

Single atom alloy catalyst for SO₃ decomposition: Enhancement of platinum catalyst's performance by Ag atom embedding

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Supporting Information

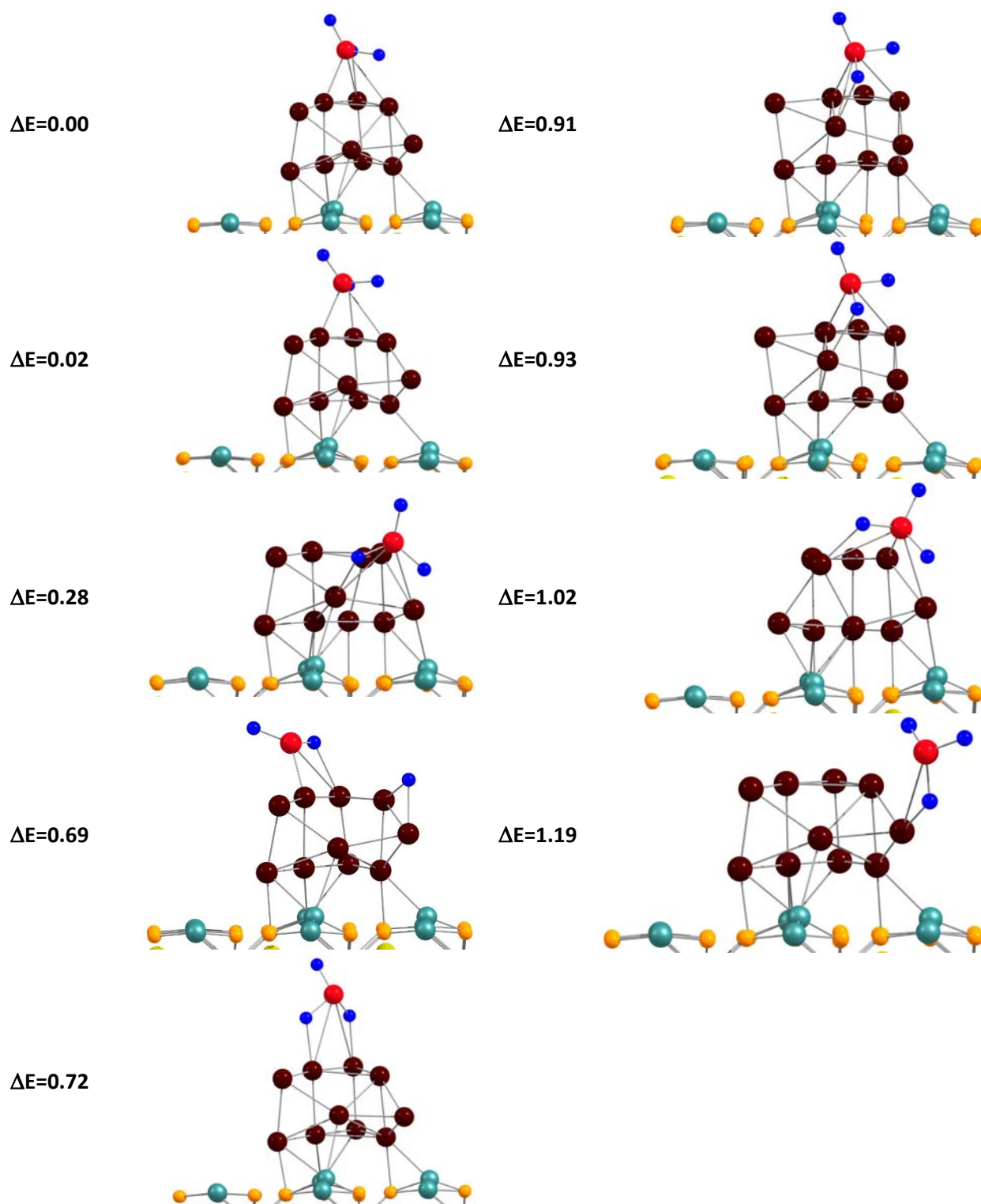


Figure-S1: Low lying isomer of SO₃-(Pt₁₀@Al₂O₃) complex. ΔE indicates relative stability.

