

Supplementary Information

PbTe Quantum Dots as Electron Transfer Intermediates for Enhanced Hydrogen Evolution Reaction of Amorphous MoS_x/TiO₂ Nanotube Arrays

Shiyuan Gao,^a Bin Wang,^a Xinyu Liu,^b Zhanhu Guo,^c Zhongqing Liu^{*a} and Yichao Wang^{*d,e}

^a School of Chemical Engineering, Sichuan University, Chengdu 610065, Sichuan, P. R. China

^b Qiushi Honors College, Tianjin University, Tianjin 300072, P. R. China

^c Chemical and Biomolecular Engineering Department, University of Tennessee Knoxville, Tennessee 37996 USA

^d School of Life and Environmental Sciences, Deakin University, Geelong, Victoria 3216, Australia

^e College of Materials and Textiles, Zhejiang Sci-Tech University, Hangzhou 310018, P.R. China

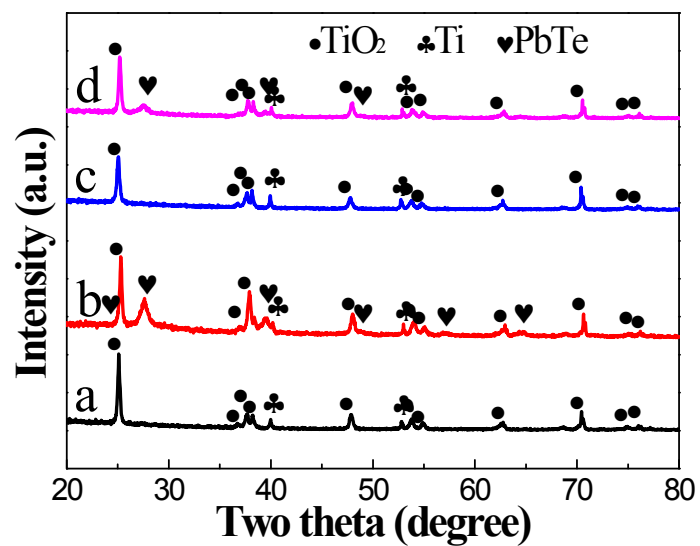


Figure S1 XRD patterns of pristine TNAs (a), PbTe/TNAs (b), MoS_x /TNAs(c) and MoS_x @PbTe/TNAs (d).

