

Supporting Information

Two-Dimensional Bimetallic Phosphide Ultrathin Nanosheets as Non-Noble Electrocatalysts for Highly Efficient Oxygen Evolution Reaction

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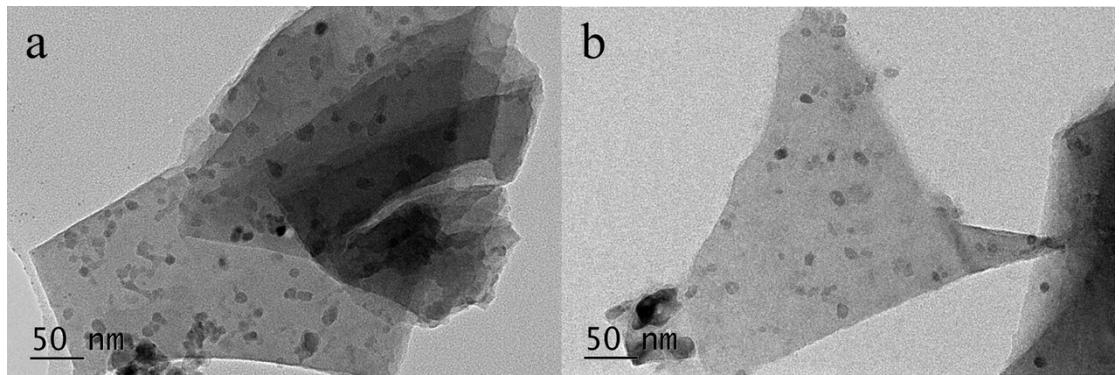


Fig. S1 TEM images of (a) $\text{Co}_{0.9}\text{Fe}_{0.1}\text{P}$ and (b) $\text{Co}_{0.7}\text{Fe}_{0.3}\text{P}$ nanosheets.

