

Supporting Information

Multi-components composed $\text{Cu}_2\text{O}@\text{FePO}_4$ core-cage structure to jointly promote fast electron transfer toward highly sensitive *in situ* detection of nitric oxide

Yuhuan Zhang,^{a,c} Shi-Yu Lu,^{a,c} Zhuanzhuan Shi,^{a,c} Zhi Liang Zhao,^{a,c} Qian Liu,^{a,c} Jie-Chang Gao,^{a,c} Taotao Liang,^{a,c} Zhuo Zou,^{a,c} Chang Ming Li^{*a,b,c}

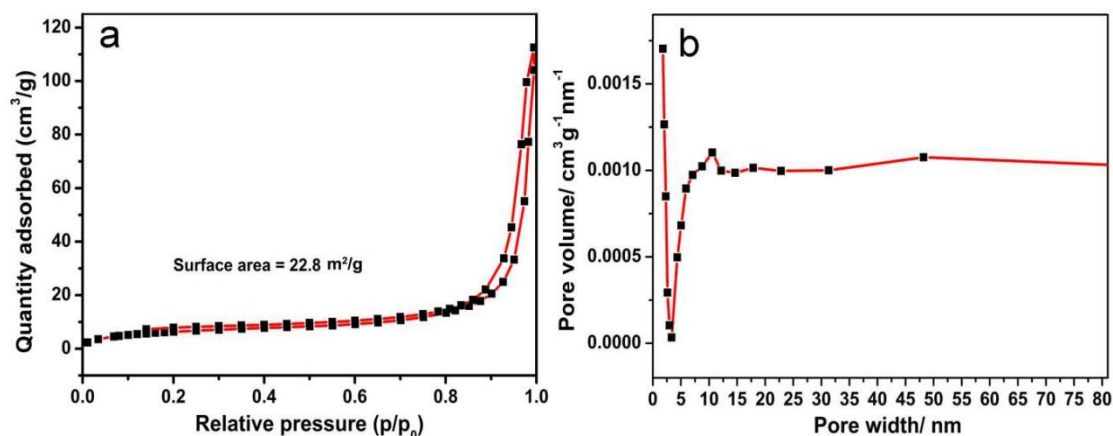


Fig. S1. (a) Nitrogen adsorption-desorption and (b) pore-size distribution curves of $\text{Cu}_2\text{O}@\text{FePO}_4\text{CC}$.

*Corresponding author at: Institute for Clean Energy and Advanced Materials,
Southwest University, Chongqing 400715, China. Fax: +86 23 68254969.
E-mail address: ecmli@swu.edu.cn (C.M. Li).

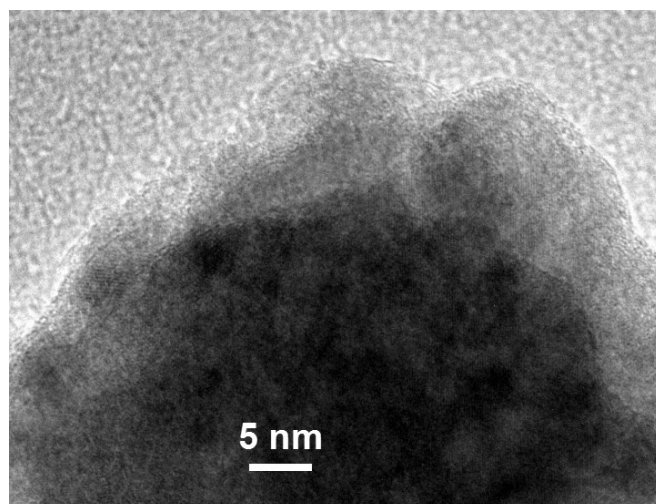


Fig. S2. HRTEM image of $\text{Cu}_2\text{O}@\text{FePO}_4\text{CC}$.

