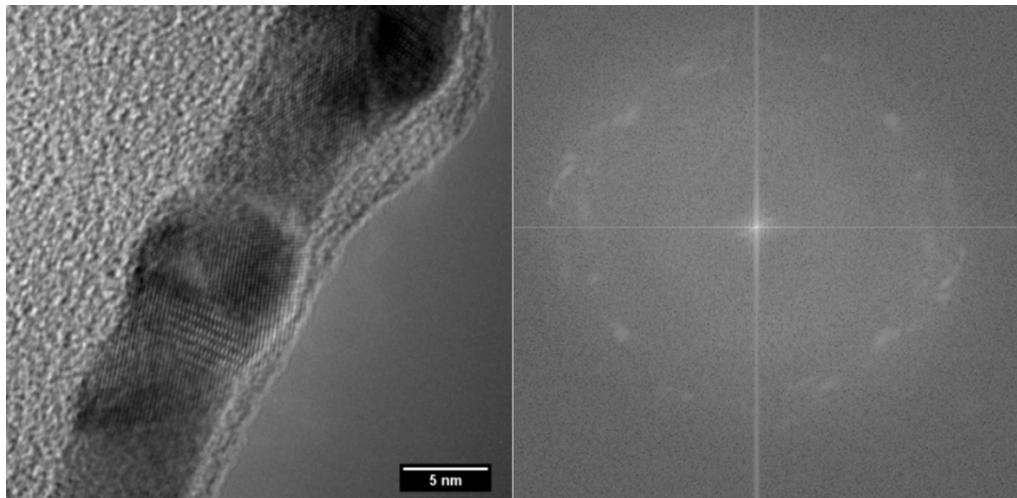


Supplementary Information

Tunable synthesis of water soluble ultrathin AuAg nanowires and their catalytic applications

Daniel K. Kehoe, Sarah A. McCarthy and Yurii K. Gun'ko*



S1: HR-TEM (left) and FFT (right) of ultrathin AuAg NWs produced following dilution after 1 day of aging.

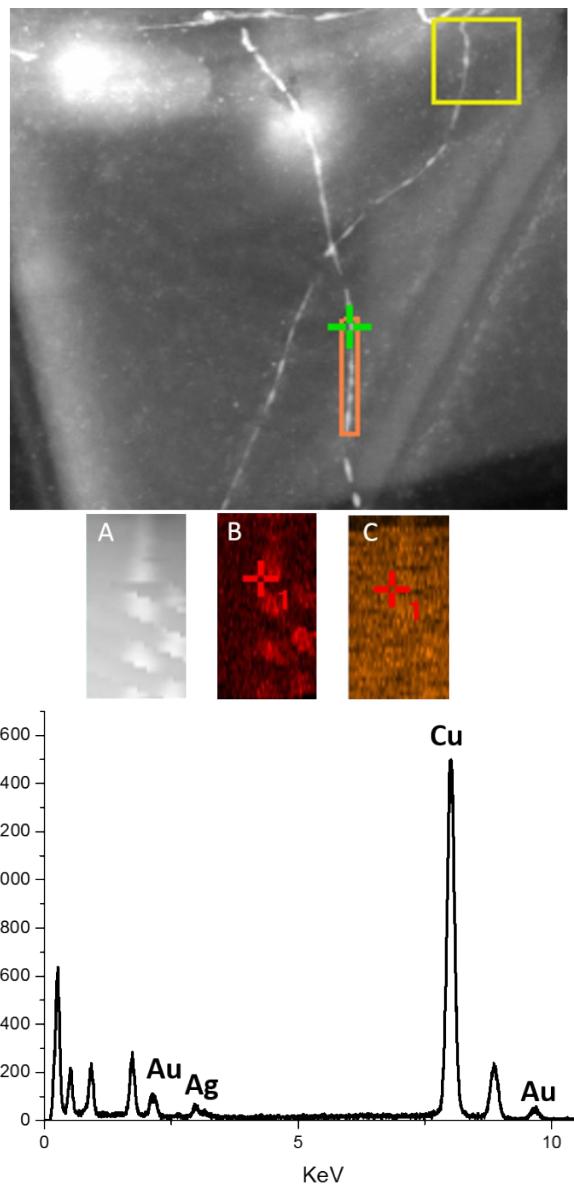


Figure S2: HAADF STEM (A) and EDX map of pre-diluted AuAg NWs were and B and C denotes the L-line of Ag and M-line of Au respectively. The associated EDX spectrum (bottom) is also detailed.

