# **Electronic Supplementary Information**

## **High-performance gas sensors based on functionalized**

#### single-wall carbon nanotube random networks for detection

## of nitric oxide down to ppb-level

Jun-Young Jeon<sup>1,2+</sup>, Byeong-Cheol Kang<sup>1+</sup>, Young Tae Byun<sup>2\*</sup>, and Tae-Jun Ha<sup>1\*</sup>

 <sup>1</sup> Department of Electronic Materials Engineering, Kwangwoon University Seoul 139-701, Republic of Korea
<sup>2</sup> Sensor System Research Center, Korea Institute of Science and Technology (KIST) Seoul 136-

791, Korea

\*Corresponding author. Tel.: +82 2 940 8678.

E-mail address: taejunha0604@gmail.com, byt427@kist.re.kr



**Figure S1.** Morphology of SWCNT random networks (a) before and (b) after functionalization with PEI (concentration of 10 wt%), obtained by AFM measurements



**Figure S2.** XPS spectra of elements of C, N, and O in the chemical bonding of the SWCNT random networks (a) before and (b) after functionalization with PEI (concentration of 10 wt%)



**Figure S3.** The normalized response of SWCNT random networks after functionalization with PEI as the recovery time was increased and almost 100% of resistance was recovered to the initial



**Figure S4.** The normalized resistance of the proposed NO gas sensors with different PEI concentrations (0, 2, 4, 6, 8, and 10 wt%)



Figure S5. The normalized resistance of SWCNT random networks after functionalization with PEI when exposed to the  $NO_2$  gas at 100 ppb



**Figure S6.** The theoretical detection limit of the proposed NO gas sensors by linear extrapolation from the response slope in the linear regime at low ppb level

Table S1. 5<sup>th</sup> order polynomial fitting implemented within data-point range

*Time (sec)	$(Y_i - Y)$	$(Y_i - Y)^2$
200	0.00130	$1.70 \times 10^{-6}$
210	-0.00418	$1.74 \times 10^{-5}$
220	0.00331	$1.09 \times 10^{-5}$
230	0.00199	$3.97 \times 10^{-6}$
240	-0.00113	$1.28 \times 10^{-6}$
250	-0.00462	2.13 × 10 <sup>-5</sup>
260	0.00224	$5.02 \times 10^{-6}$
270	0.00298	$8.88 \times 10^{-6}$
280	-0.00144	$2.09 \times 10^{-6}$
290	-0.000904	$8.18 \times 10^{-7}$
300	-0.000462	$2.14 \times 10^{-7}$

#### \* The time when the resistance is measured.

- 1. We carried out the linear fitting on the sensitivity of the proposed NO gas sensors as a function of NO concentration, as shown in Fig. S4,
- 2. We extracted the measured values of electrical resistance in the proposed NO gas sensors (from 11 points).
- 3. We conducted a 5th order polynomial fitting implemented within data-point range (see Table S1) where Yi is the measured data point and Y is the corresponding value calculated from the curve-fitting equation

$$V_{x^2} = \sum (Y_i - Y)^2$$

4. We calculated the RMSnoise from the following equation

$$RMS_{noise}(ppm^{-1}) = \sqrt{\frac{V_{x^2}}{(N-1)}}$$

where N is the number of data points used in the curve fitting

5. We extracted the detection limit by using the following equation.

Detection limit =  $3 \times \frac{RMS_{noise}}{slope}$ 

**Table S2.** Sensing performance of NO gas sensors in previous studies based on various materials, such as metal oxide, polymer, or carbon-based material

material	Response (%)	NO concentration (ppm)	Response time (min)	Reference
zinc oxide	7.3	10	5	1
Nickel phthalocyanine	0.4	5	20	2
WO <sub>3</sub> /Cr <sub>2</sub> O <sub>3</sub>	10	5	60	3
poly[N-9'-heptadecanyl-2,7-	5	5	5	4
thienyl-2',1',3'-				
benzothiadiazole)](PCDTBT)				
Multi wall carbon nanotube	1	2	30	5
Single wall carbon nanotube	12	1	10	6
Tin dioxide (SnO <sub>2</sub> )	28.7	1	6	7
PEDOT:PSS	2.2	0.35	10	8
zinc oxide	8.25	0.25	15	9
indium gallium zinc oxide	0.02	0.1	2	10
Single wall carbon nanotube	9	0.005	1	11
Single wall carbon nanotube	50	0.1	30	Thiswork

#### References

- 1. C.-C. Liu, J.-H. Li, C.-C. Chang, Y.-C. Chao, H.-F. Meng, S.-F. Horng, C.-H. Hung and T.-C. Meng, *J. Phys. D: Appl. Phys.*, 2009, **42**, 155105.
- 2. K.-C. Ho and Y.-H. Tsou, Sensor. Actuat. B-Chem., 2001, 77, 253-259.
- C. Sun, G. Maduraiveeran and P. Dutta, Sensor. Actuat. B-Chem., 2013, 186, 117-125.
- 4. A. Gusain, N. J. Joshi, P. Varde and D. Aswal, *Sensor. Actuat. B-Chem.*, 2017, **239**, 734-745.
- 5. T. Ueda, S. Katsuki, N. H. Abhari, T. Ikegami, F. Mitsugi and T. Nakamiya, *Surf. Coat. Tech*, 2008, **202**, 5325-5328.
- 6. J. Mäklin, T. Mustonen, K. Kordás, S. Saukko, G. Tóth and J. Vähäkangas, phys. status solidi B, 2007, 244, 4298-4302.
- 7. C.-Y. Kim, H. Jung, H. Choi and D.-k. Choi, J. Korean Phys. Soc., 2016, 68, 357-362.
- 8. C.-Y. Lin, J.-G. Chen, C.-W. Hu, J. J. Tunney and K.-C. Ho, *Sensor. Actuat. B-Chem.*, 2009, **140**, 402-406.
- 9. C.-Y. Lin, J.-G. Chen, W.-Y. Feng, C.-W. Lin, J.-W. Huang, J. J. Tunney and K.-C. Ho, Sensor. Actuat. B-Chem., 2011, **157**, 361-367.
- 10. H.-W. Zan, C.-H. Li, C.-C. Yeh, M.-Z. Dai, H.-F. Meng and C.-C. Tsai, Appl. Phys. Lett., 2011, 98, 253503
- 11. O. Kuzmych, B. L. Allen and A. Star, *Nanotechnology*, 2007, **18**, 375502.