

Electronic Supplementary Information

Au decoration of graphene microchannel for self-activated chemoresistive flexible gas sensors with substantially enhanced response to hydrogen

Yeonhoo Kim,^{ab} Yong Seok Choi,^c Seo Yun Park,^a Taehoon Kim,^a Seung-Pyo Hong,^a Tae Hyung Lee,^a Cheon Woo Moon,^d Jong-Heun Lee,^e Donghwa Lee,^{*f} Byung Hee Hong,^{*c} and Ho Won Jang^{*a}

^aDepartment of Materials Science and Engineering, Research Institute of Advanced Materials, Seoul National University, Seoul 08826, Republic of Korea. E-mail: hwjang@snu.ac.kr

^bCenter for Integrated Nanotechnologies, Los Alamos National Laboratory, Los Alamos, NM 87544, United States of America.

^cDepartment of Chemistry, Seoul National University, Seoul 08826, Republic of Korea. E-mail: byunghee@snu.ac.kr

^dDepartment of Chemistry and Nanoscience, Ewha Womans University, Seoul 03760, Republic of Korea.

^eDepartment of Materials Science and Engineering, Korea University, Seoul 02841, Republic of Korea.

^fDepartment of Materials Science and Engineering, and Division of Advanced Materials Science, Pohang University of Science and Technology, Pohang, Gyeongbuk 37673, Republic of Korea. E-mail: donghwa96@postech.ac.kr

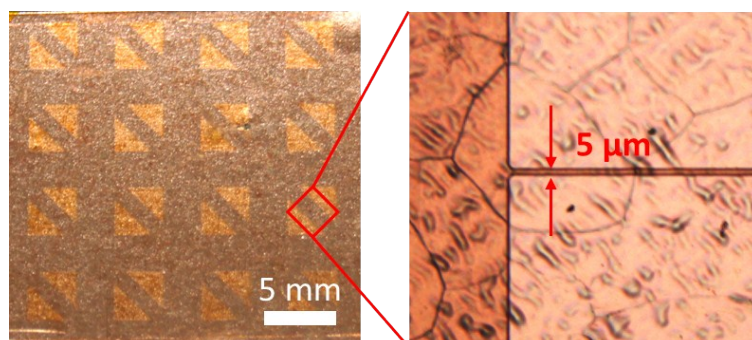


Figure S1. Micro patterning of 3LG grown on a Cu foil.

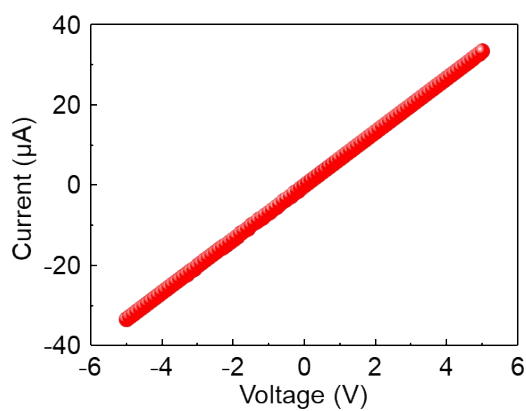


Figure S2. I-V characteristic of Au decorated graphene sensor

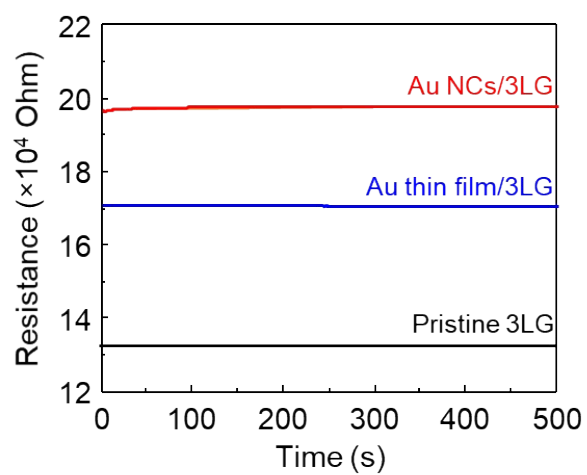


Figure S3. Resistance of pristine 3LG, Au thin film/3LG (before formation of Au NCs), and Au NCs/3LG.

