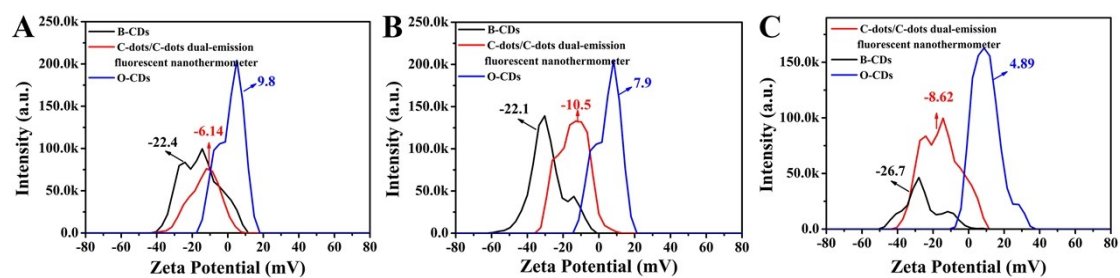


## Supplementary Information

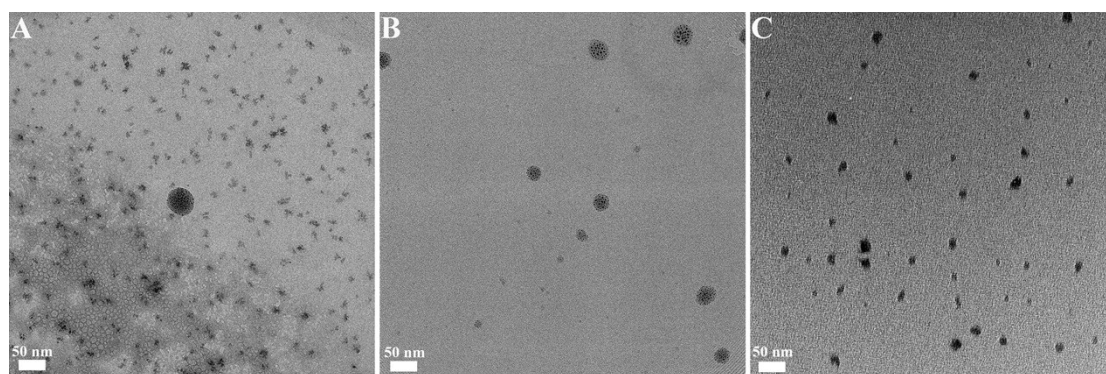
### **Surface state-controlled C-dots/C-dots based dual-emission fluorescent nanothermometer for intra-cellular thermometry**

Chan Wang<sup>§</sup>, Tantan Hu<sup>§</sup>, Tiju Thomas<sup>✉</sup>, Shanliang Song<sup>\*</sup>, Zhuoqi Wen<sup>\*</sup>,

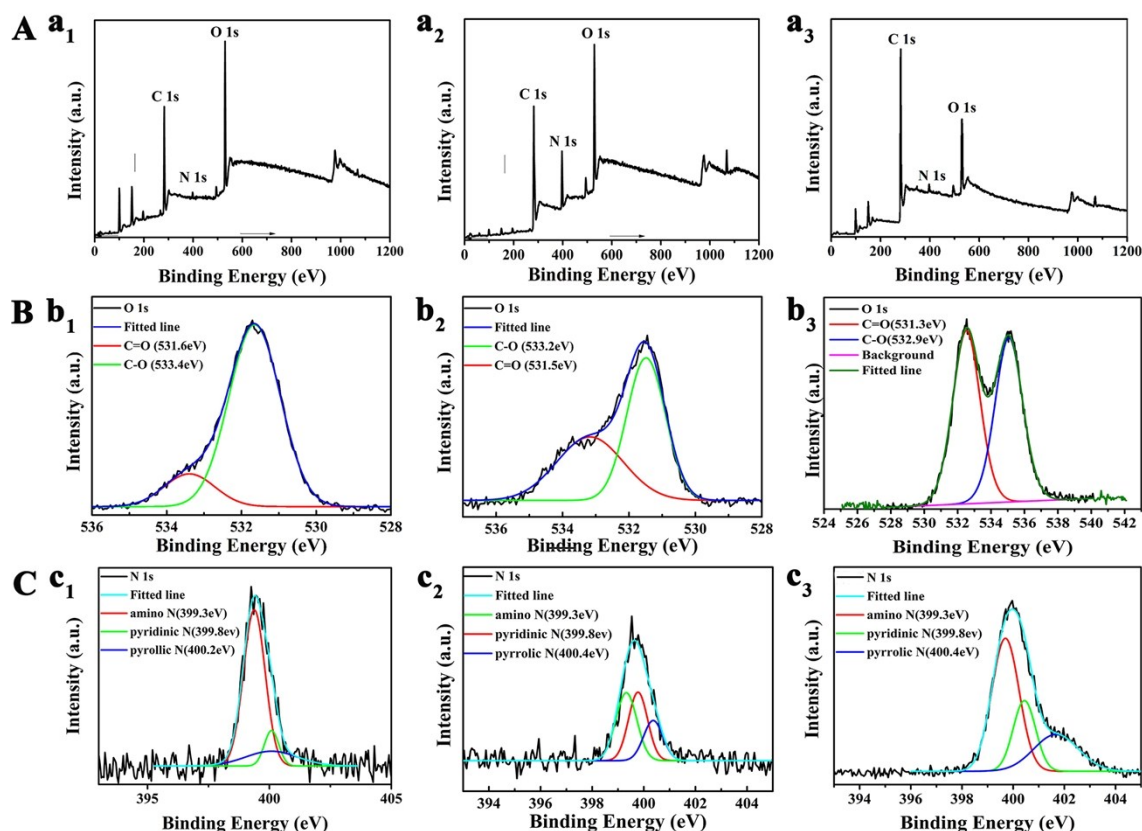
Chuanxi Wang,<sup>\*</sup> Qijun Song<sup>\*, §</sup> and Minghui Yang<sup>\*,\*</sup>



