

Supporting Information

Three-dimensional Porous Microspheres Comprising Hollow Fe₂O₃ Nanorods/CNT Building Blocks with Superior Electrochemical Performance for Lithium Ion Batteries

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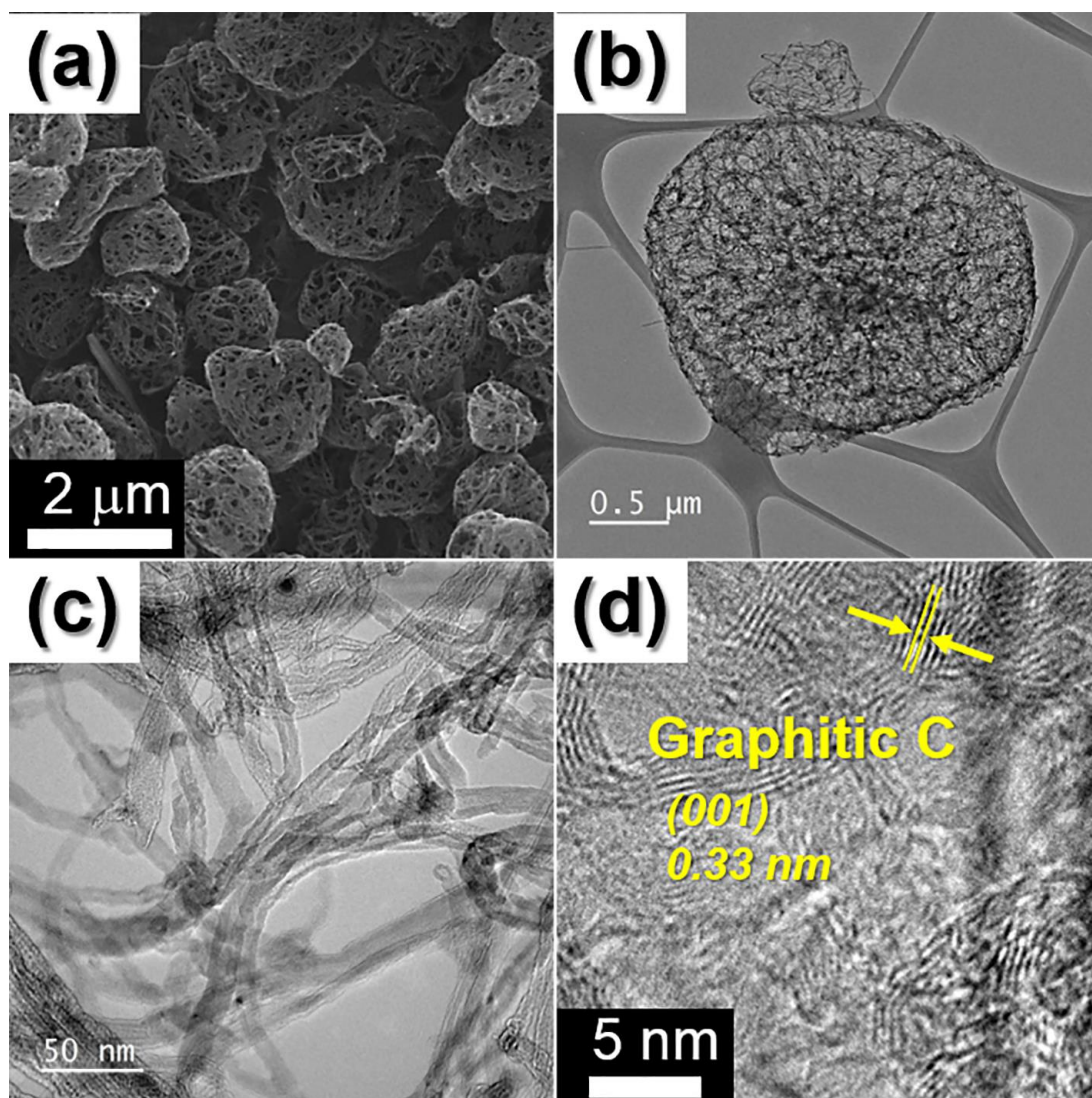


Fig. S1 (a) SEM, (b,c) TEM and (d) HR-TEM images of CNT porous microspheres.