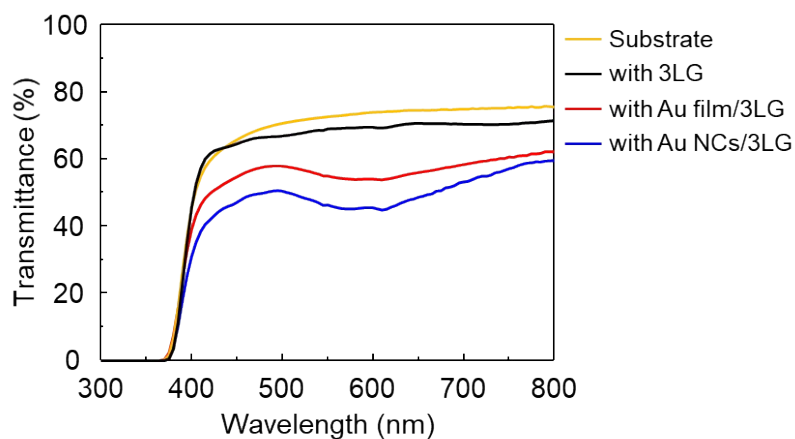


Figure S4. UV-vis transmittance spectra of the PI substrate, the substrate with 3LG, Au film/3LG (before formation of Au NCs), and Au NCs/3LG.

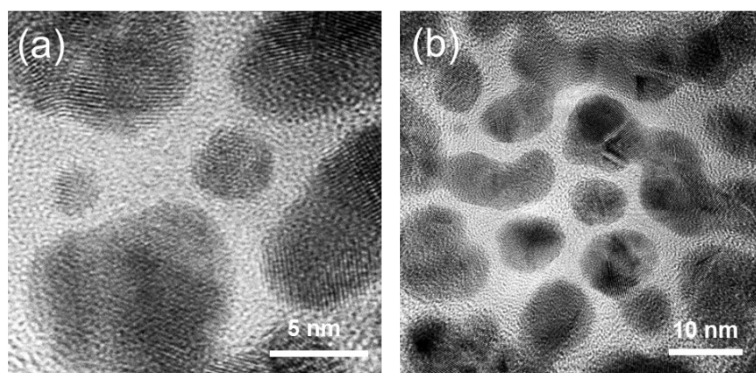


Figure S5. TEM images of Au decorated 3LG at different magnifications.

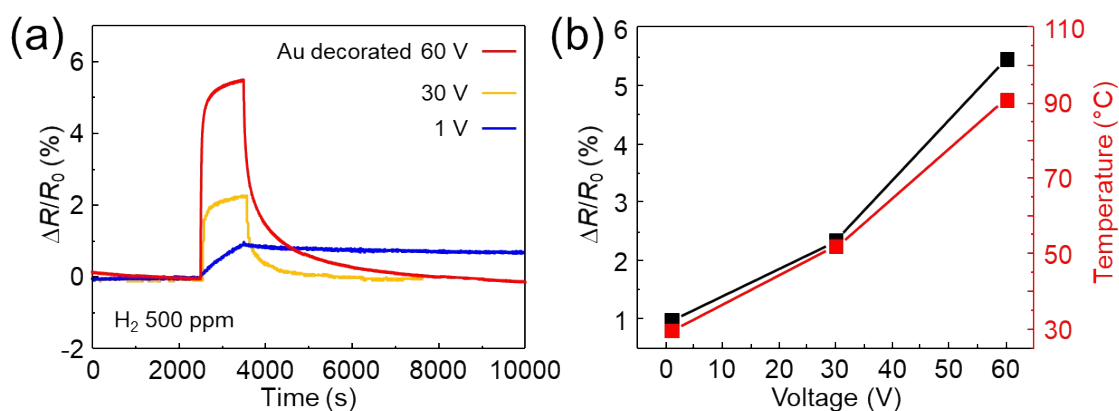


Figure S6. (a) H₂ gas sensing curves at different voltages. (b) Relationship between gas responses and temperatures according to the applied bias voltages.