**Fig S1.** Zeta potential distribution of B-CDs (black line), O-CDs (blue line), and C-dots/C-dots dual-emission nanospheres (red line) at different pH, respectively. A: pH=5.0, B: pH=7.0, C: pH=9.0.



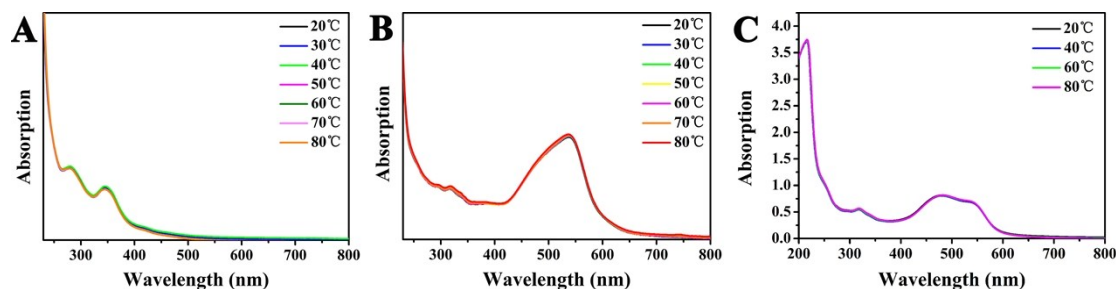
**Fig S2.** TEM of C-dots/C-dots dual-emission nanospheres at a varying feeding ratio of O-CDs to B-CDs: A, 3:1; B, 1:1; C, 1:2, respectively.



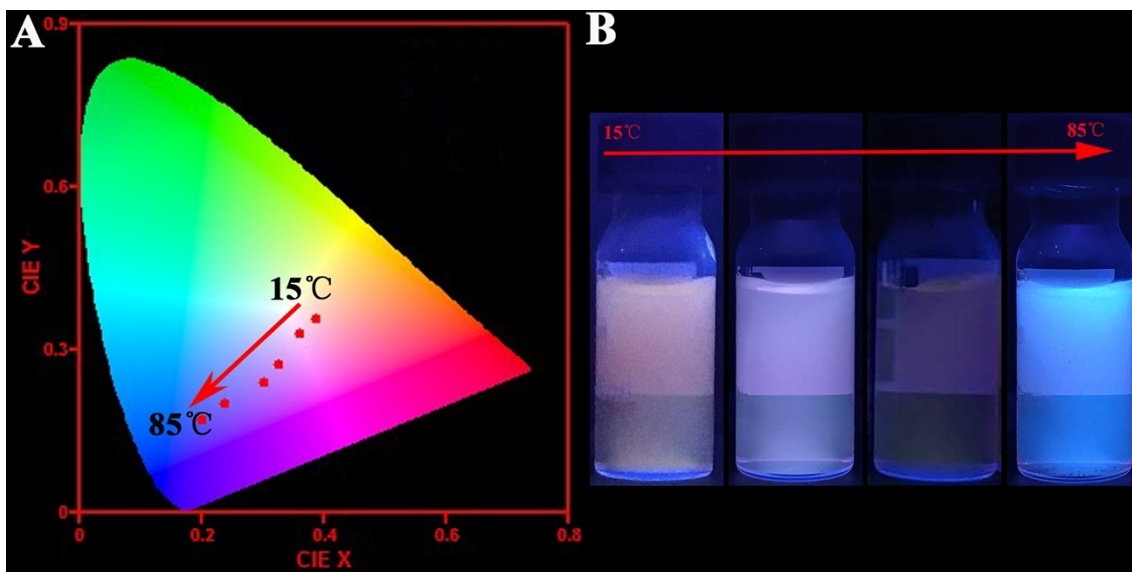
**Fig S3.** A: Full-survey X-ray photoelectron spectroscopy (XPS) of B-CDs ( $a_1$ ), O-CDs ( $a_2$ ), C-dots/C-dots dual-emission nanospheres ( $a_3$ ); B: high-resolution XPS O1s spectra of B-CDs 120 ( $b_1$ ), O-CDs ( $b_2$ ), CDs-based dual emissive nanoparticles ( $b_3$ ); C: high-resolution XPS N1s spectra of B-CDs 120 ( $c_1$ ), O-CDs ( $c_2$ ), CDs-based dual emissive nanoparticles ( $c_3$ ). Each band was deconvoluted following the literature.