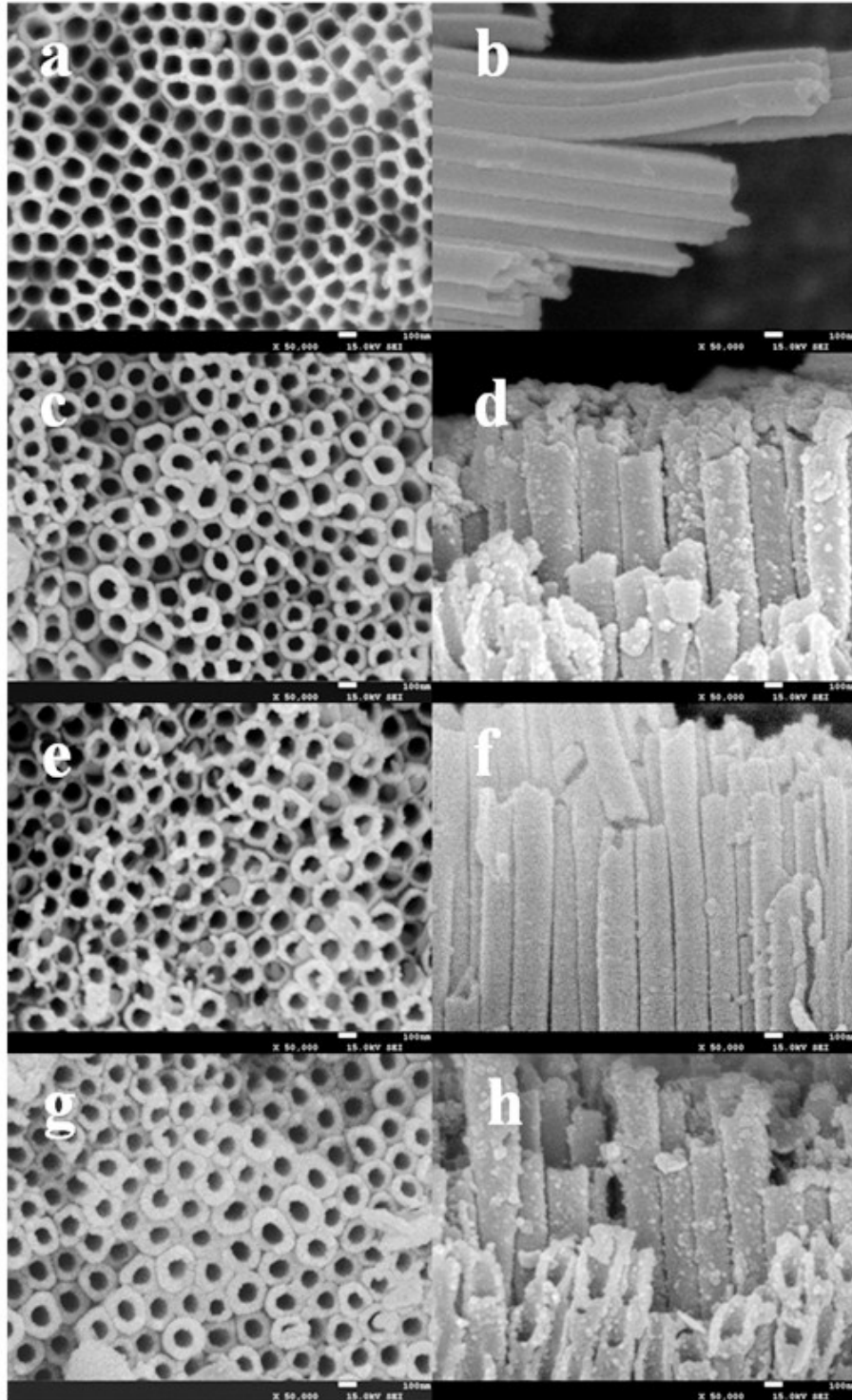


Figure S2 FE-SEM images of TNAs (a, b), PbTe/TNAs (c, d), MoS_x/TNAs (e, f) and PbTe@MoS_x/TNAs (g,h)