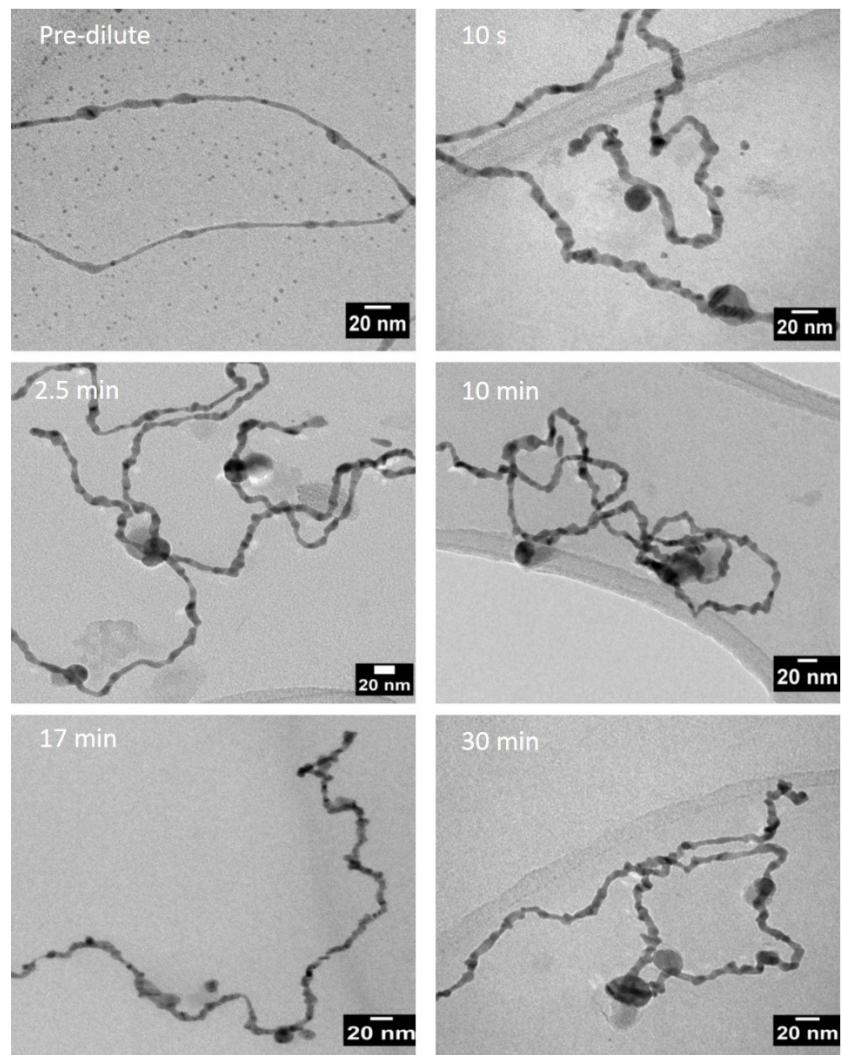


Figure S3: TEM analysis monitoring the growth of AuAg nanowires following dilution with water after 18 hrs of aging.

