

Electronic Supplementary Information

Exploring the Surface Chemistry of Cesium Lead Halide Perovskite Nanocrystals

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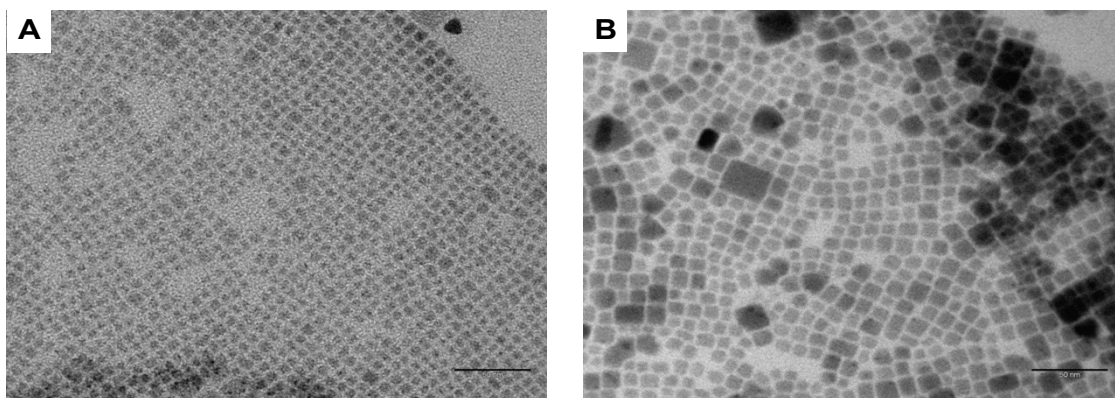


Figure S1. TEM images of as-prepared CsPbCl_3 (A) and CsPbI_3 (B) NCs. Scale bar = 100 nm.

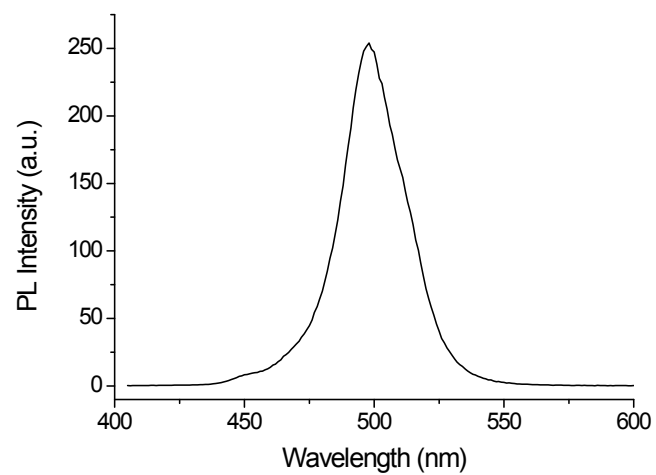


Figure S2. PL spectrum (recorded in cyclohexane) of the supernatant recovered after the purification step for CsPbBr_3 NCs.