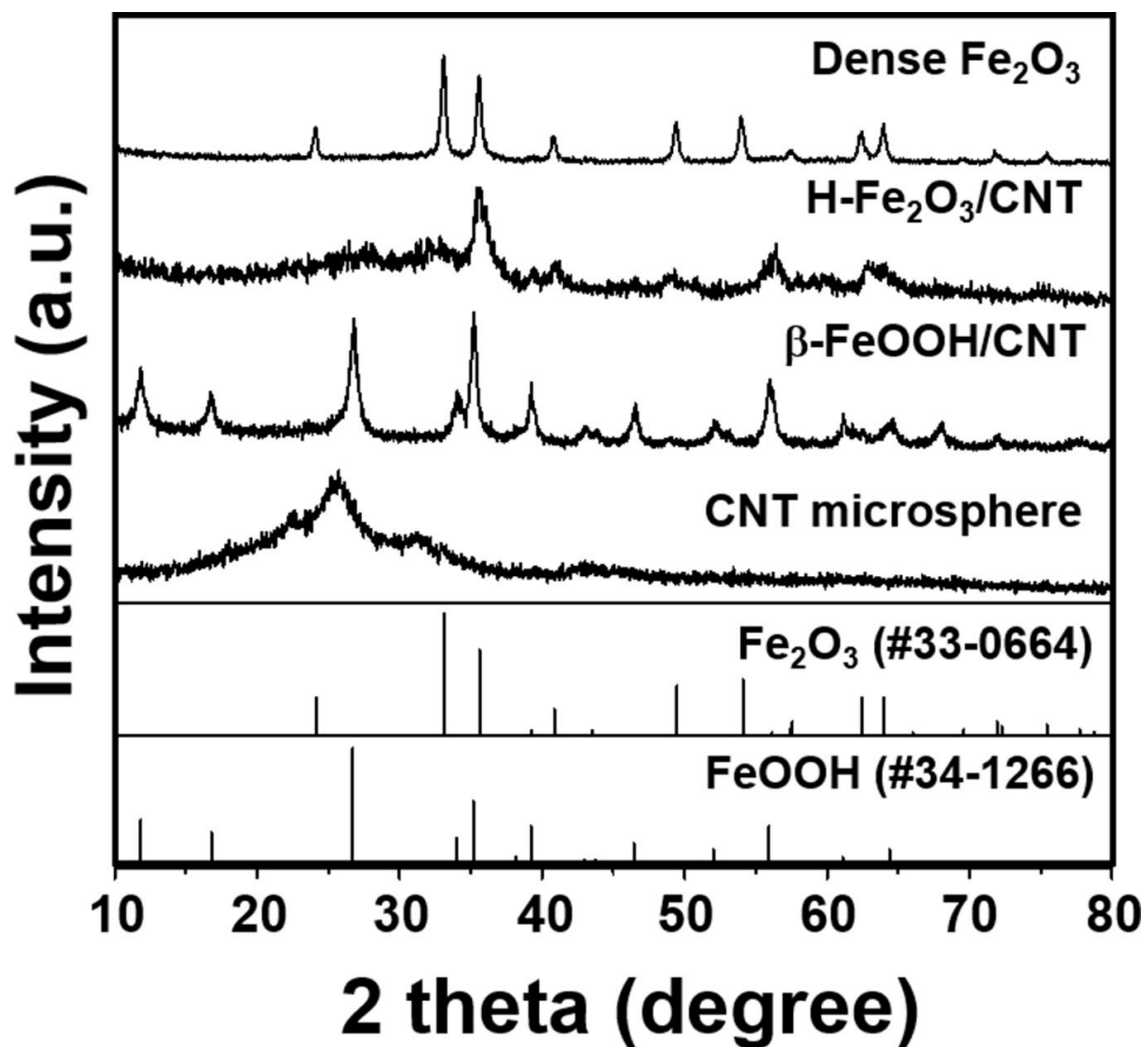


Fig. S2 XRD patterns of β -FeOOH/CNT microspheres, H- Fe_2O_3 /CNT microspheres, dense Fe_2O_3 nanorods and CNT porous microspheres.