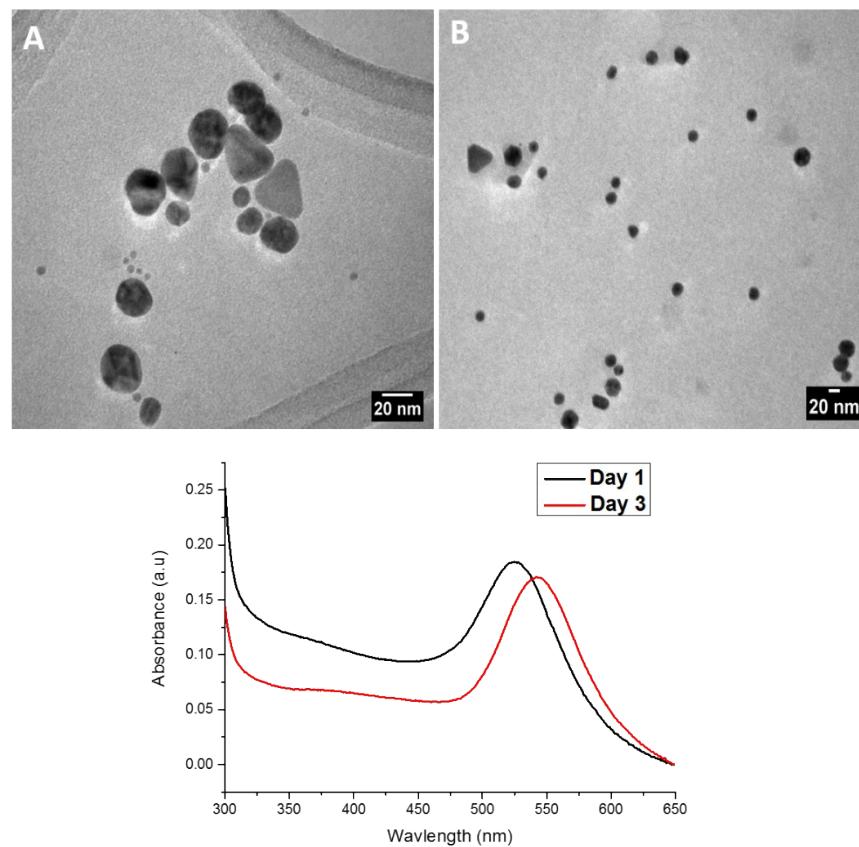


Figure S4: TEM image of Au nanoparticles after 1 (A) and 3 (B) days aging following dilution and UV-Vis spectra of products produced following dilution after each day.

