

Supporting Information

**Preparation of high-yield and ultra-pure Au₂₅ nanoclusters:
towards their implementation in real-world applications**

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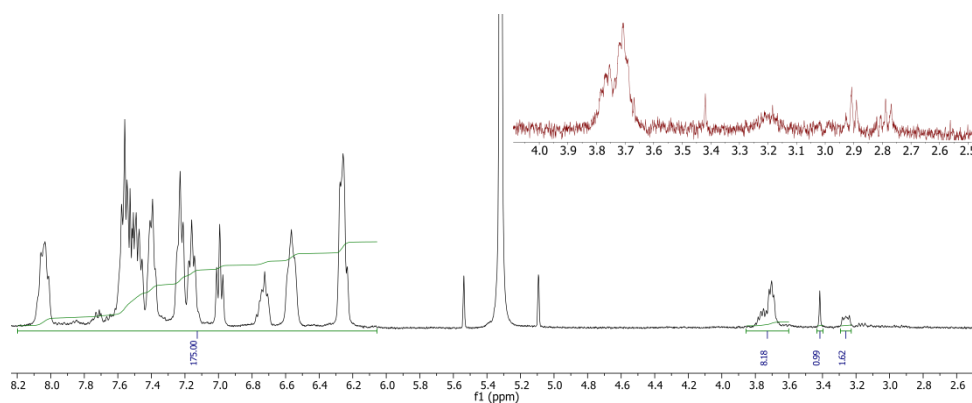


Fig. S1: ¹H NMR spectrum of the “purified” Au₂₅ nanocluster in CD₂Cl₂. The set of peaks from 6.3 to 8.1 ppm correspond to phenyl groups from PPh₃ and SCH₂CH₂Ph (integral normalized to 175). The inset shows a ¹H NMR spectrum of a different sample of Au₂₅ NCs in CD₂Cl₂ with unbound SCH₂CH₂Ph present in solution giving rise to two triplets at 2.8 and 2.9 ppm. These signals originate from alpha and beta CH₂ groups in SCH₂CH₂Ph. Proton signals from such alpha/beta group and are known to shift downfield, be broadened or even become undetectable in ¹H or ¹³C NMR spectra upon binding to a Au NC surface.¹ All three effects were observed for the CH₂ groups in Au₂₅ NC-bound SCH₂CH₂Ph: We propose that the signal from beta CH₂ is shifted from 2.8 to 3.7 ppm as well as broadened whereas the alpha CH₂ group, closest to the Au core becomes undetectable.

In accordance to a previous report,² alpha CH₂ groups of tetraoctylammonium (TOA) molecules are assigned to a signal at 3.25 ppm. Its integral compared to the phenyl regions’ is remarkably low compared to a previous report.² The singlet at 3.4 ppm is assigned to the CH₃ group of methanol residually present from the purification.³

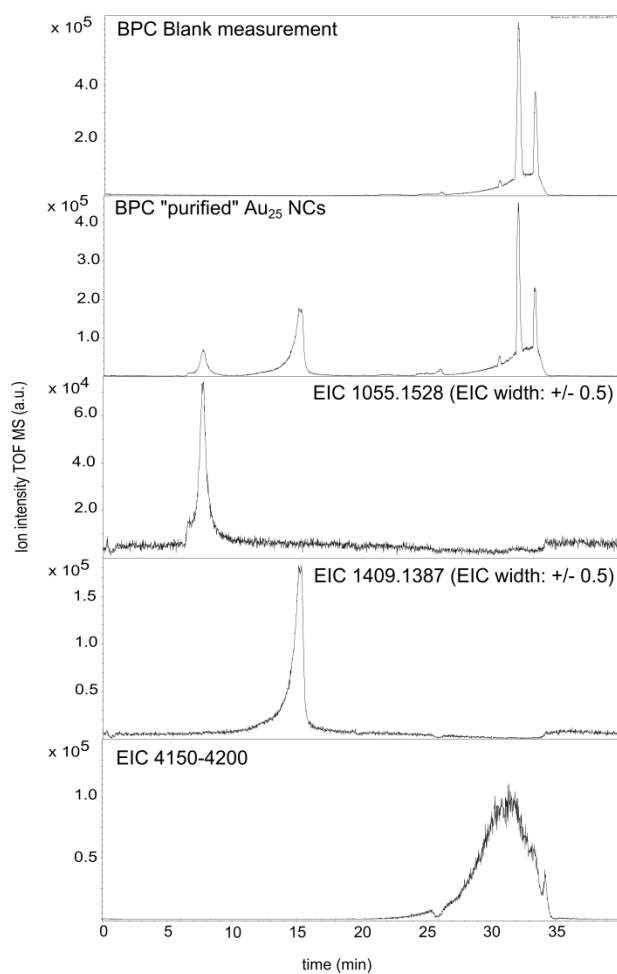


Fig. S2: LC-MS measurement of Au₂₅ NCs. Base peak and extracted ion chromatograms (BPC and EIC, respectively) obtained by electrospray ionization mass spectrometry (ESI-MS).

References

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