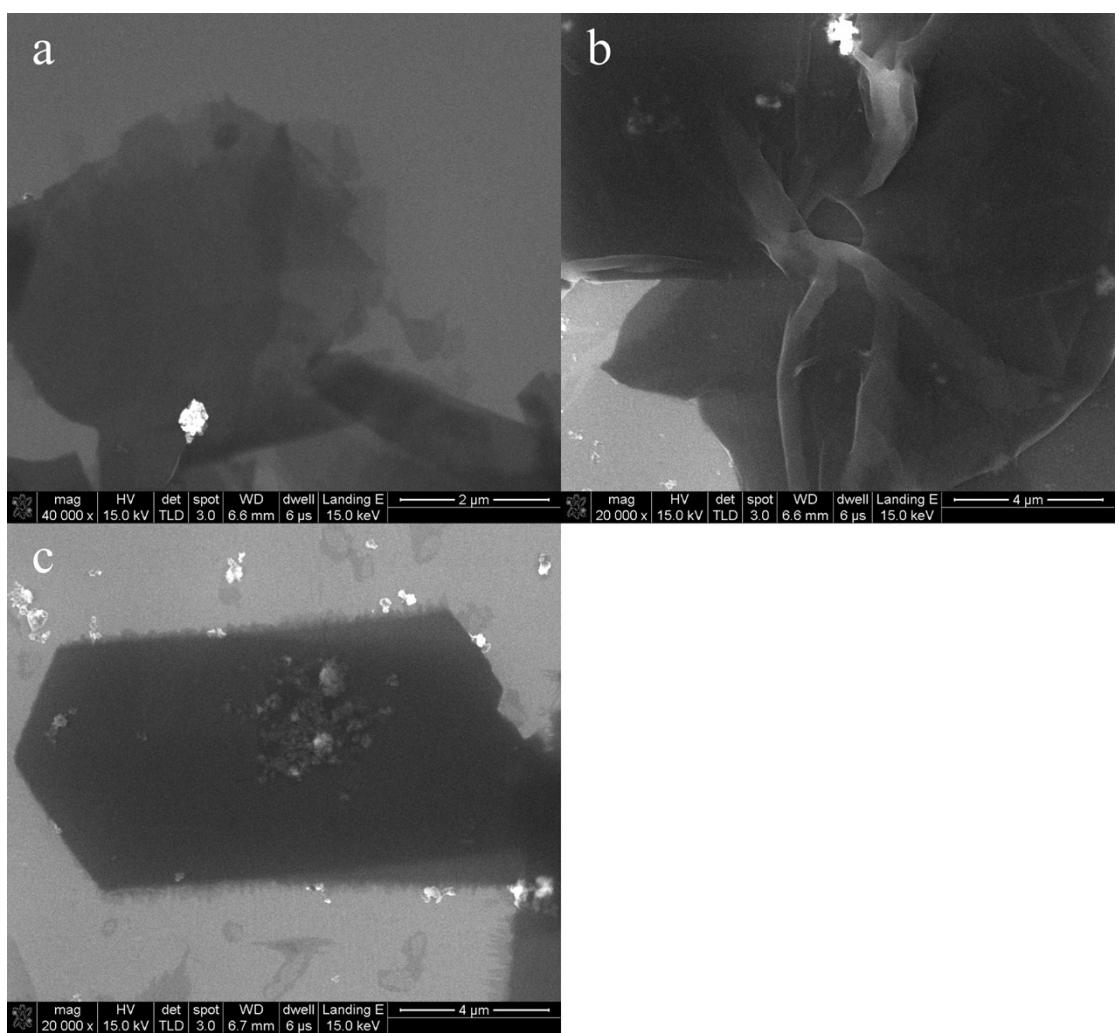


Fig. S2 SEM images of (a) CoP, (b) $\text{Co}_{0.9}\text{Fe}_{0.1}\text{P}$ and (c) $\text{Co}_{0.7}\text{Fe}_{0.3}\text{P}$ nanosheets.