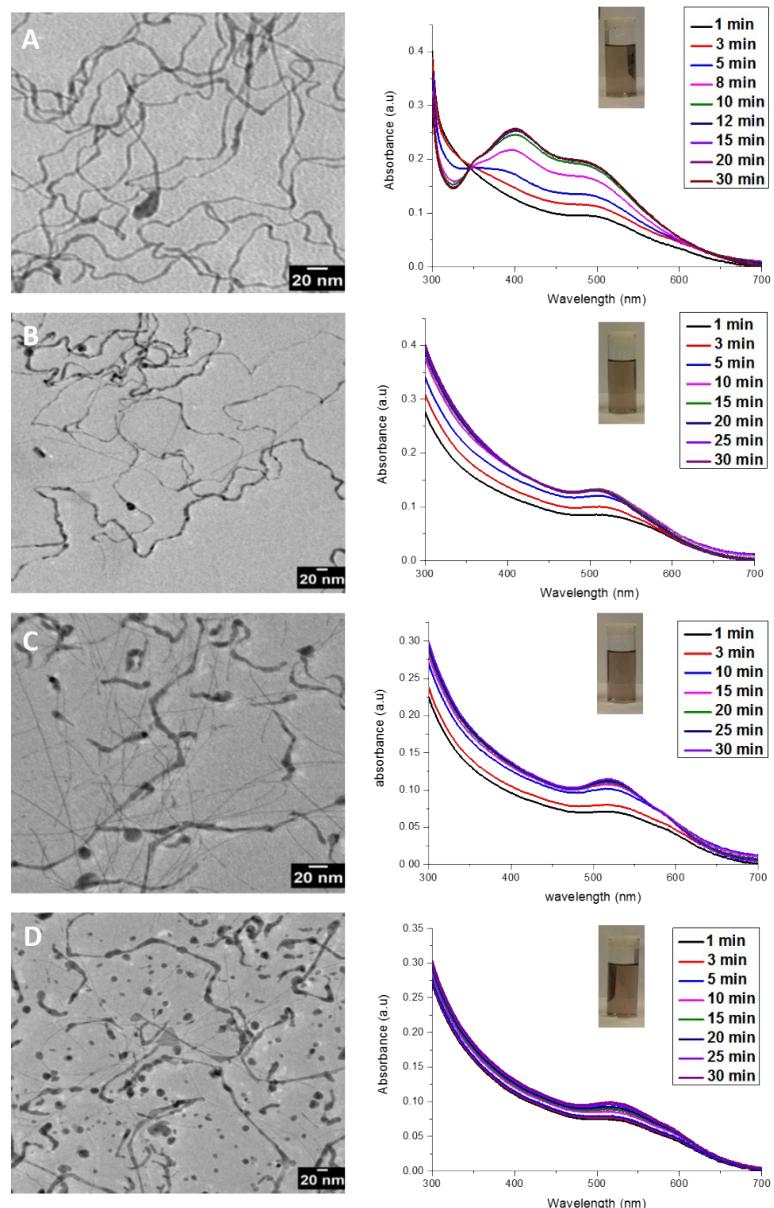


Figure S5: TEM images (left) and UV-Vis (right) analysis of nanowires produced following dilution with water (A), methanol (B), ethanol (C) and isopropanol (D) after 2 days of aging. Photograph inset of final solution.

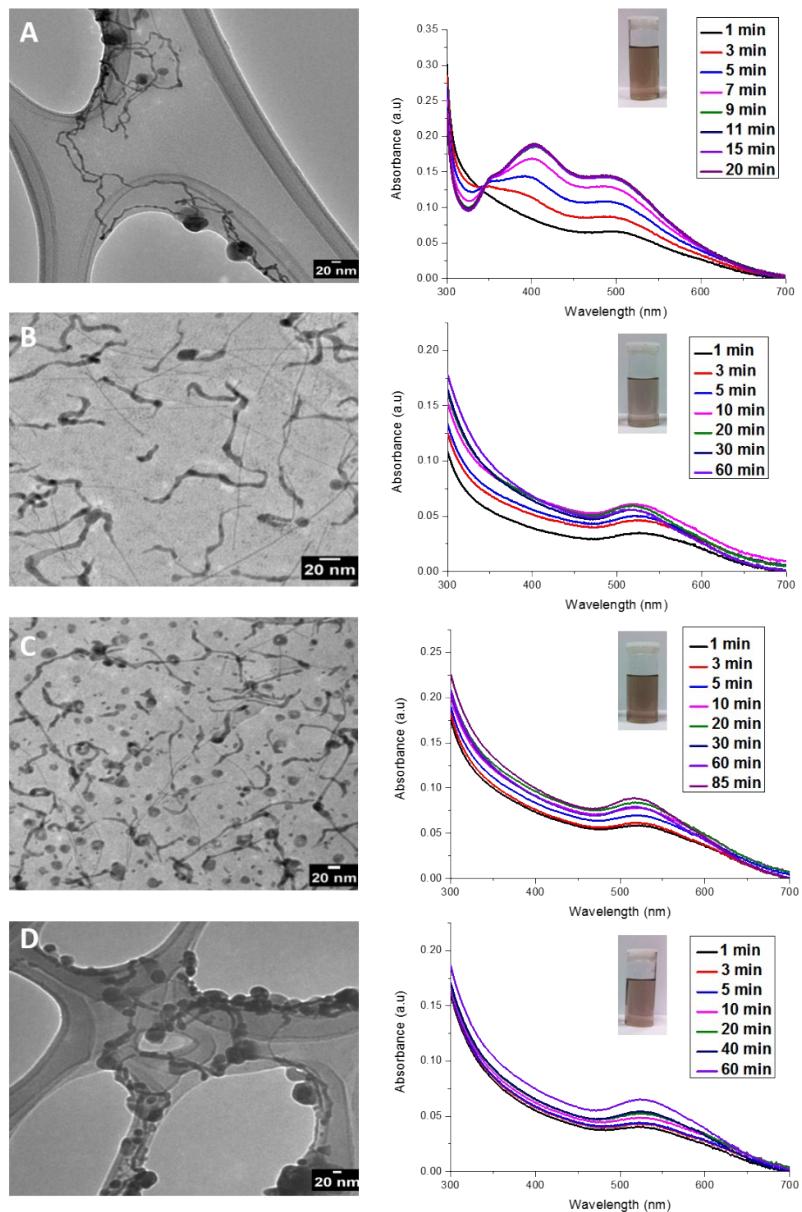


Figure S6: TEM images (left) and UV-Vis (right) analysis of nanowires produced following dilution with water (A), methanol (B), ethanol (C) and isopropanol (D) after 3 days of aging. Photograph inset of final solution.

