

Supporting Information

Ligand dynamics control structure, elasticity, and high-pressure behavior of nanoparticle superlattices

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Force Field

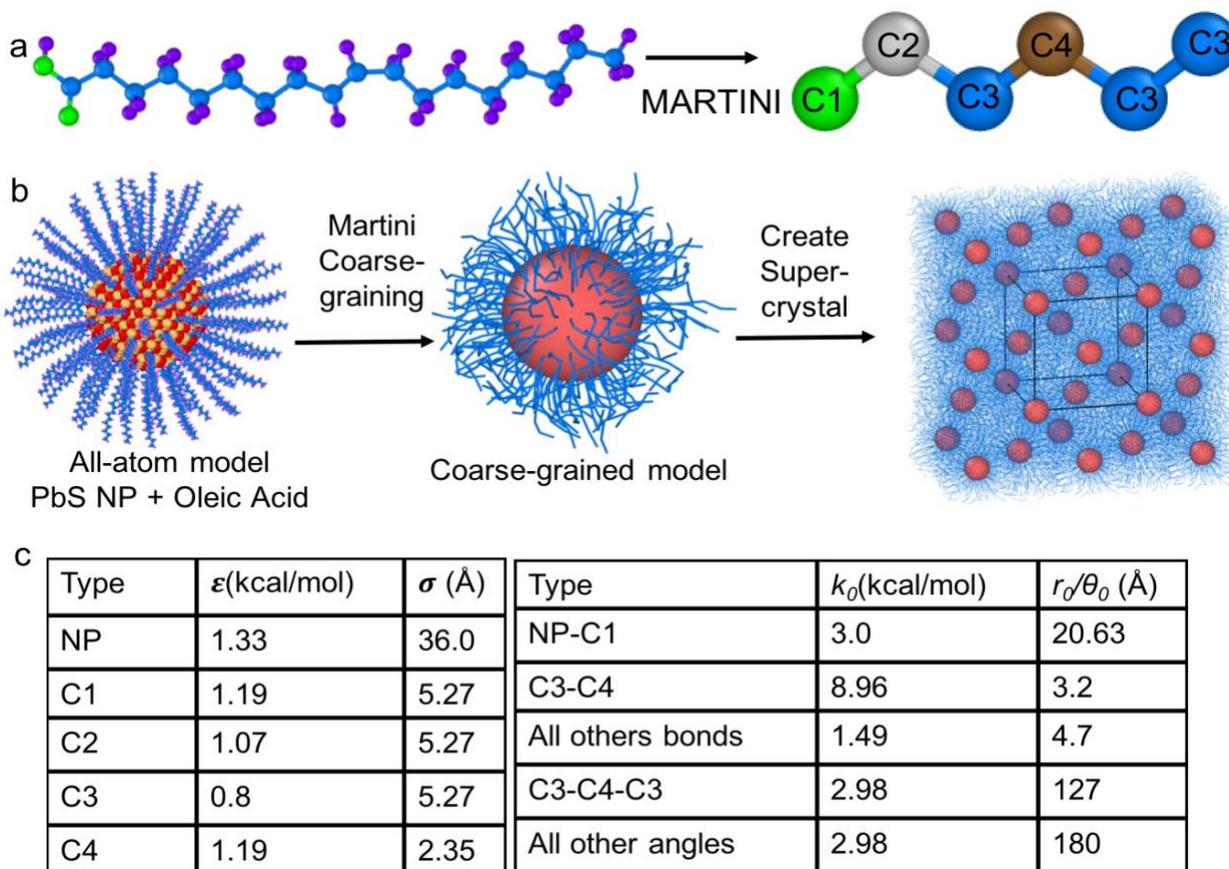


Figure S1. Details of coarse-grained model employed to simulate nanoparticle superlattices. Each PbS nanoparticle is modelled as an indivisible bead called NP. (a) MARTINI mapping scheme is used to obtain coarse-grained model (composed of beads C1-C4) from all-atom configuration of oleic acid. (b) Illustration of the steps used to prepare a CG configuration of NPSL. (c) MARTINI force field parameters used in this study are tabulated.