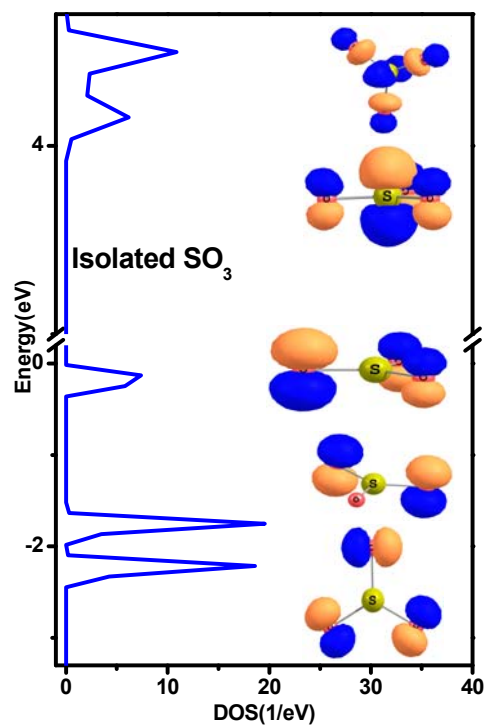


Figure-S2: Electronic density of states of pristine SO_3 molecule. Molecular Orbital pictures has been obtained via all electron calculations at B3LYP/6-311+g(d, p) level of theory.

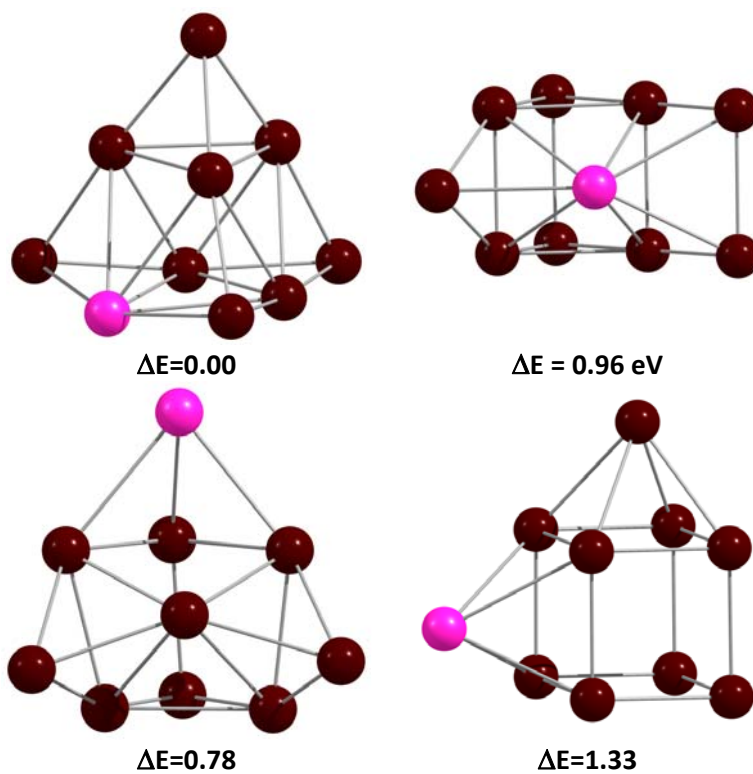


Figure-S3: Low lying isomer of gas phase AgPt_9 cluster. ΔE indicates relative stability.