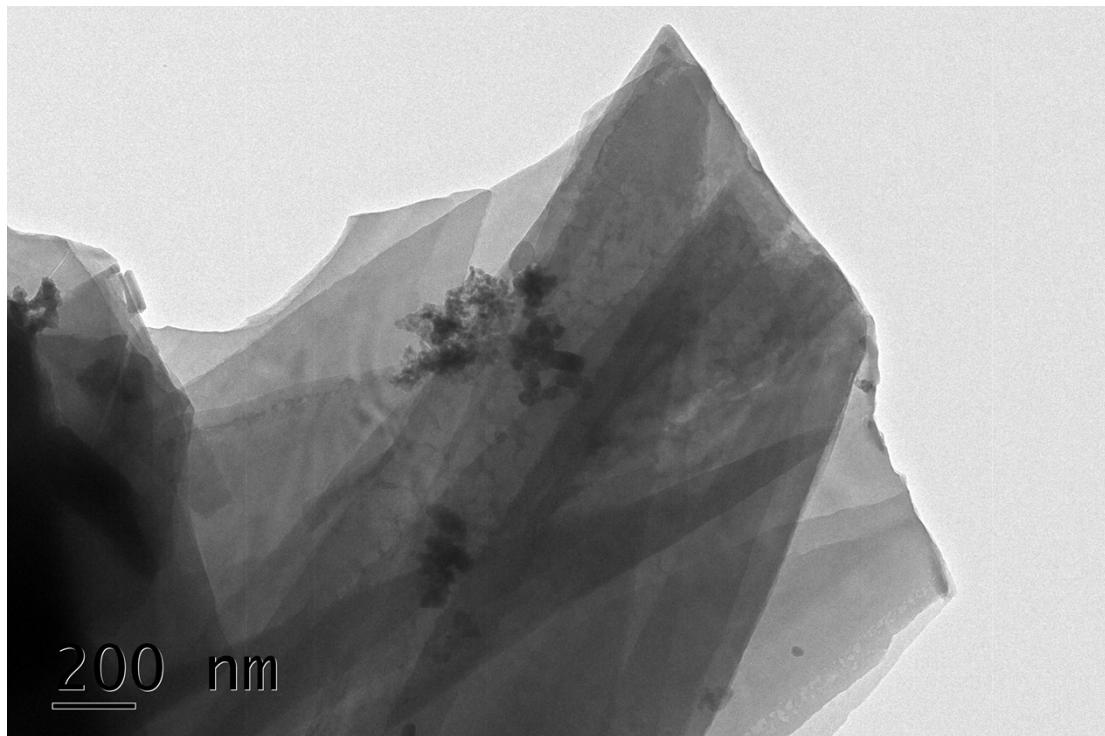


Fig. S3 TEM image of $\text{Co}_{0.8}\text{Fe}_{0.2}\text{P}$ nanosheets at low magnification.



Fig. S4 SEM image of $\text{Co}_{0.8}\text{Fe}_{0.2}\text{P}$ nanosheets at low magnification.

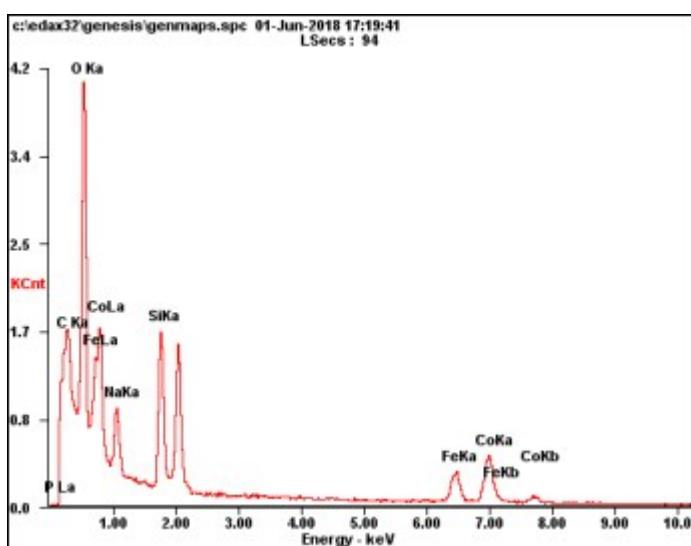


Fig. S5 EDX patterns of $\text{Co}_{0.8}\text{Fe}_{0.2}\text{P}$ nanosheets.

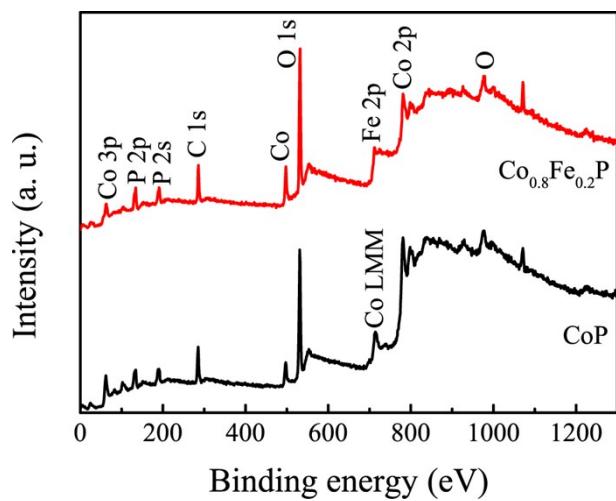


Fig S6 XPS survey scans of CoP and $\text{Co}_{0.8}\text{Fe}_{0.2}\text{P}$ nanosheets.