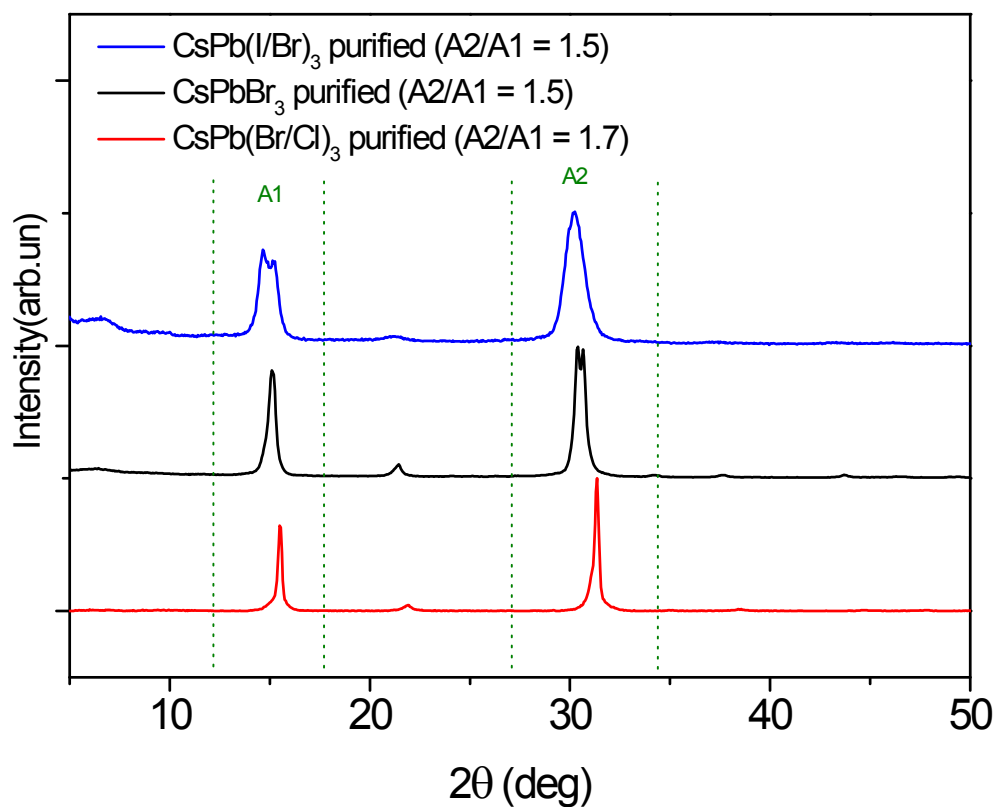


Figure S3. XRD $\theta/2\theta$ scans from purified samples, showing the intensity ratio (in the legend) between the integrated areas under the two main peaks (A1 and A2).



Figure S4. As-prepared (left) and purified (right) CsPbBr_3 NCs samples in the solid state after three weeks storage under ambient conditions and room light.

Table S1. Element concentration (atomic %) and obtained molar ratios of the studied samples analyzed via FEG-SEM-EDX (n=3).

Sample	Element concentration (atomic %)										Molar ratios				
	Cs		Pb		Cl		Br		I		Cs	Pb	Cl	Br	I
	Conc	SD	Conc	SD	Conc	SD	Conc	SD	Conc	SD					
CsPbI ₃	21.33	0.58	16.88	0.64	-	-	-	-	61.79	0.07	1.3	1.0	-	-	3.7
CsPb(I/Br) ₃	23.90	0.85	17.40	0.33	-	-	34.95	1.44	23.74	0.64	1.4	1.0	-	2.0	1.4
pur. CsPb(I/Br) ₃	24.23	0.26	18.63	0.29	-	-	48.05	0.79	9.09	0.30	1.3	1.0	-	2.6	0.5
CsPb(Br/Cl) ₃	25.00	0.21	19.10	0.13	29.21	0.37	26.69	0.41	-	-	1.3	1.0	1.5	1.4	-
pur. CsPb(Br/Cl) ₃	22.13	0.28	19.48	0.45	30.80	0.09	27.59	0.45	-	-	1.1	1.0	1.6	1.4	-
CsPbBr ₃	24.16	0.77	19.99	0.20	-	-	55.85	0.85	-	-	1.2	1.0	-	2.8	-
pur. CsPbBr ₃	25.02	0.26	19.60	0.20	-	-	55.37	0.05	-	-	1.3	1.0	-	2.8	-
CsPbCl ₃	21.10	0.70	19.36	0.82	59.54	0.61	-	-	-	-	1.1	1.0	3.1	-	-