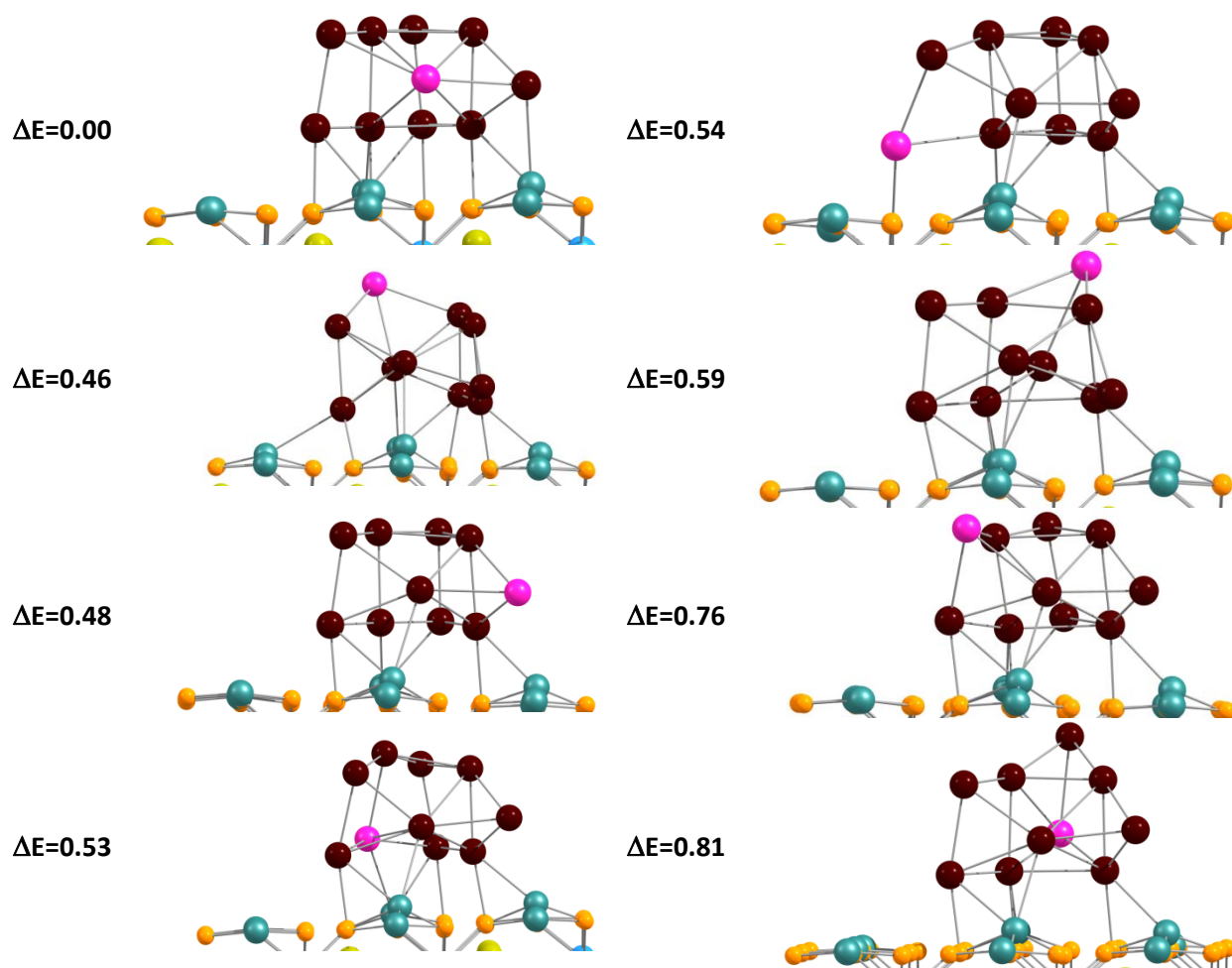


Figure-S4: Low lying isomer of $\text{AgPt}_9@\text{Al}_2\text{O}_3$ cluster. ΔE indicates relative stability.