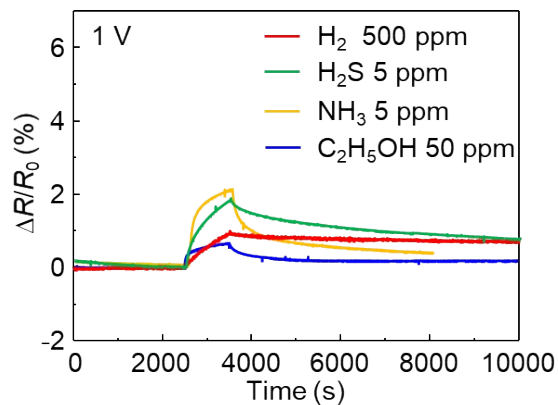


Figure S7. Sensing curves of the Au decorated graphene sensor at 1 V

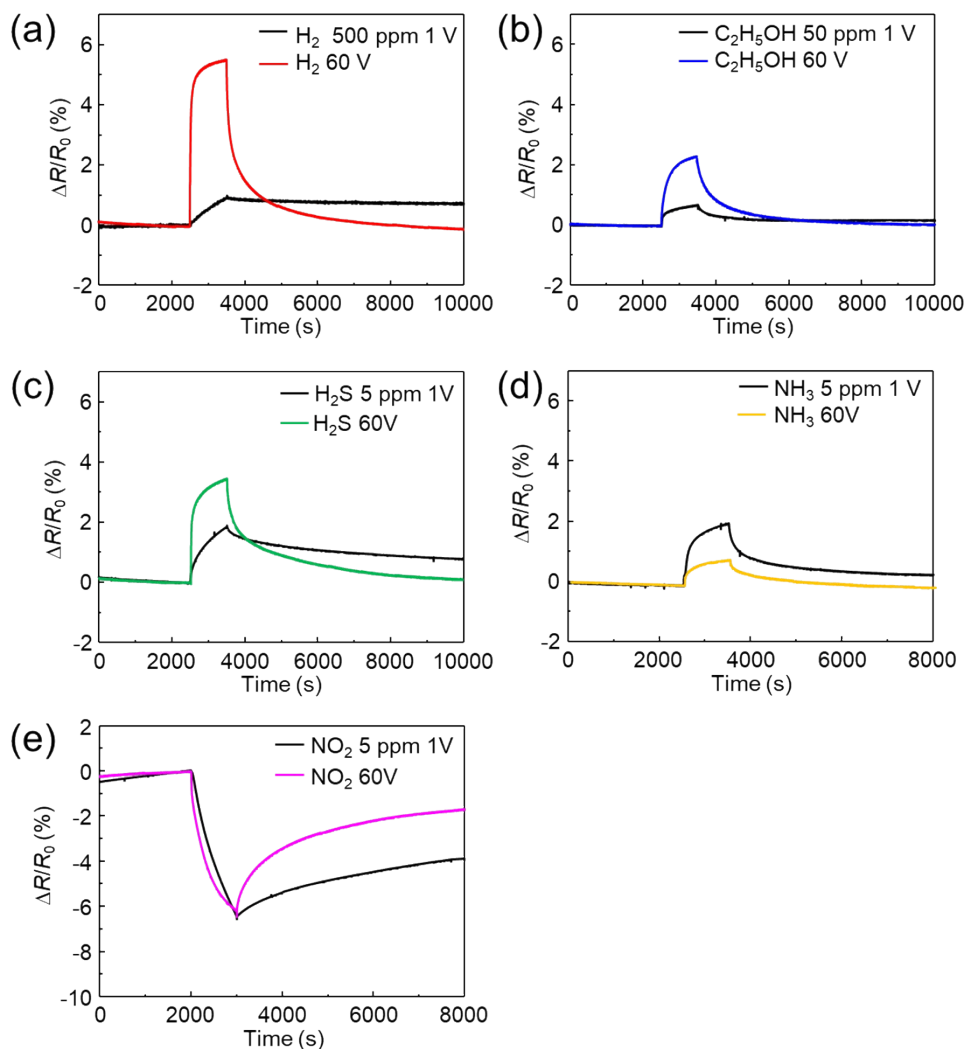


Figure S8. Comparison of the sensing properties of Au decorated graphene sensor at bias voltage of 1 and 60 V to H₂, C₂H₅OH, H₂S, NH₃, and NO₂ gas.

