Supporting Materials

Improved Efficiency and Thermal Stability of Ternary All-Small-Molecule Organic Solar Cells by NCBA as a Third Component Material

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2. Experimental details

2.1 Materials

p-DTS-(FBTTh₂)₂ was purchased from 1-Material Inc. NCBA and PC₇₁BM was purchased from Luminescence Technology Co; *o*-dichlor-obenzene (ODCB) and 1,8diiodooctane (DIO) were purchased from Sigma-Aldrich Co. MoO₃ and Ag were purchased from Alfa Aesar Co. A mixture of *p*-DTS-(FBTTh₂)₂ and blend acceptor was dissolved in ODCB solvent (3:2) with stirring overnight, and then diluted to 25 mg mL⁻¹ in total without stirring and kept for over 3 h. ZnO solution was synthesized by sol-gel method ^{1, 2}.

2.2 Device preparation and characteristics

The ITO glasses were pre-cleaned by ultrasonic bath and UV ozone treat in a plasma asher (100 W, 20 min). A thin layer (20 nm) of ZnO was spin-coated onto the ITO glass and baked at 150 °C for 20 min in air. A blend solution of *p*-DTS-(FBTTh₂)₂, NCBA and PC₇₁BM was spin-coated on the ZnO layer to form an photoactive layer and thermal annealing treatment were carried out in a N₂-filled glove box and the nominal thickness of ~100 nm (with a variation of ~15 nm over the entire film). A MoO₃ layer and Ag electrode were evaporated under vacuum through a shadow mask to define the active area of the devices (3×3 mm²). The current density versus voltage (*J*–*V*) characteristics of PSCs were performed in a glove box with a computer-controlled Keithley 236 Source Measure Unit under illumination at 100 mW cm⁻² using an AM1.5 G solar simulator. The external quantum efficiency (*EQE*) spectrum was measured with the Stanford Research Systems model SR830 DSP lock-in amplifier coupled with a

WDG3 monochromator and a 500 W xenon lamp.

The hole-only space-charge-limited-current (SCLC) devices and electron-only SCLC devices were fabricated with structures of ITO/PEDOT:PSS/PBDB-T:ITIC:F-N2200/Au and Al/PBDB-T:ITIC:F-N2200/Al, respectively. The charge carrier mobilities were calculated using the equation ^{3, 4}:

$$J=\frac{9}{8}\varepsilon_{r}\varepsilon_{0}\mu\frac{V^{2}}{d^{8}},$$

where *J* is the current density, μ is the charge carrier mobility, ε_0 (8.85×10₋₁₄ F/cm) and ε_r the permittivity of free space and relative permittivity of the material (ε_r was assumed to be 3), respectively, and *V* is the SCLC effective voltage. The charge carrier mobility were calculated using the equation ⁵:

$$\mu = \mu_0 \exp[0.89\gamma \sqrt{\frac{V}{L}}]$$

where μ_0 is the charge mobility under zero electric field and γ is a constant. Then, the Mott-Gurney equation can be described by ⁶:

$$J = \frac{9}{8} \varepsilon_r \varepsilon_0 \mu_0 \frac{V^2}{L^3} \exp[0.89\gamma \sqrt{\frac{V}{L}}]$$

In this case, the charge mobility were estimated using the following equation ⁶:

$$\ln(\frac{JL^3}{V^2}) = 0.89\gamma \sqrt{\frac{V}{L}} + \ln(\frac{9}{8}\varepsilon_r \varepsilon_0 \mu_0)$$



Figure S1. Energy band diagram of the ternary OSC for figure (a) and the chemical

structure of NCBA and p-DTS-(FBTTh₂)₂ for figure (b).



Figure S2. (a) Absorption spectra of neat p-DTS-(FBTTh₂)₂, NCBA, and PC₇₁BM, and

(b) binary thin films of *p*-DTS-(FBTTh₂)₂:PC₇₁BM and *p*-DTS-(FBTTh₂)₂:NCBA and ternary thin films of *p*-DTS-(FBTTh₂)₂:NCBA:PC₇₁BM (3:0.24:1.76).



Figure S3. XRD patterns of *p*-DTS-(FBTTh₂)₂:NCBA:PC₇₁BM ternary thin films with



NCBA concentrations in the blend acceptor of 0, 12, 18, and 100 wt.%.

Figure S4. 3D and phase images of AFM images of the binary and ternary thin films.

p-DTS-(FBTTh₂)₂:PC₇₁BM=3:2 for (a and b); (b) p-DTS-

$$(FBTTh_2)_2:NCBA:PC_{71}BM=3:0.24:1.76$$
 for (c and d); (c) *p*-DTS-
(FBTTh_2)_2:NCBA:PC_{71}BM=3:0.36:1.64 for (e and f); (d) *p*-DTS-



(FBTTh₂)₂:NCBA=3:2 for (g and h).

Figure S5. (a, c, and e) Current-voltage characteristics of binary *p*-DTS- $(FBTTh_2)_2$:PC₇₁BM- and ternary *p*-DTS- $(FBTTh_2)_2$:NCBA:PC₇₁BM (3:0.24:1.76) and *p*-DTS- $(FBTTh_2)_2$:NCBA:PC₇₁BM (3:0.36:1.64)-based OSCs dependent on incident light intensity; (b, d and f) normalized photocurrents at -0.5 V measured for various intensities.



Figure S6. XRD curves of binary *p*-DTS-(FBTTh₂)₂:PC₇₁BM, ternary *p*-DTS-(FBTTh₂)₂:NCBA:PC₇₁BM (3:0.24:1.76), and ternary *p*-DTS-(FBTTh₂)₂:NCBA:PC₇₁BM (3:0.36:1.64) thin films after 100 h of thermal annealing treatment at 90°C.





Figure S7. 3D and phase images of AFM images of binary *p*-DTS-(FBTTh₂)₂:PC₇₁BM for (a and b), ternary *p*-DTS-(FBTTh₂)₂:NCBA:PC₇₁BM (3:0.24:1.76) for (c and d), and ternary *p*-DTS-(FBTTh₂)₂:NCBA:PC₇₁BM (3:0.36:1.64) thin films for (e and f) after 100 h of thermal annealing treatment at 90°C.

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