Electronic Supplementary Inofrmation

Freestanding CoSeO₃·H₂O Nanoribbon/Carbon Nanotube Composite Paper for 2.4 V High-Voltage, Flexible, Solid-State Supercapacitors

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Figure S1. SAED patterns taken from the same place of an individual $CoSeO_3 \cdot H_2O$ nanoribbon at initial stage (a) and (b) after a few seconds of electron beam irradiation .



Figure S2. The SEM images of the product with the (a) presence and (b) absence of 3.0 g of PVP surfactant.



Figure S3. The photographs of the as-prepared $CoSeO_3 \cdot H_2O$ paper with high flexibility.



Figure S4. The photograph depicts excellent flexible performance of the as-prepared $CoSeO_3$ ·H₂O, HWCNTs paper by a glass rod.



Figure S5. Typical XRD patterns of CoSeO₃·H₂O /HWCNTs paper and CoSeO₃·H₂O power.



Figure S6. The electronic conductivity of pure HWCNTs paper, CSO/H-1, CSO/H-2, CSO/H-3 composite papers, and $CoSeO_3 \cdot H_2O$ paper.



Figure S7. In-plane SEM images of representative $CoSeO_3$ ·H₂O/HWCNTs paper with different mass ratio at (a) 9:1 and (b) 10:1.



Figure S8. Typical strain–stress curves of pure $CoSeO_3 \cdot H_2O$ paper, $CoSeO_3 \cdot H_2O$ /HWCNTs (9:1) and $CoSeO_3 \cdot H_2O$ /HWCNTs (8:1) composite papers.



Figure S9. Electrochemical impedance spectra of the $CoSeO_3 \cdot H_2O/HWCNTs$ paper electrode at the initial state and after 5,000 cycles, respectviely.



Figure S10. (a) Typical TEM image of HWCNTs. (b, c) The photograph of the as-prepared HWCNTs film with high flexiblity.



Figure S11. (a) Characteristic CV curves of the HWCNTs@Ni foam electrode in aqueous KOH electrolyte at different scan rates. (b) Galvanostatic charge–discharge curves. (c) C_s of HWCNTs@Ni foam electrode at different current densities. The capacitance of the activated carbon is 186 F g⁻¹ for the applied current density of 1 A g⁻¹.



Figure S12. Cyclic voltametry (CV) of the $CoSeO_3 \cdot H_2O$ @ Ni foam and HWCNTs@ Ni foam in aqueous KOH electrolyte at a scan rate of 15 mV s⁻¹.

Table S1. Summary of the performance metrics of various freestanding paper-based asymmetrical supercapacitors previously reported and the present device.

Electrode materials (paper-like ASCs)	working Potential (V)	Specific capacitance (F g ⁻¹)	maximum Energy density (Wh kg ⁻¹)	maximum power density	Cyclic performance	Ref.
				(kW kg ⁻¹)	(retention)	
RGO/MnO ₂ // RGO	1.5	30.5	9.48	1.027	3600 (93%)	[1]
CoO@NiO/Activated carbon textiles // Activated carbon textiles /graphene	1.6	147.6	52.26	9.53	2000 (97.53%)	[2]
3D MnO ₂ network// Activated carbon	1.7	762	40.2	6.23	8000 (90%)	[3]
MnO ₂ /graphene // Carbon nanotubes/grphene	1.8	69.4	31.8	9.19	10000 (84.4%)	[4]
H-TiO ₂ @Ni(OH) ₂ // N-doped carbon Nanowires	1.8	150.6	70.9	20.9	5000 (90%)	[5]
NiMn-LDH/ Carbon nanotubes // RGO/ Carbon nanotubes	1.7	221	88.3	17.2	1000 (94%)	[6]
NiCo ₂ O ₄ / Carbon cloth// 3D porous graphene paper	1.8	71.32	60.9	11.36	5000 (96.8%)	[7]
RuO2-IL-CMG//						
IL-CMG(ionic liquid functionalized- chemically modified graphene)	1.8	175	19.7	6.8	2000 (79.4%)	[8]
CoSeO3·H2O/HWCNTs //HWCNTs	2.4	165.3	132.3	13.17	10000 (94.5%)	our work

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