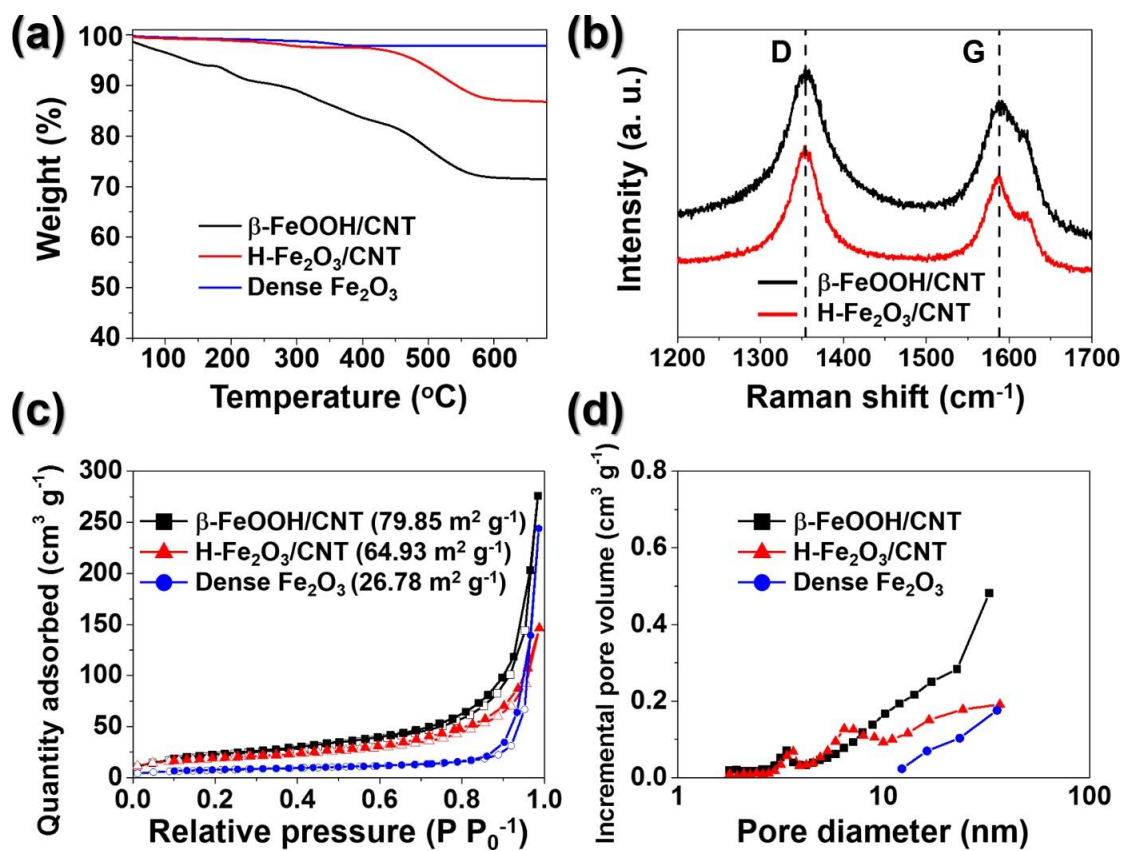


Fig. S3 (a) TGA curves, (b) Raman spectra, (c) N₂ adsorption and desorption isotherms and (d) BJH pore size distributions of β -FeOOH/CNT microspheres, H-Fe₂O₃/CNT microspheres and dense Fe₂O₃ nanorods.

