Electronic Supplementary Information

Multidimensional and hierarchical carbon-confined cobalt phosphide

nanocomposite as advanced anodes for lithium and sodium storage

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Figure S1 (a and b) SEM and TEM images of (a and b) Co-NAT precursor and (c and d) Co-NTA@hydrotherm carbon nanofibers.



Figure S2 SEM and TEM images of (a and b) Co@C, (c and d) Co@DC and (e and f) Co@DC@GR.



Figure S3 XRD patterns of Co@C, Co@DC and Co@DC@GR.



Figure S4 (a and b) low magnification and (c and d) cross-section images of CoP@DC@GR.



Figure S5. Surface wetting of $LiPF_6$ electrolyte droplet on (a-c) CoP@C, (d-f)

CoP@DC and (g-i) CoP@DC @GR.



Figure S6 Electrical conductivities of the pressed pellets of CoP@C, CoP@DC and

CoP@DC@GR.



Figure S7 (a) Schematic illustration of the synthesis of CoP@C@GR nanocomposite; (b and c) SEM and (d and e) TEM images of CoP@C@GR nanocomposite.



Figure S8 (a) Schematic illustration of the synthesis of pure CoP nanofibers; SEM images of (b) Co_3O_4 nanofibers and (d) pure CoP nanofibers; XRD patterns of (c) Co_3O_4 nanofibers and (e) pure CoP nanofibers.



Figure S9 (a) SEM image, (b) XRD pattern and (c) Nitrogen adsorption/desorption

isotherms of DC@GR nanocomposite.



Figure S10 Discharge and charge profiles of (a) CoP@DC and (b) CoP@C

electrodes.



Figure S11 Cycling performance and Coulombic efficiencies of CoP@C@GR, pure CoP and DC@GR at a current density of 0.25 A g^{-1} .



Figure S12 (a) Nyquist plots for the electrodes made of CoP@C, CoP@DC and CoP@DC@GR after 70 cycles at a current density of 0.25 A g⁻¹; (b) Liner fits (relationship between Z' and $\omega^{-1/2}$) in low-frequency region of CoP@C, CoP@DC and CoP@DC@GR electrodes.



Figure S13 (a) HRTEM image of the fully charged electrode in LIBs; (b) TEM image of the fully discharged electrode in LIBs.



Figure S14 (a) Schematic representation of the CoP@DC@GR//LiCoO₂ full-cell; (b) Charge and discharge profiles of the CoP@DC@GR//LiCoO₂ full-cell; (c) Cycle performance the CoP@DC@GR//LiCoO₂ full-cell.

Table S1 A comparison study of the CoP@DC@GR hybrid nanofibers with othertypical phosphide anode materials for SIBs.

Materials	Current density	Specific capacity, cycle number	Ref.
CoP@DC@GR	0.1 A g ⁻¹	650 mAh g ⁻¹ ,100 cycles	This work
	0.5 A g ⁻¹	398 mAh g ⁻¹ ,300 cycles	
CoP@C-RGO	0.1 A g ⁻¹	473 mAh g ⁻¹ , 100 cycles	49
СоР	0.1 A g ⁻¹	315 mAh g ⁻¹ , 25 cycles	50
Ni₂P⊂pGN	0.2 A g ⁻¹	161 mAh g ⁻¹ , 100 cycles	29
H-FeP@C@GR	0.1 A g ⁻¹	400 mAh g ⁻¹ , 250 cycles	16
Sn ₄ P ₃ @C	0.1 A g ⁻¹	580 mAh g ⁻¹ , 120 cycles	51
CuP ₂ /C	0.15 A g ⁻¹	430 mAh g ⁻¹ , 30 cycles	52
Ni ₂ P@C/GA	0.15 A g ⁻¹	254 mAh g ⁻¹ , 100 cycles	13

Table S2 The comparison study of lithium and sodium storage for the

CoP@DC@GR nanocomposite.

	Lithium storage (0.25 A g ⁻¹)	Sodium storage (0.1 A g ⁻¹)
Initial reversible capacity	855 mAh g ⁻¹	801 mAh g ⁻¹
Initial Coulombic efficiency	73%	64%
Reversible capacity	754 mAh g ⁻¹ after 300 cycles	650 mAh g ⁻¹ after 100 cycles
Capacity retention	88%	81%



Figure S15 (a) Nyquist plots for the electrodes made of CoP@C, CoP@DC and CoP@DC@GR after 3 cycles for SIBs; (b) Liner fits (relationship between Z' and $\omega^{-1/2}$) in low-frequency region of CoP@C, CoP@DC and CoP@DC@GR electrodes.

Table S3 The comparison study of Rct for CoP@C, CoP@DC andCoP@DC@GR electrodes.

	Electrode	For LIBs	For SIBs
	CoP@C	217 Ω	740 Ω
Rct	CoP@DC	127 Ω	498 Ω
	CoP@DC@GR	45 Ω	257 Ω



Figure S16. (a) CV curves at different scan rates from 0.2 to 2.0 mV s⁻¹ for SIBs; (b) Log i vs. log v plots at oxidation and reduction state; (c) Capacitive- and diffusion-controlled contribution to charge storage at 0.6 mV s⁻¹; (d) Normalized contribution ratio of capacitive- and diffusion-controlled capacities at different scan

rates.



Figure S17 SEM and TEM images of CoP@DC@GR electrode after 100 cycles for SIBs.



Figure S18 Projected density of states (PDOS) of the CoP surfaces. The Fermi level (dashed line) is set as zero.