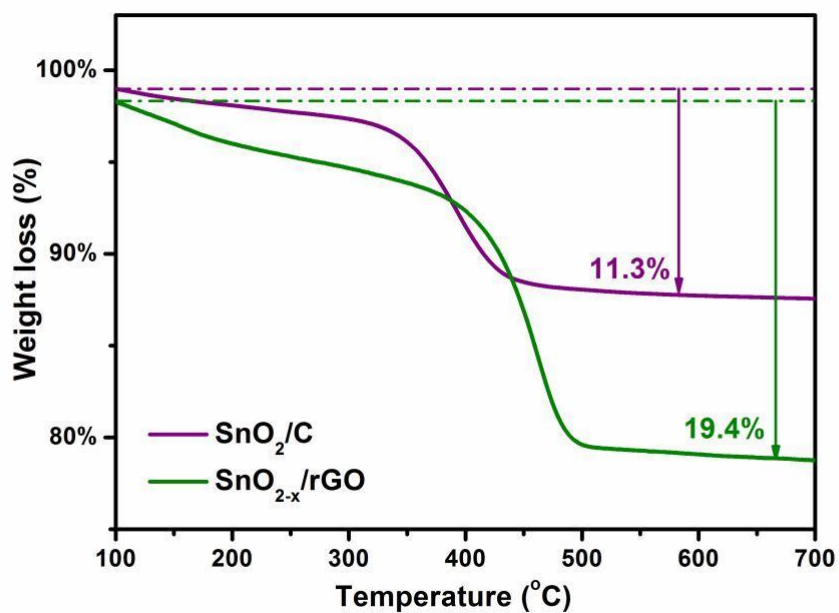


Fig. S4 TG curves of SnO₂/C and SnO_{2-x}/N-rGO samples

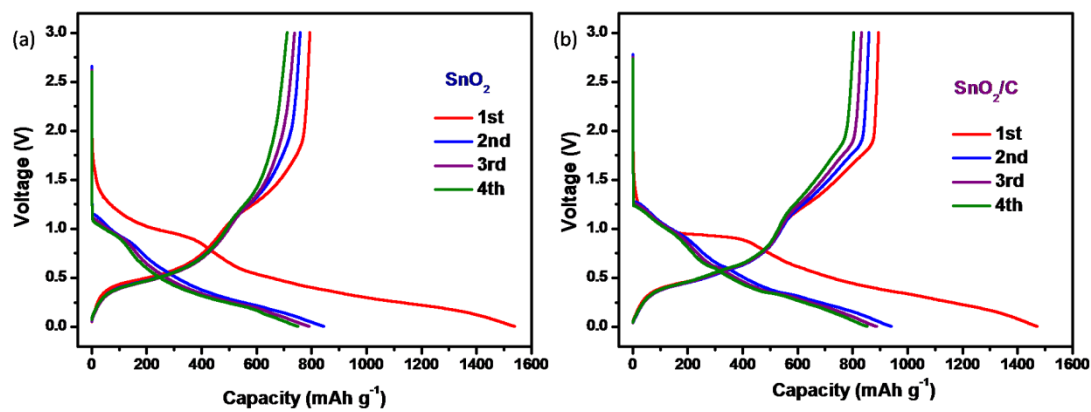


Fig. S5 The first four discharge-charge curves of (a) SnO₂ and (b) SnO₂/C electrode at a current density of 0.1 A g⁻¹