**Table S1.** XPS data analyses of the C 1s spectra of B-CDs, O-CDs and C-dots/C-dots dual-emission nanospheres.

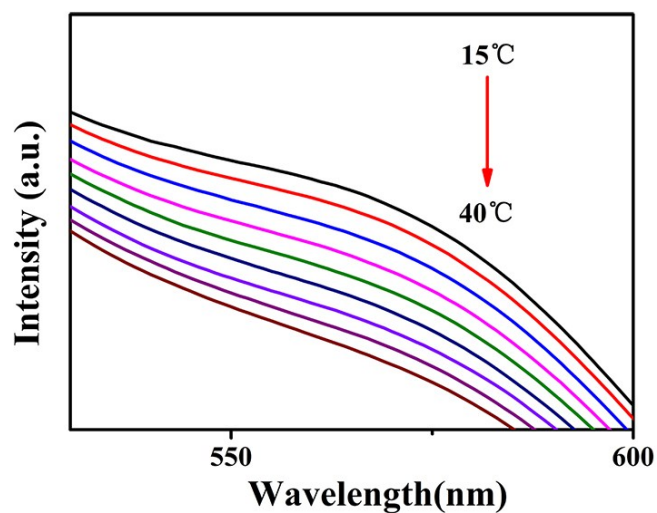
Sample	C-C/C=C	C-N/C-O	C=O	COOH
B-CDs	0.404	0.313	0.255	0.028
O-CDs	0.375	0.357	0.231	0.037
C-dots/C-dots dual-emission nanospheres	0.408	0.244	0.303	0.046



**Fig S4.** Absorption spectra of B-CDs (A), O-CDs (B), C-dots/C-dots dual-emission nanospheres (C) under varying temperature, respectively.



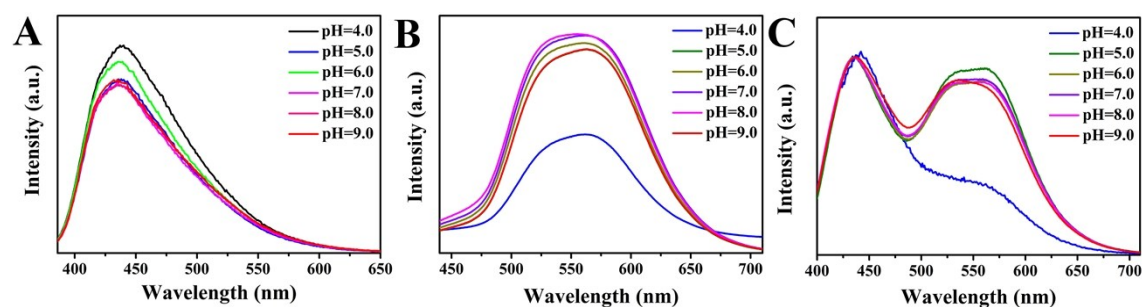
**Fig S5.** A: variation in the color coordinates of the C-dots/C-dots dual-emission nanospheres with increasing temperature from 15 to 85 °C; B: corresponding photograph of the C-dots/C-dots dual-emission nanospheres under increasing temperature.



**Fig S6.** The magnified pattern at 590 nm of Figure 4A.

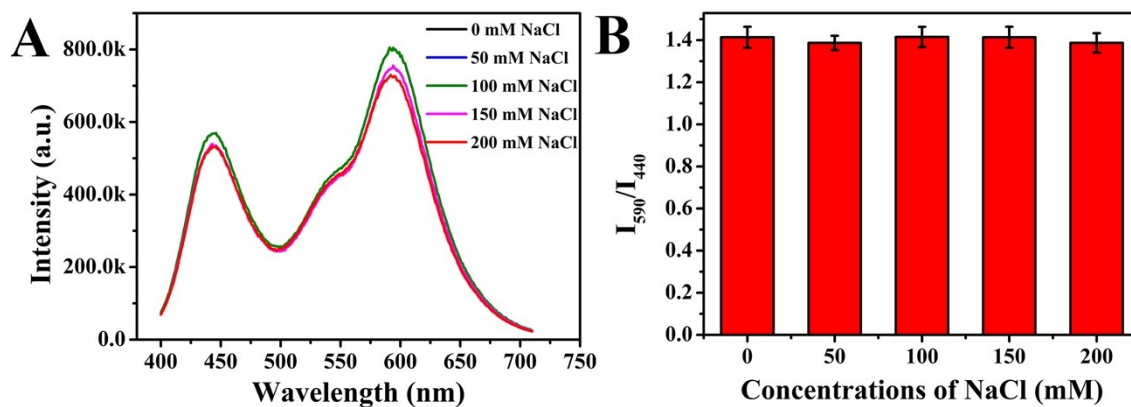
**Table S2.** A contrast of temperature responsive properties for recently reported CDs based nanothermometer.