Conc = concentration

SD = standard deviation

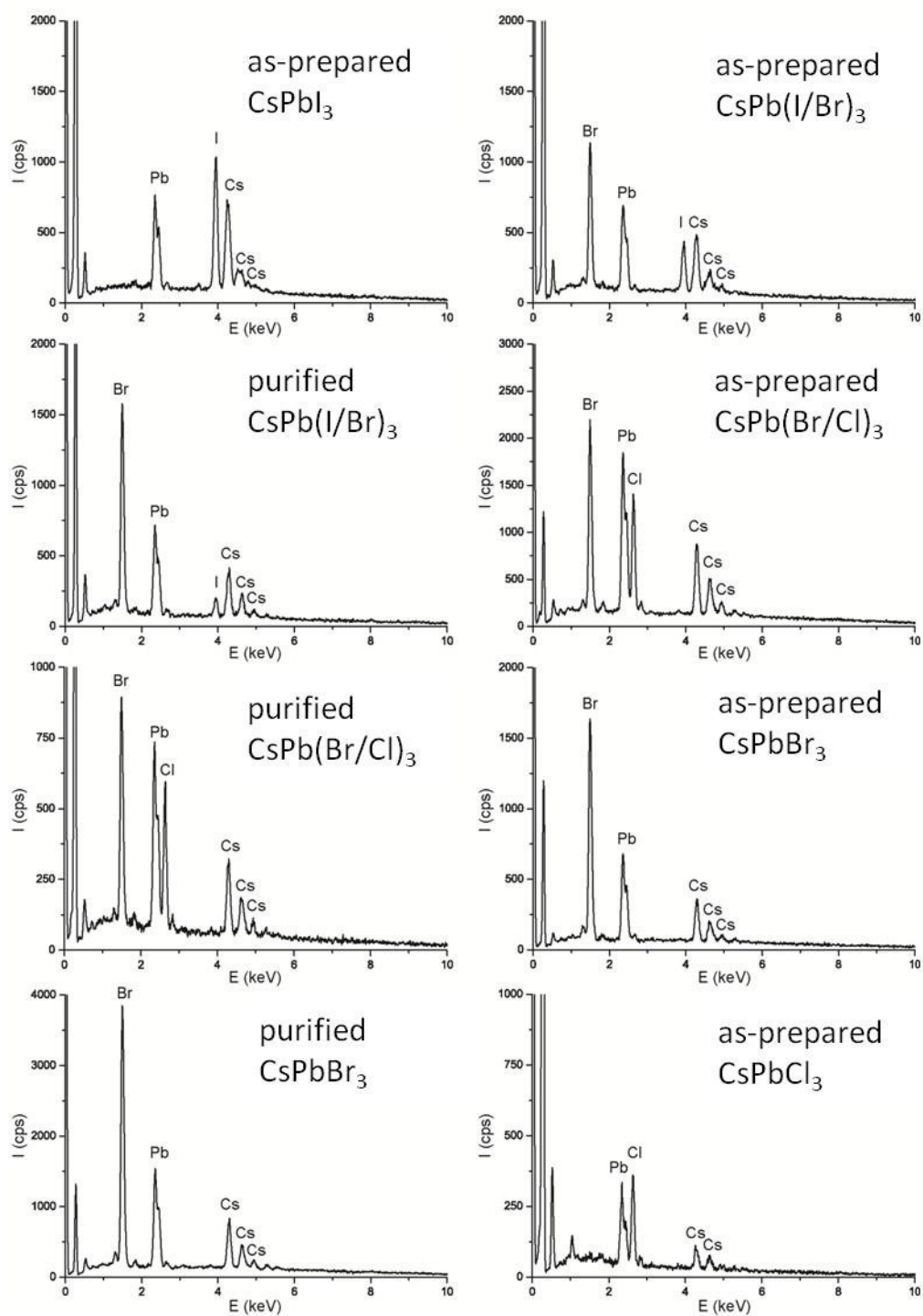


Figure S5. EDX profiles of as-prepared and purified NCs.

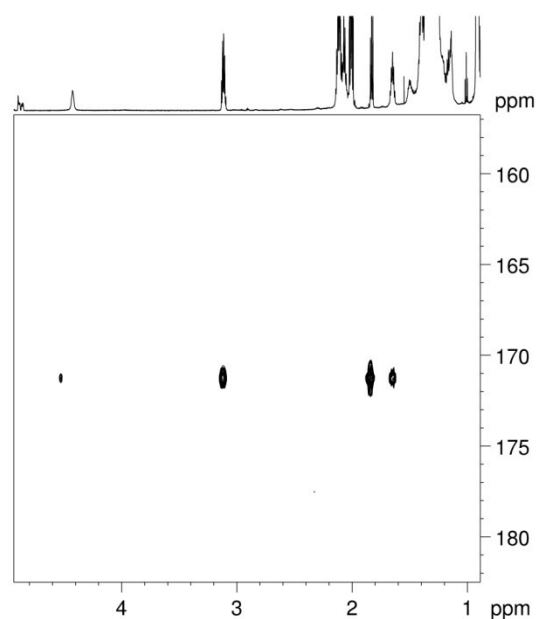


Figure S6. HMBC spectrum in C_6D_6 of the supernatant solution containing the amide, highlighting the presence of the $-CONH-$ group at $\delta = 171.8$ ppm.