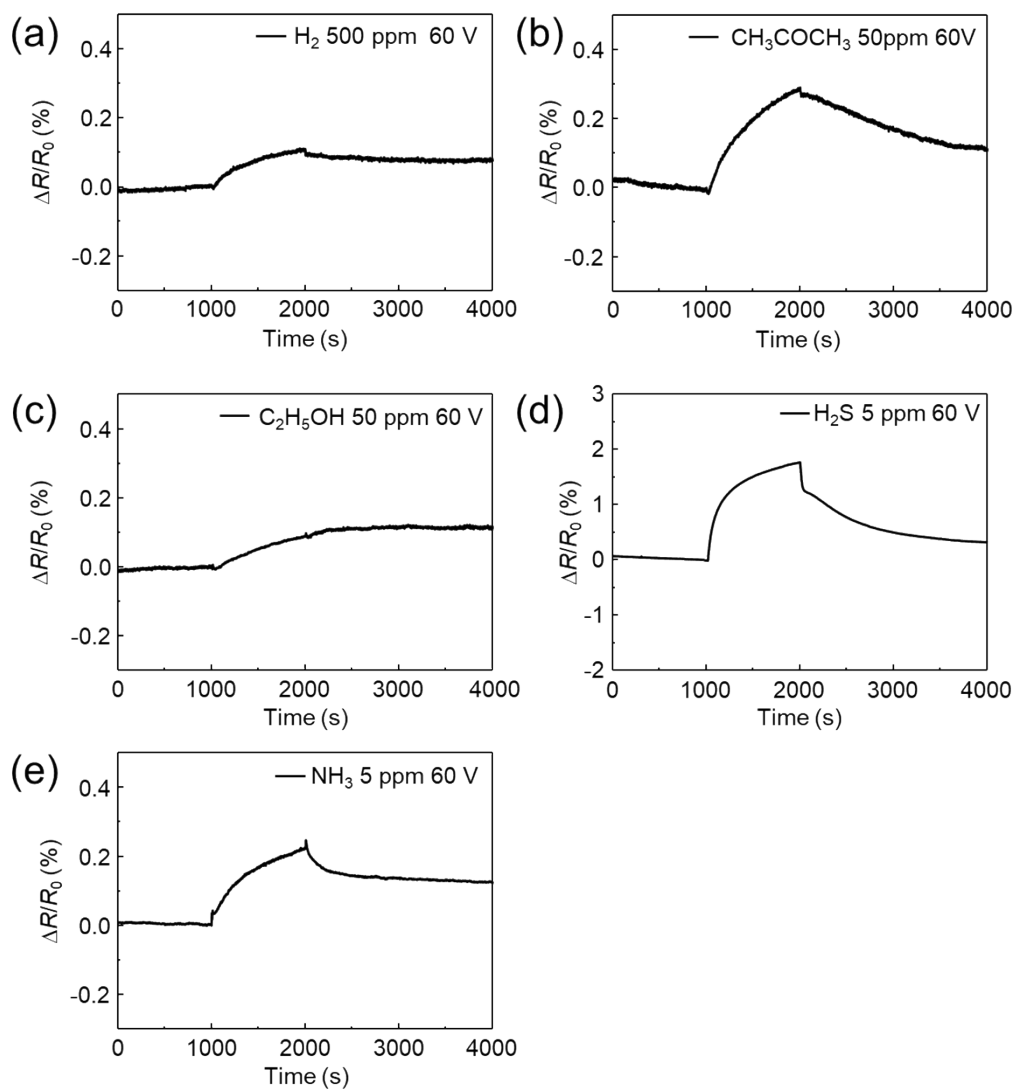


Figure S9. Response curves of pristine graphene sensor upon exposure to different gases.