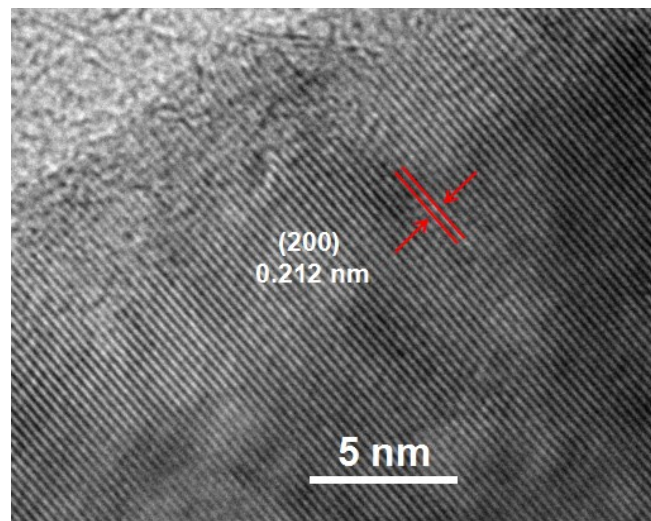


Fig. S3. HRTEM image of Cu_2O .

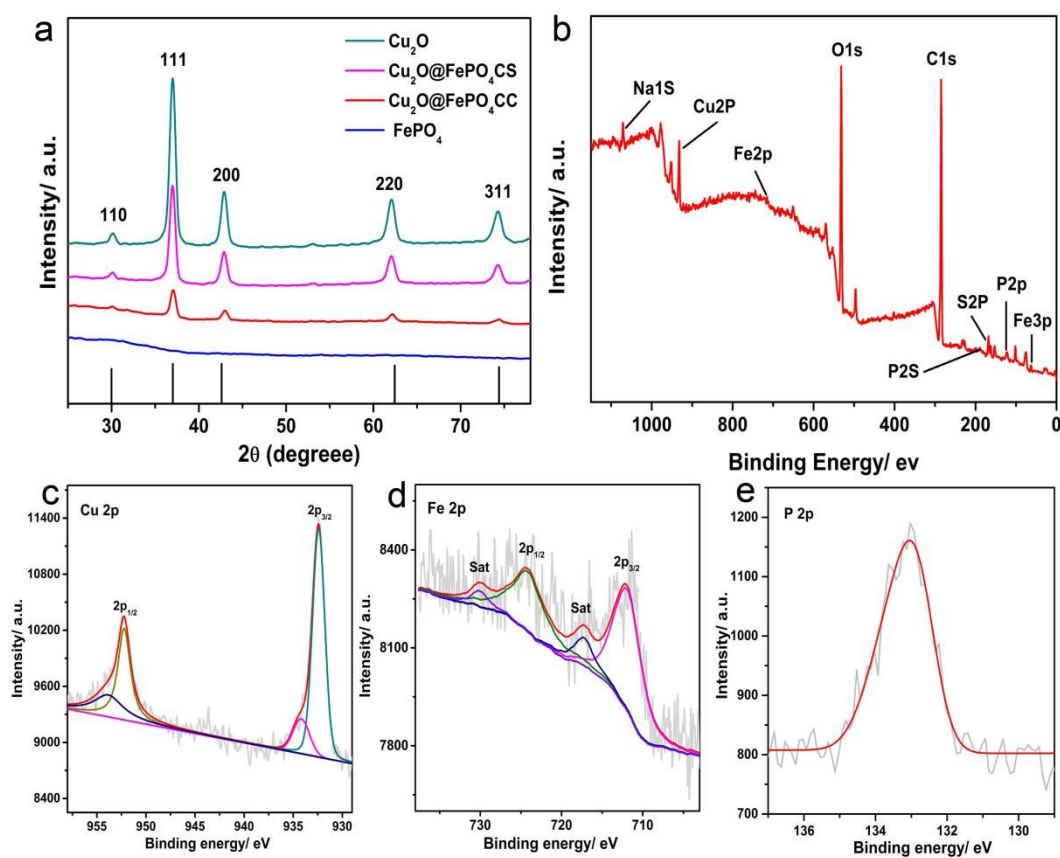


Fig. S4. (a) XRD patterns of Cu_2O , $\text{Cu}_2\text{O@FePO}_4\text{CS}$, $\text{Cu}_2\text{O@FePO}_4\text{CC}$ and FePO_4 . (b)

XPS survey spectrum of $\text{Cu}_2\text{O@FePO}_4\text{CC}$ and corresponding high-resolution spectra

in (c) Cu 2p, (d) Fe 2p and (e) P 2p regions.