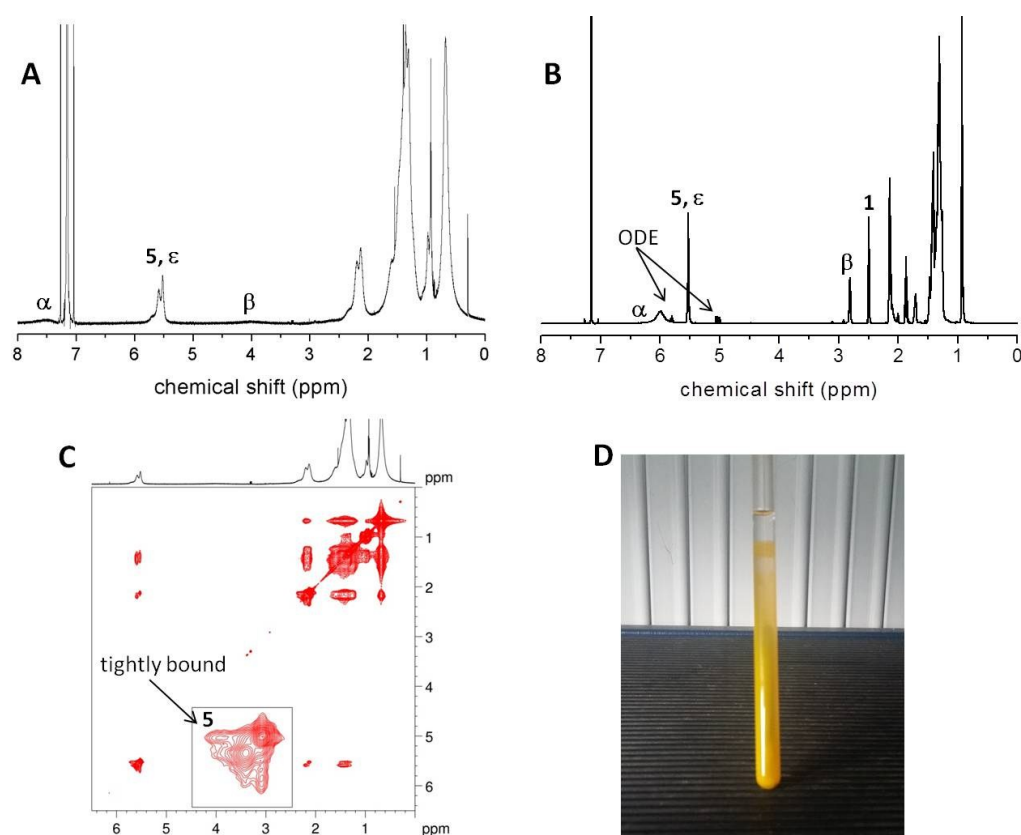


Figure S7. (A) 1H -NMR spectrum of the twice-purified $CsPbBr_3$ NCs in C_6D_6 . (B) 1H -NMR spectrum of the supernatant solution obtained from the relevant centrifugation in C_6D_6 . (C) 2D-NOESY spectrum of twice-purified $CsPbBr_3$ NCs in C_6D_6 ; the inset shows an expansion of the vinylene region. (D) Picture of the same solution stored for one week in ambient conditions.

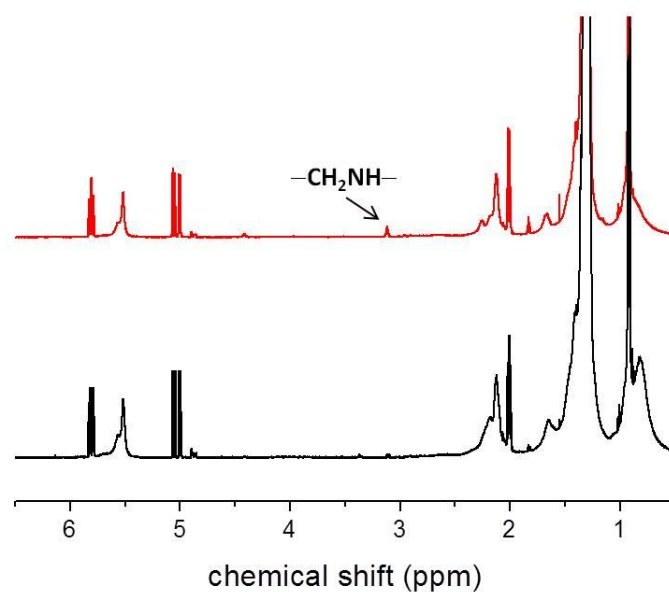


Figure S8. ^1H -NMR spectra of purified CsPbBr_3 NCs (bottom) and the relevant sample stored in ambient conditions for three weeks in C_6D_6 . The arrow points to one of the typical methylene signals ($-\text{CH}_2\text{NHCO}-$) attributable to the amide presence.

