

Supporting Information

Stable, Color-Tunable 2D SCN-Based Perovskites: Revealing the Critical Influence of Asymmetric Pseudo-Halide on Constituent Ions

Chia-Hsin Li,^a Chang-Che Tsai,^a Ming-Yun Liao,^a Yu-An Su,^a Shiang-Tai Lin,^a and Chu-Chen Chueh^{a,b}*

^a C.-H. Li, C.-C. Tsai, M.-Y. Liao, Dr. Y.-A. Su, Prof. S.-T. Lin, and Prof. C.-C. Chueh
Department of Chemical Engineering, National Taiwan University, Taipei 10617, Taiwan
E-mail: cchueh@ntu.edu.tw

^b Prof. C.-C. Chueh

Advanced Research Center for Green Materials Science and Technology, National Taiwan University, Taipei 10617, Taiwan

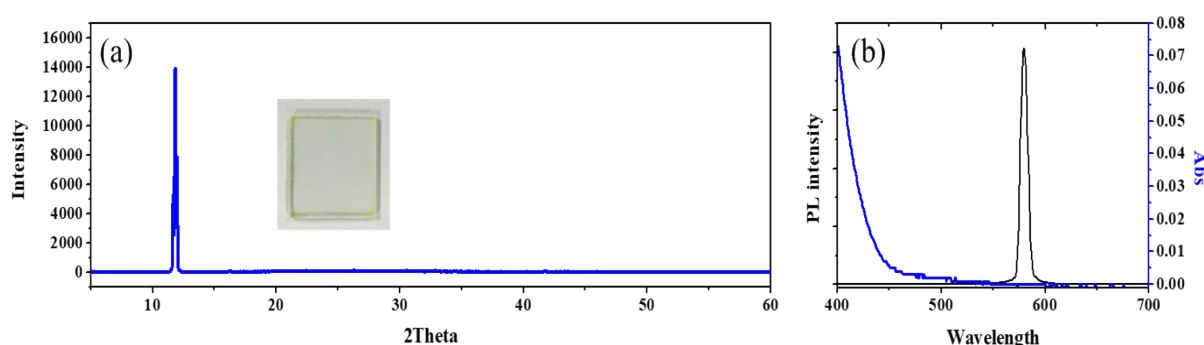


Figure S1. (a) The XRD pattern and (b) UV-vis absorption and PL spectrum of the as-cast film made from the precursor solution containing FAI and Pb(SCN)₂ with the aim to form 2D FA₂PbI₂(SCN)₂. Insert is the real picture of the as-cast film, displaying matt white color. The result indicates the failure in forming 2D FA₂Pb(SCN)₂I₂, as compared to the results of MA₂Pb(SCN)₂I₂ and Cs₂Pb(SCN)₂I₂ (Figure 1) prepared by same deposition technique.

Table S1. The optimal unit cell parameters and energy of the studied SCN-based Cs-perovskites.

Systems	a	b	c	volume	Total energy
	Å	Å	Å	Å ³	eV
Cs-I	18.64	6.39	6.38	760.61	-114.85
Cs-bb	18.64	6.14	6.38	729.90	-116.30
Cs-bc	18.69	6.32	6.20	733.33	-116.27
Cs-cb	18.67	6.33	6.20	732.73	-116.27
Cs-cc	18.91	6.41	6.04	732.04	-116.34
Cs-Br	18.62	6.15	6.06	693.86	-117.55

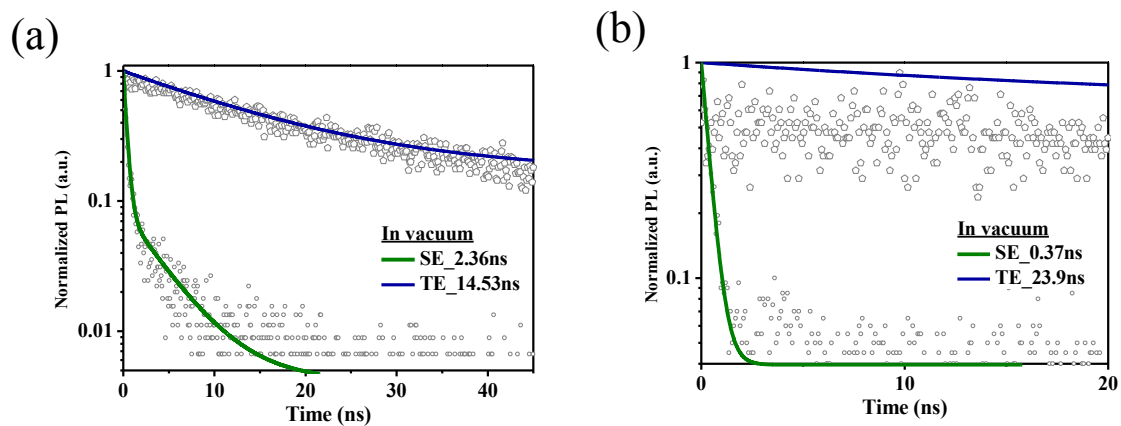


Figure S2. TRPL spectra of the (a) $\text{Cs}_2\text{Pb}(\text{SCN})_2\text{I}_2$ and (b) $(\text{MA})_2\text{Pb}(\text{SCN})_2\text{I}_2$ films measured in vacuum.