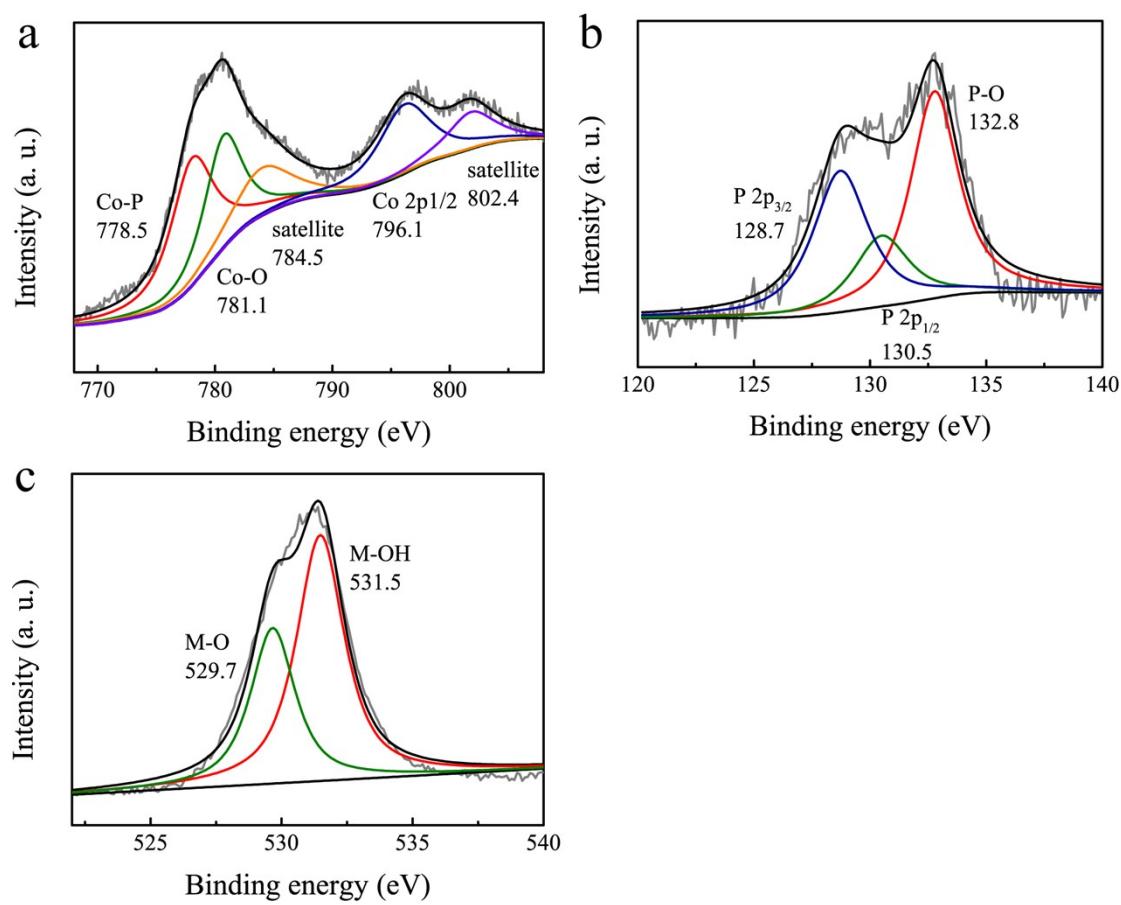


Fig. S7 XPS spectra of CoP nanosheets. (a) Co 2p, (b) P 2p and (c) O 1s.

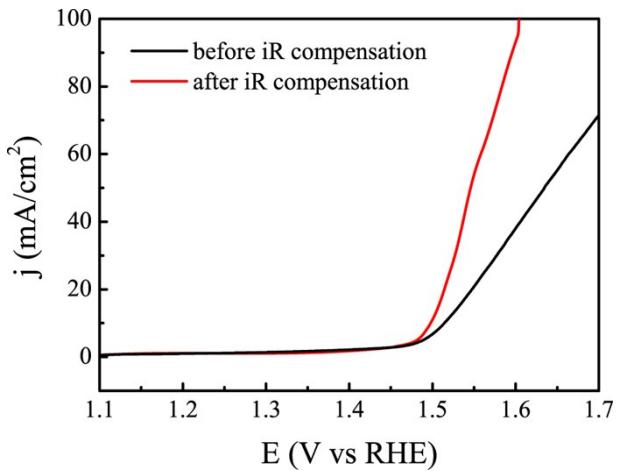


Fig. S8 The LSV curves of $\text{Co}_{0.8}\text{Fe}_{0.2}\text{P}$ nanosheets before and after iR compensation.