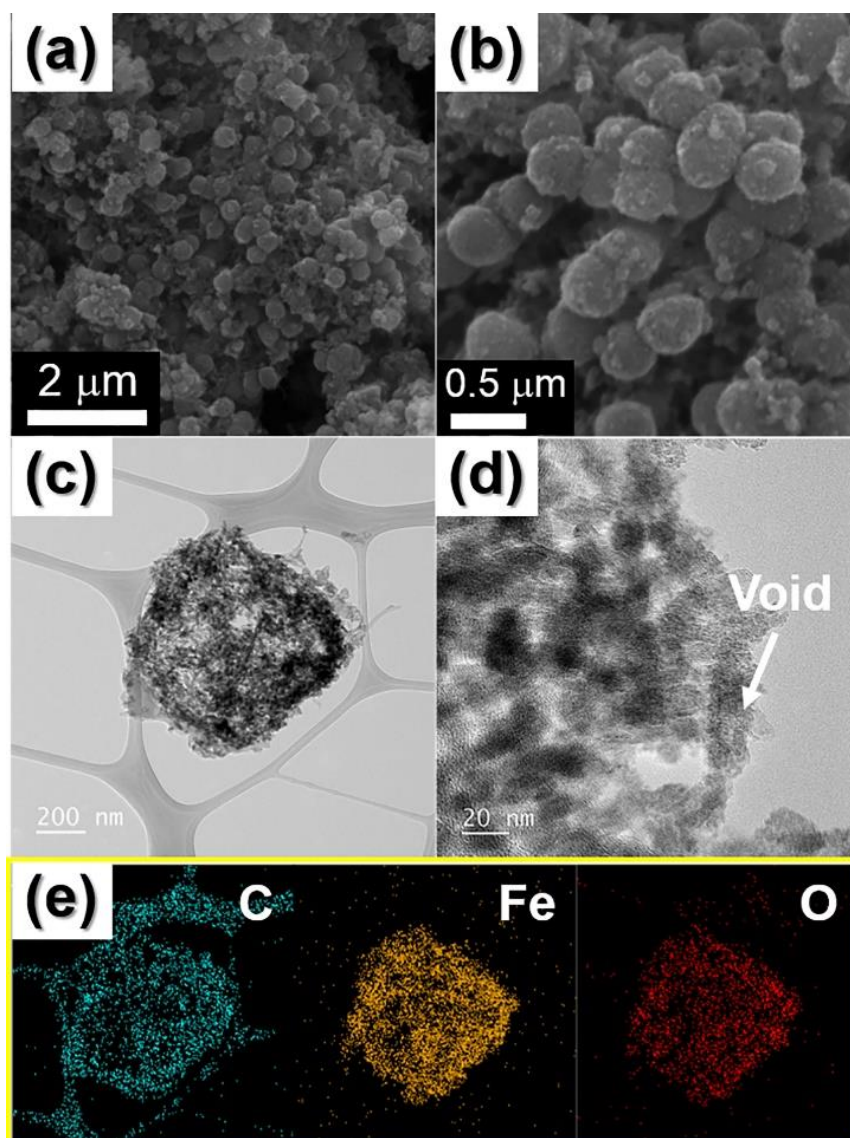


Fig. S4 (a and b) SEM, (c and d) TEM and (e) EDX mapping images of H-Fe₂O₃/CNT microspheres after 200 cycles.

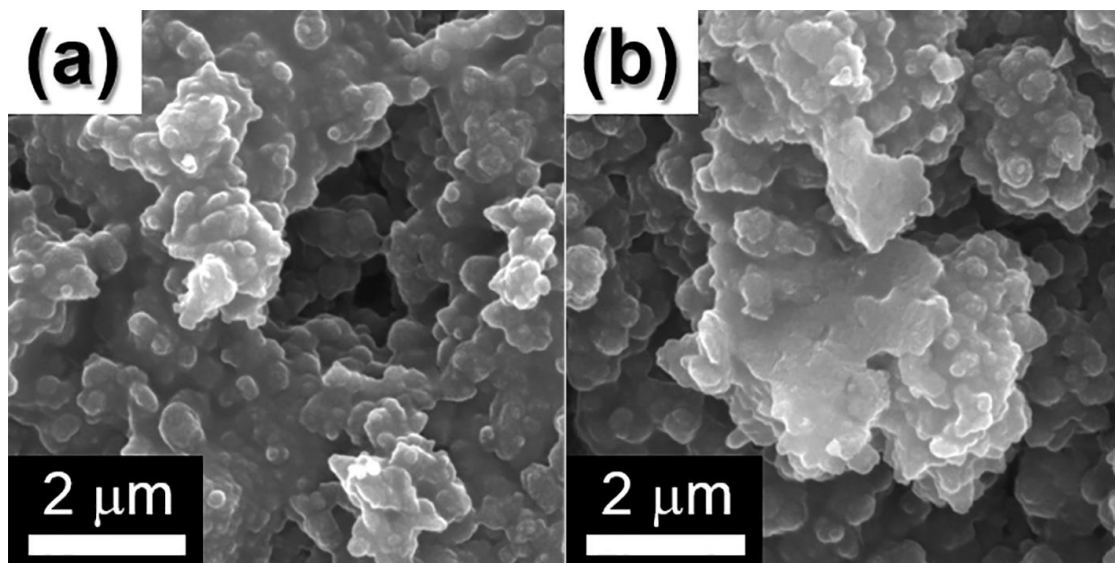


Fig. S5 (a and b) SEM images of dense Fe₂O₃ nanorods obtained after 200 cycles.