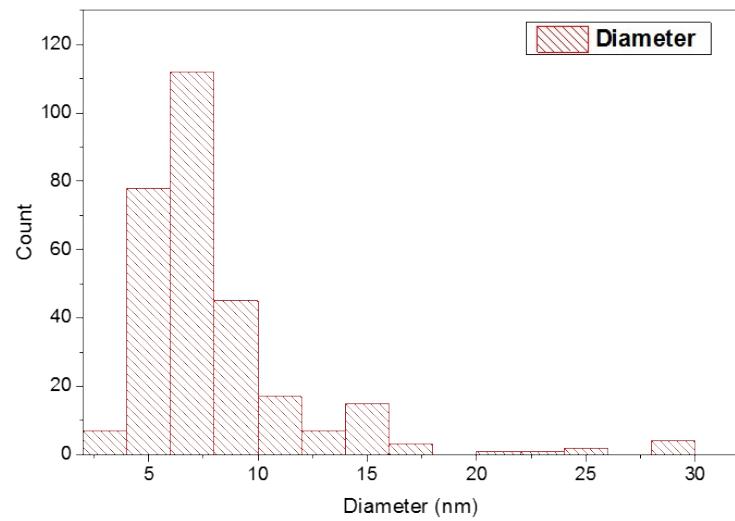


Figure S7: Size distribution of ultrathin AuAg NWs produced using a dilution factor of 5 after 1 day of aging with average diameter of 8 nm.

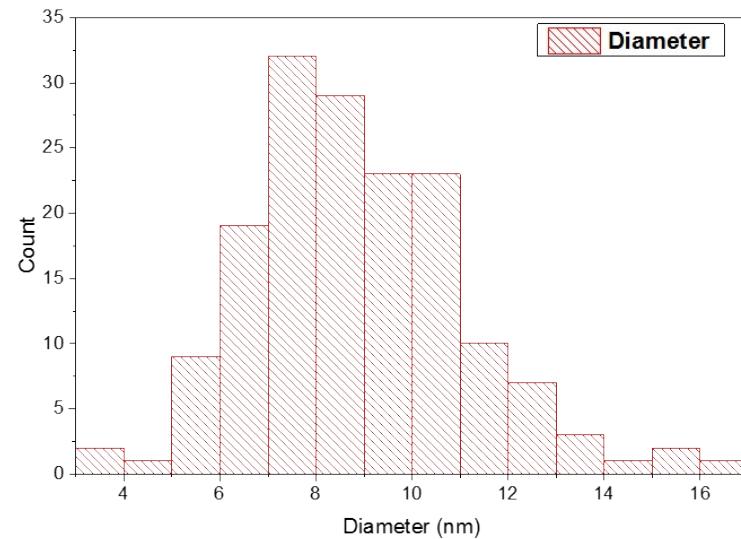


Figure S8: Size distribution of ultrathin AuAg NWs produced using a dilution factor of 10 after 1 day of aging with average diameter of 8.8 nm.