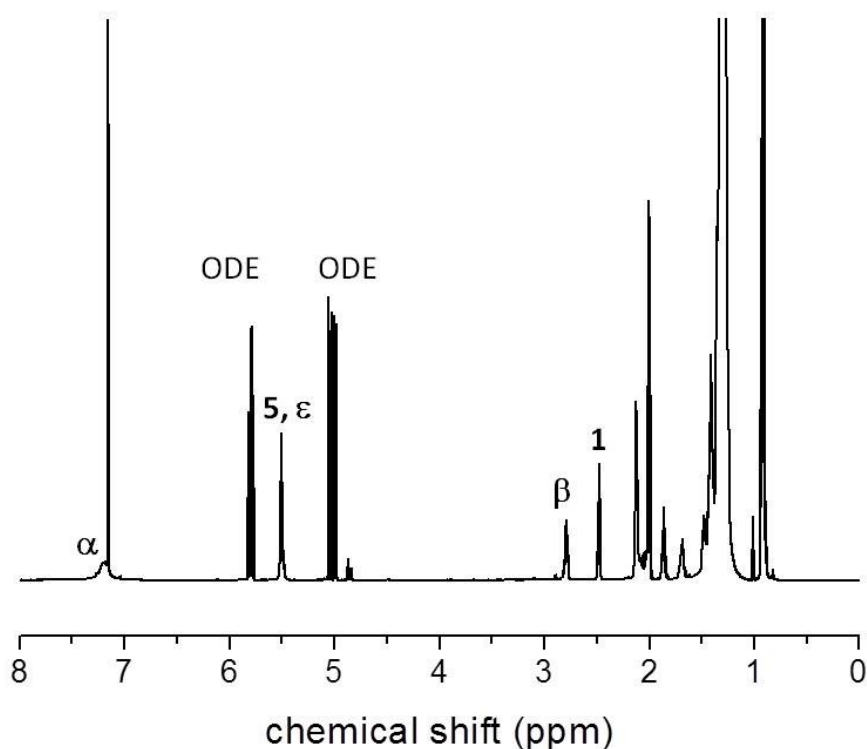


Figure S9. ^1H -NMR spectrum in C_6D_6 of $\text{PbI}_2/\text{OLA}/\text{OLAm}/\text{ODE}$ mixture used for the synthesis of CsPbI_3 NCs immediately before the addition of cesium-oleate. No signals attributable to the amide are present.

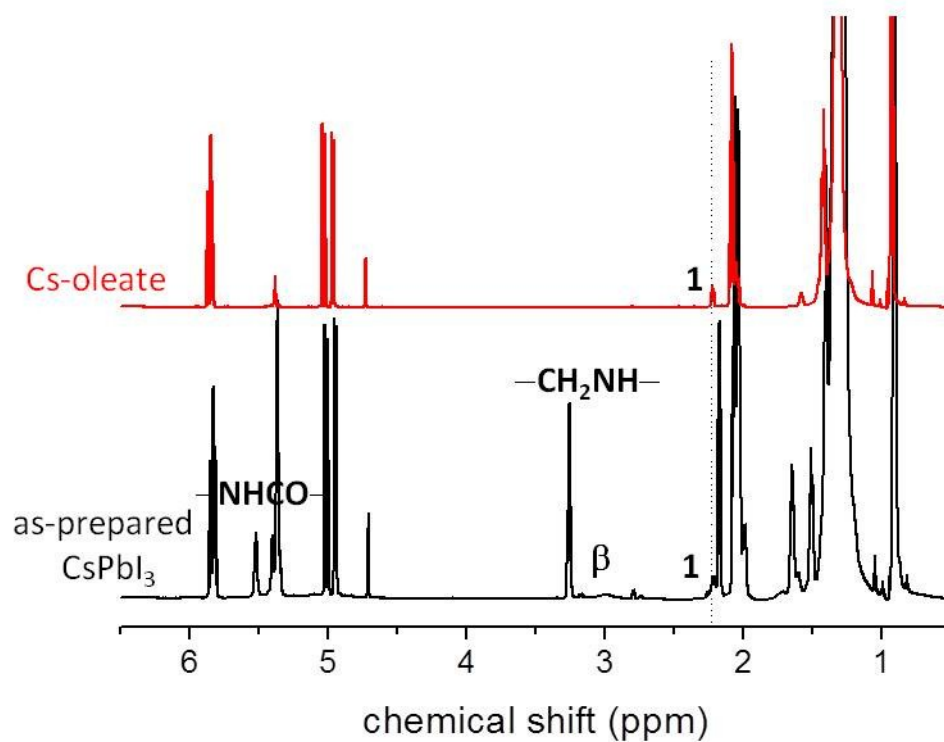


Figure S10. Comparison between the ^1H -NMR spectra of as-prepared CsPbI_3 NCs and cesium-oleate in CDCl_3 .

