

$M_xP(M=Co/Ni)@Carbon$ Core–Shell Nanoparticles Embedded in 3D Cross-linked Graphene Aerogels Derived from Seaweed Biomass for Hydrogen Evolution Reaction

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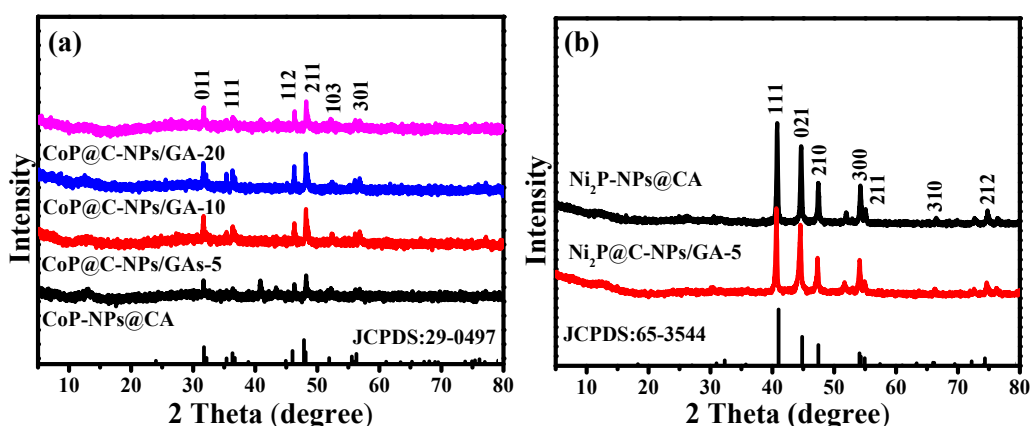


Figure S1 (a) XRD patterns of CoP-NPs@CA and CoP@C-NPs/GA-x, (b) Ni₂P-NPs@CA and Ni₂P@C-NPs/GA-x

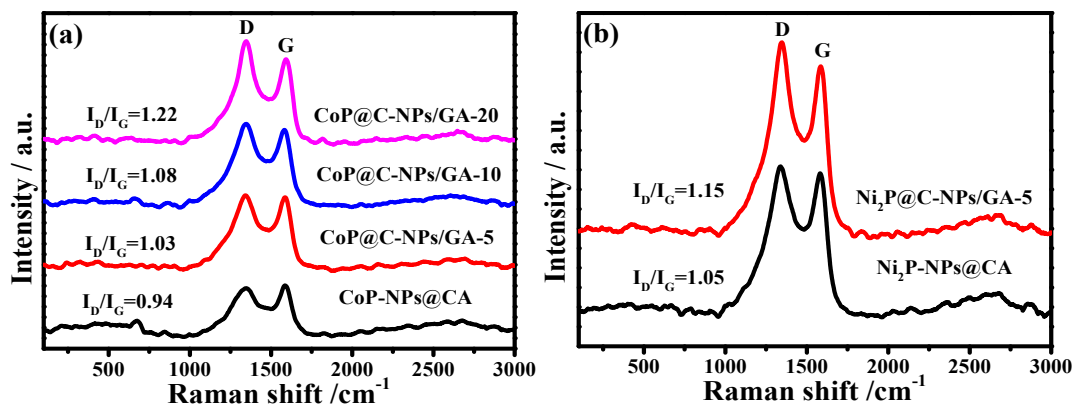


Figure S2 (a) Raman spectra of CoP-NPs@CA and CoP@C-NPs/GA-x, (b) Ni₂P-NPs@CA and Ni₂P@C-NPs/GA-5.

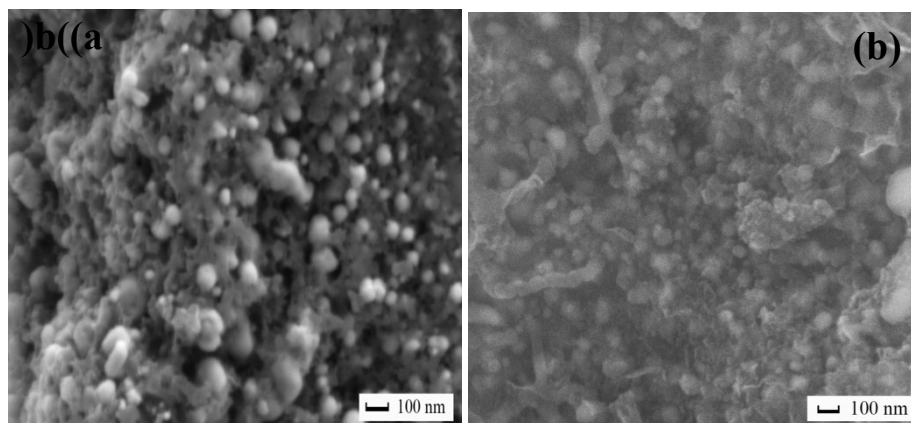


Figure S3 FESEM images of different mass ratio of graphene oxide (a) $\text{Ni}_2\text{P-NPs@CA}$, (b) $\text{Ni}_2\text{P@C-NPs/GA-5}$.