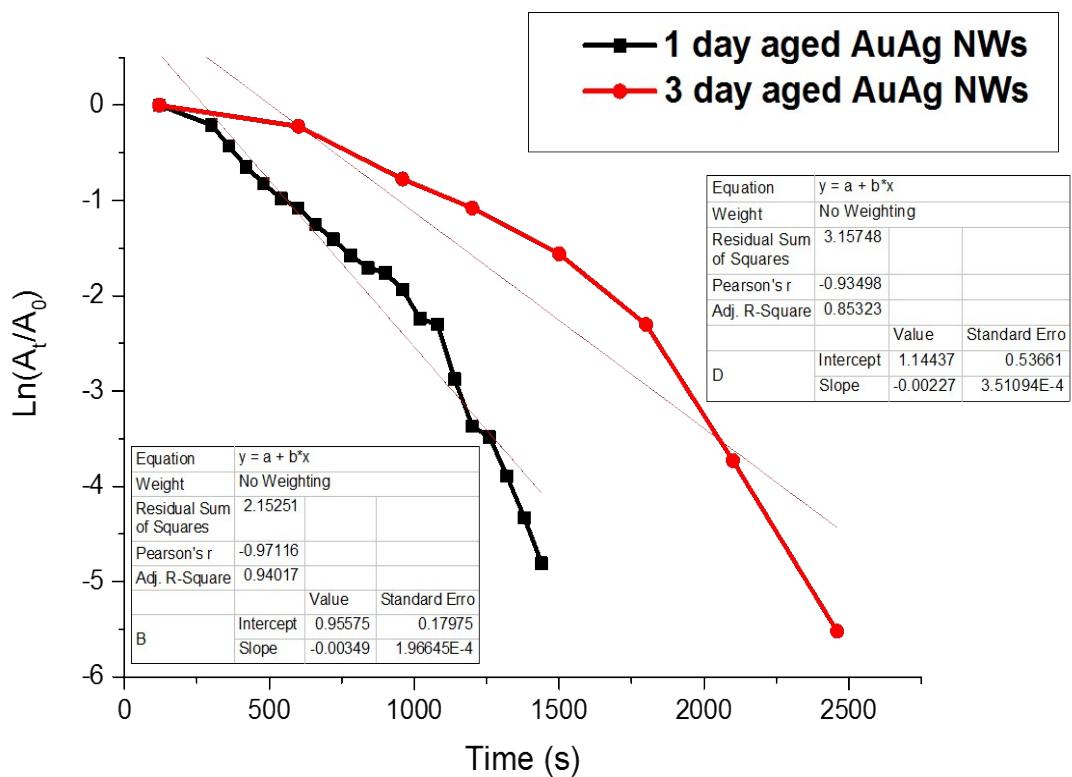


Figure S9: kinetic curves for the rate of reduction 4-nitrophenol using ultrathin AuAg nanowires produced after 1 day and 3 days at 400 nm detailing linear fitting.

