

Supporting information

Novel tunneled phosphorus-doped WO₃ films achieved by ignited red phosphorus for stable and fast switching electrochromic performance

Bon-Ryul Koo^a, Kue-Ho Kim^b, and Hyo-Jin Ahn^{a,b*}

^aProgram of Materials Science & Engineering, Convergence Institute of Biomedical Engineering and Biomaterials, Seoul National University of Science and Technology, Seoul 01811, Korea

^bDepartment of Materials Science and Engineering, Seoul National University of Science and Technology, Seoul 01811, Korea

*Corresponding author.

E-mail address: hjahn@seoultech.ac.kr (H.-J. Ahn)

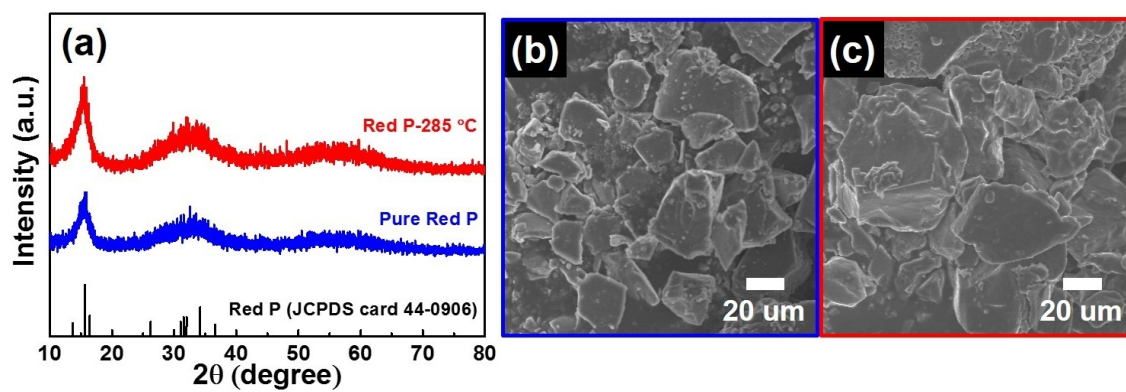


Fig. S1 (a) XRD patterns of pure red P particles (blue line) and red P particles annealed at 285 °C (red line) and their resultant SEM images ((b) and (c), respectively).

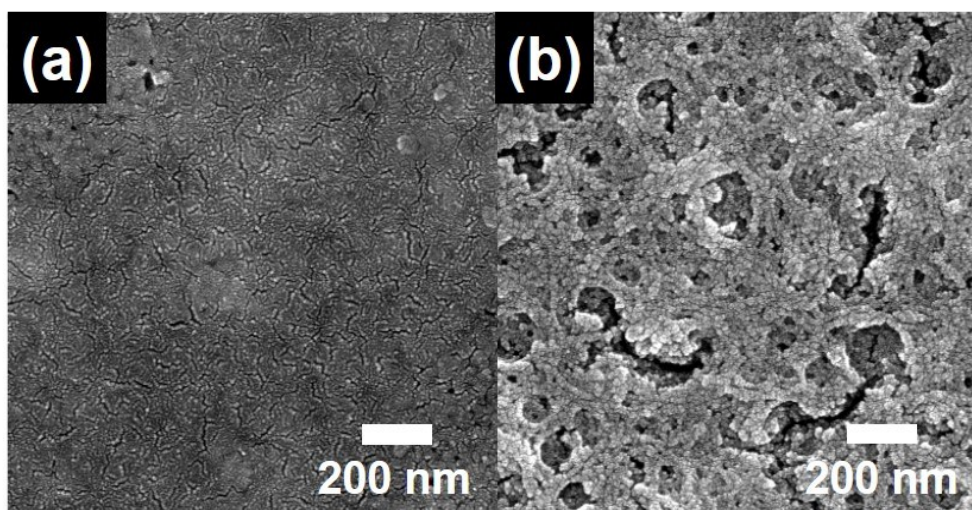


Fig. S2 Top-view SEM images of WO_3 -1RP annealed at (a) 240 °C and (b) 340 °C.