Table S2. TRPL carrier lifetime for the studied 2D perovskite films.

		$\text{Cs}_2\text{Pb}_2(\text{SCN})_2\text{I}_2$	$(\text{MA})_2\text{Pb}_2(\text{SCN})_2\text{I}_2$
In vacuum	SE	2.36 ns	0.368 ns
	TE	14.53 ns	23.9 ns
In ambient air	SE	3.97 ns	0.359 ns
	TE	12.37 ns	17.09 ns

Table S3. Fitting parameters for the TRPL dynamics for the studied 2D perovskite films.

$\text{Cs}_2\text{Pb}_2(\text{SCN})_2\text{I}_2$					$(\text{MA})_2\text{Pb}_2(\text{SCN})_2\text{I}_2$			
In vacuum		In air			In vacuum		In air	
	SE	TE	SE	TE	SE	TE	SE	TE
A1	31.75	163	41.55	152.55	187	7.29	177	28.5
τ_1 (ns)	4.19	14.53	6.34	12.38	0.369	23.9	0.359	5.45
A2	358	-	417	-	-	-	-	37.1
τ_2 (ns)	0.34	-	0.42	-	-	-	-	20.75

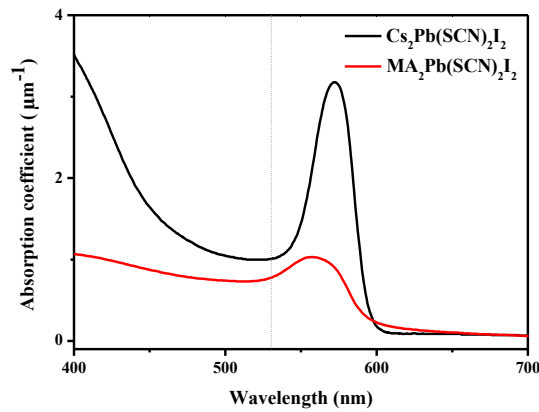
**Figure S3.** Absorption coefficients of $\text{Cs}_2\text{Pb}(\text{SCN})_2\text{I}_2$ and $\text{MA}_2\text{Pb}(\text{SCN})_2\text{I}_2$.

Table S4. Enthalpies of decomposition of each system calculated from Eqn (1), (2), and (3).

	Compound	CS-I	CS-bb	CS-bc	CS-cb	CS-cc	CS-Br
Enthalpy of decomposition (eV)	Eqn 1	12.69	13.32	13.37	13.34	13.75	13.52
	Eqn 2	3.52	3.84	3.81	3.79	3.76	3.94
	Eqn 3	12.98	13.66	13.63	13.57	13.76	14.00

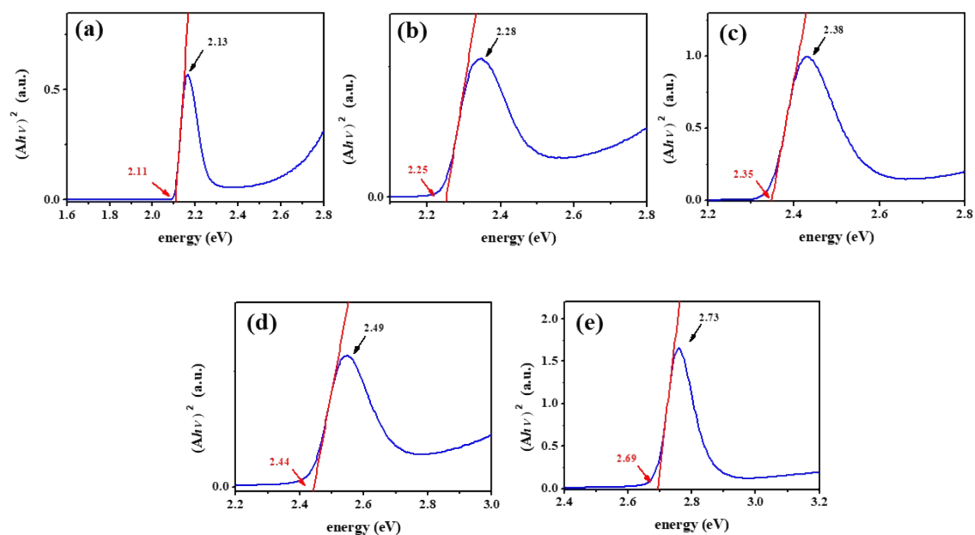


Figure S4. The Tauc plots of the studied $\text{Cs}_2\text{Pb}(\text{SCN})_2(\text{I}_{1-x}\text{Br}_x)_2$ films: (a) $x = 0$, (b) $x = 1/3$, (c) $x = 1/2$, (d) $x = 2/3$, and (e) $x = 1$.