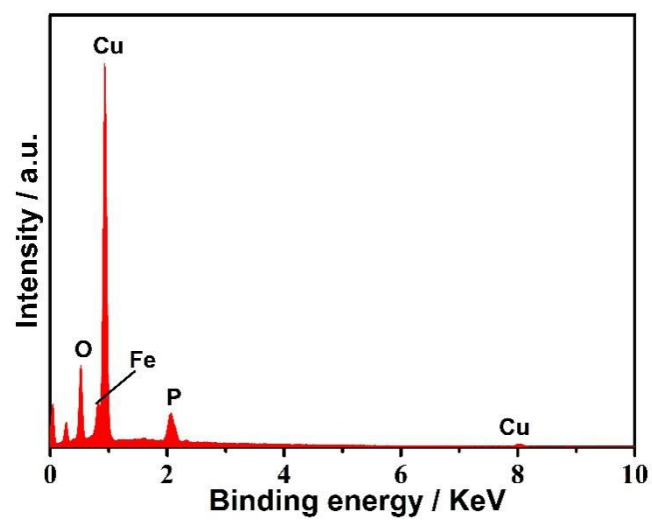


Fig. S5. EDS spectrum of $\text{Cu}_2\text{O}@ \text{FePO}_4\text{CC}$.

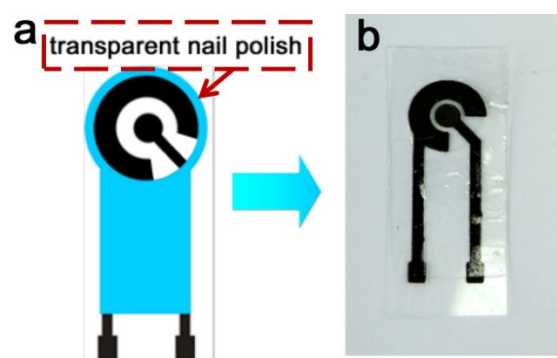


Fig. S6. (a) Scheme and (b) optical image of SPE.