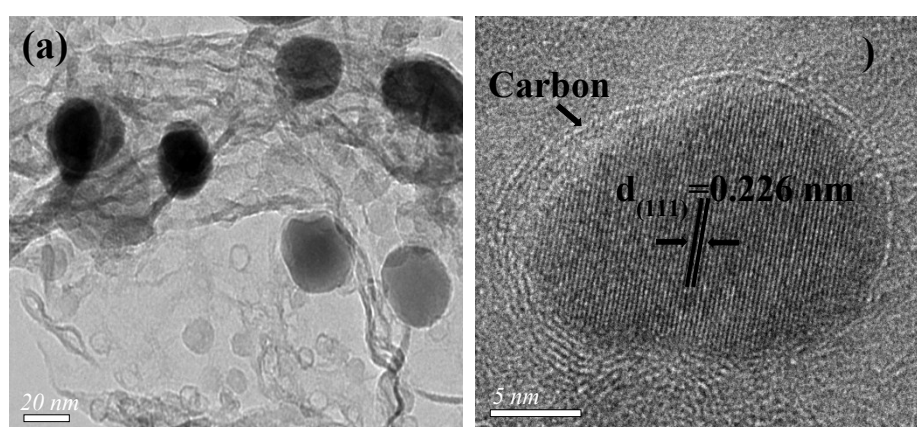


Figure S4 (a) High-magnification TEM images of the synthesized $\text{Ni}_2\text{P@C-NPs/GA-5}$, (b) HRTEM images of $\text{Ni}_2\text{P@C-NPs/GA-5}$.

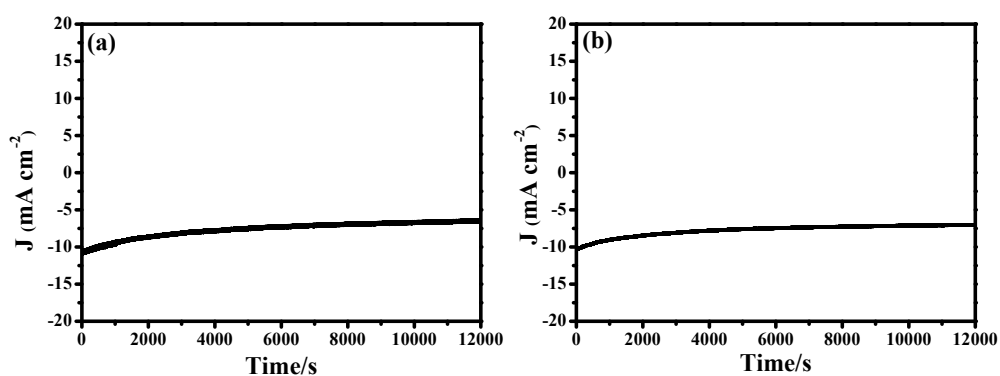


Figure S5 Stability tests of $\text{Ni}_2\text{P@C-NPs/GA-5}$ using the chronopotentiometric measurements (without iR corrections) with the fixed current density of 10 mA cm^{-2} in (a) $0.5 \text{ M H}_2\text{SO}_4$ and (b) 1 M KOH solutions.

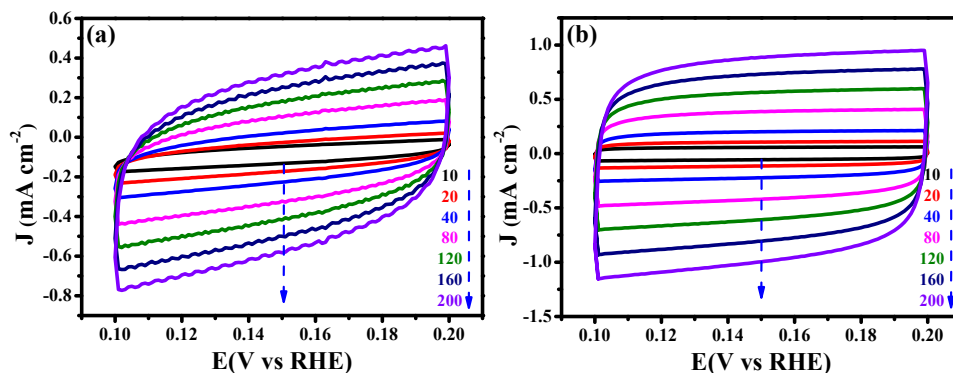


Figure S6 CVs performed at various scan rates in the region of 0.10–0.20 V versus RHE for CoP-NPs@CA electrocatalyst (a) in 0.5 M H₂SO₄ electrolytes and (b) in 1 M KOH electrolytes.