Materials	Synthesis Method	Propertie	Temperature range	Sensitivity	Comment	Reference
Zn-CQDs	Zinc reduction method	single emission	10-100°C	_____	Low sensitivity and poor accuracy	1
CDs	Hydrothermal method	single emission	15-90°C	0.69%/°C	Poor accuracy and repeatability	2
CDs@UiO-66	In-situ synthesis	single emission	25-110°C	_____	Low sensitivity and poor accuracy	3
CDs-Au NCs	Chemical crosslinking	dual emission	20-75°C	1.8%/°C	Complicated experiment procedure	4
CDs@(PSS/LDH) <sub>n</sub> UTFs	Layer-by-layer assembly	dual emission	0-80°C	0.68%/°C	Cumbersome process and low sensitivity	5
MSCDs	Multi-step synthesis	dual emission	20-50°C	1.29%/°C	CDs are not temperature sensitive	6
C-dots/C-dots	Electrostatic self-assembly	dual emission	15-85°C	0.93%/°C	Simple, fast preparation and relatively high sensitivity	This work

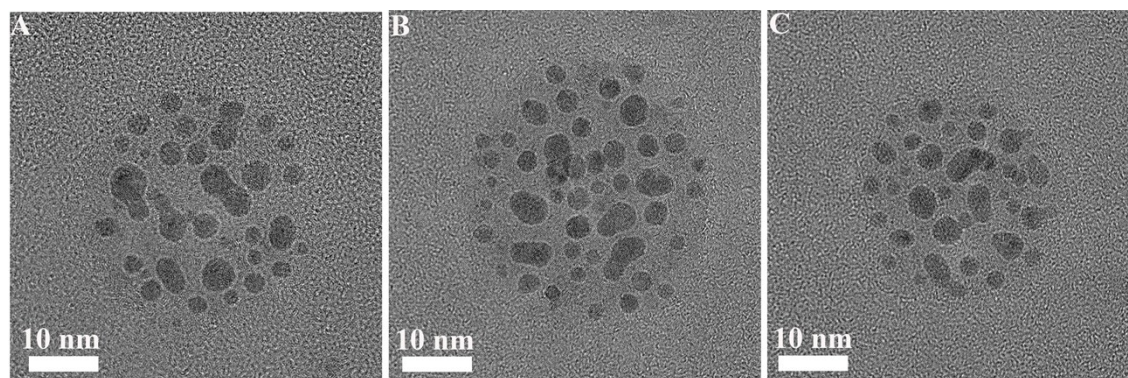


**Fig S7.** The PL spectra of B-CDs (A), O-CDs (B) and C-dots/C-dots dual-emission

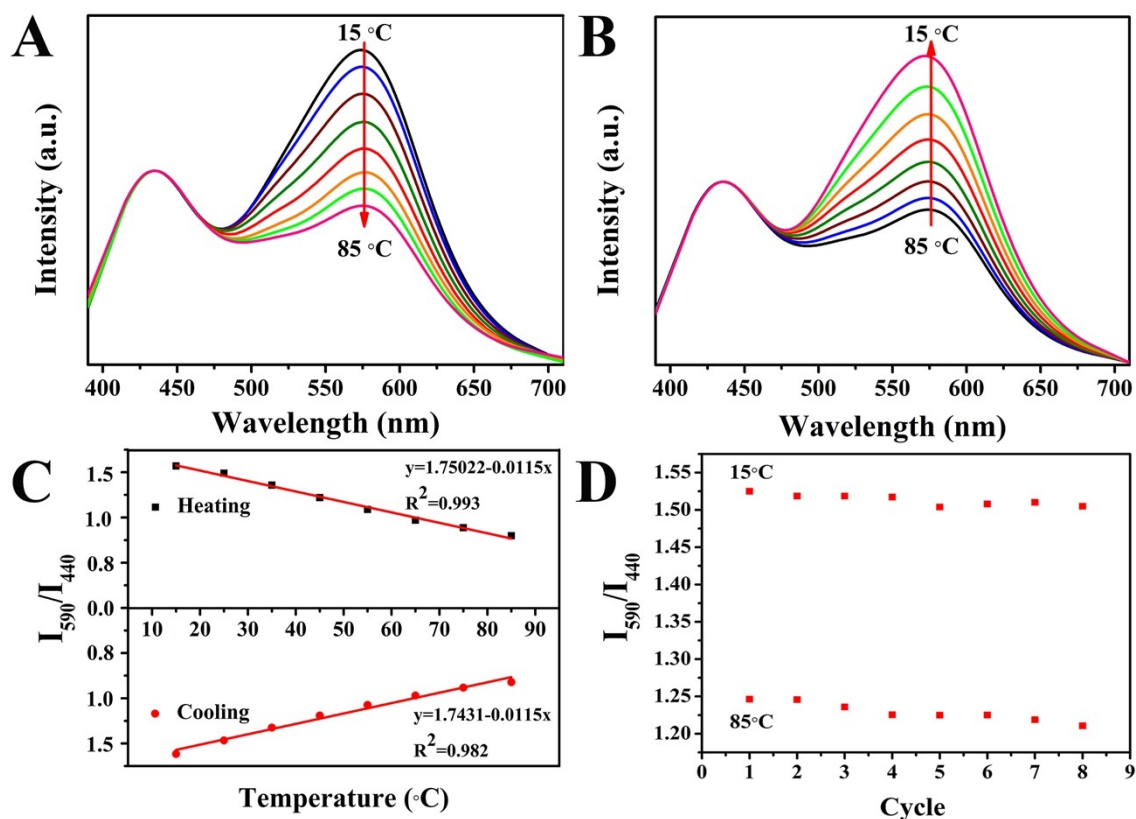
nanospheres (C) under different pH ranges (4.0-9.0).