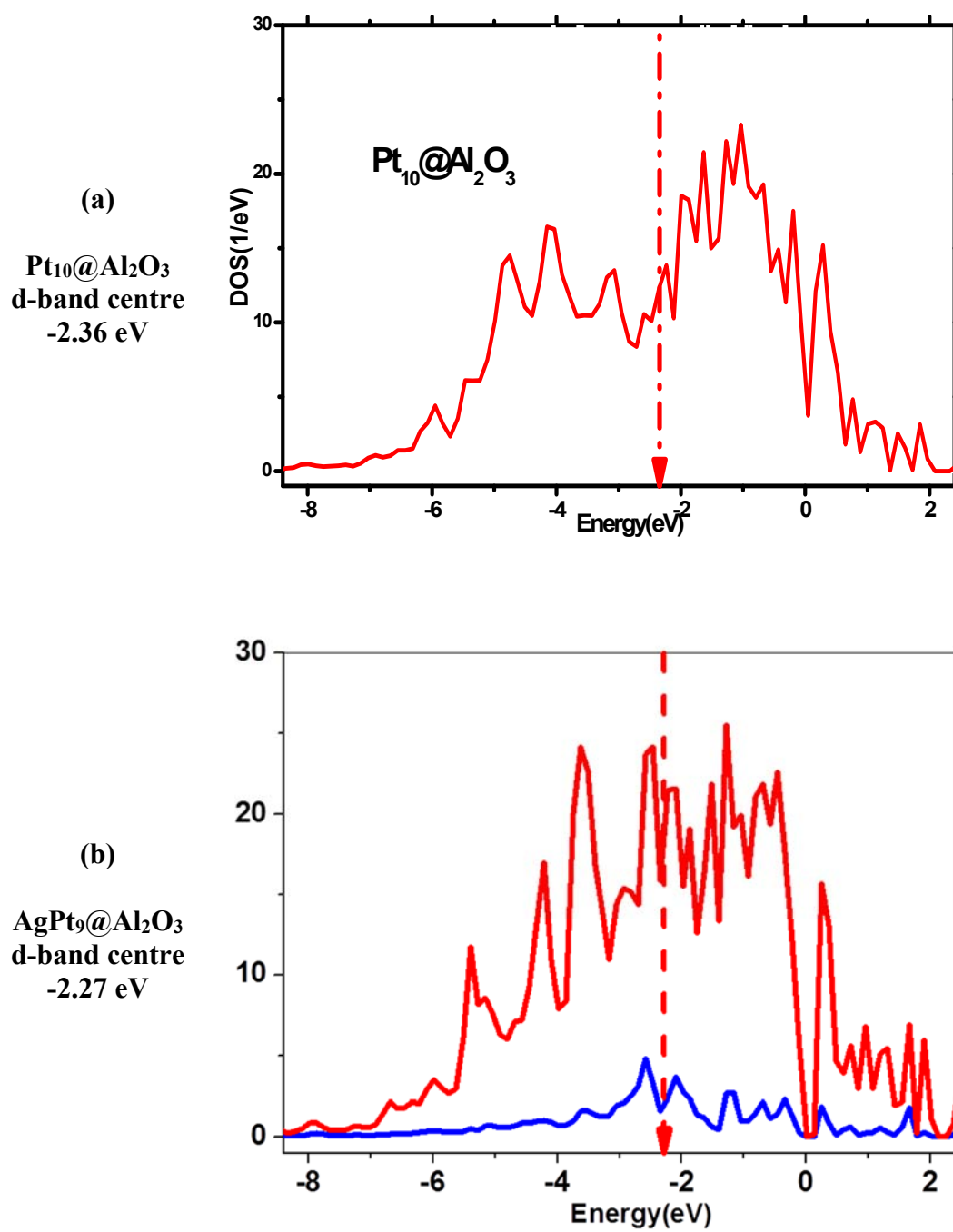


Figure-S5: Comparative DOS spectrum of (a) $\text{Pt}_{10}@ \text{Al}_2\text{O}_3$ and (b) $\text{AgPt}_9@ \text{Al}_2\text{O}_3$.