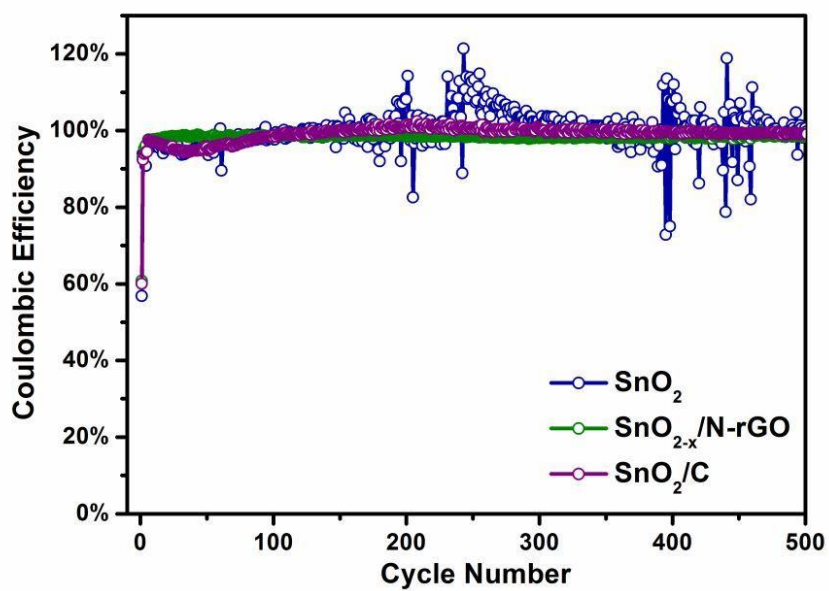


Fig. S6 Coulombic efficiency curves of as-prepared SnO₂ samples under 0.5 A g⁻¹

Table S1 Cyclic performance comparison of as-prepared SnO₂/rGO with previous work about SnO₂/graphene-based anode materials

Material	Cycle number	Current density (A g ⁻¹)	Specific capacity retention (mAh g ⁻¹)	Reference
SnO ₂ /rGO	500	1.0	552	<i>J. Mater. Chem. A</i> , 2017, 5, 4535-4542
SnO ₂ /graphene@C	100	0.1	820	<i>J. Mater. Chem. A</i> , 2016, 4, 362-367
SnO ₂ @graphene/Pani	100	0.1	770	<i>ACS Appl. Mater. Interfaces</i> , 2015, 7, 2444-2451
SnO ₂ /rGO/C foam	100	0.13	717	<i>ChemElectroChem</i> , 2016, 3, 1063-1071
SnO ₂ /rGO	100	0.1	536	<i>Electrochimica Acta</i> , 2016, 207, 9-15
SnO ₂ quantum dots@GO	100	0.1	1121	<i>Small</i> , 2016, 12, 588-594
SnO ₂ @polyaniline@graphene	100	1.0	560	<i>J. Power. Sources</i> , 2015, 290, 61-70
SnO ₂ /C/graphene sheets	100	0.2	830	<i>ACS Appl. Mater. Interfaces</i> , 2014, 6, 7434-7443
SnO ₂ @double graphene	120	0.08	591	<i>Nano Energy</i> , 2014, 3, 80-87
SnO ₂ /GO composite	200	0.1	800	<i>J. Mater. Chem. A</i> , 2013, 1, 7558-7562
SnO ₂ /graphene nanobelts	50	0.1	825	<i>ACS Nano</i> , 2013, 7, 6001-6006
SnO ₂ /graphene composite	200	0.1	830	<i>Adv. Mater.</i> 2013, 25, 3307-3312
Mesoporous SnO ₂ @graphene	50	0.078	848	<i>Adv. Funct. Mater.</i> , 2013, 23, 3570-3576
SnO ₂ with oxygen vacancies wrapped in N-doped rGO	500	0.5	912	Our work

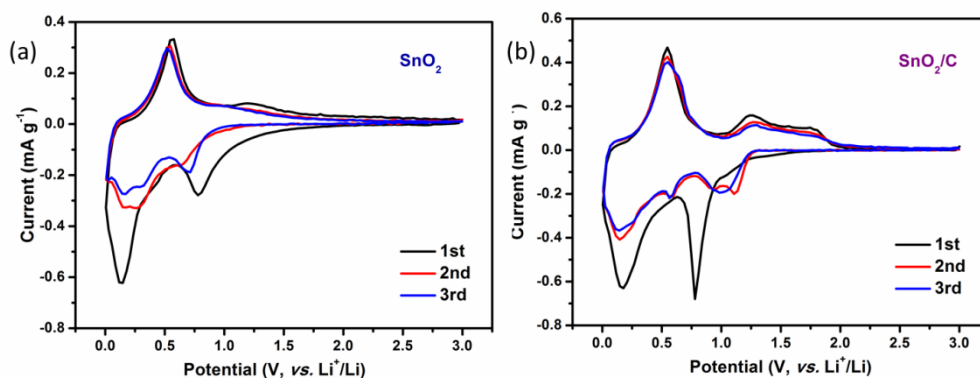


Fig. S7 CV curves of (a) SnO₂ and (b) SnO₂/C electrode in the voltage of 0.005 to 3.0 V at a scanning rate of 0.1 mV s⁻¹