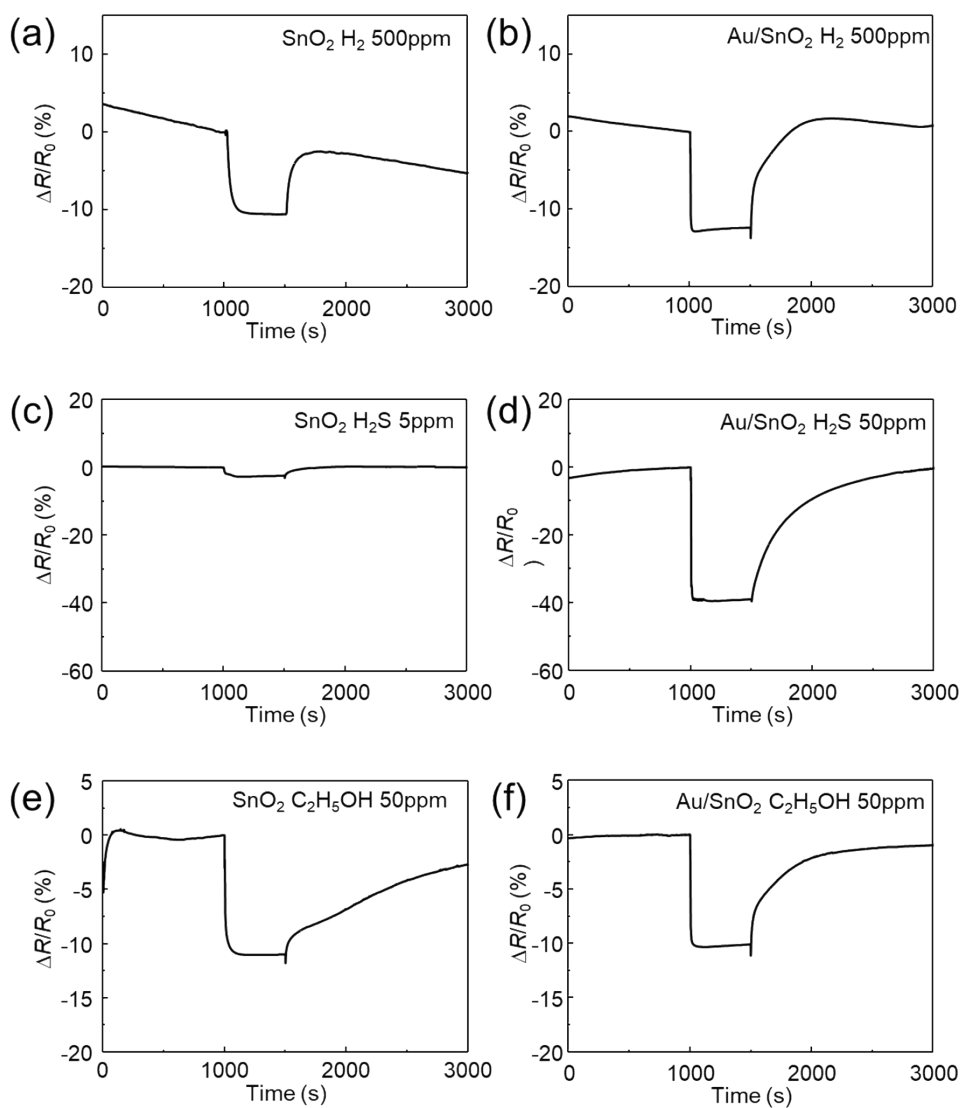


Figure S10. Response curves of SnO₂ thin film sensor and the sensor with Au decoration upon exposure to H₂, H₂S, and C₂H₅OH.

