

## Supporting Information

### **Morphology-controlled synthesis of 3D mesoporous rosette-like CeCoO<sub>x</sub> catalyst by pyrolysis of Ce[Co(CN)<sub>6</sub>] and applied for the catalytic combustion of toluene**

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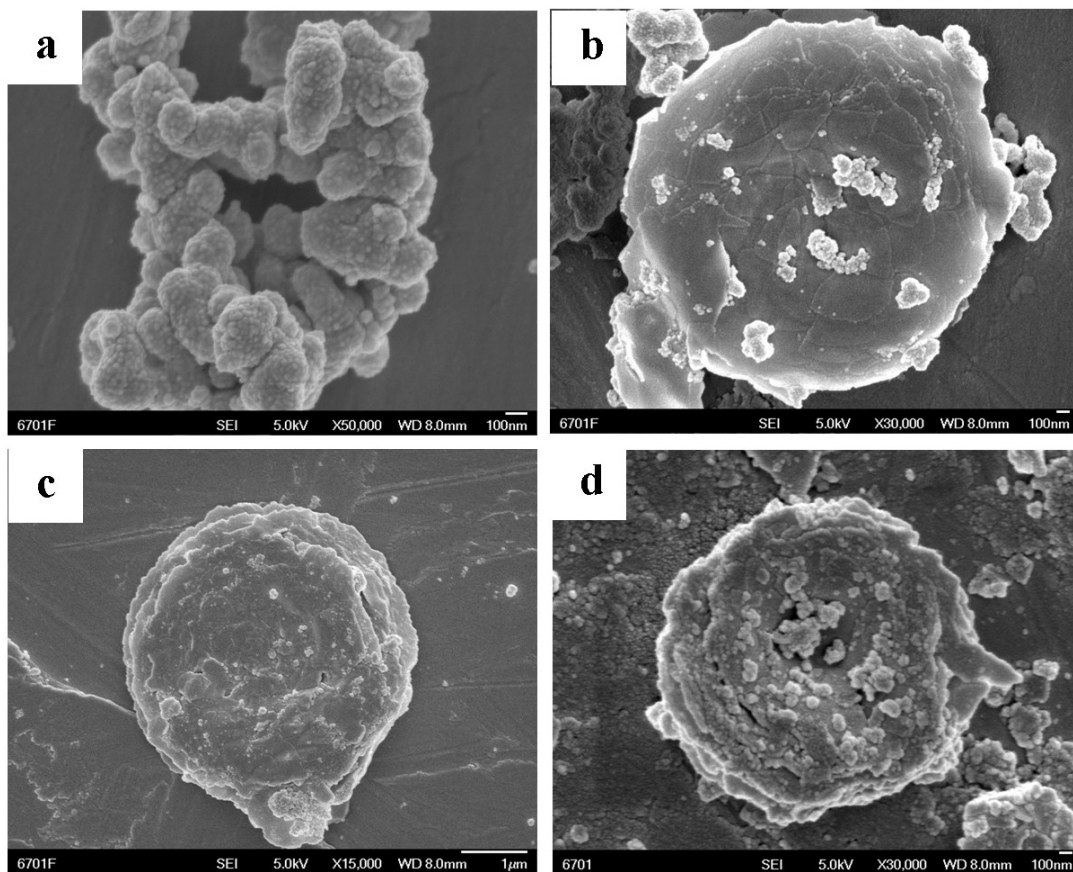
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**Fig.S1 SEM image of Ce[Co(CN)<sub>6</sub>]-200 precursors with different hydrothermal time: (a) 6h, (b) 12h, (c) 18h (d) 24h.**

