

Electronic Supplemental Information

Anisotropic Manganese Antimonide Nanoparticle Formation by Solution-Solid-Solid Growth Mechanism : Consequence of Sodium Borohydride Addition Towards Reduced Surface Oxidation and Enhanced Magnetic Moment

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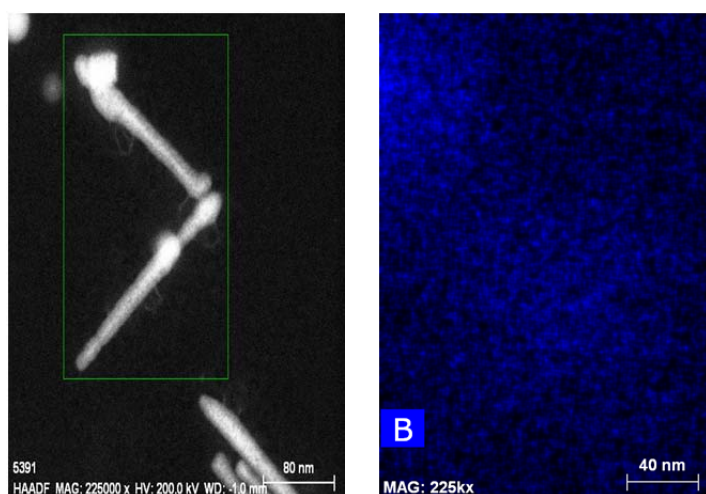


Figure S1: HAADF-STEM of MnSb nanoparticles synthesized in the presence of NaBH₄ and EDS mapping for B. The scale bar corresponds to 40 nm.

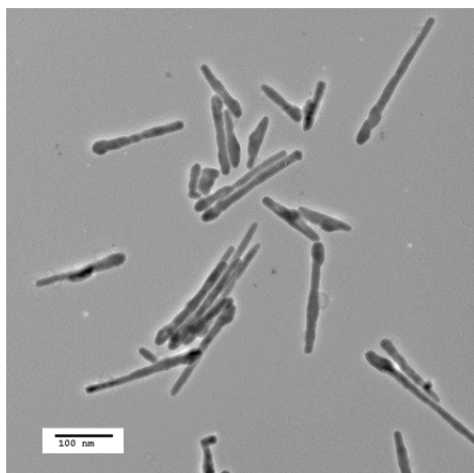


Figure S2: TEM image of MnSb nanoparticles reminiscent of peas in a pod synthesized in the presence of 2.64 mmol NaBH₄

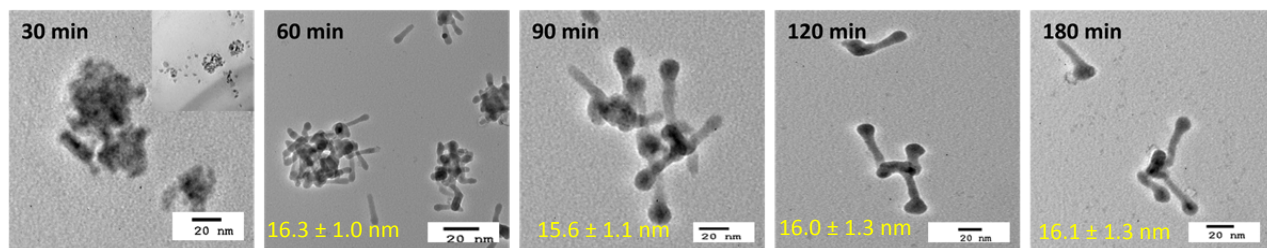


Figure S3 (a): TEM images of the fractions obtained at different time intervals (30, 60, 90, 120, 180 min) and quenched in cold chloroform from the reaction using near-stoichiometric $\text{Mn}_2(\text{CO})_{10}$ and 0.53 mmol NaBH_4

