## Electronic Supplementary Information

## NaYbF<sub>4</sub>@CaF<sub>2</sub> Core–Satellite Upconversion Nanoparticles: One-Pot Synthesis and Sensitive Detection of Glutathione

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Figure S1. Barrette-Joynere-Halenda (BJH) pore-size distribution curve of the  $\alpha$ -NaYbF<sub>4</sub>@CaF<sub>2</sub> core-satellite nanoparticles.



**Figure S2.**  $\alpha$ -NaYbF<sub>4</sub> core nanoparticles synthesized by (a) one-shot quick injection (4 mL) and step-by-step injection at (b) 1.33 mL×3 times, (c) 1 mL×4 times, and (d) 0.8 mL×5 times.



**Figure S3.** HRTEM images of the  $\alpha$ -NaYbF<sub>4</sub>@CaF<sub>2</sub> core–satellite nanoparticles (a) at the end of core reaction, (b) right after the first shell injection, and (c) at the end of the second shell reaction, respectively.



Figure S4. Comparison of upconversion performance of the  $\alpha$ -NaYbF<sub>4</sub>:Er (2%)@CaF<sub>2</sub> coresatellite nanoparticles and  $\beta$ -NaYbF<sub>4</sub>:Er (2%)@NaYF<sub>4</sub> core-shell nanoparticles. The emission intensity of the  $\alpha$ -NaYbF<sub>4</sub>:Er (2%)@CaF<sub>2</sub> core-satellite nanoparticles is about one fifth the intensity of typical  $\beta$ -NaYbF<sub>4</sub>:Er (2%)@NaYF<sub>4</sub> core-shell nanoparticles under the same testing condition.



**Figure S5.** (a) Synthesis scheme and TEM images, (b) XRD pattern, and (c) photoluminescence spectra of the  $\alpha$ -NaYbF<sub>4</sub>:Tm (1%)@CaF<sub>2</sub> core-satellite UCNPs prepared by a step-by-step injection of excessive CaF<sub>2</sub> shell precursors (4 mL). The TEM images show a continuous enlargement in particle size as increasing precursor content. However, the comparison of PL spectra show that the emission intensity is unlikely to be further enhanced by growing a thick shell (4 injections of shell precursor).



**Figure S6**. TEM images of (a)  $\alpha$ -NaYbF<sub>4</sub>:Tm (1%)@CaF<sub>2</sub> core–satellite nanoparticles and (b)  $\alpha$ -NaYbF<sub>4</sub>:Tm (1%)@CaF<sub>2</sub> core–shell nanoparticles after removal of oleate ligands by acidic treatment. (c) Upconversion spectra of the  $\alpha$ -NaYbF<sub>4</sub>:Tm (1%)@CaF<sub>2</sub> core–satellite nanoparticles and core–shell nanoparticles in aqueous solution. The results show that the satellite CaF<sub>2</sub> coatings provide equivalent protection of the upconversion core as dense shells in aqueous media.



Figure S7. TEM images of (a)  $\alpha$ -NaYF<sub>4</sub>@CaF<sub>2</sub> core-satellite nanoparticles and (b)  $\alpha$ -NaYbF<sub>4</sub>@SrF<sub>2</sub> core-satellite nanoparticles.



**Figure S8**. Absorption spectrum of  $MnO_2$  showing the spectral overlapping with the emission spectrum of the  $\alpha$ -NaYbF<sub>4</sub>:Tm (1%)@CaF<sub>2</sub> UCNPs excited at 980 nm.



Figure S9. Basic reaction principle of GSH detection.



**Figure S10.** TEM images of the  $\alpha$ -NaYbF<sub>4</sub>:1%Tm@CaF<sub>2</sub>@MnO<sub>2</sub> nanohybrids synthesized by mixing the  $\alpha$ -NaYbF<sub>4</sub>:Tm (1%)@CaF<sub>2</sub> core–satellite UCNPs with CTAB, MES, and different amount of KMnO<sub>4</sub> (50–500 µL, 10 mM) for 30 min.



**Figure S11.** (a) TEM image of  $MnO_2$  nanosheets synthesized by mixing of MES and KMnO<sub>4</sub> with ultrasound treatment and centrifugation for 30 min. (b) TEM image of UCNP +  $MnO_2$  hybrids synthesized by mixing the  $\alpha$ -NaYbF<sub>4</sub>:Tm (1%)@CaF<sub>2</sub> core–satellite UCNPs, MES, and KMnO<sub>4</sub> with ultrasound treatment and centrifugation.



Figure S12. TEM images of UCNPs@MnO<sub>2</sub> hybrids using (a)  $\alpha$ -NaYbF<sub>4</sub>:Tm (1%)@CaF<sub>2</sub> core-dense shell UCNPs and (b)  $\beta$ -NaYbF<sub>4</sub>:Tm (1%)@NaYF<sub>4</sub> UCNPs, respectively. Note that the layered MnO<sub>2</sub> in b is highlighted with red arrow.



**Figure S13.** (a) Nitrogen adsorption-desorption isotherms of  $\alpha$ -NaYbF<sub>4</sub>:Tm (1%)@CaF<sub>2</sub> core–dense shell UCNPs. (b) Photoluminescence response of MnO<sub>2</sub>-modified  $\alpha$ -NaYbF<sub>4</sub>:Tm (1%)@CaF<sub>2</sub> core–dense shell UCNPs as a function of GSH content. (c) Calibration curve of GSH detection using the MnO<sub>2</sub>-modified  $\alpha$ -NaYbF<sub>4</sub>:Tm (1%)@CaF<sub>2</sub> core–dense shell nanohybrids. LOD = 0.88  $\mu$ M.