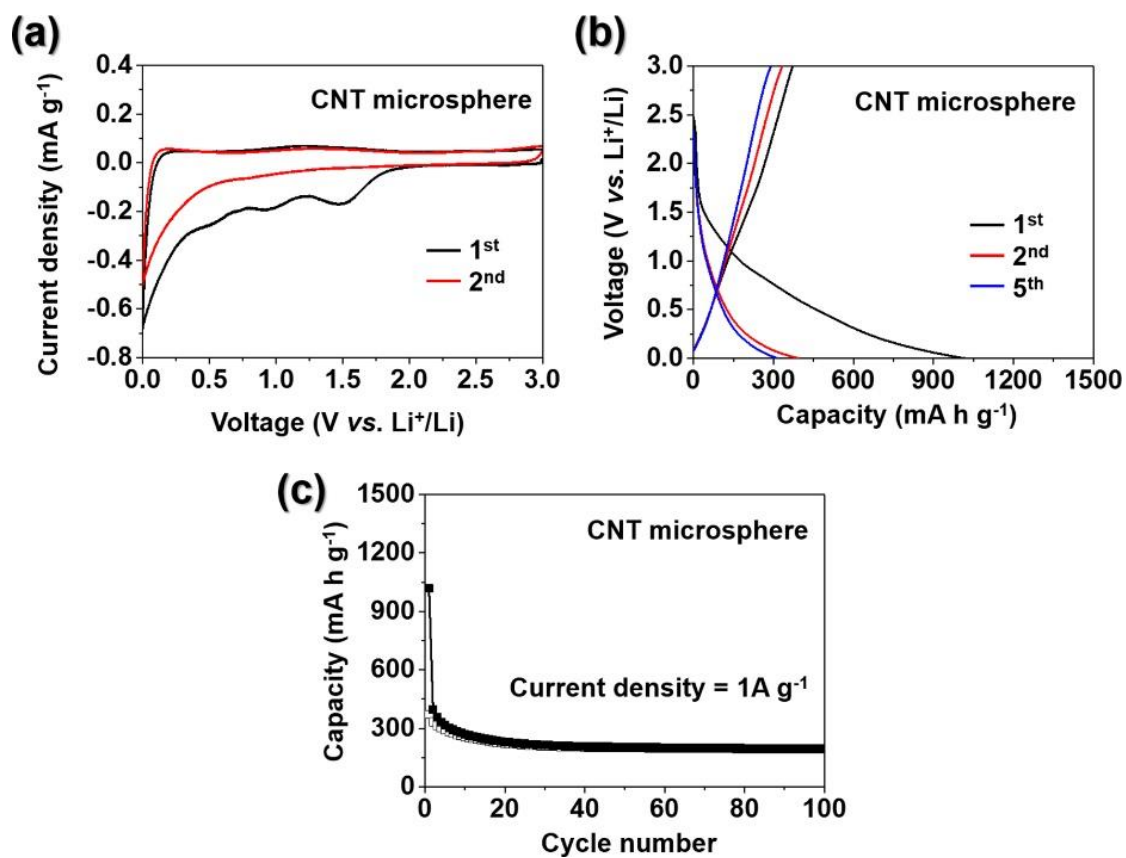


Fig. S6 (a) CV curves, (b) charge-discharge profiles and (c) cycling performance of CNT microspheres.