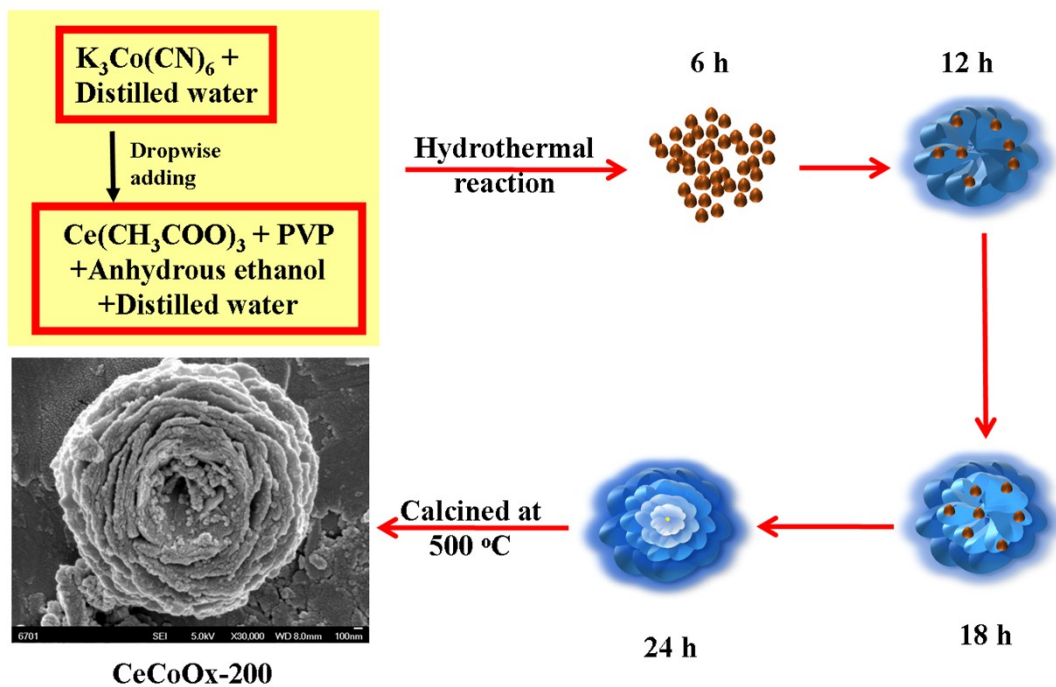
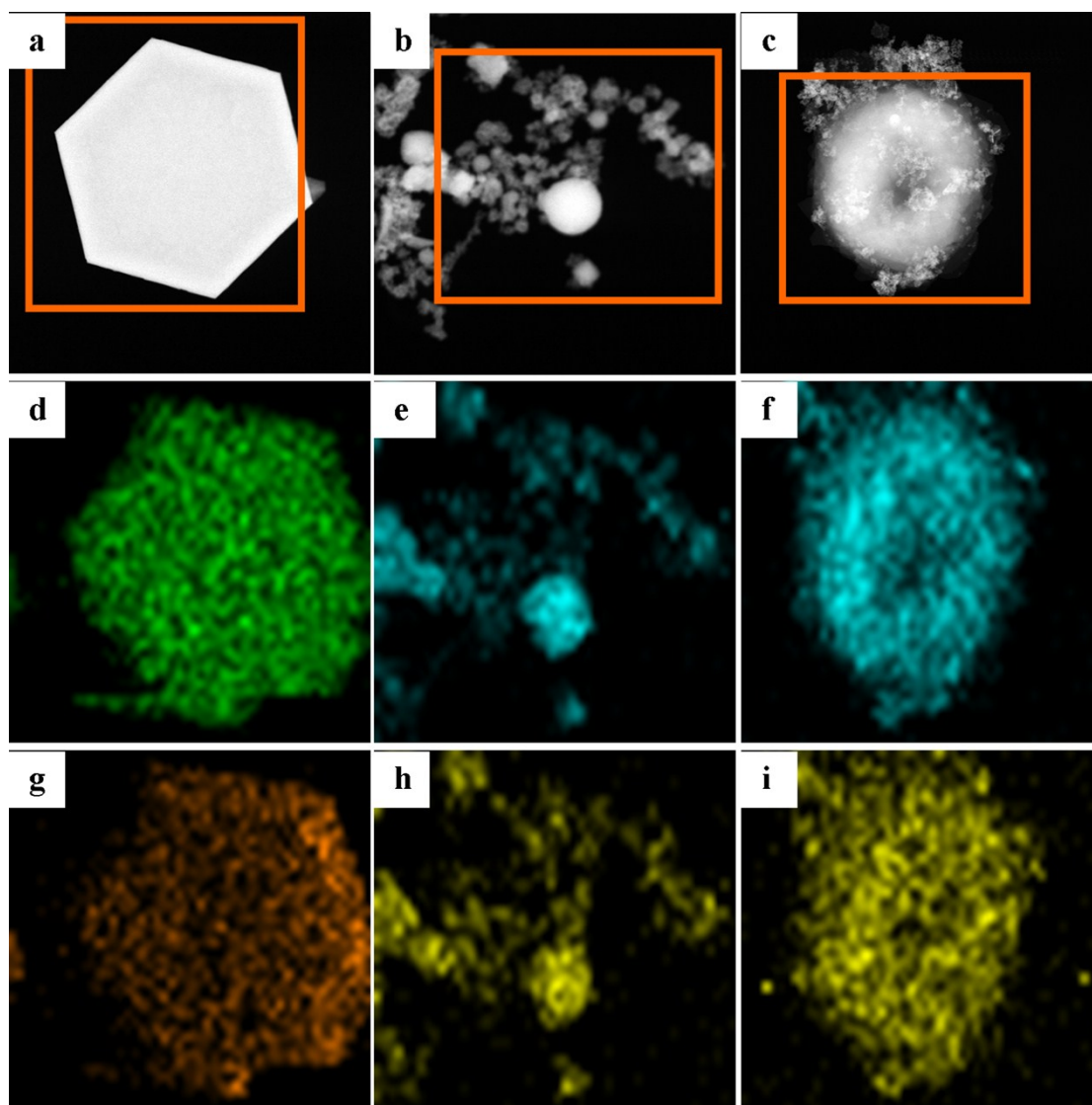
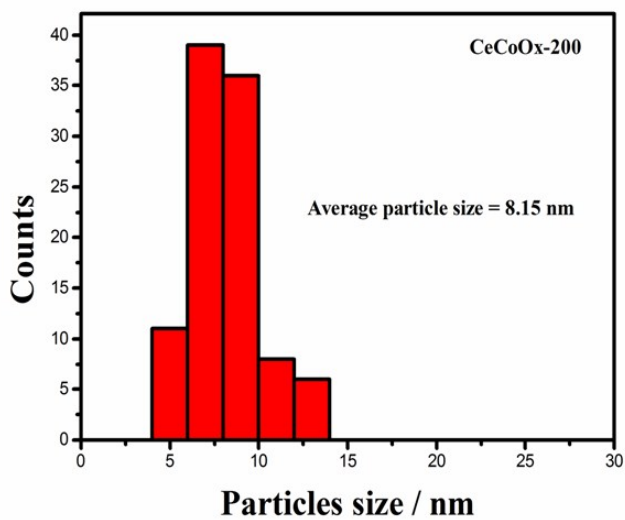