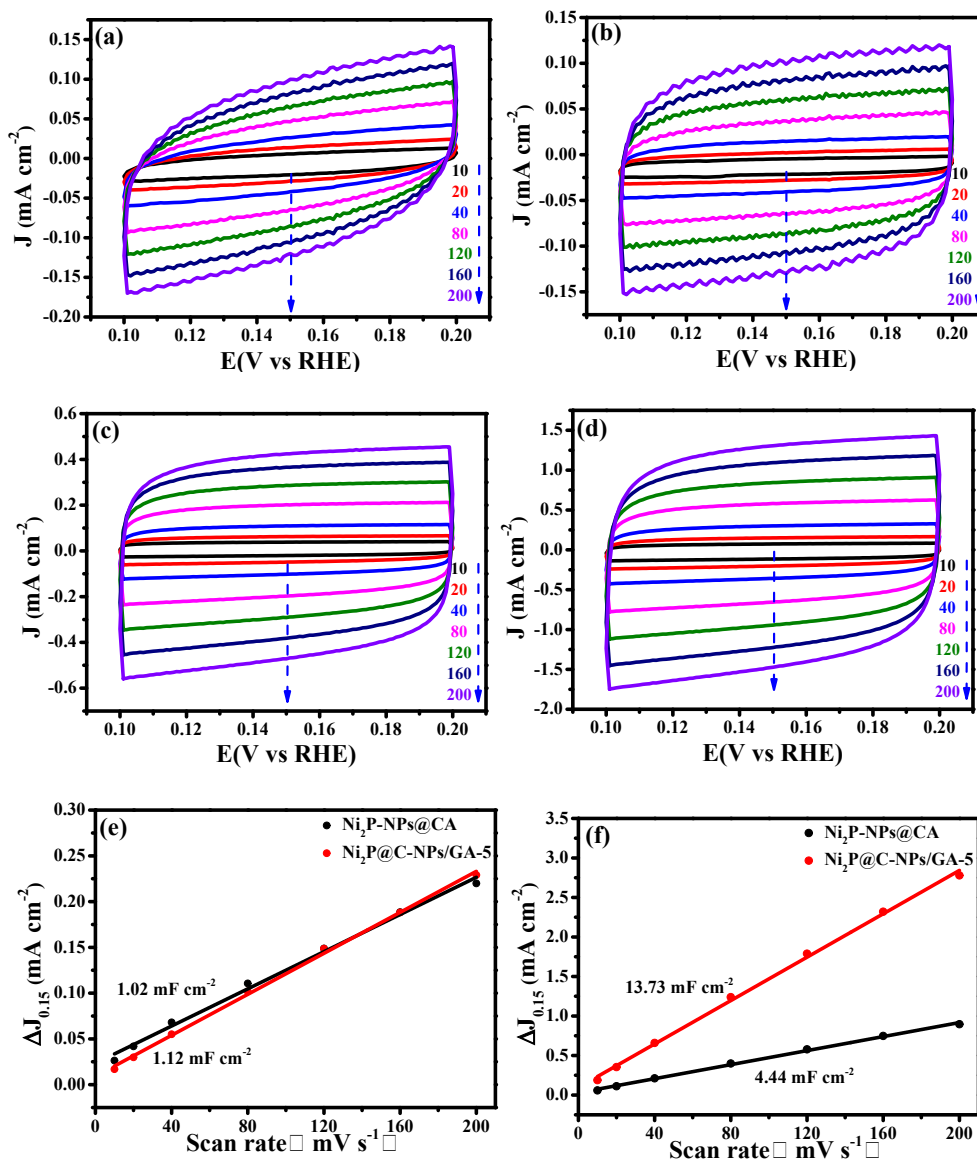


Figure S7 CVs performed at various scan rates in the region of 0.10–0.20 V versus RHE for Ni₂P-NPs@CA (a) in 0.5 M H₂SO₄ electrolytes and (c) in 1 M KOH electrolytes, Ni₂P@C-NPs/GA-5 (b) in 0.5 M H₂SO₄ electrolytes and (d) in 1 M KOH electrolytes. Ni₂P-NPs/CA and Ni₂P@C-NPs/GA-5 in current density at 0.15 V versus RHE

plotted against the scan rate and fitted to a linear regression allows for the estimation of C_{dl} (e) in 0.5 M H_2SO_4 electrolytes and (f) in 1 M KOH electrolytes.