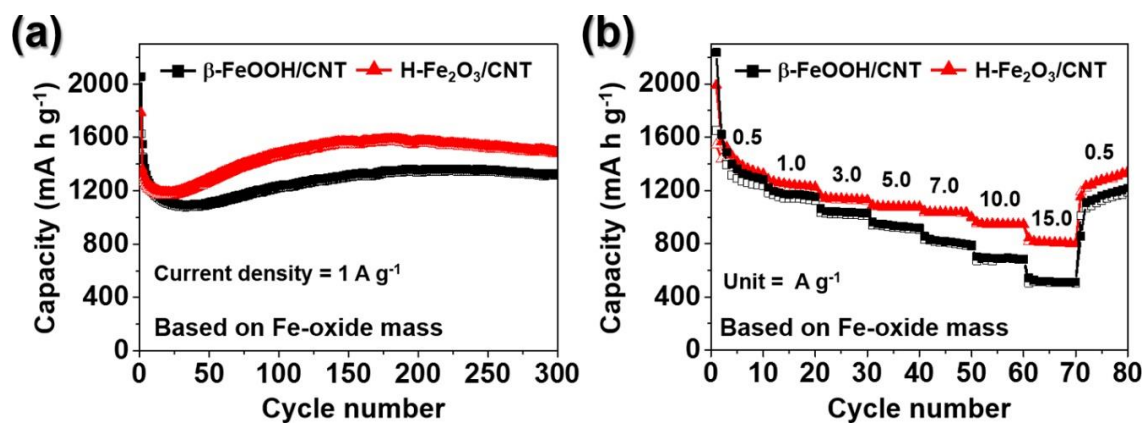
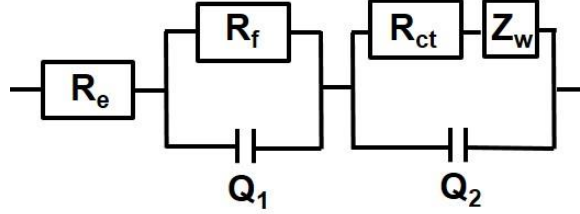


Fig. S7 (a) cycling and (b) rate performances of β -FeOOH/CNT and H-Fe₂O₃/CNT microspheres based on Fe-oxide mass.

Equivalent circuit model



R_e : the electrolyte resistance, corresponding to the intercept of high frequency semicircle at Z' axis

R_f : the SEI layer resistance corresponding to the high-frequency semicircle

Q_1 : the dielectric relaxation capacitance corresponding to the high-frequency semicircle

R_{ct} : the denote the charger transfer resistance related to the middle-frequency semicircle

Q_2 : the associated double-layer capacitance related to the middle-frequency semicircle

Z_w : the Li-ion diffusion resistance

Fig. S8 Randle-type equivalent circuit model used for AC impedance fitting.

Table S1. Electrochemical properties of the Fe₂O₃ materials with various structures as anode materials for LIBs.

Materials	Voltage range [V]	Current rate [A g ⁻¹]	Discharge capacity after cycling [mA h g ⁻¹]	Rate capability [mA h g ⁻¹]/[A g ⁻¹]	Ref
<i>H-Fe₂O₃/CNT microspheres</i>	0.001 – 3.0	1.0	1307 (300th)	703 (15.0 A g⁻¹)	Our work
Hollow Fe ₂ O ₃ spheres	0.05-3.0	0.2	710 (100 th)	-	[1]
Hierarchical hollow spheres composed of Fe ₂ O ₃ nanosheets	0.01-3.0	0.5	815 (200 th)	330 (5.0 A g ⁻¹)	[2]
Hierarchical Fe ₂ O ₃ microboxe	0.005-3.0	0.2	945 (30 th)	-	[3]
Hollow Fe ₂ O ₃ nanospheres	0.01-3.0	0.25	490 (50 th)	-	[4]
Hollow Fe ₂ O ₃ nanobarrels	0.01-3.0	0.5	916 (100 th)	403 (10.0 A g ⁻¹)	[5]
Multi-shelled hollow Fe ₂ O ₃ spheres	0.05-3.0	0.4	861 (50 th)	294 (4.0 A g ⁻¹)	[6]
Graphene-constructed hollow Fe ₂ O ₃ spheres	0.01-3.0	0.1	950 (50 th)	640 (1.0 A g ⁻¹)	[7]
Carbon coated hollow Fe ₂ O ₃ sphere	0.01-3.0	0.3	950 (100 th)	-	[8]
Fe ₂ O ₃ nanorods	0.005-3.0	0.5	970 (100 th)	300 (5.0 A g ⁻¹)	[9]
Spindle-like Fe ₂ O ₃	0.01-3.0	0.2	911 (50 th)	424 (10.0 A g ⁻¹)	[10]
Fe ₂ O ₃ nanoparticle-loaded carbon nanofibers	0.05-2.8	0.05	488 (75 th)	288 (0.5 A g ⁻¹)	[11]
Fe ₂ O ₃ nano-assembled spindle	0.005-3.0	0.1	~900 (40 th)	430 (1.0 A g ⁻¹)	[12]

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