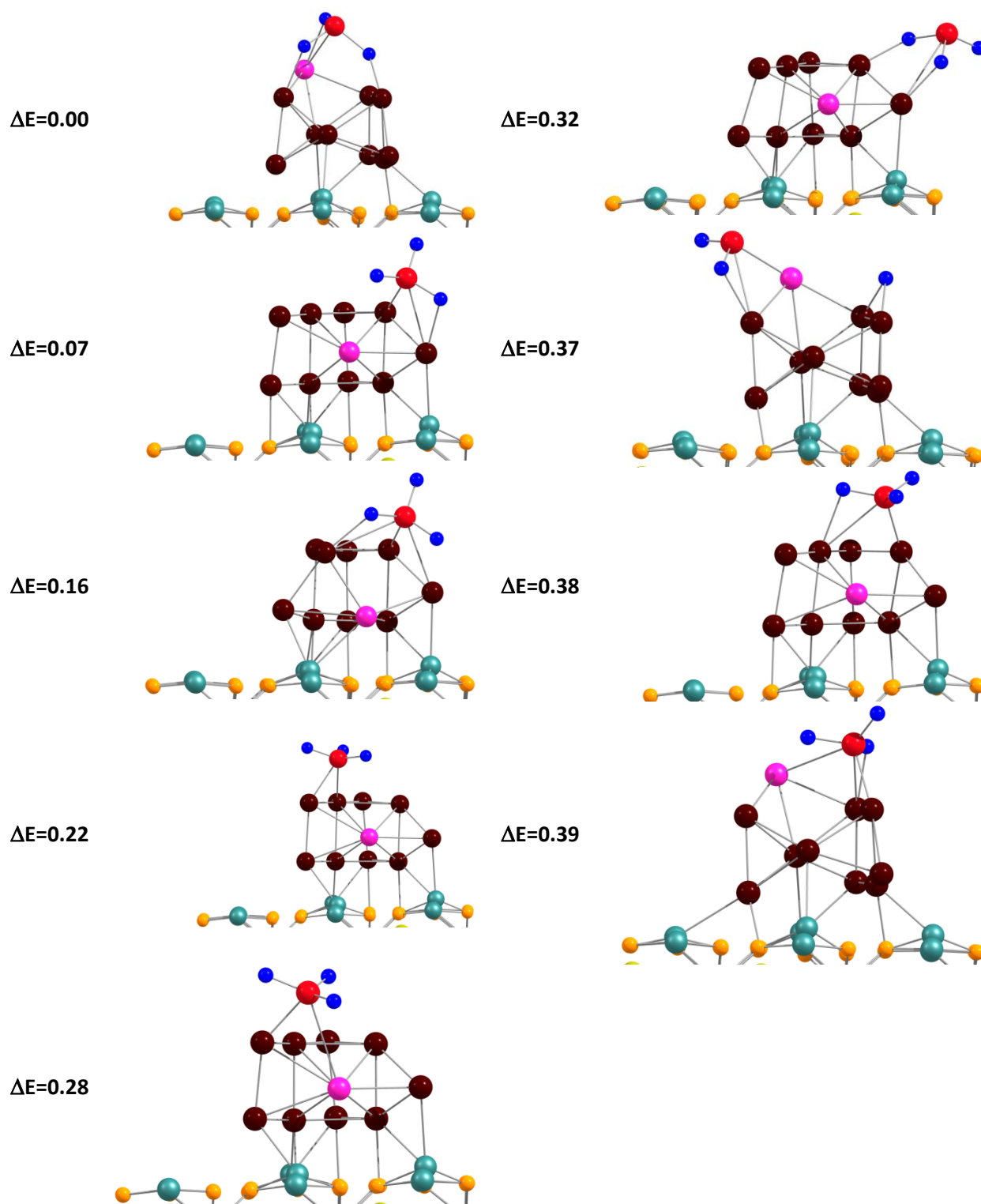


Figure-S6: Low lying isomer of $\text{SO}_3\text{-(AgPt}_9\text{@Al}_2\text{O}_3\text{)}$ complex. ΔE indicates relative stability.