We employed the well-established MARTINI scheme to coarse-grain the oleic acid ligand molecules; each nanoparticle is modeled as an indivisible bead (Figure S1). The MARTINI force-field parameters are used to model interactions between different ligand molecules, and the nanoparticles (the parameters are summarized in Figure S1c). For unlike interactions, we employed Lorentz-Berthelot mixing rule where the LJ parameters for interactions between beads i and j are given by: $\epsilon_{ij} = (\epsilon_{ii}\epsilon_{jj})^{1/2}$, and

$$S_{ij} = \frac{1}{2}(S_{ii} + S_{jj}),$$

where, (ϵ_{ii}, S_{ii}) , and (ϵ_{jj}, S_{jj}) are the LJ parameters for i - i and j - j interactions respectively.

Equation of State (EOS)

We have computed the equation state of the NPSL from our coarse-grained simulation and compare with experiment. At each pressure, the system is equilibrated for 10ns in an isothermal isobaric ensemble at temperature $T = 300\text{K}$. The equilibrium volume is computed by averaging over 10000 frames collected over a period of 1 ns. The MARTINI force-field describes the equation of state of NPSLs in good qualitative accordance with experiments (Figure S2). This suggests that the coarse-grained model qualitatively capture the thermophysical properties of the experimental system.

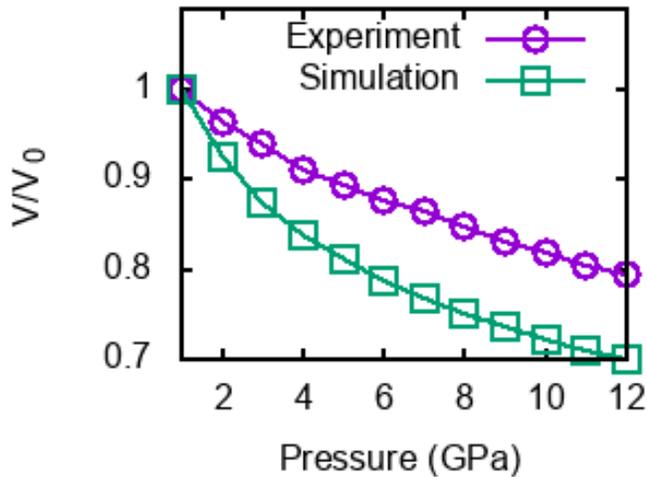


Figure S2. Equation of state of a ligand-rich nanoparticle superlattice ($\rho=4.9/\text{nm}^2$) from coarse-grained molecular dynamics and experiments at 300 K. The unit cell volume (V) at a given pressure is normalized by its value at pressure $P = 1\text{GPa}$ (V_0).

Radial Distribution Function (RDF)

Figure S3 shows the NP-NP radial distribution in NPSL (for various ligand coverage densities) at selected pressures during a compression-release cycles. It clearly suggests the NPSL sustain high applied pressure without losing their crystallinity at high ligand coverage densities ($\rho = 1.8 - 5.5 \text{ nm}^{-2}$).