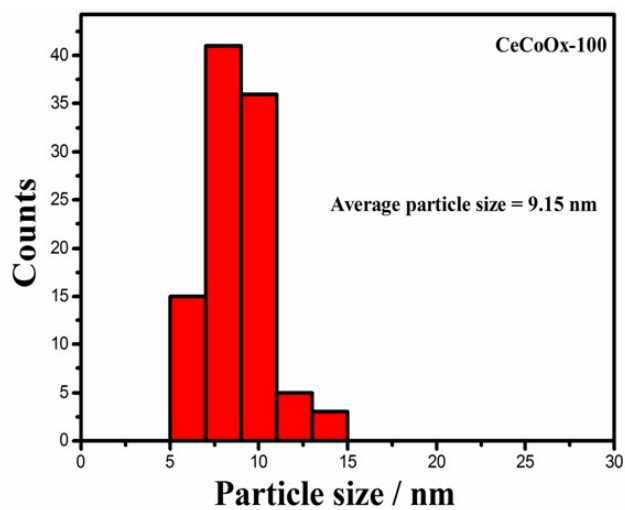
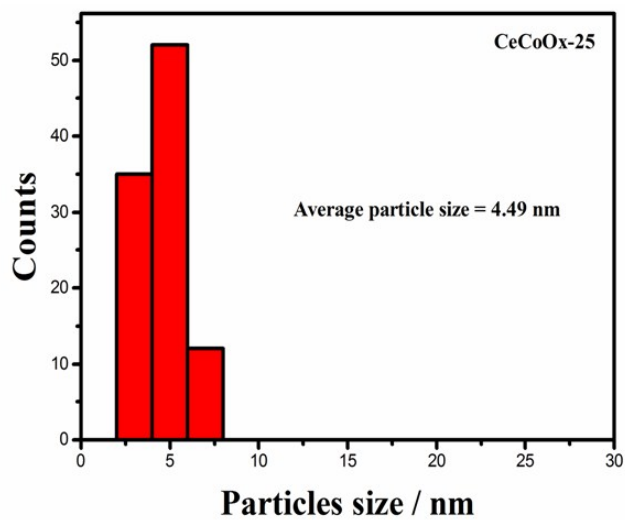