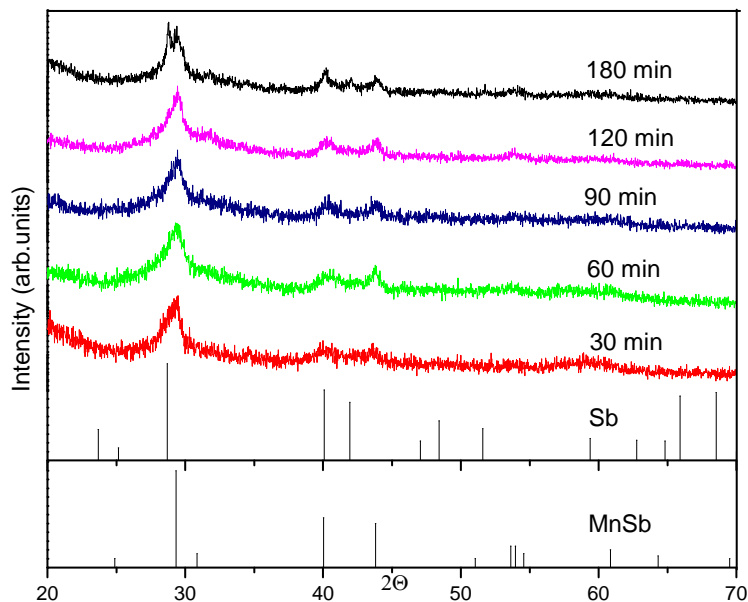


Figure S3(b): PXRD patterns of the fractions obtained at different time intervals (30, 60, 90, 120, 180 min) and quenched in cold chloroform from the reaction that used near-stoichiometric $\text{Mn}_2(\text{CO})_{10}$ in the presence of NaBH_4 (0.53 mmol)

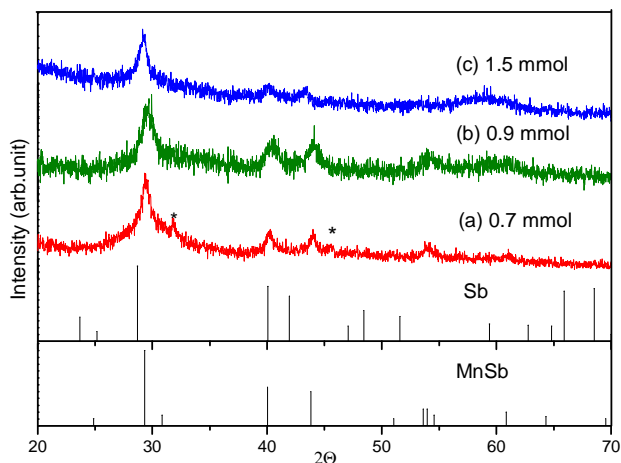


Figure S4 (a-c): PXRD patterns of MnSb nanoparticles synthesized in the presence of NaBH_4 (0.79 mmol), using (a) 0.7 (b) 0.9 (c) 1.5 mmol (40, 80, 200% mole excess) of $\text{Mn}_2(\text{CO})_{10}$. MnO_2 peaks are marked with asterisks (PDF #00-42-1169)

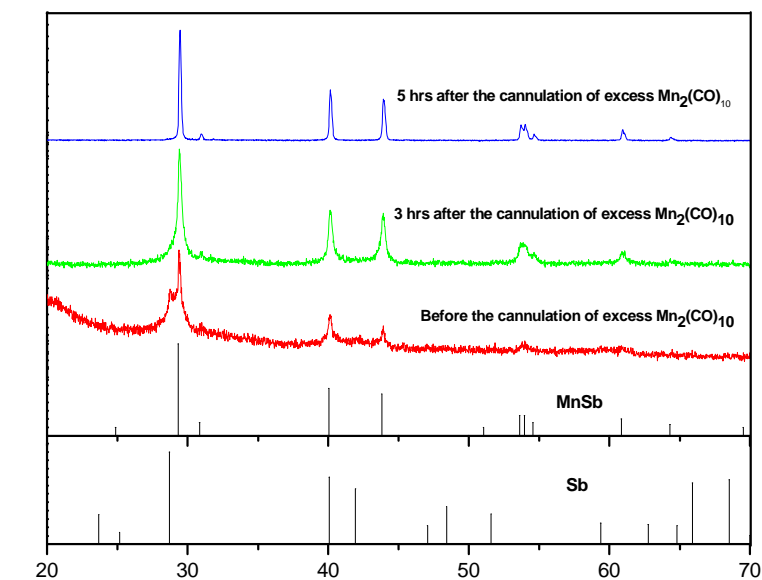


Figure S5: PXRD patterns of near-stoichiometric $\text{Mn}_2(\text{CO})_{10}$ containing reaction mixture and upon the introduction of excess $\text{Mn}_2(\text{CO})_{10}$ to the near-stoichiometric $\text{Mn}_2(\text{CO})_{10}$ containing the reaction mixture after 3 and 5 h carried out in the presence of 1.59 mmol NaBH_4 .