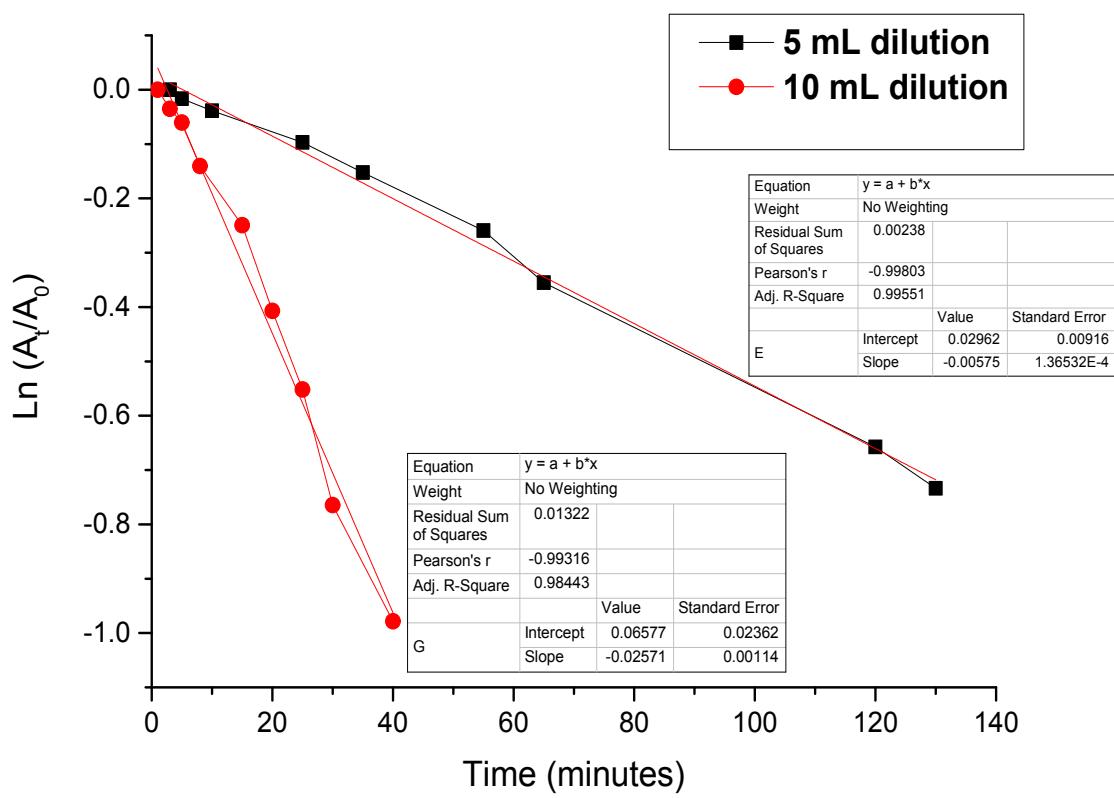


Figure S10: Kinetic curves for rate of formation of AuAg nanowires following dilution with 5 and 10 mL at 324 nm detailing linear fitting.

Table S1: Catalytic reduction of 4-Nitrophenol by various noble metal nanomaterials

Catalyst	Rate constant (10^{-3} s^{-1})	Catalyst loading
Ag NW @ Au NP ¹ 0.1% Au/Ag	0.87	0.5 mg mL ⁻¹
1% Au/Ag	1.4	0.5 mg mL ⁻¹
5% Au/Ag	3.4	0.5 mg mL ⁻¹
Ag NWs ¹ (30 nm)	0.47	0.45 mg mL ⁻¹
Au NWs (2 nm) on alumina beads ²	8.8	2.28 μg
Pt ₉₃ Bi ₇ NWs ³ (4.1 nm)	0.46	15 μg
Pt ₉₂ Bi ₈ NWs ³ (4.7 nm)	2.3	15 μg
Pt ₅₅ Pd ₃₈ Bi ₇ NWs ³ (4.4 nm)	4.3	15 μg
Au@Ag NP ⁴ (10 nm core)	0.09	-
Au NP (6 nm)@GO ⁵	2	-
Ag NP (7.5 nm)@GO ⁶	3.4	$1.8 \times 10^{-4} \text{ M}$
Au NP@Tween 80 ⁷ (14 nm)	1.5	1 mmol/L
Ag NP@SiO ₂ nanorattles ⁸	3.3	4 mg
Au@Ag NP on MOF ⁹	4.97	3.46 mg
AuAg (8 nm) (this work)	3.4	0.45 mg mL ⁻¹
AuAg (3nm) (this work)	2.2	0.45 mg mL ⁻¹

GO = graphene oxide NP = nanoparticle NW = nanowire MOF = metal organic framework

References

1. H. Fu, X. Yang, X. Jiang, A. Yu, *Langmuir : the ACS journal of surfaces and colloids*, 2013, **29**, 7134-7142.
2. S. Kundu, A. Leelavathi, G. Madras, N. Ravishankar, *Langmuir*, 2014, **30**, 12690-12695.
3. Y. Y. Shen, Y. Sun, L. N. Zhou, Y. J. Li, E. S. Yeung, *Journal of Materials Chemistry A*, 2014, **2**, 2977-2984.
4. K. K. Haldar, S. Kundu, A. Patra, *ACS Applied Materials & Interfaces*, 2014, **6**, 21946-21953.

5. Y. Choi, H. S. Bae, E. Seo, S. Jang, K. H. Park, B.-S. Kim, *Journal of Materials Chemistry*, 2011, **21**, 15431-15436.
6. T. Wu, L. Zhang, J. Gao, Y. Liu, C. Gao, J. Yan, *Journal of Materials Chemistry A*, 2013, **1**, 7384-7390.
7. P. Suchomel, L. Kvitek, R. Prucek, A. Panacek, A. Halder, S. Vajda, R. Zboril, *Scientific Reports*, 2018, **8**, 1-11.
8. J. Hou, B. Yu, E.-g. Liu, W.-b. Dong, P.-c. Lu, Z. Wang, V. C. Yang, J.-b. Gong, *RSC Advances*, 2016, **6**, 95263-95272.
9. H.-L. Jiang, T. Akita, T. Ishida, M. Haruta, Q. Xu, *Journal of the American Chemical Society*, 2011, **133**, 1304-1306.