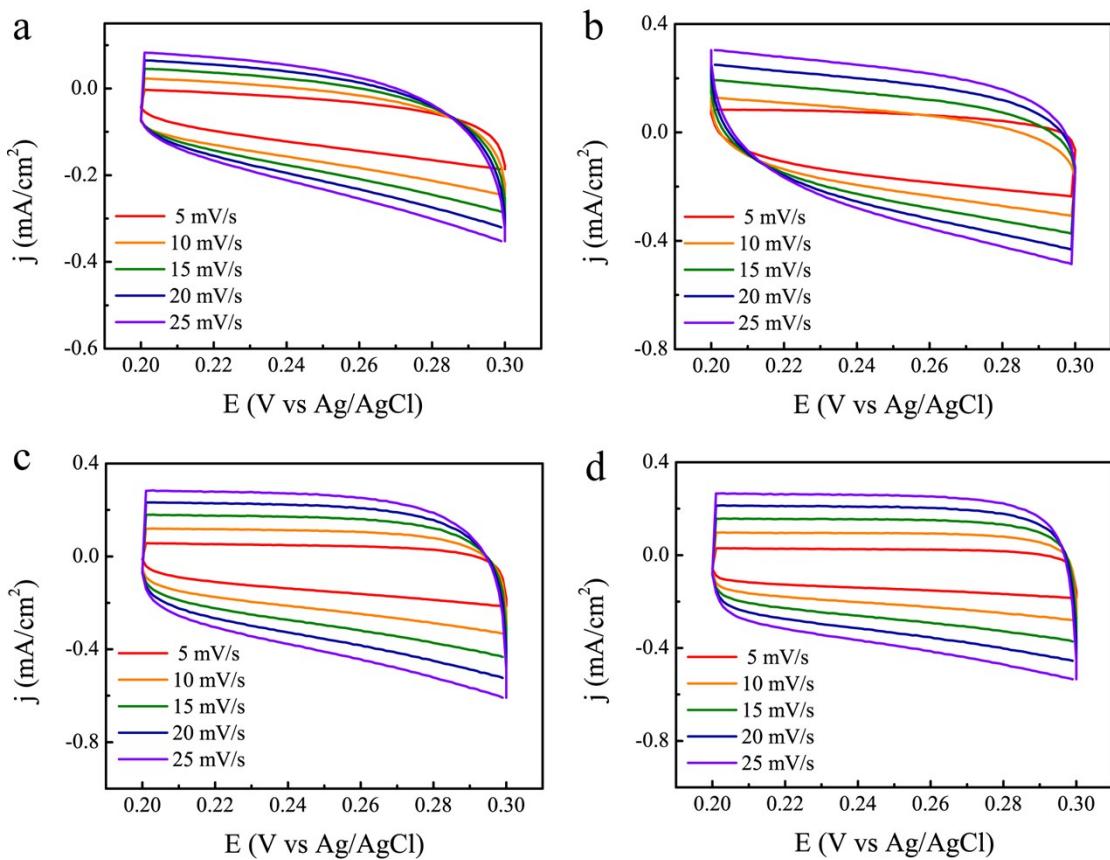


Fig. S9 Cyclic voltammetry of (a) CoP, (b) $\text{Co}_{0.9}\text{Fe}_{0.1}\text{P}$ (c) $\text{Co}_{0.8}\text{Fe}_{0.2}\text{P}$ and (d) $\text{Co}_{0.7}\text{Fe}_{0.3}\text{P}$ nanosheets at different scan rate.