Fig.S2 The formation mechanism of CeCoOx-200.



**Fig.S3 STEM and Energy dispersive X-ray spectrometry elemental maps of CeCoOx-25 (a, d, g), CeCoOx-100 (b, e, h) and CeCoOx-200 (c, f, i)**



**Fig.S4 The Size distribution of CeCoOx-100 and CeCoOx-200.**

**Table S1 Raman Shifts of the Characteristic Peaks of three CeCoOx catalysts.**

<b>Phonon Mode</b>	<b>F<sub>2g</sub>(a)</b>	<b>2TA</b>	<b>F<sub>2g</sub>(b)</b>	<b>E<sub>g</sub></b>	<b>F<sub>2g</sub>(c)</b>	<b>O<sub>v</sub></b>	<b>F<sub>2g</sub>(d)</b>	<b>A<sub>1g</sub></b>
<b>CeCoOx-25</b>	<b>187</b>	<b>-</b>	<b>460</b>	<b>479</b>	<b>520</b>	<b>597</b>	<b>-</b>	<b>680</b>
<b>CeCoOx-100</b>	<b>-</b>	<b>245</b>	<b>445</b>	<b>-</b>	<b>-</b>	<b>-</b>	<b>-</b>	<b>-</b>
<b>CeCoOx-200</b>	<b>181</b>	<b>-</b>	<b>446</b>	<b>-</b>	<b>510</b>	<b>591</b>	<b>605</b>	<b>665</b>
<b>Literature<sup>20, 23, 24</sup></b>	<b>194</b>	<b>256</b>	<b>465</b>	<b>482</b>	<b>522</b>	<b>595</b>	<b>618</b>	<b>691</b>