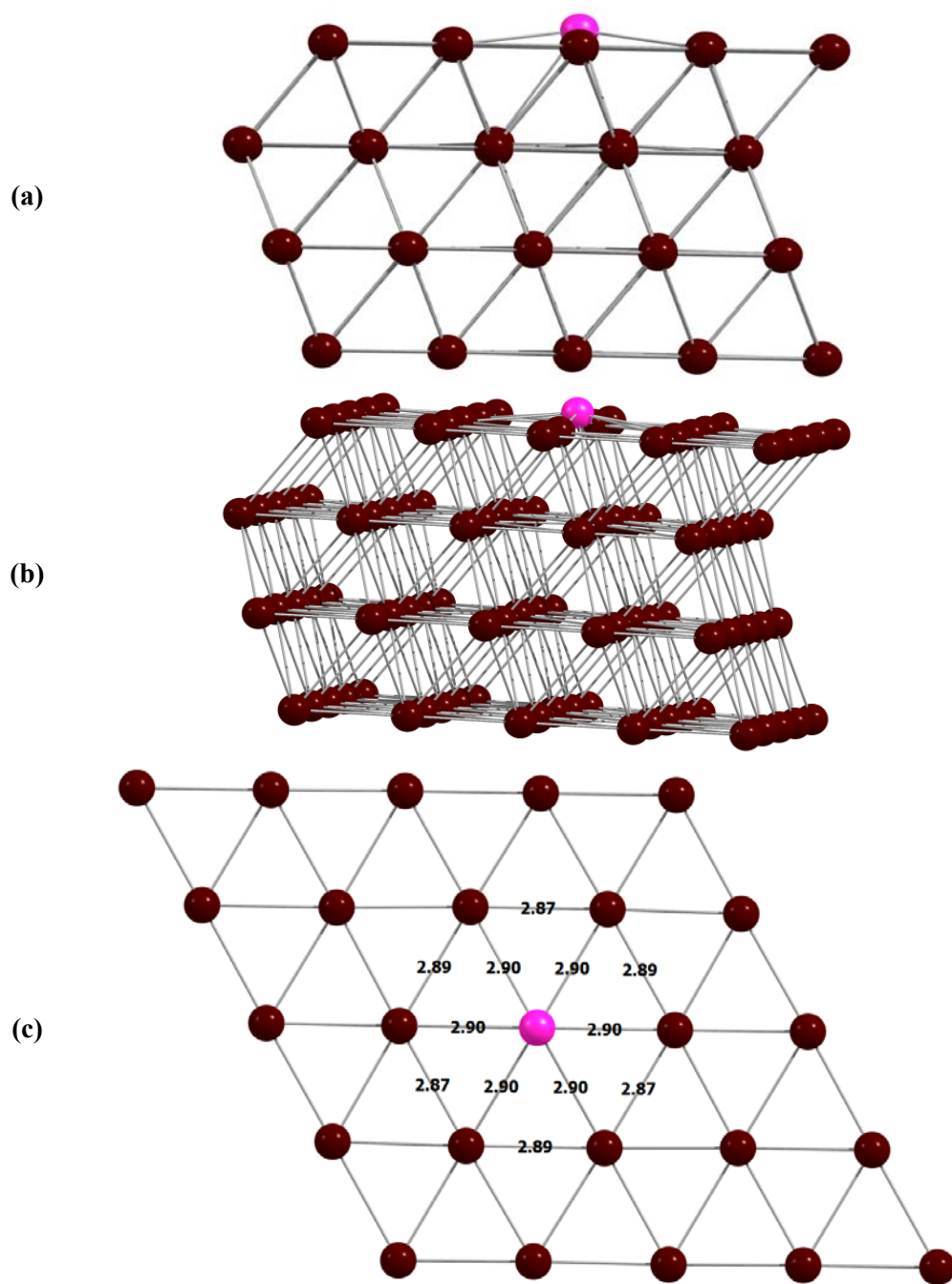
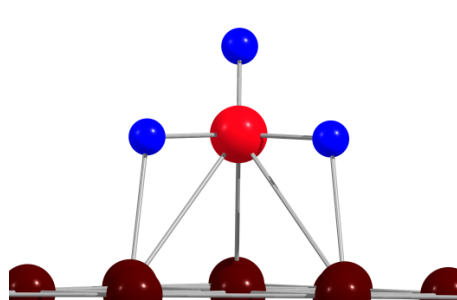
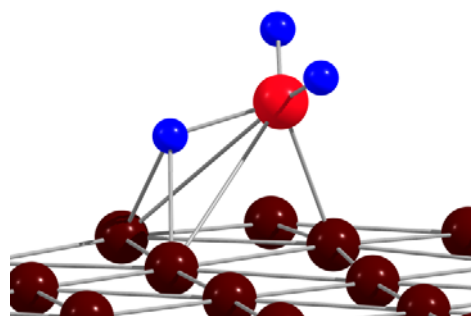