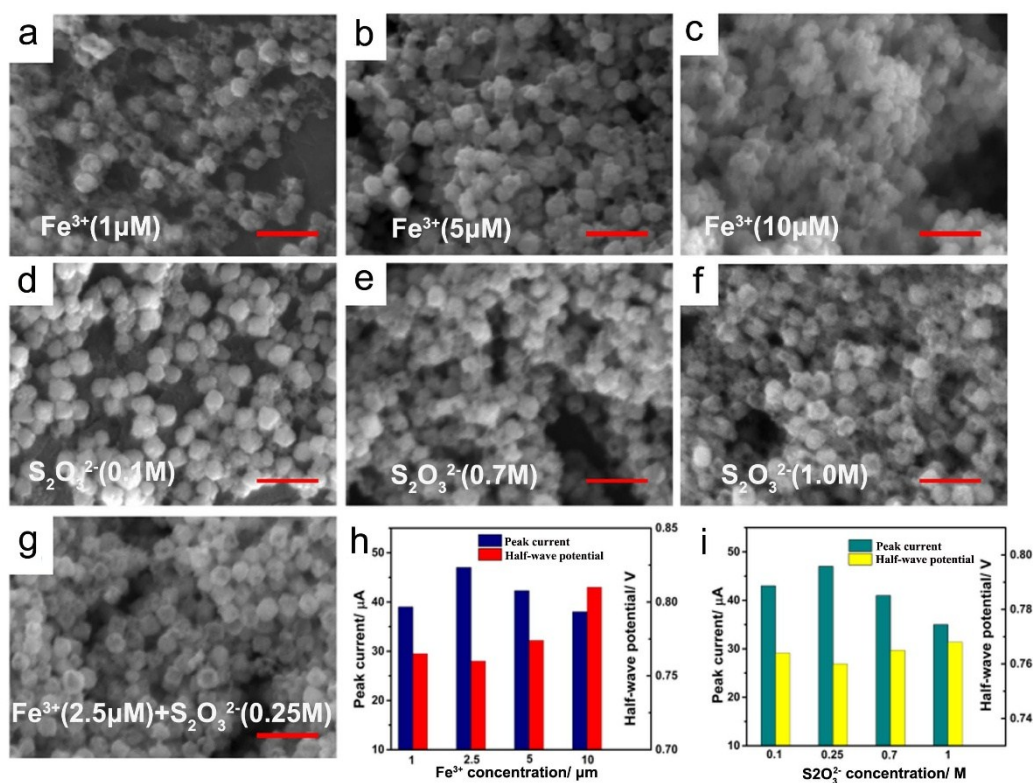


Fig. S7. SEM images of Cu₂O@FePO₄CC materials synthesized with various Fe³⁺ concentrations: (a) 1.0 μM, (b) 5.0 μM, (c) 10.0 μM, (g) 2.5 μM. And the SEM images of Cu₂O@FePO₄CC materials synthesized with different S₂O₃²⁻ concentrations: (d) 0.1 M, (e) 0.7 M, (f) 1 M, (g) 0.25 μM. Effect of the different addition of Fe³⁺ (h) and S₂O₃²⁻ (i) on the performance of Cu₂O@FePO₄CC/SPE in CV responses, peak currents or half-wave potentials. (Scale bar is equal to 200 nm for every image)

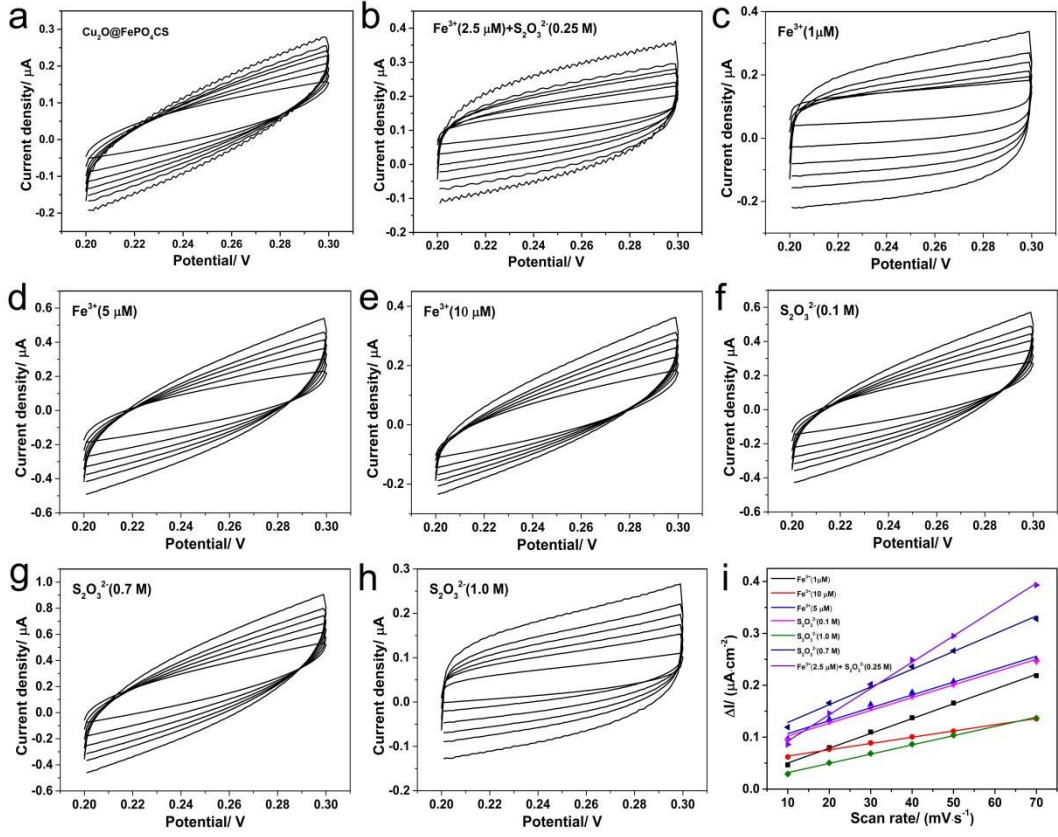


Fig. S8. CV curves of $\text{Cu}_2\text{O@FePO}_4\text{CS}$ (a) and $\text{Cu}_2\text{O@FePO}_4\text{CC}$ synthesized under different conditions of (b) 2.5 μM , (c) 1 μM , (d) 5 μM and (e) 10 μM Fe^{3+} concentrations; (f) and (g) and (h) are the CV curves for $\text{Cu}_2\text{O@FePO}_4\text{CC}$ synthesized under 0.1 M, 0.7 M and 1.0 M $\text{S}_2\text{O}_3^{2-}$; the potential window is 0.2–0.3 V and the scan rates vary from 10, 20, 30, 40, 50 to 70 mV s^{-1} . (i) Capacitive currents at 0.25 V as a function of scan rates for different materials.

