Supporting Information

MOF-derived coral-like NiSe@NC nanohybrid: an efficient electrocatalyst for hydrogen evolution reaction under all-pH values

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Fig. S1. The XRD patterns of Ni-MOFs.

Table S1. The approximate contents of NiSe and carbon for NiSe@NC-500, NiSe@NC-550,

Catalysts	NiSe (%)	C (%)
NiSe@NC-500	88.08	11.92
NiSe@NC-550	62.33	37.67
NiSe@NC-600	67.27	32.73
NiSe@NC-650	86.25	13.75
NiSe@NC-700	90.33	9.67

NiSe@NC-600, NiSe@NC-650, NiSe@NC-700.



Fig. S2. The SEM image of Ni-MOFs.



Fig. S3. The SEM images of (a) NiSe@NC-500, (b) NiSe@NC-550, (c) NiSe@NC-650, (d)

NiSe@NC-700.



Fig. S4. XPS survey of NiSe@NC-600.

Table S2. Element contents of NiSe@NC-600 obtained by XPS spectrum.

Elements	Se (At. %)	Ni (At. %)	C (At. %)	N (At. %)
Contents	24.1	26.5	44.5	4.9



Fig. S5. XRD patterns of (a) Ni-MOFs-1, (b) NiSe@C-600, (c) NiO, (d) Ni@NC-600 and (e)

NiSe.



Fig. S6. Cyclic voltammograms (CV) at various scan rate of (a) NiSe@NC-500 (b) NiSe@NC-550. (c) NiSe@NC-600. (d) NiSe@NC-650. (e) NiSe@NC-700 for HER.

Table S3. Comparison of catalytic performance of different HER electrocatalysts in 0.5 M H₂SO₄.

Electrocatalyst	η ₀ (mV)	Tafel (mV dec ⁻¹)	C _{dl} (mF·cm⁻²)	ECSA (cm²)	Rct (Ω)
NiSe@NC-500	140	62.1	1.0	28.6	100.2
NiSe@NC-550	80	61.3	4.4	125.7	44.1
NiSe@NC-600	60	53.3	13.4	383	38.5
NiSe@NC-650	100	62.7	3.5	100	65.2
NiSe@NC-700	150	76.8	0.4	11.4	167.1



Fig. S7. SEM image of NiSe@NC-600 after 2000 CV cycles.



Fig. S8. SEM image of NiSe@NC-600 after 20h at 10 mA \cdot cm⁻² in 0.5 M H₂SO₄ solution.



Fig. S9. SEM image of NiSe@NC-600 after 12h at 20 mA \cdot cm⁻² in 0.5 M H₂SO₄ solution.



Fig. S10. SEM image of NiSe@NC-600 after 12h at 35 mA \cdot cm $^{-2}$ in 0.5 M H_2SO_4 solution.

Electrocatalyst	Electrolyte	η ₁₀ (mA·cm ⁻²)	References
NiSe@NC-600	0.5 M H ₂ SO ₄	123	This work
CoSe ₂ NP/CP	0.5 M H ₂ SO ₄	139	1 [1]
NiSe nanofibers	0.5 M H ₂ SO ₄	270	2 ^[2]
NiSe ₂ NPs	0.5 M H ₂ SO ₄	180	3 [3]
CoSe ₂ NPs	0.5 M H ₂ SO ₄	193	4 [4]
NiSe ₂ NSs	0.5 M H ₂ SO ₄	173	5 [5]
CoSe ₂ NP/CP	0.5 M H ₂ SO ₄	139	6 ^[6]
NiSe ₂ /CNT	0.5 M H ₂ SO ₄	200	7 [7]
CoSe ₂ NP	0.5 M H ₂ SO ₄	250	8 [8]
Co _{0.85} Se	0.5 M H ₂ SO ₄	180	9 [9]
NiSe ₂ /C	0.5 M H ₂ SO ₄	200	10 ^[10]

Table S4. Comparison of electrocatalytic activity for recently reported HER catalysts.



Fig. S11. Tafel plots of NiSe@NC-600 at different pH range.



Fig. S12. Electrochemical impedance spectroscopy (EIS) of NiSe@NC-600 in neutral and

alkaline solutions.

References

- 1 D. Kong, H. Wang, Z. Lu and Y. Cui, J. Am. Chem. Soc., 2014, 136, 4897-4900.
- 2 M. R. Gao, Z. Y. Lin, T. T. Zhuang, J. Jiang, Y. F. Xu, Y. R. Zheng and S. H. Yu, *J. Mater. Chem.*, 2012, **22**, 13662-13668.
- 3 D. Kong, J. J. Cha, H. Wang, H. R. Lee and Y. Cui, *Energy Environ. Sci.*, 2013, **6**. 3553-3558.
- 4 H. Zhang, L. Lei and X. Zhang, *RSC Adv.*, 2014, 4, 54344-54348.
- 5 S. Lu and Z. Zhuang, Sci. China Mater., 2016, 59, 217-238.
- H. Zhang, B. Yang, X. Wu, Z. Li, L. Lei and X. Zhang, ACS Appl. Mater. Interface, 2015, 7, 1772-1779.
- 7 B. Wang, X. Wang, B. Zheng, B. Yu, F. Qi, W. Zhang, Y. Li and Y. Chen, *Electrochem*. *Commun.*, 2017, **83**, 51-55.
- 8 J. Lin, J. He, F. Qi, B. Zheng, X. Wang, B. Yu, K. Zhou, W. Zhang, Y. Li and Y. Chen, *Electrochim. Acta*, 2017, **247**, 258-264.
- 9 B. Yu, F. Qi, X. Wang, B. Zheng, W. Hou, Y. Hu, J. Lin, W. Zhang, Y. Li and Y. Chen, *Electrochim. Acta*, 2017, **247**, 468-474.
- 10 D. Song, H. Wang, X. Wang, B. Yu and Y. Chen, *Electrochim. Acta*, 2017, **254**, 230-237.