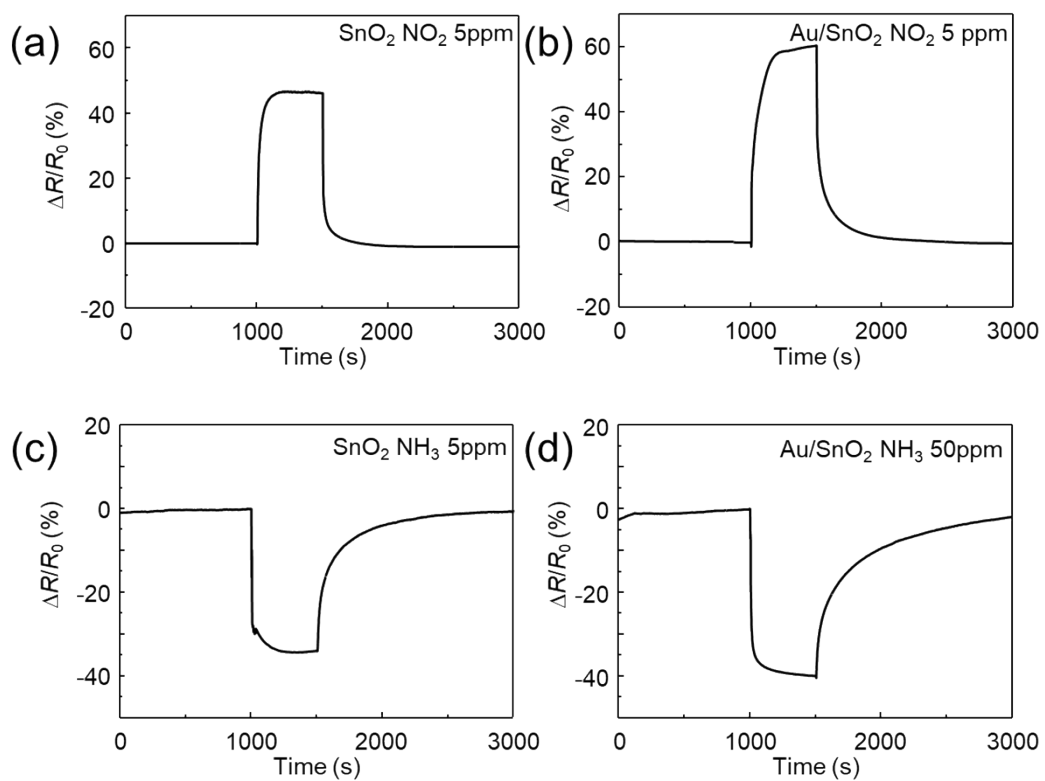


Figure S11. Response curves of SnO₂ thin film sensor and the sensor with Au decoration upon exposure to NO₂, and NH₃

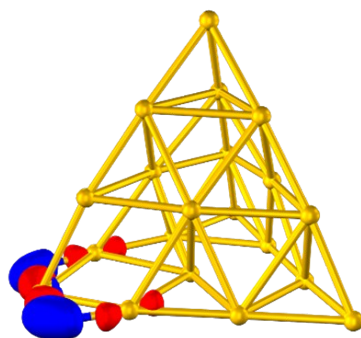


Figure S12. Variation of charge density after 2H adsorption; blue and red colors show the regions where charge density is increased and decreased the adsorption.

Sensing materials	Temp. (°C)	H ₂ concentration (ppm)	Ambient	Gas response	Response time (s)	Recovery time (s)	Flexibility & transparency	Ref.
Au/graphene	27	500	Air	~6 %	16	274	Flexible and transparent	This work
Pt/CNT	27	40000	Air	~16 %	~540	~5000	Non	20
Pt/rGO	27	1000	N ₂	~100 %	~600	~700	Non	23
Pt-Pd composites	22	20000	Ar	~4.1%	~1	~10	Non	28
Pd/graphene	27	500	Air	~4 %	~250	~250	Non	50
Pd/GNR	27	500	N ₂	~2.5 %	~30	~90	Non	51
Pd/graphene	27	250	Air	~3.5 %	~100	~330	Non	52

Table S1. Comparison with hydrogen gas sensing properties of noble metal decorated graphene-based sensors in previous literatures.^{20, 23, 28, 53-56}