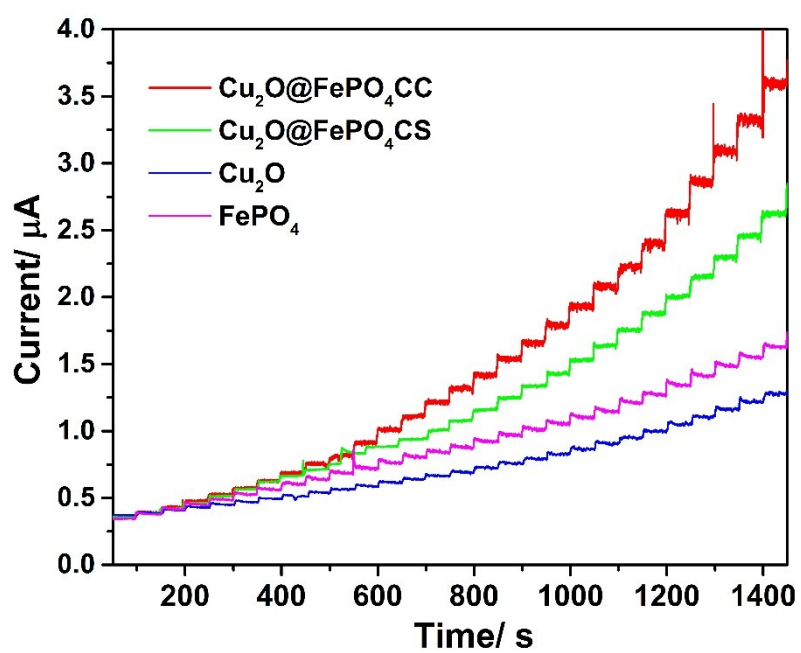


Fig. S9. Amperometric curve of $\text{Cu}_2\text{O}@ \text{FePO}_4 \text{CC}/\text{SPE}$, $\text{Cu}_2\text{O}@ \text{FePO}_4 \text{CS}/\text{SPE}$, $\text{Cu}_2\text{O}/\text{SPE}$, FePO_4/SPE upon continuous injection of NO at 0.97 V in 0.01 M PBS (pH 7.4).

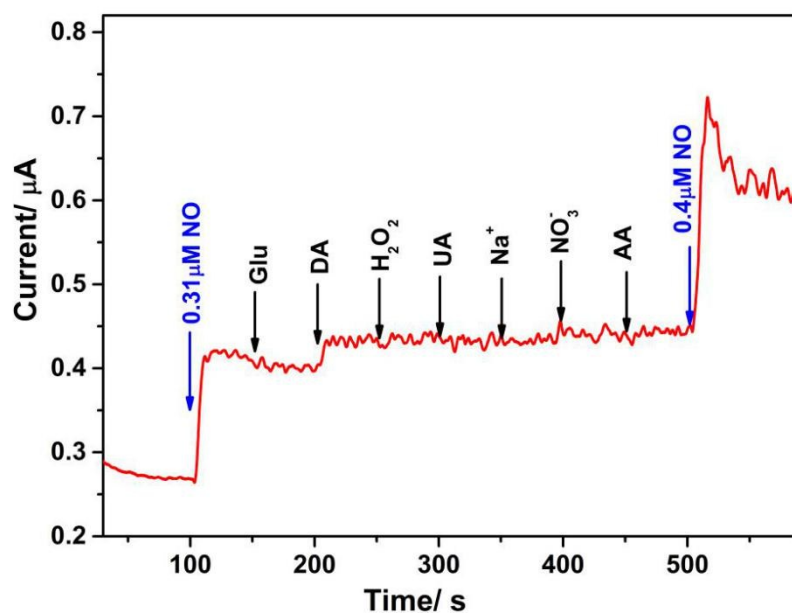


Fig. S10. Selectivity of Cu₂O@FePO₄CC/SPE.