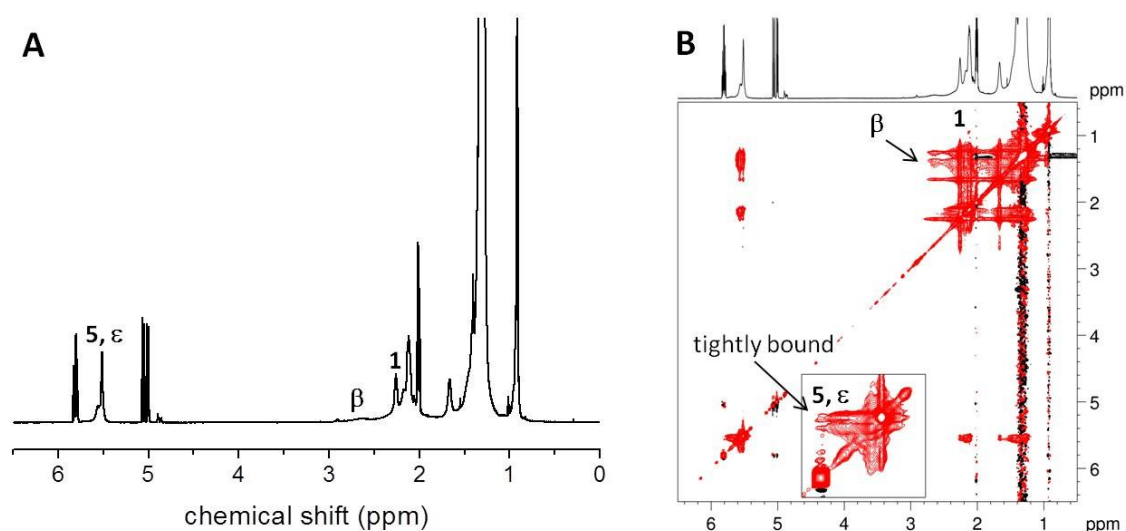


Figure S11. (A) ^1H -NMR spectrum of as-prepared $\text{CsPb}(\text{Br/Cl})_3$ NCs in C_6D_6 and (B) the relevant 2D-NOESY spectrum.

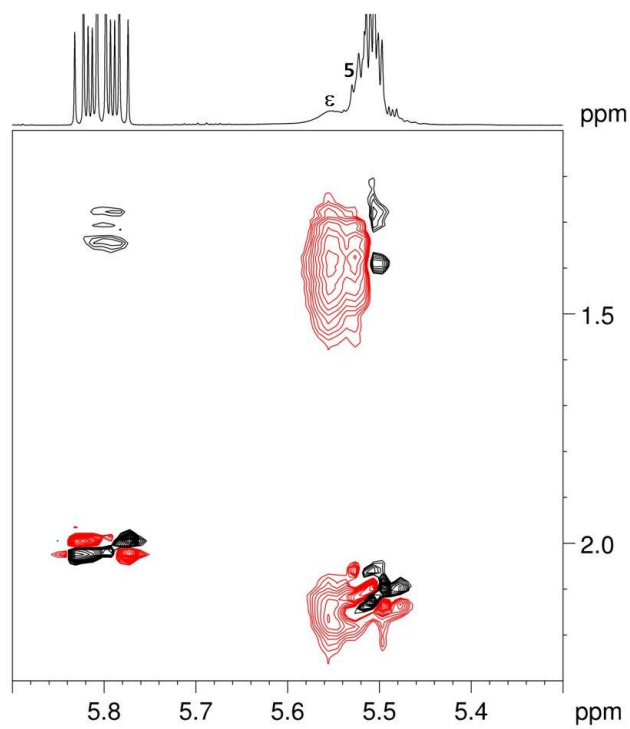


Figure S12. Expansion of the vinylene region of the 2D-NOESY spectrum of as-prepared CsPb(I/Br)₃ NCs in C₆D₆ evidencing the absence of tightly bound ligands.