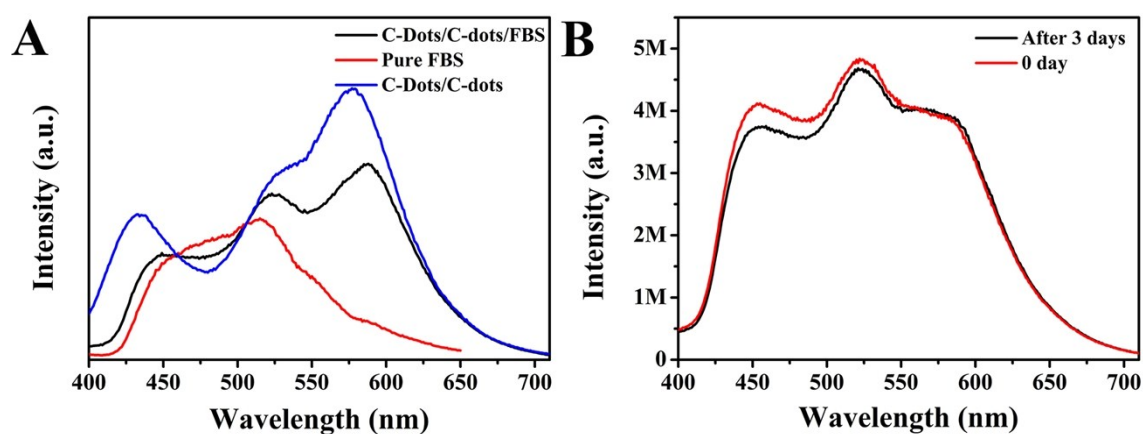
**Fig S8.** The PL intensity variation of C-dots/C-dots dual-emission nanospheres under different concentrations of NaCl (0-200mM).



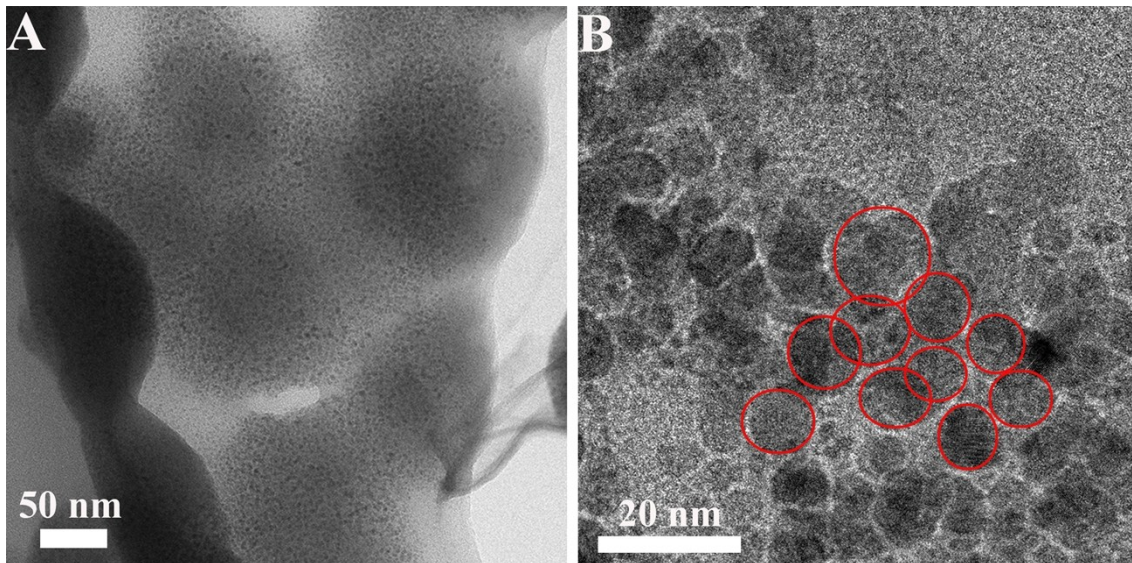
**Fig S9.** The HRTEM of C-dots/C-dots dual-emission nanospheres under a pH of 5.0 (A), 7.0 (B) and 9.0 (C), respectively.