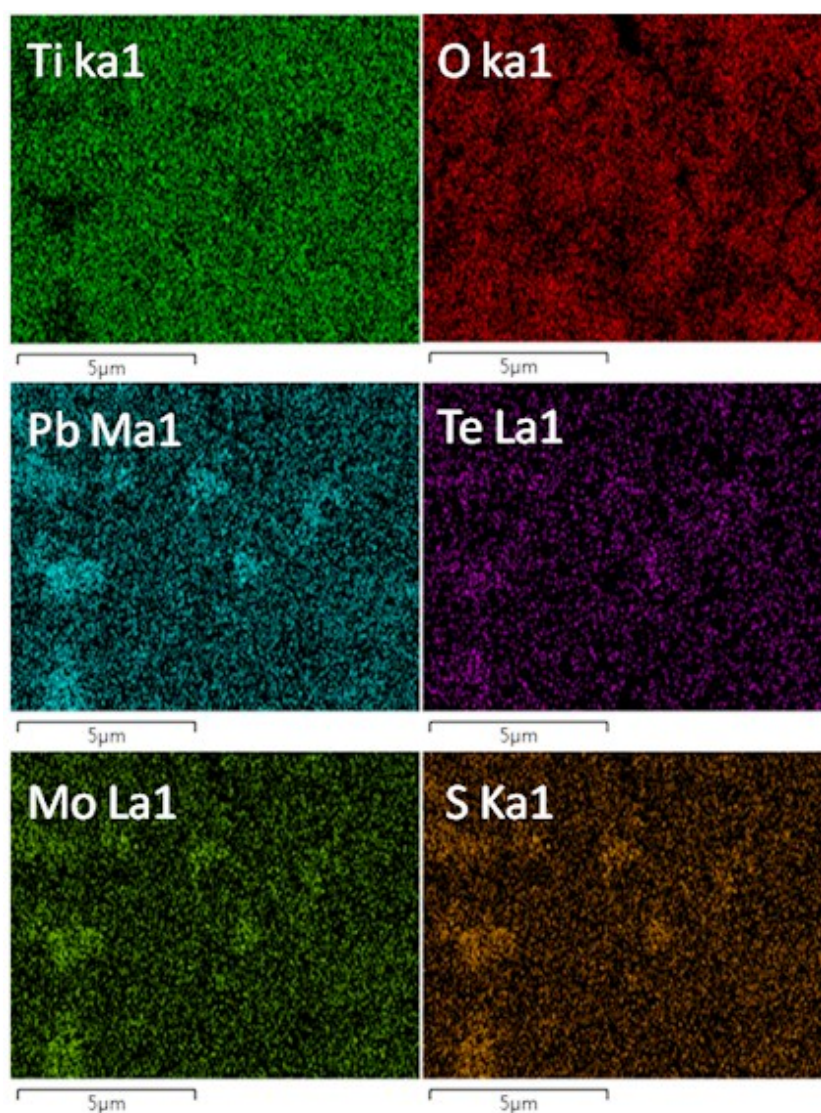


Figure S3 EDS elemental mapping images of the PbTe@MoS_x/TNAs.