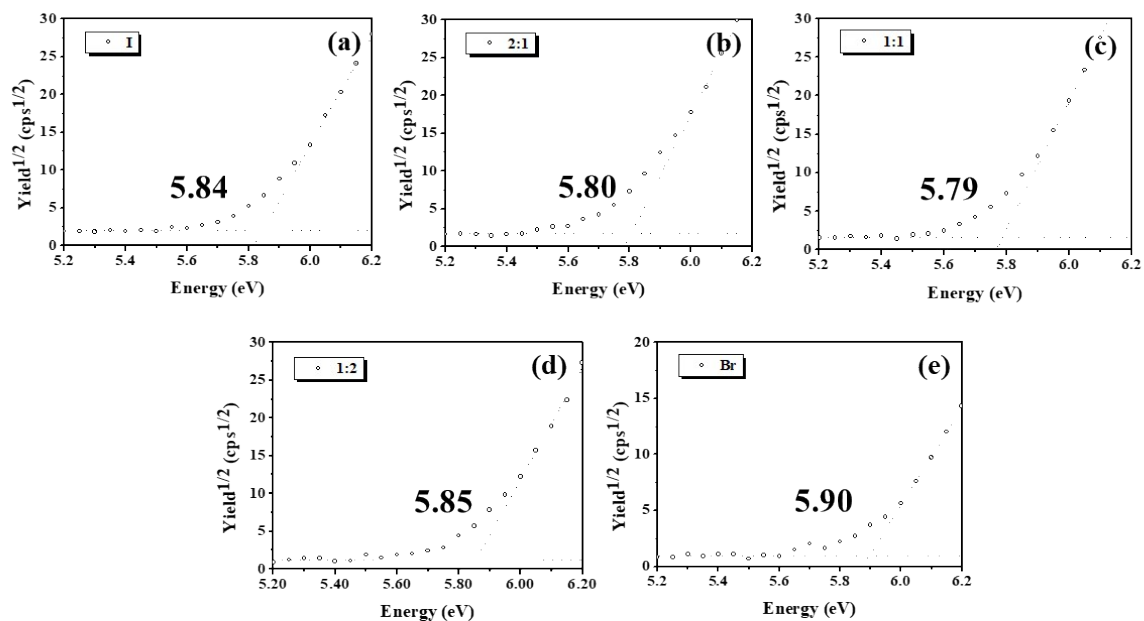


Figure S5. The HOMO level of the studied $\text{Cs}_2\text{Pb}(\text{SCN})_2(\text{I}_{1-x}\text{Br}_x)_2$ films measured by photoelectron spectroscopy: (a) $x = 0$, (b) $x = 1/3$, (c) $x = 1/2$, (d) $x = 2/3$, and (e) $x = 1$.

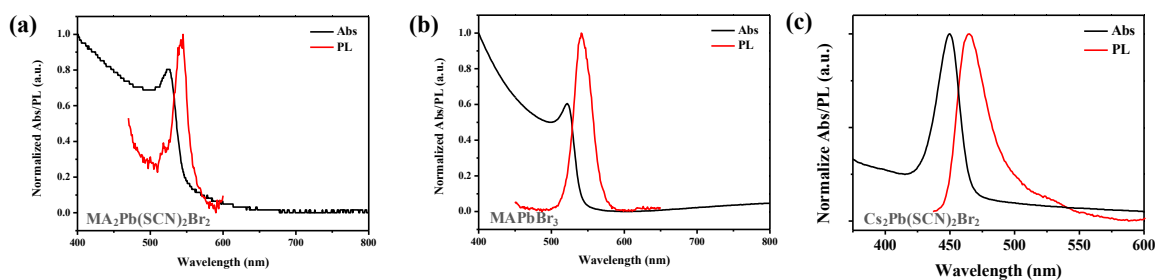


Figure S6. The UV-vis absorption and PL spectra of (a) the prepared $(\text{MA})_2\text{Pb}(\text{SCN})_2\text{Br}_2$, (b) MAPbBr_3 , and (c) $\text{Cs}_2\text{Pb}(\text{SCN})_2\text{Br}_2$ films.

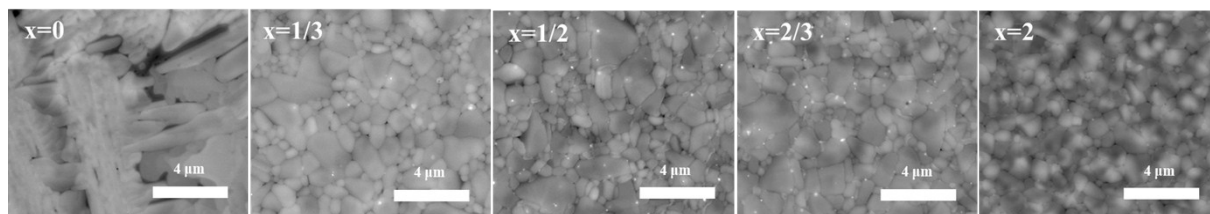


Figure S7. The SEM images (10000x) of the studied $\text{Cs}_2\text{Pb}(\text{SCN})_2(\text{I}_{1-x}\text{Br}_x)_2$ films.