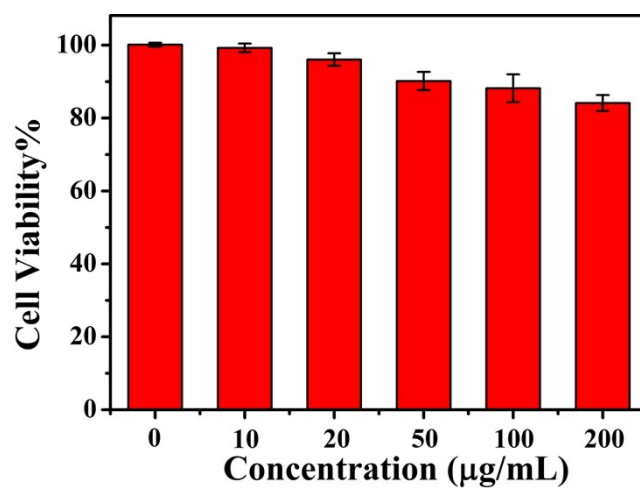
**Fig. S10** Temperature dependence of the fluorescence intensity from C-dots/C-dots dual-emission nanospheres in simulated physiological solution (pH~6.5-7.0, concentration of NaCl is 200 mM). (A) Fluorescence spectra measured under excitation of 380 nm with increasing temperature from 15 to 85 oC at steps of 10 °C. (B) Fluorescence spectra (excitation 380 nm ) for the decrease of temperature from 85 to 15 °C. (C) the fitted curve of intensity ratio of 440 to 590 nm vs. temperature. (D) eight cycles of intensity variations measured at 15–85 °C.



**Fig. S11** A: the PL spectra of C-dots/C-dots dual-emission nanospheres (blue), pure FBS (red line), and C-dots/C-dots/FBS composites (black line); B: the PL intensity variation of C-dots/C-dots/FBS composites after placed 3 days at room temperature.

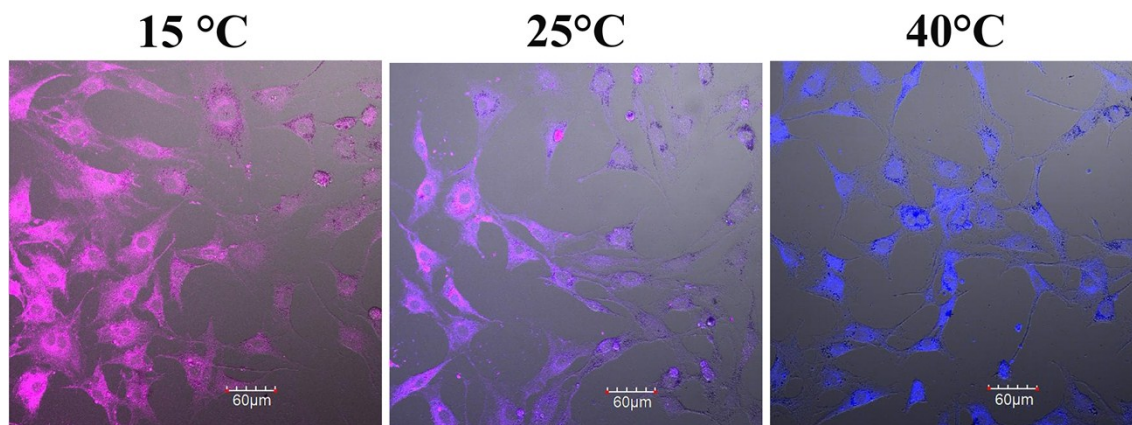


**Fig. S12** The TEM and magnified images of C-dots/C-dots/FBS composites, respectively.

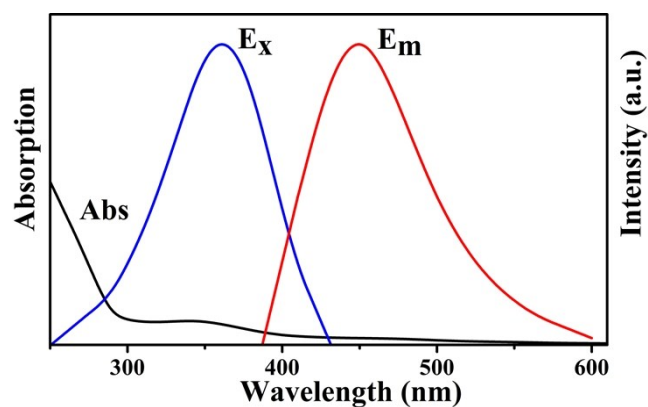


**Fig S13.** Cytotoxicity of the C-dots/C-dots dual-emission fluorescent nanothermometer toward of MC3T3-EI cells, as assessed using the MTT method.