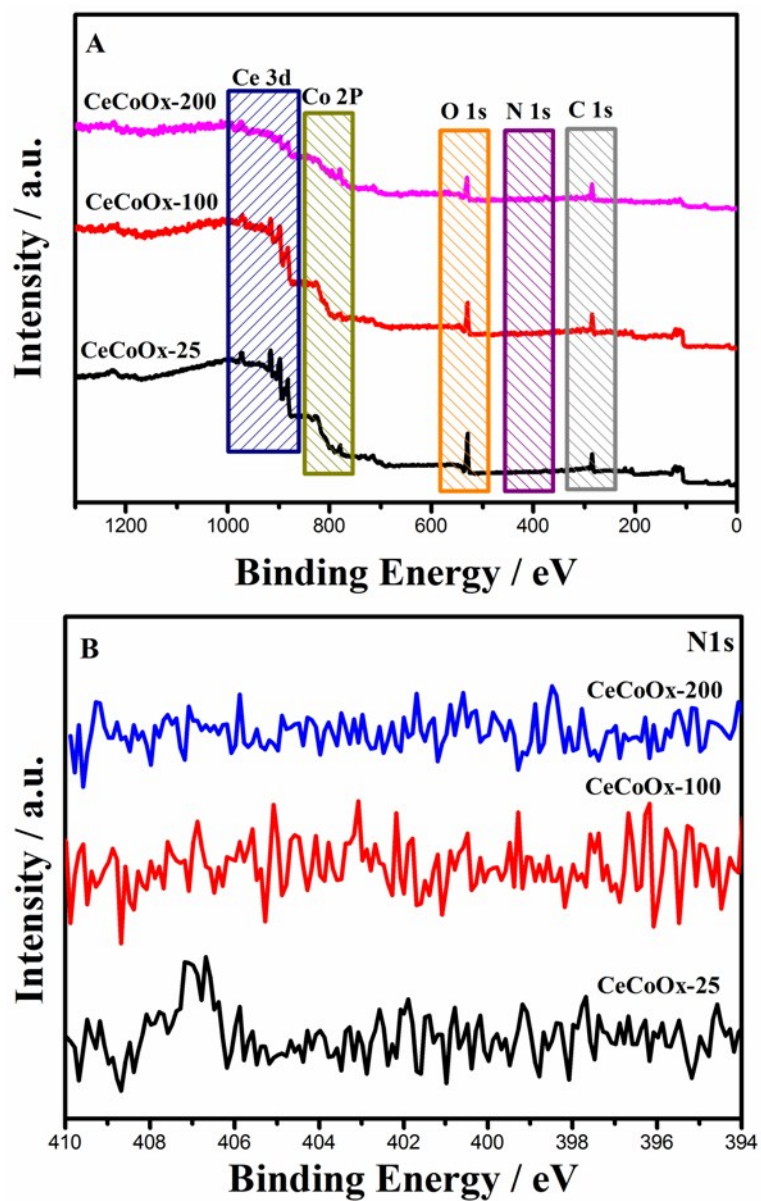


Fig.S5 The XPS survey and N1s of CeCoOx catalysts.

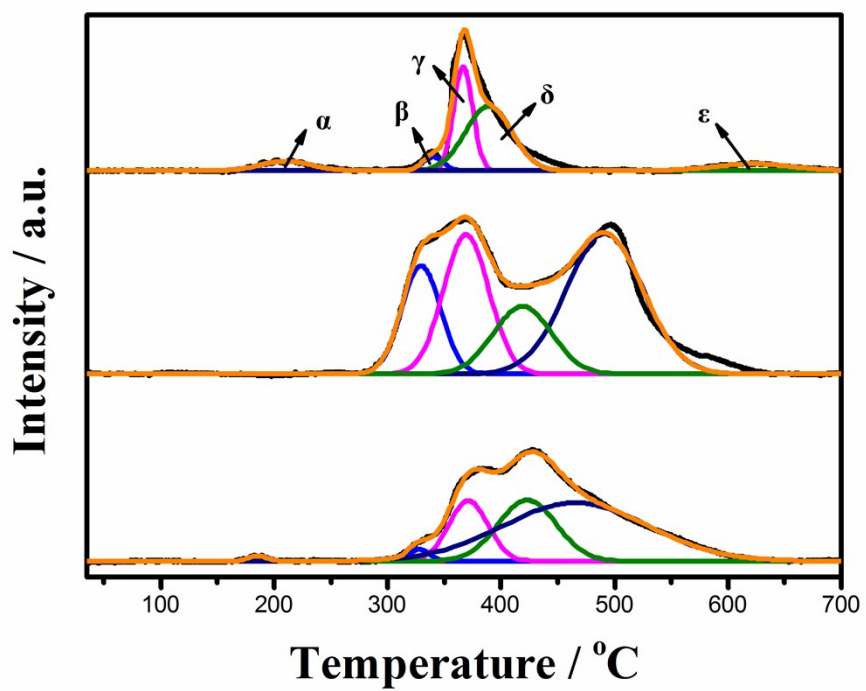


Fig.S6 CO-TPR of CeCoO<sub>x</sub> catalysts.



**Table S2 The H<sub>2</sub>-TPR consumption data of three CeCoO<sub>x</sub> catalysts**

Catalysts	Peak $\alpha$		Peak $\beta$		Peak $\gamma$		Peak $\delta$		Peak $\varepsilon$	
	Position	Area	Position	Area	Position	Area	Position	Area	Position	Area
	/ °C	a.u.	/ °C	/ a.u.	/ °C	a.u.	/ °C	a.u.	/ °C	a.u.
CeCoO <sub>x</sub> -25	-	-	277	770	308	2000	480	3500	520	8000
CeCoO <sub>x</sub> -100	-	-	298	451	369	2112	431	1054	473	616
CeCoO <sub>x</sub> -200	136	275	297	9840	325	3745	415	41468	483	15179