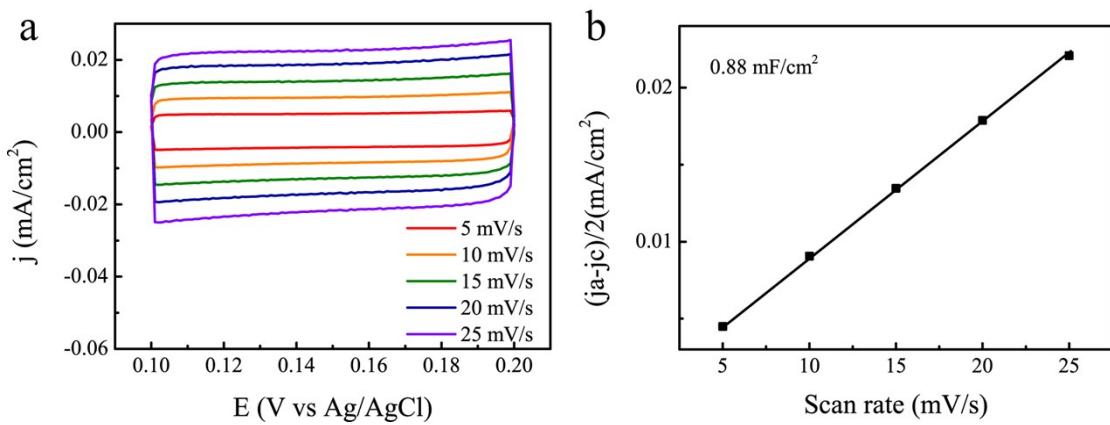


Fig. S10 (a) Cyclic voltammetry of commercial RuO₂. (b) Electrical double-layer capacitance C_{DL} of commercial RuO₂.

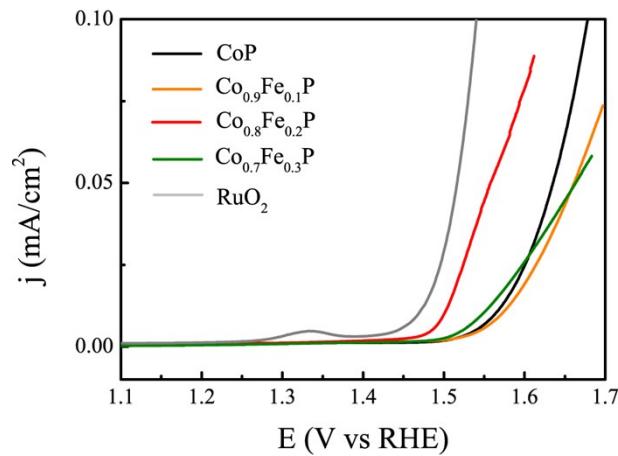


Fig. S11 LSV curves normalised by ECSA.