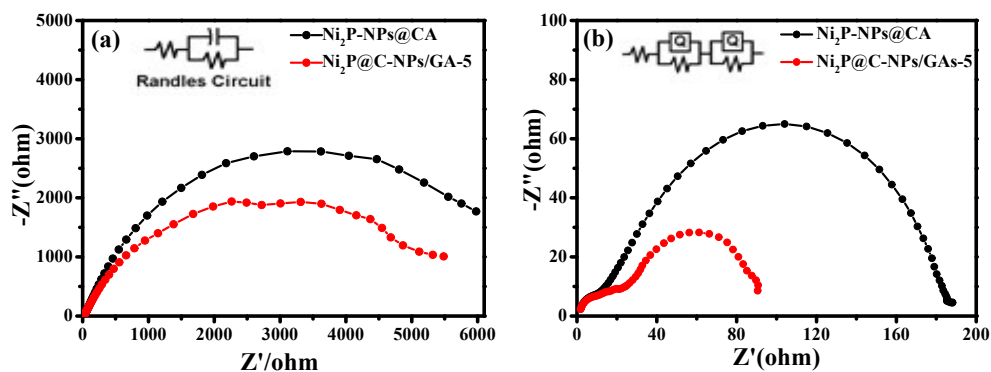


Figure S8 Nyquist plots of $Ni_2P-NPs@CA$ and $Ni_2P@C-NPs/GA-5$ over the frequency range from 100 kHz to 0.1Hz (a) at $\eta=135$ mV with an AC voltage of 10 mV in in 0.5 M H_2SO_4 electrolytes,(b) at $\eta=377$ mV with an AC voltage of 10 mV in 1 M KOH electrolytes.

Table S1 Comparison of catalytic parameters of different HER catalysts in 0.5 M H₂SO₄.

	Loading (mg.cm ⁻²)	η_{10} (mV)	Tafel slop (mV dec ⁻¹)	C_{dl} (mF cm ⁻²)	Reference
Porous CoP concave polyhedron	0.35	133	51	/	1
CoP@C	0.31	130	64	9.30	2
CoP@NC-T	0.31	124	59	9.90	2
CoP Hollow Polyhedron	0.10	159	59	/	3
CoP/CNTs	/	139	52	/	4
Co@NC/NG	0.29	/	79	37.30	5
Co@NGF	0.28	125	94	/	6
Ni₂P/N-doped Graphene	0.18	102	59	/	7
CoSe₂@DC	0.36	132	82	51.10	8
Ni₅P₄-Ni₂P nanosheet	68.2 (electrode)	120	79	/	9
Ni₁₂P₅/CNT	0.18	240	81	/	10
MoP	0.36	125	54	/	11
Mo₂C@NC	0.28	124	60	73.20	12
SV-MoS₂	/	170	60	/	13
CoP NPs	0.35	221	87	/	14
Free CoP NPs	/	167	88	2.57	15
CoP-RGO	/	157	70	5.23	15
Mo₃S₁₃@rGO-CNTs	/	179	67	/	26
aerogels					
GCA-MoSe₂-2 aerogels	/	228	68	1.31	27
MoSe₂/carbon fiber aerogel	/	179	62	/	28
CoP@C-NPs/GA-5	0.28	120	57	55.83	Our work

Table S2 Comparison of catalytic parameters of different HER catalysts in 1 M KOH.

	Loading (mg.cm⁻²)	η_{10} (mV)	Tafel slop (mV dec⁻¹)	C_{dl} (mF cm⁻²)	Reference
CoP@C	0.31	170	83		2
CoP/NF	/	182	109	/	16
CoP nanowire arrays/CC	0.92	209	129	/	17
CoP nanoparticle/Ti	0.18	170	66		18
NiMoN/CC	1.10	109	95	21	19
Ni-P/CP	25.80	117	85		20
FeP nanorod arrays/CC	1.50	218	146	/	21
NiFeVS/NF	38.25	161	96	32.30	22
	(electrode)				
WP₂	/	225	84		23
Lepidocrocite VOOH hollow nanospheres	0.80	164	104		24
Fe₂O₃/Fe@CN	0.28	330	114	16.20	25
CoP@C-NPs/GA-5	0.28	225	66	18.85	Our work

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