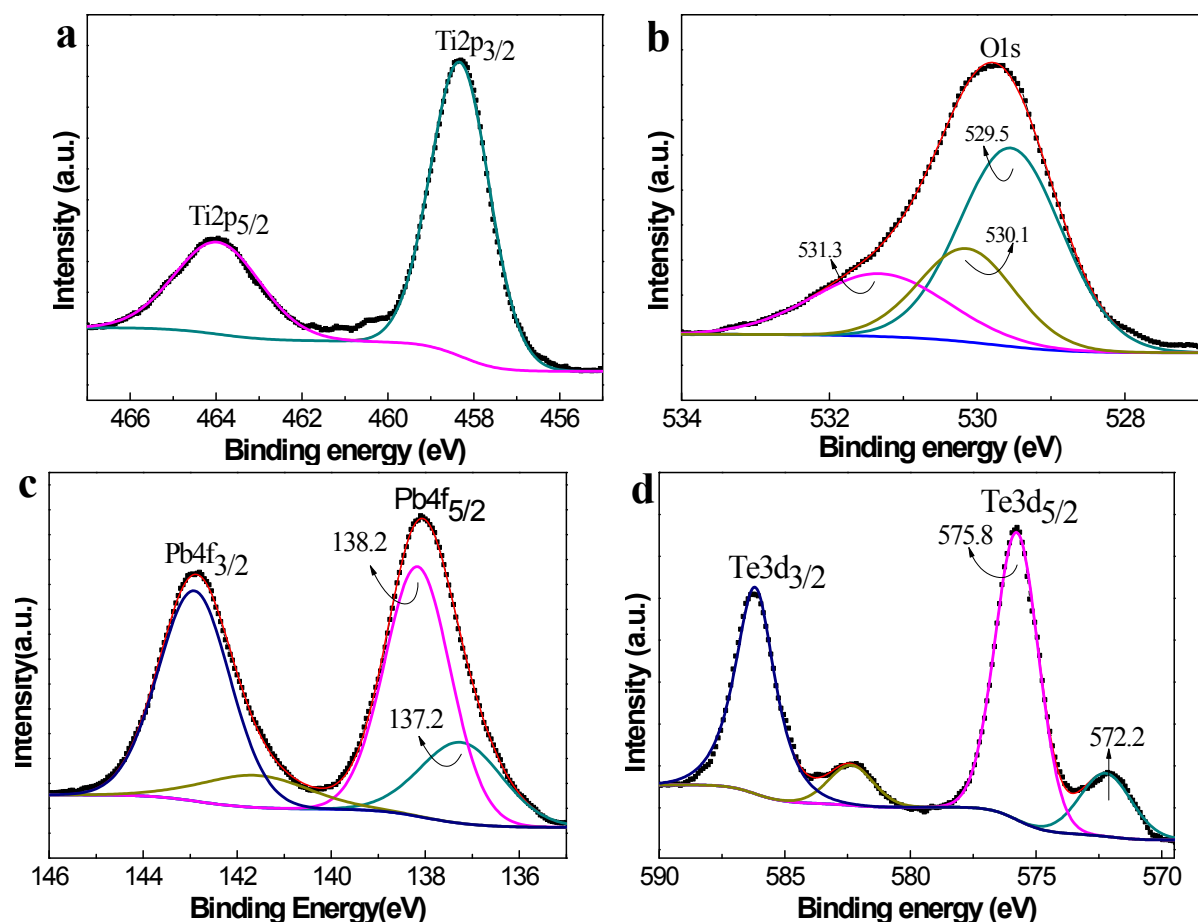


Figure S4 High resolution XPS spectra of (a) Ti2p, (b) O1s, (c) Pb4f and (d) Te3d of the PbTe/TNAs

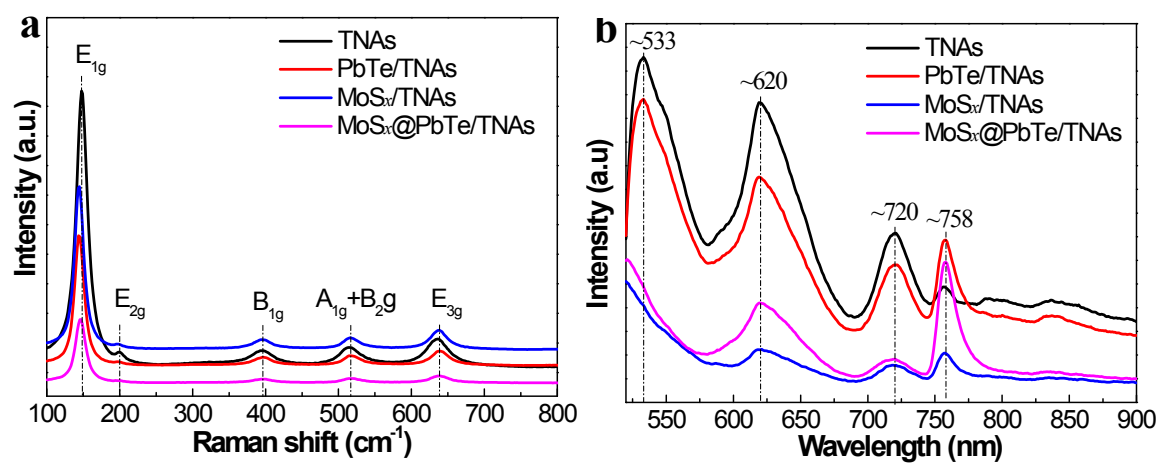


Figure S5 Raman spectroscopy (a) and PL emission spectra (b) of the samples.

Table S1 Chemical composition of the sample from EDS (at. %)

Sample	Te L	Pb M	S K	Mo	O K	Ti K
PbTe/TNAs	1.48	1.02	—	—	61.42	36.08
MoS _x @PbTe/TNAs	6.69	9	4.5	2.56	60.09	17.15

Table S2 Simulated values of the devices in equivalent circuits of the samples

Sample	R_s ($\Omega \cdot \text{cm}^{-2}$)	R_1 ($\Omega \cdot \text{cm}^{-2}$)	R_2 ($\Omega \cdot \text{cm}^{-2}$)	C_1 ($\text{F} \cdot \text{cm}^{-2}$)	C_2 ($\text{F} \cdot \text{cm}^{-2}$)
TNAs	0.5786	6.036	924.2	0.02556	0.009913
PbTe/TNAs	2.541	50.05	183.5	0.0008415	0.01547
MoS _x /TNAs	1.015	3.340	344.4	0.01453	0.01047
MoS _x @PbTe/TNA	1.013	3.291	103.8	0.02385	0.01572

s