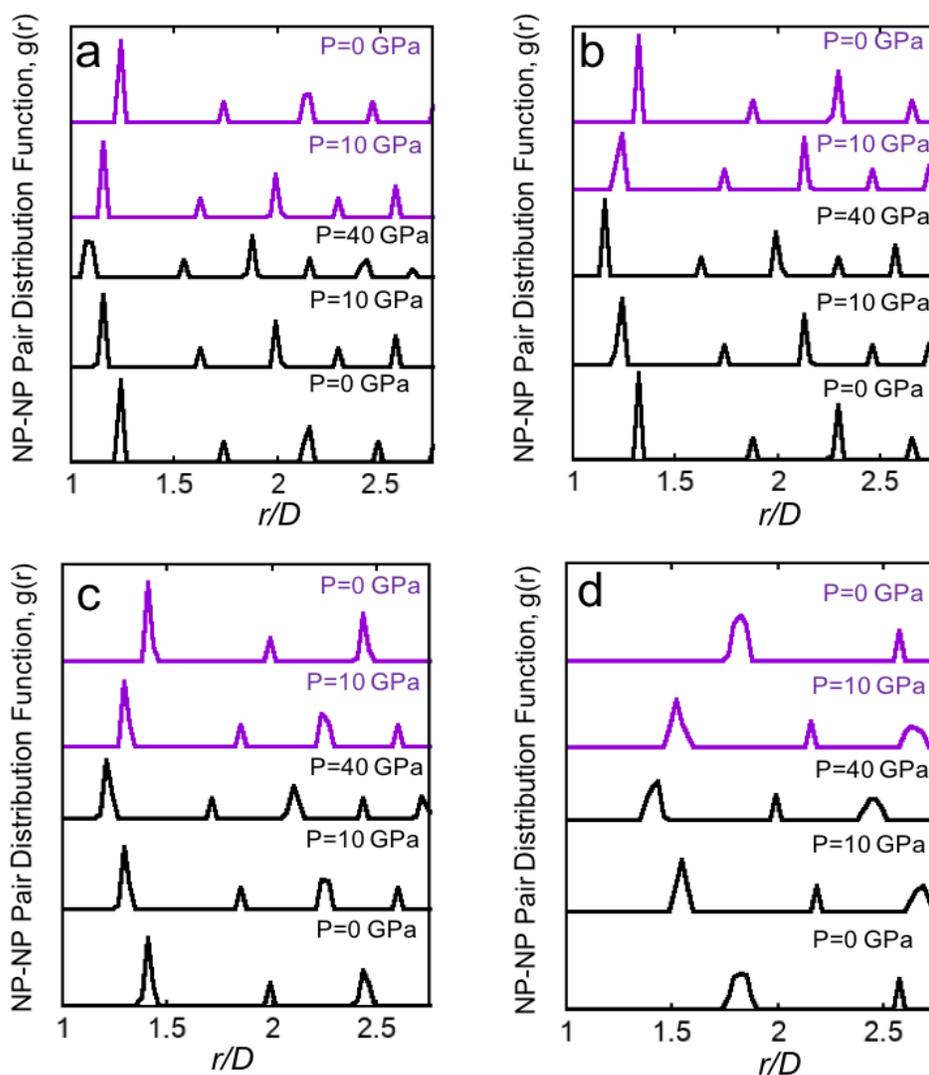


Figure S3. Radial distribution functions: a, b, c and d correspond to ligand density $\rho = 1.8 \text{ nm}^{-2}$, 2.4 nm^{-2} , 3.6 nm^{-2} and 5.5 nm^{-2} , respectively.

Thermogravimetric Analysis

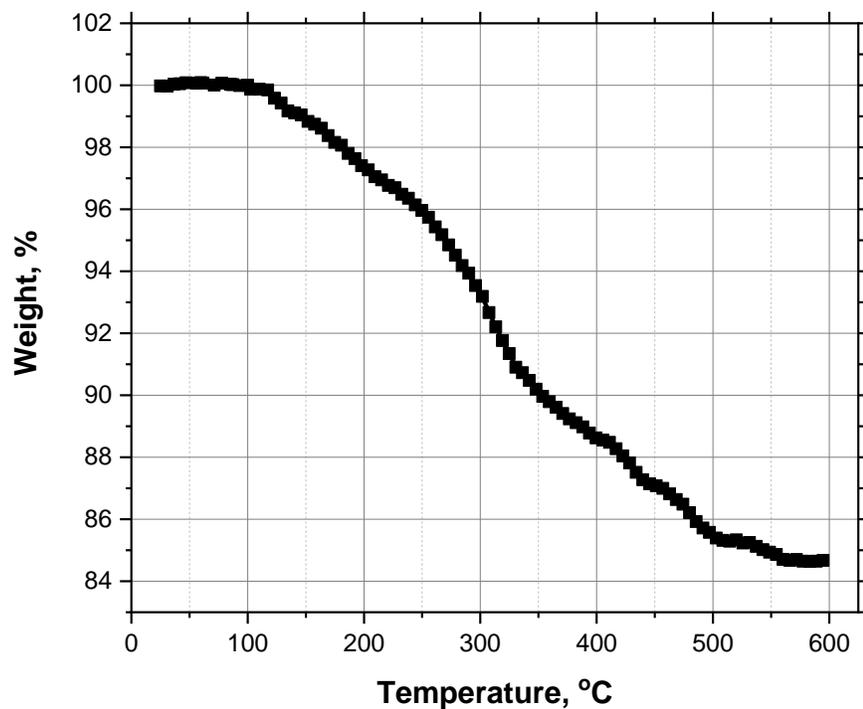


Figure S4. The thermogravimetric analysis (TGA) data obtained for the PbS NPSLs obtained by destabilization of the toluene solution of 7 nm PbS by isopropanol.

