

Journal Name

ARTICLE

Electronic Supplementary Material (ESI) for Nanoscale.

This journal is The Royal Society of Chemistry 2018

Supplementary Information for

Core-Shell Structure of Polydopamine-Coated Phosphorus-Carbon Nanotube Composite for High-Performance Sodium-Ion Batteries

Weili Liu^a, Xianxia Yuan^b, Xuebin Yu^{a*}

^aDepartment of Materials Science, Fudan University, Shanghai 200433, China

^bDepartment of Chemical Engineering, Shanghai Jiao Tong University, Shanghai 200240, China

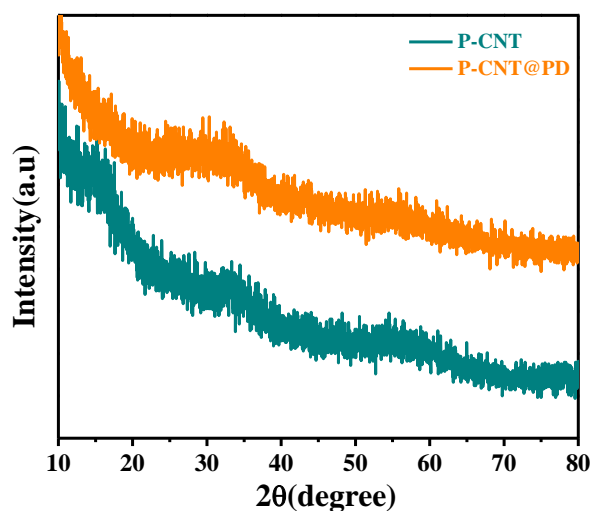


Figure S1. XRD patterns of P-CNT and P-CNT@PD composites.

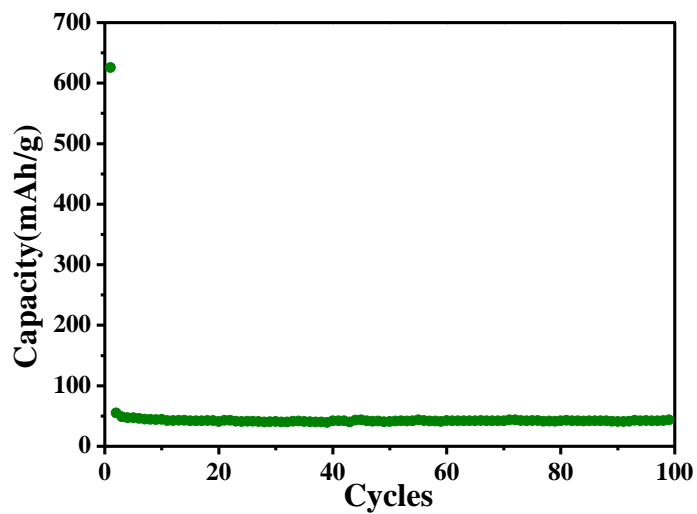


Figure S2. The specific capacity of carbon nanotube (CNT) cycling at a current density of 0.52 A/g between 0.001 and 2.0 V (vs. Na⁺/Na).

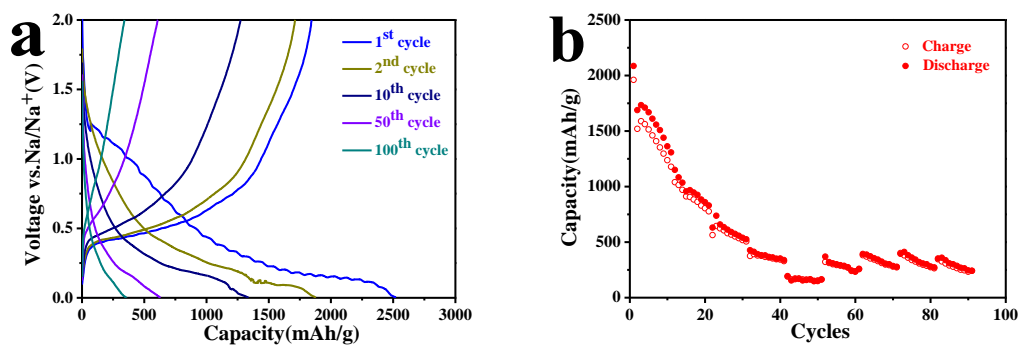
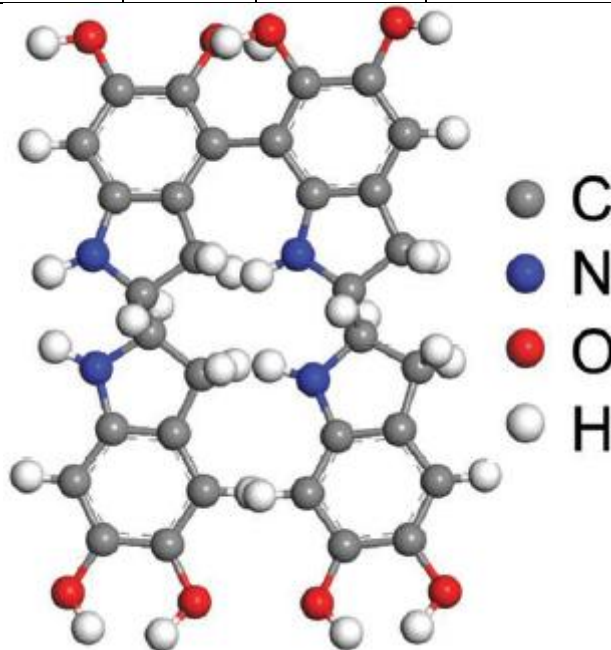


Figure S3. (a) Charge-discharge voltage profiles of P-CNT composite during cycling at a current density of 0.52 A/g between 0.001 and 2.0 V (vs. Na⁺/Na). (b) Rate performance at various current densities from 0.26 A/g to 5.2 A/g of P-CNT composite between 0.001 and 2.0 V (vs. Na⁺/Na).

Table S1. Elements in PD and atom number per PD molecular unit

Elements	N	O	H	C
Atom number per PD molecular unit	4	8	32	32



N content in PD as follows: $4 \times 14 / (4 \times 14 + 16 \times 8 + 32 + 32 \times 12) = 9.33\%$

N content in P-CNT@PD composite was 0.517% $[(0.448\% + 0.546\%) / 2]$

So the PD content in P-CNT@PD as follows: $0.517\% / 9.33\% = 5.54\%$

Table S2. Kinetic parameters of P-CNT and P-CNT@PD electrodes

	50	100	200
$R_{\text{eleP-CNT}}$	34.65	30.16	25.26
$R_{\text{intP-CNT}}$	246.2	297.6	406.6
$R_{\text{eleP-CNT@PD}}$	6.249	6.65	9.536
$R_{\text{intP-CNT@PD}}$	273	220	158.2