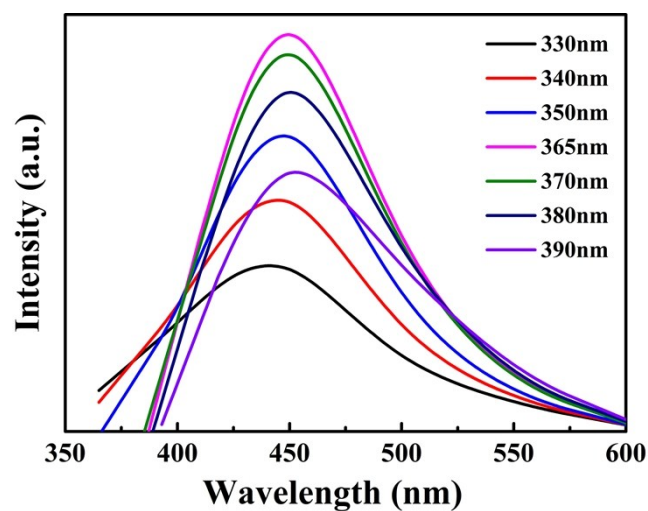




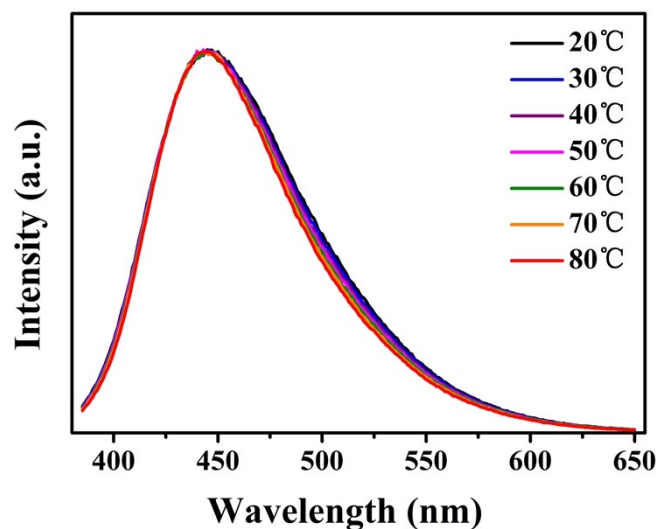
**Fig S14.** Confocal fluorescence images of MC3T3–E1 cells under different physiological temperature, incubated with 20  $\mu\text{g/mL}$  of C-dots/C-dots dual-emission nanospheres for 12 h. All images are obtained using an excitation of wavelength 405 nm. The emissions are recorded in the same range of 420–750nm.



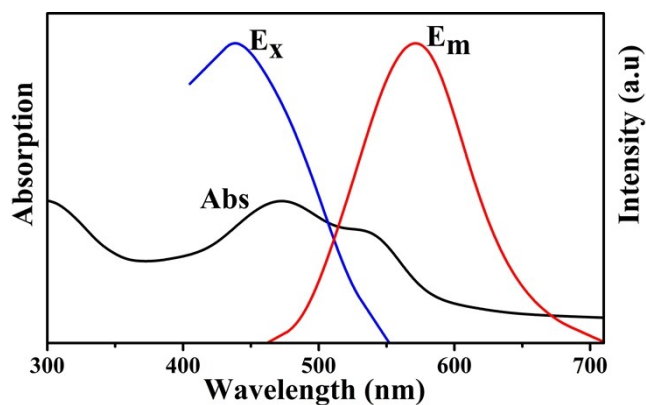
**Fig S15.** The absorption curves, excitation spectra and emission spectra of B-CDs.



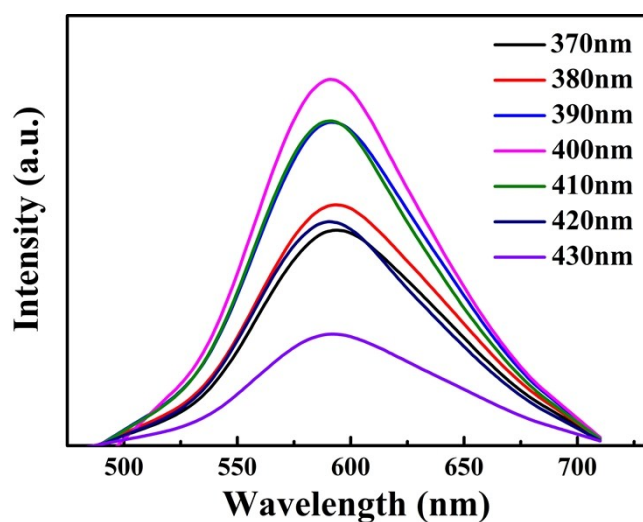
**Fig S16.** The PL emission spectra of B-CDs with different excitation wavelengths.