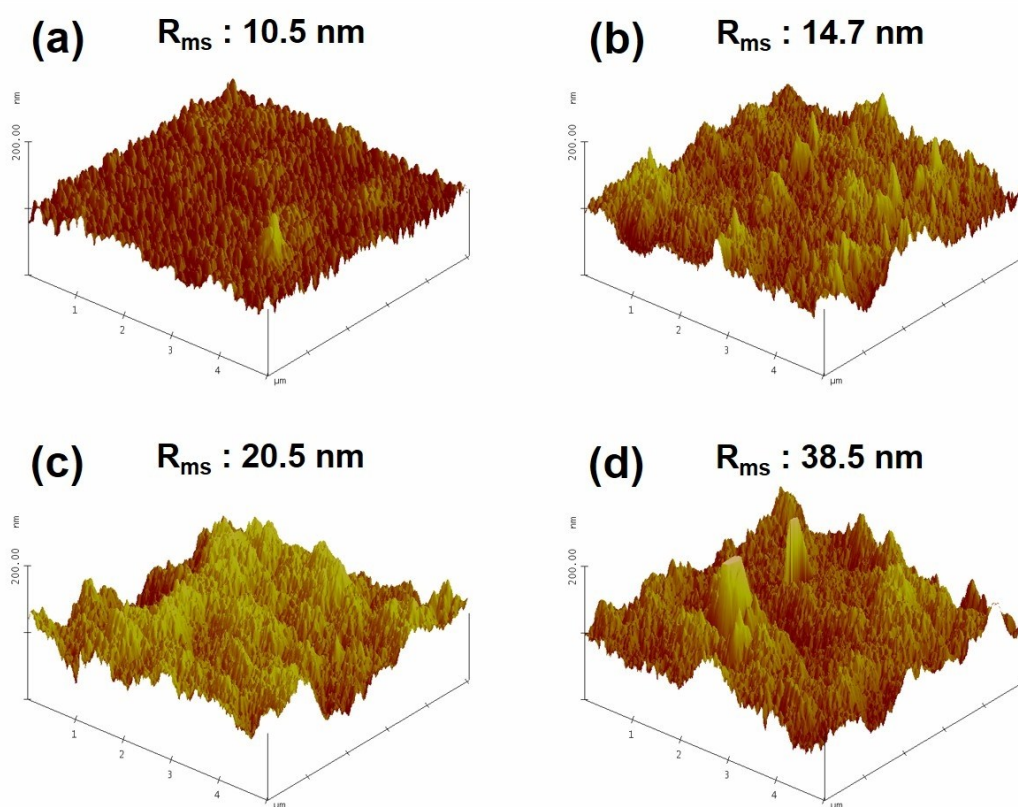


Fig. S3 AFM images obtained from (a) bare WO_3 , (b) WO_3 -0.5RP, (c) WO_3 -1RP, and (d) WO_3 -5RP.

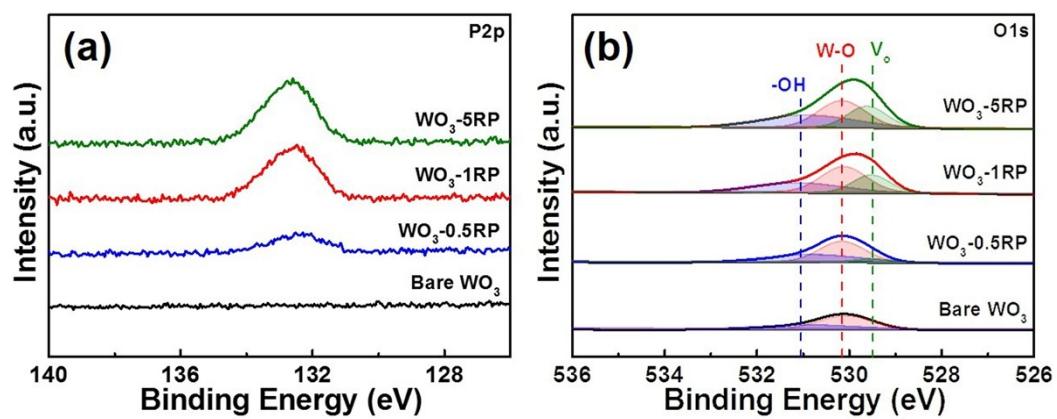


Fig. S4 XPS core-level spectra of (a) P $2p$ and (b) O $1s$ obtained from all samples.