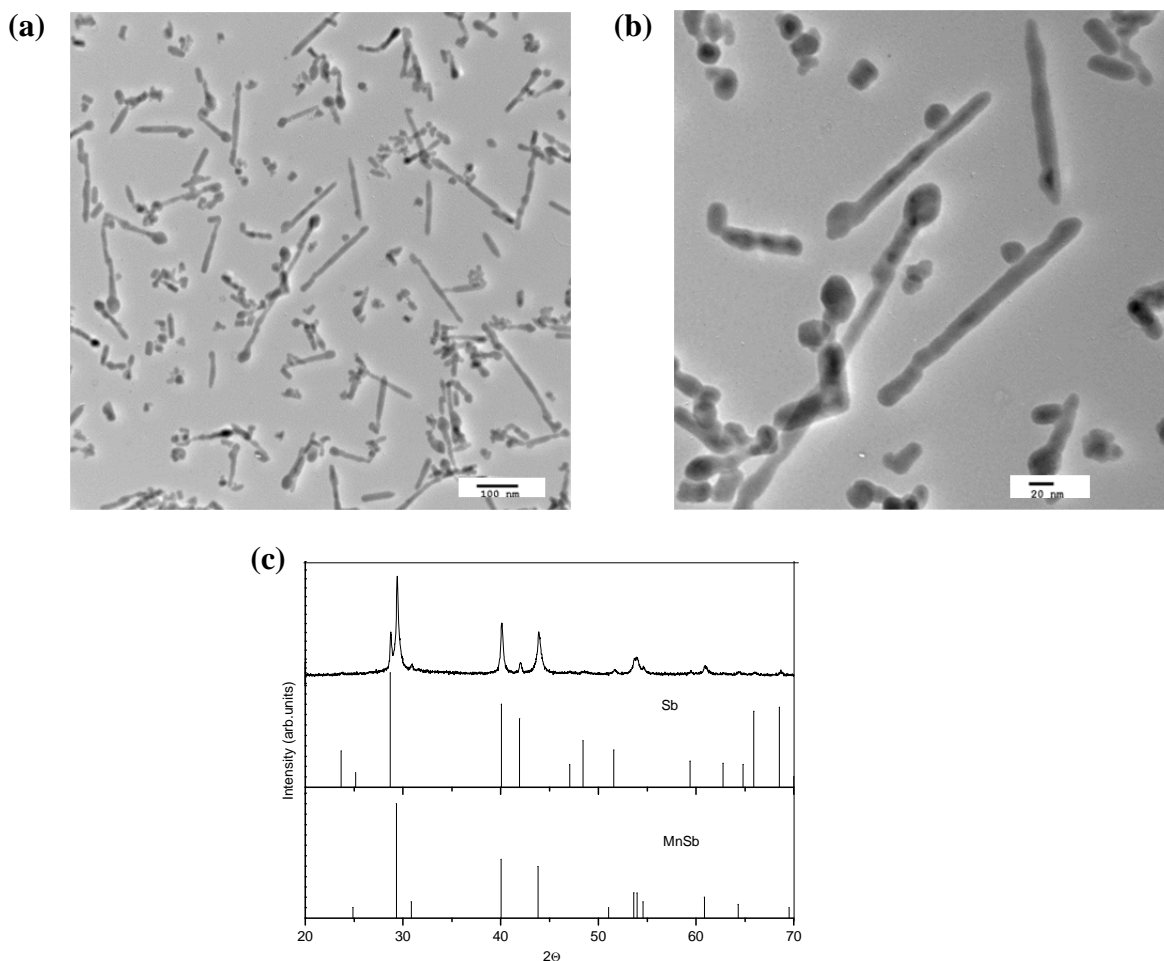


Figure S6: (a) Low magnification and (b) high magnification TEM images (c) PXRD pattern of the MnSb nanoparticles synthesized under near stoichiometric $\text{Mn}_2(\text{CO})_{10}$ amount with NaBH_4 (1.59 mmol) after heating for a total of 8 h