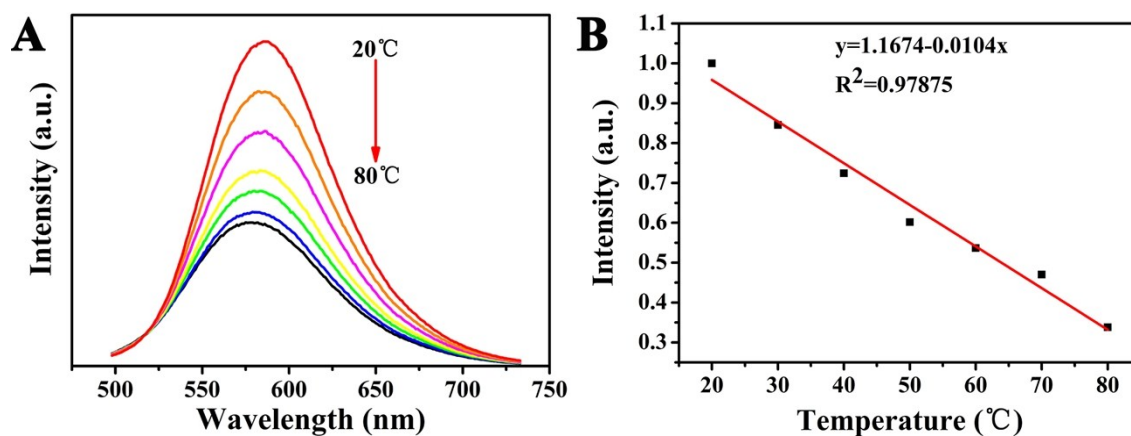
**Fig S17.** The PL emission spectra of B-CDs under different temperature, excited by the optimal excitation wavelength of 370nm.



**Fig S18.** The absorption curves, excitation spectra and emission spectra of O-CDs.



**Fig S19.** The PL emission spectra of O-CDs with different excitation wavelengths.



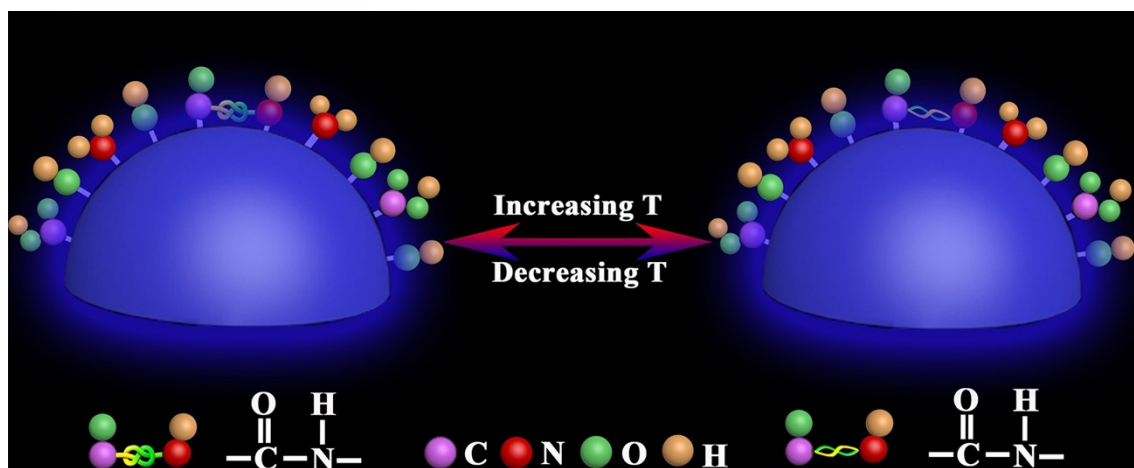
**Fig S20.** Temperature dependence of the fluorescence intensity from O-CDs in aqueous solution. A: Fluorescence emission spectra measured under excitation of 400 nm with the increasing temperature from 20 to 80 °C at a step of 10 °C; B: the fitted curve of intensity of 585 nm vs. temperature.

**Table S3.** Average fluorescence lifetime of B-CDs under different temperatures.

T(°C)	$\tau_{ave}$ (ns)
20	4.9
40	5.2
60	5.1
80	5.1

**Table S4.** Average fluorescence lifetime of O-CDs under different temperatures.

T(°C)	$\tau_{ave}$ (ns)
20	3.8
40	3.8
60	3.7
80	3.6



**Fig S21.** Scheme showing the surface state change of B-CDs.

## Reference

1. R. Miao, S. Zhang, J. Liu and Y. Fang, *Chem. Mater.* 2017, **29**, 5957-5964.
2. C. Wang, Z. Xu, H. Cheng, H. Lin, M. G. Humphrey and C. Zhang, *Carbon* 2015, **82**, 87-95.
3. C. Yao, Y. Xu and Z. Xia, *J. Mater. Chem. C* 2018, **6**, 4396-4399.
4. C. X. Wang, H. H. Lin, Z. Z. Xu, Y. Huang, M. G. Humphrey and C. Zhang, *ACS Appl. Mater. Interfaces* 2016, **8**, 6621-6628.
5. W. Liu, S. Xu, Z. Li, R. Liang, M. Wei, D. G. Evans and X. Duan, *Chem. Mater.* 2016, **28**, 5426-5431.
6. X. Liu, X. Tang, Y. Hou, Q. Wu and G. Zhang, *RSC Adv.* 2015, **5**, 81713–81722.