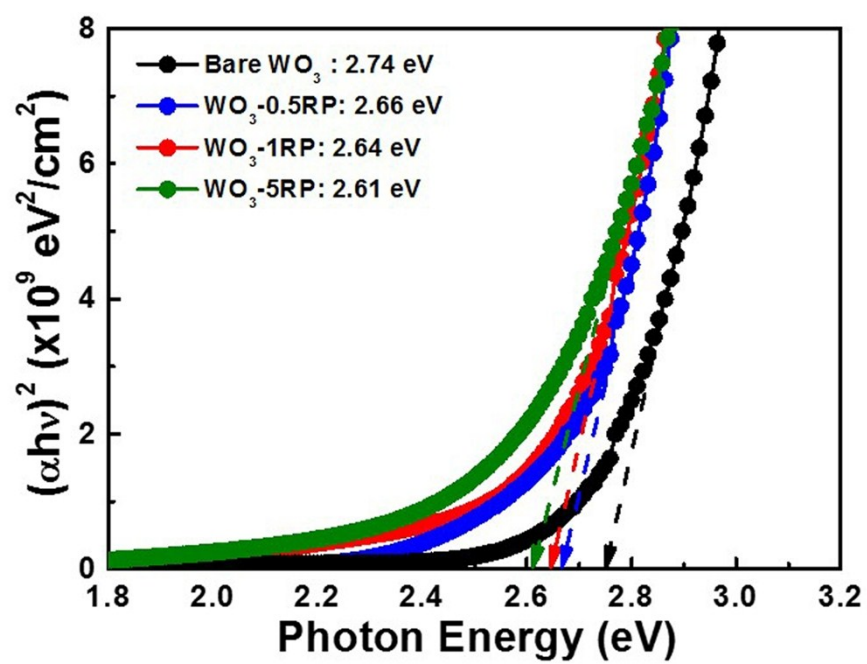


Fig. S5 Plots of $(\alpha h\nu)^2$ versus photon energy for all samples.

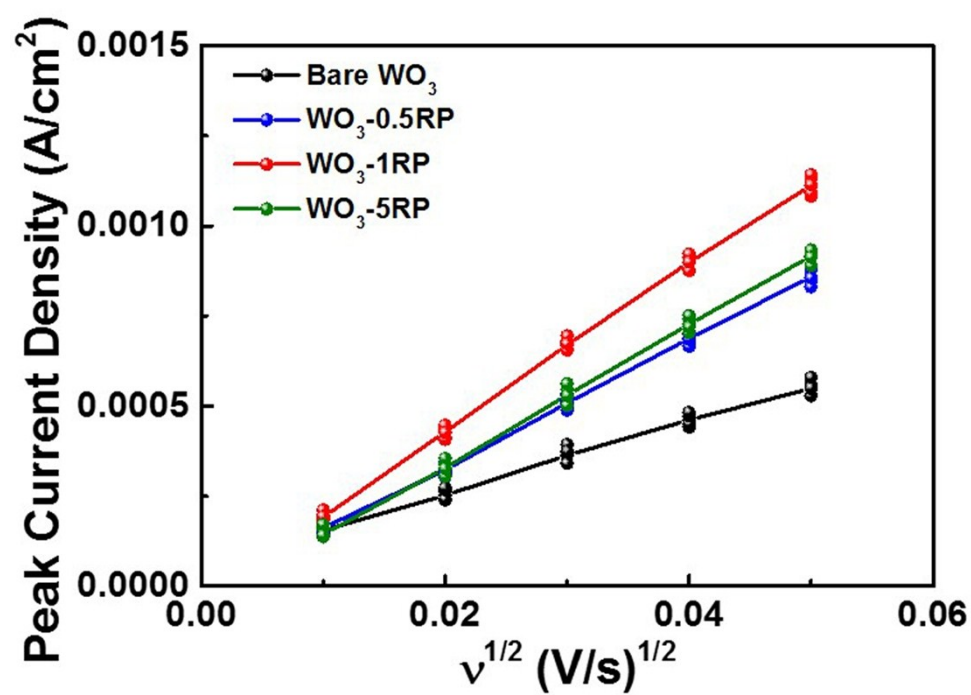


Fig. S6 Plots of peak current density as a function of the scan rates for all samples.