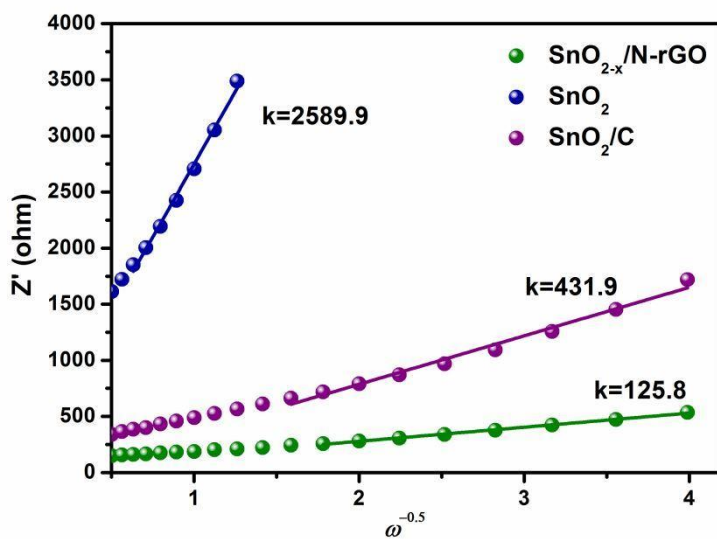


Fig. S8 The relationship between Z' and $\omega^{-0.5}$ of as-prepared SnO₂ samples

Table S2 The values of R_f , R_{ct} and D_{Li} calculated from the Nyquist plots

	R_s (Ω)	R_{ct} (Ω)	σ
SnO_2	20.3	534.4	2589.9
SnO_2/C	19.5	124.6	431.9
$\text{SnO}_{2-x}/\text{N-rGO}$	6.1	55.7	125.8

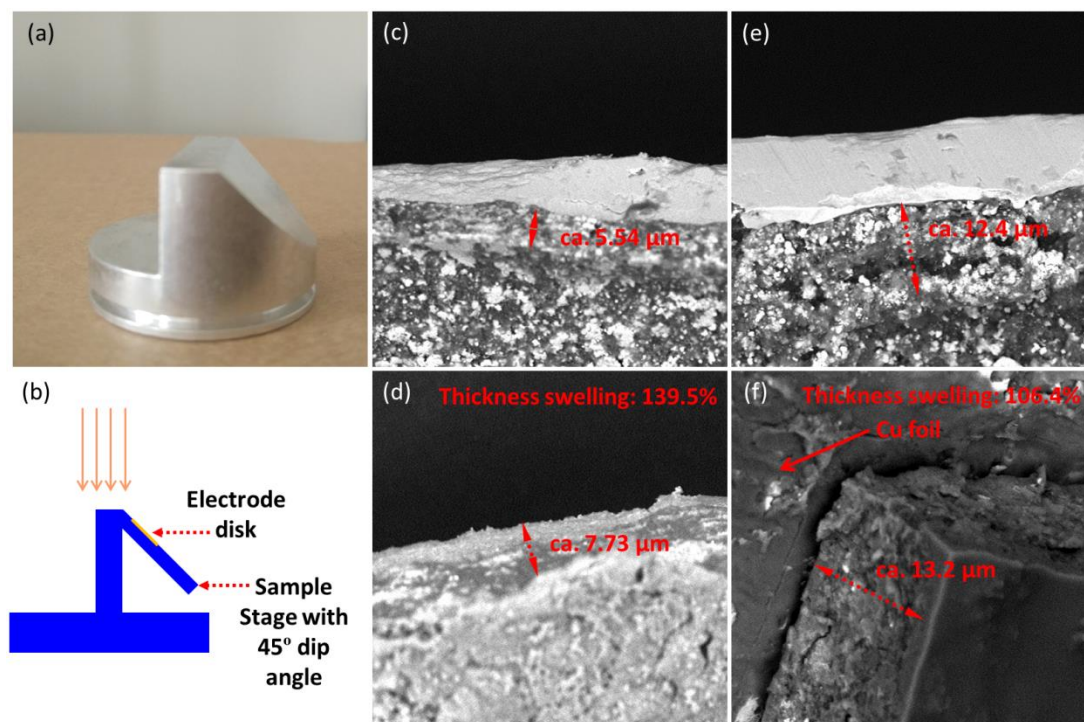


Fig. S9 (a) Photo of the sample stage; (b) working schematic of the sample stage with dip angle; SEM images of SnO_2 electrode disk (c) at initial status and (d) after cycles, and $\text{SnO}_{2-x}/\text{N-rGO}$ electrode disk (e) at initial status and (f) after cycles.