Transmission Electron Microscopy (TEM)

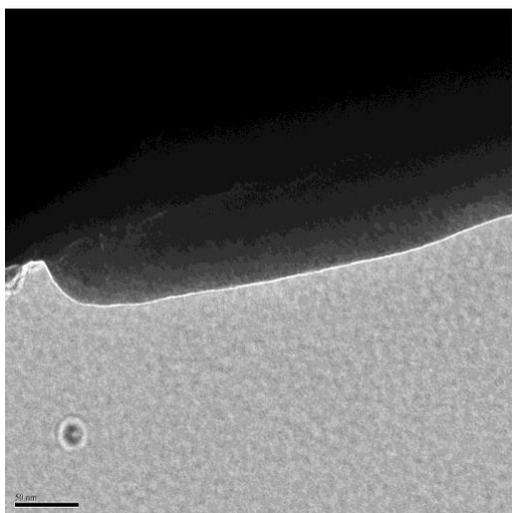


Figure S5. The TEM image of a fragment of PbS NPSL after compression.

Small-Angle X-ray Scattering (SAXS) experiments

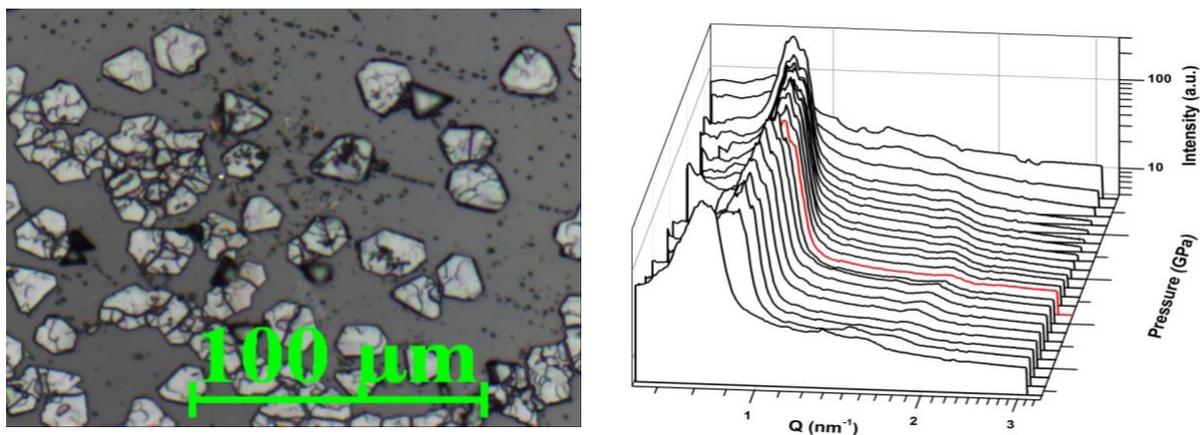


Figure S6. (Left) Optical micrograph and (Right) SAXS data on face-centered cubic NPSLs grown on Si substrate from 9.5 nm PbS NPs compressed up to 55.86 GPa and after pressure release.

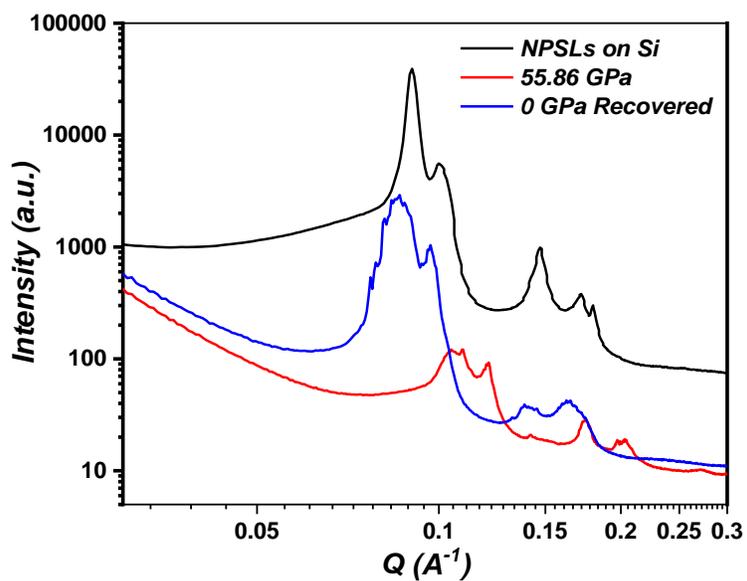


Figure S7. SAXS of face-centered cubic NPSLs (assembled from 9.5 nm PbS NPs) grown on Si substrate (black), and of individual NPSL at 55.86 GPa (red) and after pressure release (blue).