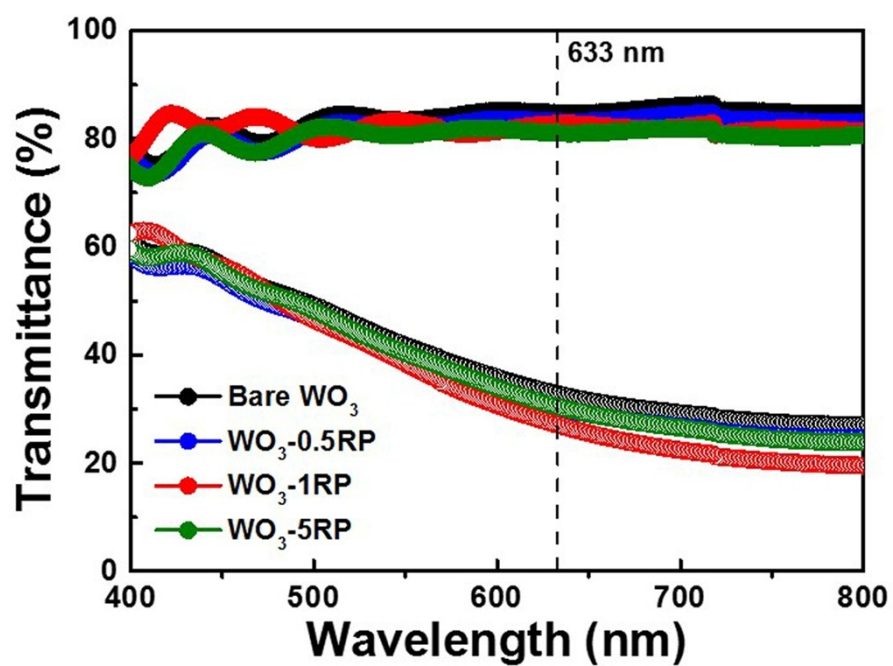


Fig. S7 Transmittance spectra of all samples measured at different working states (-0.7 V for coloration state (dotted line) and 1.0 V for bleached state (solid line)).

Table S1 EC performance comparison of previously reported WO₃-based materials.

Material	Coloration speed (s)	Bleaching speed (s)	CE (cm²/C)
Porous WO ₃ films ¹	37.0	5.0	36.0
Porous WO ₃ films ²	-	-	38.0
Macroporous WO ₃ films ³	5.1	8.7	50.1
Cylinder-like WO ₃ nanorod arrays ⁴	6.0	5.0	61.0
Mesoporous WO ₃ films ⁵	10.0	10.0	50.0
Disordered porous semicrystalline WO ₃ films ⁶	4.2	5.5	32.3
WO ₃ nanotree films ⁷	-	-	43.6
Vertically aligned Ni- doped WO ₃ films ⁸	7.8	6.0	60.5
Hydrogenated WO ₃ nanosheet ⁹	8.3	6.8	-
Stacked WO ₃ nanosheet ¹⁰	-	-	32.0
Our study	6.1	2.5	55.9

Notes and references

1. J. P. Cronin, D. J. Tarico, J. C. L. Tonazzi, A. Agrawal, S. R. Kennedy, *Sol. Energy Mater. Sol. Cells.*, 1993, **29**, 371.
2. C.-P. Li, C. A. Wolden, A. C. Dillon, R. C. Tenent, *Sol. Energy Mater. Sol. Cells.*, 2012, **99**, 50.
3. L. Yang, D. Ge, J. Zhao, Y. Ding, X. Kong, Y. Li, *Sol. Energy Mater. Sol. Cells.*, 2012, **100**, 251.
4. F. Zheng, W. Man, M. Guo, M. Zhang, Q. Zhen, *CrystEngComm.*, 2015, **17**, 5440.
5. C. P. Li, F. Lin, R. M. Richards, C. Engtrakul, A. C. Dillon, R. C. Tenent, C. A. Wolden, *Electrochem. Commun.*, 2012, **25**, 62.
6. F. Lin, J. Cheng, C. Engtrakul, A. C. Dillon, D. Nordlund, R. G. Moore, T.-C. Weng, S. K. R. Williams, R. M. Richards, *J. Mater. Chem.*, 2012, **22**, 16817.
7. J. P. Cronin, D. J. Tarico, J. C. L. Tonazzi, A. Agrawal, S. R. Kennedy, *Sol. Energy Mater. Sol. Cells.*, 1994, **29**, 371.
8. J. Zhou, Y. Wei, G. Luo, J. Zheng, C. Xu *J. Mater. Chem. C*, 2016, **4**, 1613.
9. X. Zhou, X. Zheng, B. Yan, T. Xu, Q. Xu, *Appl. Surf. Sci.*, 2017, **400**, 6583.
10. K. Wang, P. Zeng, J. Zhai, Q. Liu, *Electrochem. Commun.*, 2013, **26**, 1203.