Supplementary Information

Effective blue-violet photoluminescence through the lanthanum and

fluorine ions co-doping in CsPbCl₃ perovskite quantum dots

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Fig. S1 The statistics on the dimensions of $CsPbCl_3$ and $CsPb(Cl/F)_3$ QDs with different Cl-to-F molar feed ratios; a, 1:0; b, 1:0.2 and c,1:0.8.



Fig. S2 EDX pattern of CsPb(Cl/F)₃ QDs prepared with Cl-to-F molar feed ratio of 1:0.8.



Fig. S3 XPS spectra for CsPbCl3 and CsPb(Cl/F)3 QDs prepared with Cl-to-F molarfeedratioof1:0.8.



Fig. S4 Excitation spectra of $CsPbCl_3$ and $CsPb(Cl/F)_3$ QDs under monitored wavelength of 410 nm.



Fig. S5 Low temperature fluorescence spectroscopy of $CsPbCl_3$ and $CsPb(Cl_{0.7}F_{0.3})_3$ QDs measured at 10K.



Fig. S6 The statistics on the dimensions of $CsPb(Cl_{0.7}F_{0.3})_3$:La³⁺ QDs prepared with La³⁺ ions doping concentration of 3.1, 7.2 and 14.4%.



Fig. S7 EDX pattern of $CsPb(Cl_{0.7}F_{0.3})_3$:La³⁺ QDs with the 14.4% La³⁺ ions doping concentration.



Fig. S8 The normalized fluorescent emission intensity as a function of UV irradiated time for the CsPbCl₃, CsPb(Cl_{0.7}F_{0.3})₃ and CsPb(Cl_{0.7}F_{0.3})₃:La³⁺QDs.

The 365 nm UV lamp with the power of 6W is used to detect the photoluminescence stability of three samples. After irradiating 48 h, the result shows that the photoluminescence stability of the perovskite NCs improved obviously as introducing of F- and La^{3+} ions.



Fig. S9 Excitation spectra of $CsPb(Cl_{0.7}F_{0.3})_3$:La³⁺ QDs under monitored wavelength of 410 nm.



Fig. S10 Low temperature fluorescence spectroscopy of $CsPb(Cl_{0.7}F_{0.3})_3$ and $CsPb(Cl_{0.7}F_{0.3})_3$:La³⁺ QDs measured at 10K.

Table S1	The lattice	lattice cons	tant of CsI	PbCl ₃ and	$CsPb(Cl/F)_3$	QDs with	n different
Cl-to-F m	olar feed ra	tios.					

	1:0	1:0.2	1:0.4	1:0.6	1:0.8	1:1
(110)	3.97	4.00	4.00	4.00	4.01	4.01
(200)	2.82	2.83	2.84	2.84	2.84	2.85
(211)	2.28	2.30	2.30	2.31	2.31	2.30

Table S2 Lifetime values of CsPbCl₃ and CsPb(Cl/F)₃ QDs monitored at 410 nm as well as the corresponding radiative recombination rate and nonradiative recombination rate calculated from the experimental values of photolumincescence QY and decay lifetimes.

Sample	τ_1 (ns)	a ₁ (%)	τ_2 (ns)	a ₂ (%)	$\tau_{avg} (ns)$	radiative	nonradiative
						recombination	recombination
						rate (s ⁻¹)	rate (s ⁻¹)
1:0	0.7	74.6	6.9	25.4	5.5	6.2×10 ⁶	1.7×10 ⁸
1:0.2	0.8	65.1	7.1	34.9	6.0	6.3×10 ⁶	1.6×10 ⁸
1:0.4	0.8	63.0	9.8	37.0	8.7	6.6×10 ⁶	1.1×10 ⁸
1:0.6	0.8	59.7	10.3	40.3	9.3	7.1×10 ⁶	1.0×10 ⁸
1:0.8	0.9	54.8	11.6	45.2	10.7	8.7×10 ⁶	8.5×10 ⁷
1:1	0.8	60.1	10.2	39.9	9.2	5.0×10 ⁶	1.1×10 ⁸

Table S3 The La³⁺ ion doping concentration (%) from ICP characterization, lattice constant (110) and the photoluminescence QYs under the excitation wavelength of $365 \text{ nm of } CsPb(Cl_{0.7}F_{0.3})_3:La^{3+}$ QDs prepared at different hot injection temperature.

Injection	Injection La ³⁺ ions		lattice constant	photoluminescence
temperature	concentration	concentration	(110) (Å)	QYs (%)
(°C)	(%) used in feed	(%) from ICP		
	solution	characterization		
160	65.2	3.1	3.97	22.6
180	65.2	5.2	3.95	28.6
200	65.2	14.4	3.95	36.5
220	65.2	9.2	3.96	36.2
240	65.2	7.2	3.96	31.5

Table S4 Lifetime values of $CsPb(Cl_{0.7}F_{0.3})_3$: La³⁺ QDs prepared with different La³⁺ ions doping concentration under monitored wavelength of 410 nm as well as the corresponding radiative recombination rate and nonradiative recombination rate calculated from the experimental values of photoluminescence QY and decay lifetimes.

Sample	$\begin{array}{c} \tau_1 \\ (ns) \end{array}$	a ₁ (%)	τ_2 (ns)	a ₂ (%)	τ_{avg} (ns)	radiative	nonradiative
						rate (s ⁻¹)	rate (s ⁻¹)
CsPb(Cl _{0.7} F _{0.3}) ₃ QDs	0.9	54.8	11.6	45.2	10.7	8.7×10 ⁶	8.5×10 ⁷
3.1 %	1.6	34.6	27.6	65.4	26.8	8.4×10^{6}	2.9×10 ⁷
5.2 %	1.5	36.4	26.5	63.6	25.7	1.1×10 ⁷	2.8×10 ⁷
7.2 %	1.4	35.6	26.0	64.4	25.3	1.3×10 ⁷	2.7×10 ⁷
9.2 %	1.4	36.6	24.3	63.4	23.6	1.5×10 ⁷	2.7×10 ⁷
14.4 %	1.4	36.7	23.6	63.3	22.9	1.6×10 ⁷	2.7×10 ⁷