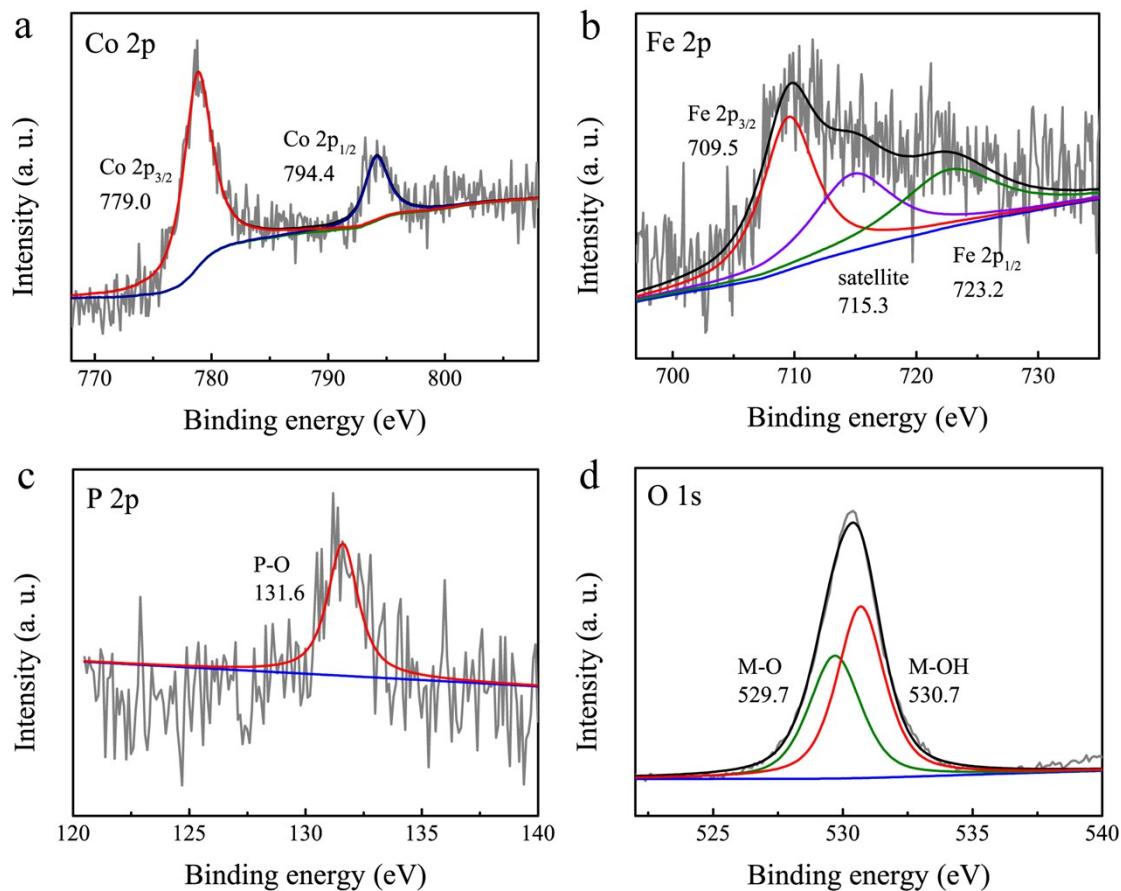


Fig. S12 XPS spectra of $\text{Co}_{0.8}\text{Fe}_{0.2}\text{P}$ nanosheets after OER test. (a) Co 2p, (b) Fe 2p, (c) P 2p and (d) O 1s.

Table S1 Summary of various metal phosphides-based catalysts for OER. The electrolyte in these works are all 1 M KOH, the electrode in these works are all rotating disk electrode (RDE).

Materials	Overpotential (mV) at 10 mA cm^{-2}	Tafel slope (mV dec $^{-1}$)	Journal, volume, first page Year,
2D $\text{Co}_{0.8}\text{Fe}_{0.2}\text{P}$ nanosheets	270	50	This work
CoP@S,N-codoped graphite carbon	350	68	Nanoscale, 2018, 10, 14613
defective carbon-	320	52.5	Adv. Energy

CoP			Mater. 2018, 8, 1703623
Co ₂ P nanocrystals	280	-	Adv. Mater. 2018, 30, 1705796
Ni _{0.6} Co _{1.4} P	300	80	Adv. Funct. Mater. 2018, 28, 1706008
NiCo ₂ P _x /CNTs	284	50.3	J. Mater. Chem. A, 2018, 6, 7420
NiCoP/C nanoboxes	330	96	Angew. Chem. 2017, 129, 3955
Co _{0.9} S _{0.58} P _{0.42}	266	48	ACS Nano 2017, 11, 11031
Co4Ni1P	245	61	Adv. Funct. Mater. 2017, 27, 1703455
activated Mn-Co oxyphosphide particles	320	52	Angew. Chem. Int. Ed. 2017, 56, 2386