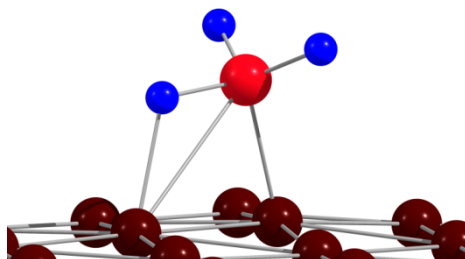
Figure-S7: Single Ag atom substituted Pt(111) surface. The substitution at 2nd and 3rd layer is found to be 0.25 and 0.26 eV higher in energy.



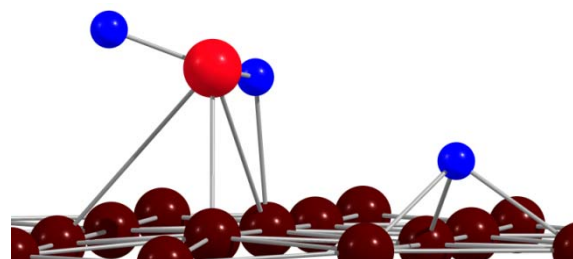
$\Delta E=0.00$



$\Delta E=0.36$

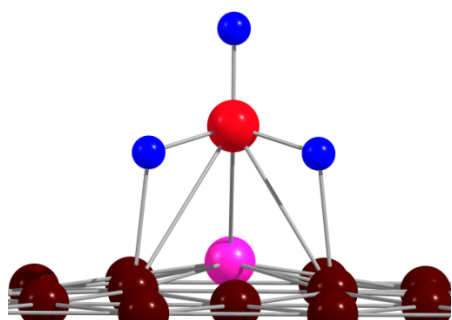


$\Delta E=0.26$

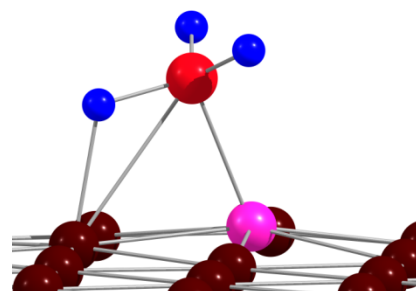


$\Delta E=1.01$

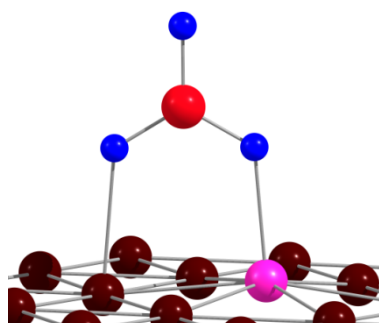
Pt(111) Surface



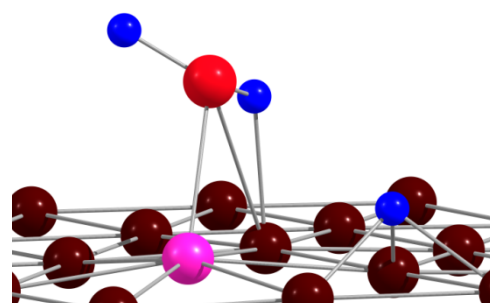
$\Delta E=0.00$



$\Delta E=0.45$



$\Delta E=0.38$



$\Delta E=0.52$

Ag-doped Pt(111) Surface

Figure-S8: Low lying isomer of SO_3 on Pt(111) and $\text{Ag}_1\text{Pt}(111)$ surface. ΔE indicates relative stability.